The Active Form of the R2F Protein of Class Ib Ribonucleotide Reductase from Corynebacterium ammoniagenes Is a Diferric Protein*

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Corynebacterium ammoniagenes contains a ribonucleotide reductase (RNR) of the class Ib type. The small subunit (R2F) of the enzyme has been proposed to contain a manganese center instead of the dinuclear iron center, which in other class I RNRs is adjacent to the essential tyrosyl radical. The nrdF gene of C. ammoniagenes, coding for the R2F component, was cloned in an inducible Escherichia coli expression vector and overproduced under three different conditions: in manganese-supplemented medium, in iron-supplemented medium, and in medium without addition of metal ions. A prominent typical tyrosyl radical EPR signal was observed in cells grown in rich medium. Iron-supplemented medium enhanced the amount of tyrosyl radical, whereas cells grown in manganese-supplemented medium had no such radical. In highly purified R2F protein, enzyme activity was found to correlate with tyrosyl radical content, which in turn correlated with iron content. Similar results were obtained for the R2F protein of Salmonella typhimurium class Ib RNR. The UV-visible spectrum of the C. ammoniagenes R2F radical has a sharp 408-nm band. Its EPR signal at g = 2.005 is identical to the signal of S. typhimurium R2F and has a doublet with a splitting of 0.9 millitesla (mT), with additional hyperfine splittings of 0.7 mT. According to X-band EPR at 77–95 K, the inactive manganese form of the C. ammoniagenes R2F has a coupled dinuclear Mn(II) center. Different attempts to chemically oxidize Mn-R2F showed no relation between oxidized manganese and tyrosyl radical formation. Collectively, these results demonstrate that enzymatically active C. ammoniagenes RNR is a generic class Ib enzyme, with a tyrosyl radical and a diferric metal cofactor.

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Deoxyribonucleotides are synthesized in all organisms from ribonucleotides, in a reaction catalyzed by the enzyme ribonucleotide reductase (RNR).1 Common to all RNRs is that they proceed via a radical mechanism (1). However, based on amino acid sequence similarities and polypeptide compositions (2), as well as metal and radical cofactors, three well-characterized RNR classes (I–III) can be distinguished. Class I RNRs consist of two homodimeric components, the larger R1 protein and the smaller R2 protein. Protein R1 contains the active site, and protein R2 contains a stable tyrosyl radical in close proximity to a diferric center. Class II RNRs are monomeric or homodimeric proteins that bind and utilize a vitamin B12 cofactor (deoxyadenosine cobalamin). Class III RNRs consist of two proteins. The larger contains its active form in a stable glycal radical close to the active site. The smaller protein generates this radical with the aid of a 4Fe-4S cluster and S-adenosylmethionine. The class I RNRs are further subdivided into class Ia and class Ib based on amino acid sequence similarities (3, 4), allosteric properties, and dependence on different physiological reductants (5). Among prokaryotes several different classes and subclasses of RNR may be encoded in the same genome. In class Ia RNRs, the R1 protein is encoded by the nrdA gene and the R2 protein by the nrdB gene (1). In class Ib RNRs, the nrdE gene codes for the R1E protein and the nrdF gene codes for the R2F protein (3–6).

The three-dimensional structures of proteins R1 and R2 of class Ia RNR from Escherichia coli are known (7, 8). Four carboxylate residues and two histidines ligate the di-iron site in protein R2. In the native, enzymatically competent form, the two ferric ions are bridged by a μ-oxo bridge and one of the carboxylates (7). The iron ligands and the stable tyrosyl radical residue are conserved among all class I sequences (1). Recently, the three-dimensional structure of the class Ib R2F protein from Salmonella typhimurium was solved (9). The ligand sphere is slightly different in S. typhimurium R2F as compared with E. coli R2.

In this study we characterize the radical and metal cofactor of enzymatically competent R2F protein of a class Ib RNR from Corynebacterium ammoniagenes. It has been suggested that R2F from C. ammoniagenes binds and is activated by manganese instead of iron, because manganese is needed for growth of the orga

1 The abbreviations used are: RNR, ribonucleotide reductase; apoR2(F), R2 proteins lacking a metal site; DTT, dithiothreitol; Fe-R2F, R2F protein reconstituted with iron; IPTG, isopropyl-1-thio-β-D-galactopyranoside; metR2(F), R2 protein with a metal site but lacking a tyrosyl radical; Mn-R2(F), R2 protein purified from bacteria grown in the presence of manganese or manganese-reconstituted apoR2; PCR, polymerase chain reaction; PAGE, polyacrylamide gel electrophoresis.
Iron-dependent Ribonucleotide Reductase from C. ammoniagenes

Iron Content, Tyrosyl Radical, and Enzyme Activity—392 mg of (NH₄)₂Fe(SO₄)₂ and 396 mg of sodium ascorbate were dissolved in 10 ml of argon-flushed 50 mM Tris-HCl, pH 7.6, resulting in a 100 mM ferrous solution with 200 mM ascorbate. To 500 μl of 0.58 mM R2F (containing approximately 1 Fe/R2F) solution (0.29 μmol) was added 11.5 μl of the 10 × reaction buffer (1.16 μmol of the reaction solution (1.16 μM)). The reaction was incubated for 5 min at 25 °C and then transferred to ice. The reaction mixture was desalted on a G-50 column (NAP-10) equilibrated with 50 mM Tris, pH 7.6, purified on MonoQ as described earlier (6), and concentrated by ultradialysis to 250 μl of R2F and a concentration of 0.14 μM.

Preparation of S. typhimurium apoR2F by Reduction of R2F—The following procedure was done at 4 °C. S. typhimurium-mixed Fe/Mn-R2F was precipitated by addition of isopropyl-1-thio-D-galactopyranoside (IPTG) to 0.5 mM and dialyzed against 4 × 50 mM Tris-HCl, pH 7.6, 50 mM EDTA, and 10 mM dithiothreitol. The material was essentially as described earlier (15). Briefly, apoR2F (0.53 mg, 6.1 nmol of apoR2F) was equilibrated with 6.9 nmol, in 0.33 ml) was treated anaerobically with a mixture of 9 μl of 10 mM Fe(H₄)SO₄, and 18 μl of 10 mM MnCl₂, followed by passage through a 10-ml column of Sephadex G-25 in buffer A. Fractions (0.5 ml) were analyzed for protein and radioactivity. A first radioactive peak coincided with the protein peak, clearly separated from a second larger radioactive peak. All protein fractions showed the same specific radioactivity. After concentration, the solution (3.2 mg/ml, 36 μM) was analyzed for enzyme activity and its EPR spectrum was recorded.

Preparation of S. typhimurium Mn-R2F—ApoR2F (1.7 mg, 19 nmol, in 0.65 ml of buffer A) was treated for 5 min with 24 μl of 10 mM MnCl₂, followed by passage through a 10-ml column of Sephadex G-25 in buffer A. The enzyme activity of Mn-R2F proteins was measured as described in Fieschi et al. (6), with 0.5 mM CDP as substrate, 1 mM dATP as allosteric effector, and 10 mM DTT as reducing agent. Unless otherwise indicated excess of R1E, purified as described in Fieschi et al. (6), was included in the assays.

The enzyme activity of S. typhimurium R2F proteins was determined in the presence of an excess of protein R1E and with 20 mM DTT as reducing agent, CDP as substrate, and dATP as allosteric effector (4). The activity found in this way amounted to approximately one third of the full activity observed with glutaredoxin or the reduced NrdH protein.

Specific activity is expressed as units/mg of protein, where one unit corresponds to the reduction of 1 nmol of CDP/min. C. ammoniagenes RNR was assayed at 30 °C and S. typhimurium RNR at 37 °C.

Analytical Methods—Protein (16) and iron (15, 17) were determined by standard methods. In spectrophotometry experiments, C. ammoniagenes protein concentration was calculated, using an extinction coefficient of 280 nm for R2F, ε = 136,000 M⁻¹ cm⁻¹. Manganese was determined by atomic absorption spectroscopy or calculated from the radioactivity of the reconstituted protein and the determined specific activity of ⁵⁹MnCl₂.

EPR Spectroscopy—EPR spectra at 77 K were obtained directly on thick slurry of harvested cells or on purified protein samples (0.25–0.77 mM R2F). X-band EPR spectra were measured either on a Bruker ESP 300 spectrometer at 77 K with the sample placed in a cold finger liquid nitrogen Dewar or on a Bruker Elexys system flowing nitrogen gas at 95 K through a Dewar flask placed in the cavity. Spectra were evaluated using Xepol 2.0 software from Bruker. The microwave saturation spectra were evaluated as described previously by the formula

\[ DI_{sat} = 1/(1 + P/M_{sat})^{(2/18), (18)} \]

where \( DI_{sat} \) is the normalized double integral of the EPR signal, \( P \) the microwave power in milliwatts (mW), \( M_{sat} \) the power of half saturation, and \( b \) the inhomogeneity factor. The value of \( b \) varies between 1, for the completely inhomogeneously broadened line, and 3, for an homogeneously broadened line.

Light Absorption Spectroscopy—The UV-visible spectra were re-
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RESULTS

Cloning and Overproduction of C. ammoniagenes

R2F in E. coli

The plasmid pIG056 contains the C. ammoniagenes nrdF gene under the control of the T7 promoter. Plasmid-containing E. coli strain JM109(DE3), which carries the T7 RNA polymerase gene under control of the Lac promoter, produced massive amounts of C. ammoniagenes R2F after IPTG induction. Three different conditions were used: LB medium, LB medium supplemented with manganese, and LB medium supplemented with iron. All three conditions resulted in similar overproduction of R2F according to SDS-PAGE analysis (with Coomassie Blue staining) (Fig. 1A). The different cell suspensions were examined by EPR. A prominent free radical signal was detected in the sample grown in LB (Fig. 2A, upper spectrum). Twice as much radical was detected when cells were grown in iron-supplemented medium (Fig. 2A, lower spectrum). In contrast, the culture with manganese-supplemented medium contained no radical EPR signal, but showed Mn(ii) with octahedral coordination (Fig. 2B). Fig. 2B covers the same magnetic field as in Fig. 2A and shows the three middle lines of the octahedrally coordinated Mn^2+. The vertical scales differ by a factor 0.62 (A/B), which is in agreement with an assignment of the Mn^2+ in Fig. 2B to unbound Mn^2+ in the medium hiding low amounts of radical features in the g = 2.00 region. The additional splittings from the forbidden transitions of the nuclear spins are not seen in the middle line. Subtraction of approximately 0.3 times the signal in Fig. 2A (upper spectrum) gives a spectrum like the Mn^2+ standard, indicative of a small amount of tyrosyl radical also in the sample from the manganese-supplemented medium. Cultures grown in ordinary or manganese-supplemented media were used for purification of R2F protein in high yield (Fig. 1 and Table 1).

Characterization of Purified C. ammoniagenes

R2F Proteins

Radical Content and Enzymatic Activity—The R2F sample purified from cells grown in rich medium without metal ion supplement had a prominent EPR signal at g = 2.005 arising from a major splitting of 0.9 millitesla (mT) and additional hyperfine couplings at 0.7 mT (Fig. 3A). The EPR spectrum can be superimposed on that from the tyrosyl radical from the S. typhimurium R2F (cf. Fig. 3B) (4, 19). The spectra in Fig. 3 differ significantly from the EPR spectrum of the tyrosyl radical of the E. coli class Ia R2 protein (20). The latter has a major splitting from one of the beta-protons of 2 mT giving a clearly resolved doublet spectrum with additional splitting from the 3.5 ring protons of 0.7 mT (20, 21). The EPR spectrum of purified C. ammoniagenes R2F in Fig. 3A thus excludes contamination of ribonucleotide reductase from the E. coli host cells.

Protein R2F purified from Mn-supplemented cells, Mn-R2F, displays a complex multiline spectrum, superimposed on a broad signal often found in dimanganese proteins (22). A small amount of free radical signal is also seen, but the concentration is lower in this sample compared with the C. ammoniagenes R2F sample without manganese addition (Fig. 4A and Table 1). The R2F sample, purified from cells grown without supplement of metal ions, had a UV-visible spectrum with a 408-nm peak, characteristic of a tyrosyl radical (20). The absorption spectrum obtained by us differs from the previously published C. ammoniagenes RNR band at 437 nm (23). The radical was sensitive to hydroxyurea, and the difference spectrum between before and after hydroxyurea scavenging (Fig. 5) agrees well with the spectrum of a tyrosyl radical (24). It is different from...
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The metal site of R2F was reconstituted with iron or manganese and studied by UV-visible spectroscopy. Iron was shown to form an oxidized metal center concomitantly with tyrosyl radical formation, because the characteristic bands from the oxidized iron center and the tyrosyl radical peak at 408 nm evolved together. It was not possible to generate an oxidized manganese center and/or tyrosyl radical by addition of manganese to the aerobic R2F.

The results presented so far suggest that the enzyme activity of C. ammoniagenes R2F correlates with the presence of a tyrosyl radical, which in turn correlates with the presence of a di-iron center. However, R2F from C. ammoniagenes has earlier been assumed to have a metal center different from other R2F proteins (6, 11–13). We therefore attempted to reconstitute the well-characterized class Ib R2F protein from S. typhimurium with iron and manganese.

### Table I

<table>
<thead>
<tr>
<th>Metal ion content, radical content, and specific activity of Corynebacterium ammoniagenes R2F proteins</th>
<th>Fe/monomer</th>
<th>Mn/monomer</th>
<th>Tyr/monomer</th>
<th>Specific activity (nmol mg⁻¹ min⁻¹)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Mn-R2F</td>
<td>0.5</td>
<td>1.6</td>
<td>0.03</td>
<td>9</td>
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<tr>
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<td>0.09</td>
<td>0.05</td>
<td>36</td>
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<tr>
<td>Fe-R2F&lt;sup&gt;a&lt;/sup&gt;</td>
<td>1.5</td>
<td>ND&lt;sup&gt;b&lt;/sup&gt;</td>
<td>0.2</td>
<td>48&lt;sup&gt;c&lt;/sup&gt;</td>
</tr>
</tbody>
</table>

<sup>a</sup> Iron-reconstituted R2F.

<sup>b</sup> ND, not determined.

<sup>c</sup> These assays contained ca. 0.2 μM R1E and 0.05 or 0.2 μM R2F.

FIG. 3. EPR spectra on purified R2F proteins. A, C. ammoniagenes R2F purified from overproducing E. coli cells grown in LB medium (16 scans). B, S. typhimurium R2F reconstituted in the presence of ferrous ions (1 scan). EPR conditions were: 0.2-mW microwave power, 60-db receiver gain, 0.1-mT modulation amplitude.

FIG. 4. EPR spectra on purified Mn-R2F proteins. A, C. ammoniagenes R2F purified from overproducing E. coli cells grown in LB medium plus 2 mM manganese and measured at 77 K. B, S. typhimurium R2F reconstituted in the presence of manganese ions. Measured at 95 K, the coupling signals in the high field part of the spectrum are more clearly seen. The corresponding spectrum recorded at 77 K has the same features, although the noise levels signals were higher. EPR conditions: A, 1 scan, 1.97-mW microwave power, 60-db receiver gain, 1-mT modulation amplitude; B, 2 scans, 1.9-mW microwave power, 63-db receiver gain, 0.5-mT modulation amplitude.

### Metal Reconstitution Studies with C. ammoniagenes R2F

The metal site of R2F was reconstituted with iron or manganese (approximately 100 μM). Subtraction of spectra obtained before (inset: solid line) and after (inset: dotted line) hydroxyurea quenching of the radical signal. For better resolution of individual spectra in the inset, the sample after hydroxyurea treatment was not corrected for dilution.

FIG. 5. UV-visible spectrum of radical in C. ammoniagenes R2F. The spectrum obtained after reconstitution of the R2F sample, which was low in metal, led to a Fe-R2F containing 1.5 Fe/monomer.

To evaluate the metal states, the features of the EPR signals were further characterized. The EPR spectrum at 77 K of the Mn-R2F sample gave rise to more than 30 resolved lines on a broad signal. The resolved lines are separated by approximately 4.6 mT and grouped into three groups of 11 lines partly overlapping with intensities 1:2:3:4:6:5:4:3:2:1 centered at g = 2.64, 2.31, and 1.69. A splitting of 4.6 mT (cf. Fig. 4A) is half of the value (9.5 mT) observed for mononuclear Mn<sup>2+</sup> and is typical of a coupled dimanganese center in the Mn<sup>2+</sup> state. Dimanganese Mn(II) is also found in arginase and manganese catalase (22, 25–28) and the hyperfine pattern observed in manganese-reconstituted C. ammoniagenes R2F resembles arginase, where the manganese center (29) is bridged by two carboxylates and one oxygen (water or hydroxyl).

The presence of an oxidized free radical signal in the Mn-R2F sample suggests that some protein molecules contained dinuclear manganese sites. Interestingly, addition of manganese to the growth medium did not affect the amount of iron in the C. ammoniagenes R2F. Iron reconstitution of the R2F sample, which was low in metal, led to a Fe-R2F containing 1.5 Fe/monomer.

The activity measurements of the two preparations showed that the R2F sample had a 4-fold higher specific activity than the Mn-R2F sample and that even higher specific activity was obtained after reconstitution of the protein in the presence of iron (Table I).

Metal Content and EPR Characteristics of the Metal Sites—

The iron content of the R2F sample was 0.5 Fe/monomer, whereas the manganese content was only 0.09 Mn/monomer (Table I). The Mn-R2F sample, however, contained 1.6 Mn/monomer (Table I), suggesting that some protein molecules contained dinuclear manganese sites. Interestingly, addition of manganese to the growth medium did not affect the amount of iron in the C. ammoniagenes R2F. Iron reconstitution of the R2F sample, which was low in metal, led to a Fe-R2F containing 1.5 Fe/monomer.
Metal Reconstitution Studies with S. typhimurium R2F

The conventional procedure for the removal of iron from class Ia ribonucleotide reductase involves dialysis of the R2 protein against a solution of 8-hydroxyquinoline in 1 mM imidazol buffer (15). This method was not suitable to remove iron from S. typhimurium R2F, as the protein precipitated. Instead, we dialyzed the protein after reduction of its iron-center against EDTA, as described under “Materials and Methods.” This gave an inactive apoR2F protein, free of iron. Treatment of R2F with either iron or manganese resulted in close to stoichiometric amounts of each metal incorporated (Table II).

Reconstituted Fe-R2F contained slightly more iron than the original R2F preparation but was only half as active. It gave an EPR spectrum typical of S. typhimurium R2F tyrosyl radical (Fig. 3B) but with less than half the amount of spins per iron (4, 20). The lower activity suggests that the protein during preparation of apoR2F was partly denatured and had lost the full ability to form the free tyrosyl radical. During manganese treatment the metal was incorporated with the same stoichiometry as during iron treatment. However, Mn-R2F was completely inactive and EPR spectroscopy showed no sign of tyrosyl radical. Instead, the EPR spectrum suggests the presence of a dinuclear Mn(II) center with similar, but somewhat different, coupling pattern to that obtained for C. ammoniagenes Mn-R2F (Fig. 4B). In this case four different multiplets of 11 lines are resolved and the background for the excited states is different from that of C. ammoniagenes Mn-R2F (Fig. 4A). A small amount of unspecifically bound mononuclear Fe(II) is seen at \( g = 4.3 \).

Treatment of apoR2F with equimolar amounts of iron and manganese gave a product that had incorporated roughly equal amounts of each ion. The reconstituted protein showed some activity, amounting to 13% of the iron-reconstituted enzyme. The low activity suggests the existence of inactive protein dimers containing both iron and manganese.

Saturation Behavior of the Tyrosyl Radicals in the C. ammoniagenes and S. typhimurium R2F Proteins

To distinguish between different environments around the tyrosyl radical, the microwave saturation properties of radical-containing R2F samples of C. ammoniagenes and S. typhimurium were compared with that of the R2 protein from E. coli. The normalized double integral of the EPR signals are plotted versus microwave power (Fig. 6). \( P_{0.5} \) or the power of half saturation, gives a measure on the magnetic interaction between the tyrosyl radical and the metal center. At 95 K, the tyrosyl radical in Fe-R2F from C. ammoniagenes has a value \( P_{0.5} = 1.3 \) mW (\( b = 1 \)). The corresponding value for S. typhimurium is \( P_{0.5} = 3.7 \) mW (\( b = 1 \)). The signal from C. ammoniagenes is thus saturating at lower power, indicating a slightly weaker coupling. Still, these signals saturate at the same order of magnitude of microwave power. At 95 K the \( P_{0.5} \) for the class Ia E. coli R2 tyrosyl radical was determined to 91 mW, which is almost two orders of magnitude larger than for C. ammoniagenes R2F, indicating a much faster relaxation of the tyrosyl radical due to interaction with the neighboring diiron site. However, also the \( b \) value has changed to \( b = 2 \), indicative of changed relaxation processes at higher temperatures. E. coli has typically \( b = 1.3 \) and \( P_{0.5} = 12 \) mW at 77 K, and an isolated tyrosyl radical has \( P_{0.5} = 0.2 \) mW (\( b = 1.0 \)) at 77 K and \( P_{0.5} = 0.36 \) mW (\( b = 1.0 \)) at 93 K (30). The determined values of \( P_{0.5} \) for C. ammoniagenes and S. typhimurium clearly show that they are interacting with the neighboring diiron center. Whether the observed difference is the result of different magnetic properties of the di-iron site and/or different geometry of the radical-diiron site will have to await magnetic susceptibility measurements and structural determinations.

Attempts to Form a Tyrosyl Radical by Oxidation of the Metal Centers in C. ammoniagenes R2F

Because the EPR spectrum of the C. ammoniagenes R2F sample showed a coupled Mn(II)-Mn(II) signal, we explored several different conditions to oxidize the metal chemically. It has been shown previously that \( \text{H}_2\text{O}_2 \) is capable of generating the tyrosyl radical in E. coli metR2, containing a diferric metal site but lacking the radical (17, 31). Addition of \( \text{H}_2\text{O}_2 \) to Mn-R2F oxidized the metal center as seen with UV-visible spectroscopy but did not generate a radical (data not shown), suggesting that oxidation of the manganese site and formation of the tyrosyl radical are uncoupled processes. Another well-characterized manganese enzyme is Mn-catalase. Its dinuclear Mn(II) center changes oxidation state transiently during the enzyme reaction, but higher valency states can also be obtained chemically, e.g. by addition of \( \text{KIO}_4 \) (25, 32). Addition of \( \text{KIO}_4 \) to Mn-R2F generated a radical but the Mn(II)-Mn(II) EPR signal was not changed, also suggesting that generation of radical is not related to oxidation of the manganese site. Addition of the reducing agent ascorbate to C. ammoniagenes Mn-R2F generated some radical and an UV-visible absorption indicative of an oxidized metal site. Again, EPR analysis of the oxidized sample showed an increase of radical but no change in the feature of the manganese center. Presumably, ascorbate did not interact directly with the metal sites but reduced ferric iron to ferrous, thereby making generation of an active metal site with radical possible.

DISCUSSION

More than two decades of literature have emphasized the typical manganese requirement of some Gram-positive bacteria (10, 33, 34). Manganese starvation studies have shown that the primary lesion in these bacteria is impaired replication (33). Growth arrest caused by starvation of manganese is reversible, and addition of manganese can resume normal
growth. This is not the case when iron is added to the manganese-starved bacteria (34). Thus iron and manganese are not interchangeable in vivo. Based on these observations, the nature of the C. ammoniagenes RNR and in particular its metal cofactor have attracted much attention, and it has been claimed that the C. ammoniagenes RNR constitutes a separate class of manganese-containing RNRs (35, 36).

The primary amino acid sequence of the C. ammoniagenes R2F protein is 67% identical to that of the S. typhimurium R2F protein. All the amino acid residues identified in the three-dimensional structure of the S. typhimurium protein that are metal ligands (9) are also found in C. ammoniagenes R2F (6). The metal site of S. typhimurium R2F can be considered to be a model for the C. ammoniagenes site, and we therefore investigated whether the naturally occurring diiron site in S. typhimurium R2F could be substituted by manganese and if the manganese protein would show activity. Reconstitution of apoR2F with manganese gave an inactive protein, whereas reconstitution with iron restored activity. Reconstitution with a mixture of iron and manganese resulted in partial, but low activity. From this it is a reasonable assumption that only the S. typhimurium R2F dimers with a full complement of four iron ions were enzymatically active. In analogy, we suggest that the low activity of R2F isolated from C. ammoniagenes after exposure to manganese also resided in a few molecules of R2F-retaining diiron sites.

Hitherto studied samples of C. ammoniagenes R2F have all been prepared from cells grown under manganese starvation and subsequently recovered in manganese-supplemented medium prior to harvest. These are of course not critical growth conditions for establishing the nature of the metal site. In preliminary experiments, we observed that a crude extract of C. ammoniagenes RNR prepared from a manganese-starved culture showed the same specific activity as a crude extract of S. typhimurium R2F (6). However, several attempts to obtain highly purified R2F from C. ammoniagenes grown in rich media without metal ion additions were unsuccessful in our hands. We therefore used the cloned C. ammoniagenes nrdF gene to construct an efficient expression system for C. ammoniagenes R2F in E. coli.

Both in vivo studies in the E. coli overproduction system and in vitro studies on overproduced R2F favor the idea that enzymatically active C. ammoniagenes R2F is an iron protein. In vivo, a high content of a typical tyrosyl radical signal was generated in iron-supplemented cultures, but not in manganese-supplemented cultures. Likewise, by studying the relation between metal content and enzyme activity in the overproduced purified protein, we can conclude that an enzymatically active iron center is self-assembled in C. ammoniagenes R2F. Formation of the manganese center, on the other hand, does not seem to correlate with enzyme activity. It was possible to reconstitute the R2F sample to higher radical content and specific activity in the presence of ferrous ions. Similar results were also obtained for the class Ib R2F protein from S. typhimurium.

Enzymatically active C. ammoniagenes R2F has a typical tyrosyl radical UV-visible absorption spectrum with a sharp 408-nm band and an EPR spectrum identical to that of S. typhimurium R2F (19, 20). The radical is susceptible to the radical scavenger hydroxyurea, and the saturation behavior of the tyrosyl radical suggests that the radical is in magnetic interaction with an antiferromagnetically coupled diferric center. Collectively, these results show that active C. ammoniagenes R2F is a typical class Ib RNR with a diferric metal cofactor coupled to a stable tyrosyl radical.

The EPR signal of the manganese-containing R2F from C. ammoniagenes has not been seen in ribonucleotide reductase before. When the class Ia R2 from E. coli is loaded with manganese, it shows a broad EPR singlet at 77 K centered at g = 2.0 (37). The manganese-reconstituted R2F from S. typhimurium used in this study showed similar features as R2F from C. ammoniagenes. Thus, the coupling between manganese atoms seems to be a property of class Ib R2F.

It has been claimed that the metal center in the class Ib enzyme of C. ammoniagenes is a mononuclear manganese center (13). The most convincing argument against a mononuclear center is the conservation of the metal ligands. Sequence alignment of C. ammoniagenes and S. typhimurium, whose structure has been determined, argues for a preserved structure of the site. We have shown that R2F from C. ammoniagenes and S. typhimurium can bind at least 1.6 manganese/monomer. The class Ia R2 from E. coli (37) can also bind approximately 2 manganese/monomer (Table III). The different class I enzymes have conserved metal ligands, and the amino acids in a sphere with a 9-Å radius around the metal center are the same (9). This means that very fine differences could regulate the properties of the metal center. As seen in Table III the enzyme activity of class Ib is lower than that of class Ia. Most likely, we have not yet found the optimal reaction conditions for class Ib.

For instance, DTT is used as a reductant in the activity assay instead of the class Ib NrdH-redoxin, which gives higher activity (6). However, because many species have more than one type of ribonucleotide reductase, class Ib might be expressed under specific conditions where a lower activity is sufficient. In addition, the enzyme activity of C. ammoniagenes class Ib RNR was measured either at a suboptimal amount of iron or a suboptimal amount of R1E. Our previous sequence homology studies have shown that C. ammoniagenes RNR is a class Ib enzyme (6). Our current demonstration that C. ammoniagenes R2F is active with iron corroborates its classification as a true class Ib enzyme.

### Table III

Radical content and specific activity of E. coli, S. typhimurium, and C. ammoniagenes R2/R2F proteins with iron or manganese metal centers

<table>
<thead>
<tr>
<th>Type of protein</th>
<th>Metal center</th>
<th>Fe/monomer</th>
<th>Mn/monomer</th>
<th>Tyr/monomer</th>
<th>Specific activitya</th>
</tr>
</thead>
<tbody>
<tr>
<td>E. coli R2</td>
<td>Fe</td>
<td>1.4–1.5</td>
<td>ND</td>
<td>0.5</td>
<td>4000–6000</td>
</tr>
<tr>
<td></td>
<td>Mn</td>
<td>ND</td>
<td>1.8–1.9</td>
<td>0</td>
<td>ND</td>
</tr>
<tr>
<td>S. typhimurium R2F</td>
<td>Fe</td>
<td>1.4–1.6</td>
<td>ND</td>
<td>0.2–0.5</td>
<td>325–660</td>
</tr>
<tr>
<td></td>
<td>Mn</td>
<td>ND</td>
<td>1.6</td>
<td>3</td>
<td></td>
</tr>
<tr>
<td>C. ammoniagenes R2F</td>
<td>Fe</td>
<td>0.5–1.5</td>
<td>0.09</td>
<td>0.05–0.2</td>
<td>36–48</td>
</tr>
<tr>
<td></td>
<td>Mn</td>
<td>0.5</td>
<td>1.6</td>
<td>9</td>
<td></td>
</tr>
</tbody>
</table>

a Specific activities were obtained at 25 °C for E. coli, at 30 °C for C. ammoniagenes, and at 37 °C for S. typhimurium.

References 1, 24, and 38.

In vivo studies in the C. ammoniagenes R2F favor the idea that enzymatically active C. ammoniagenes R2F is an iron protein. In vivo, a high content of a typical tyrosyl radical signal was generated in iron-supplemented cultures, but not in manganese-supplemented cultures. Likewise, by studying the relation between metal content and enzyme activity in the overproduced purified protein, we can conclude that an enzymatically active iron center is self-assembled in C. ammoniagenes R2F.

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1. Tulokhonova and B.-M. Sjöberg, unpublished result.
2. Y. Huque, unpublished results.
REFERENCES