

1 **Effects of steel slag application on greenhouse gas emissions and**  
2 **crop yield over multiple growing seasons in a subtropical paddy**  
3 **field in China**

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24 **ABSTRACT**

25 Asia is responsible for over 90% of the world's rice production and hence plays a key  
26 role in safeguarding food security. With China being one of the major global  
27 producers and consumers of rice, achieving a sustainable balance in maximizing crop  
28 productivity and minimizing greenhouse gas emissions from paddy fields in this  
29 country becomes increasingly important. This study examined the effects of applying  
30 steel slag, a residual product derived from the steel industry, on crop yield and CH<sub>4</sub>  
31 and N<sub>2</sub>O emissions over multiple growing seasons in a Chinese subtropical paddy  
32 field. Average CH<sub>4</sub> emission was considerably higher during the periods of rice crop  
33 growth compared to that during the periods of fallowing and vegetable crop growth,  
34 regardless of the amount of steel slag applied. When compared to the controls,  
35 significantly lower mean emissions of CH<sub>4</sub> (1.03 vs. 2.34 mg m<sup>-2</sup> h<sup>-1</sup>) and N<sub>2</sub>O (0.41  
36 vs. 32.43 μg m<sup>-2</sup> h<sup>-1</sup>) were obtained in plots with slag addition at a rate of 8 Mg ha<sup>-1</sup>  
37 over the study period. The application of slag at 8 Mg ha<sup>-1</sup> increased crop yields by  
38 4.2 and 9.1% for early and late rice crops, respectively, probably due to the higher  
39 availability of inorganic nutrients such as silicates and calcium from the slag. Slag  
40 addition had no significant effect on the concentrations of heavy metals in either the  
41 soil or the rice grains, although a slight increase in the levels of manganese and cobalt  
42 in the soil and a decrease in the levels of manganese and zinc in the rice grains were  
43 observed. Our results demonstrate the potential of steel slag as a soil amendment in  
44 enhancing crop yield and reducing greenhouse gas emissions in subtropical paddy  
45 fields in China, while posing no adverse short-term impacts on the concentrations of  
46 heavy metals in the soil or the rice grains. However, long-term implications of this  
47 management practice and the cost/benefit remain unknown, so further studies to  
48 assess the suitability at large scale are warranted.

49 *Keywords:* China; Greenhouse gas; Methane emission; Nitrous oxide emission; Paddy  
50 fields; Rice; Steel slag; Yield

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## 71 **1. Introduction**

72 Methane (CH<sub>4</sub>) and nitrous oxide (N<sub>2</sub>O) are important greenhouse gases that together  
73 account for about 20% of the global greenhouse effect (Smith et al., 2007). From 1990  
74 to 2005, agricultural CH<sub>4</sub> and N<sub>2</sub>O emissions have increased by about 17% (IPCC,  
75 2007). Rice production in particular can lead to substantial emissions of both CH<sub>4</sub> and  
76 N<sub>2</sub>O into the atmosphere, owing to the dominance of a flooded environment, and the  
77 large inputs of nitrogen from chemical fertilizers and manure, respectively (FAO,  
78 2003), which together exacerbate the problem of global climate change (van  
79 Groenigen et al., 2013). Global rice production is projected to increase from 473  
80 million tonnes in 1990 to at least 781 million tonnes by 2020 (IRRI, 1989). Given that  
81 the paddy fields in China account for 23% of all cultivated lands in the country, and  
82 nearly 20% of the world's total rice production area (Frolking et al., 2002), it is of  
83 national and global significance to examine the dynamics of greenhouse gas emissions  
84 from the Chinese paddy fields for their implications to both atmospheric chemistry  
85 and climate change (Hou et al., 2012).

86 Recently, a number of agricultural management strategies have been proposed,  
87 including the development of new rice varieties (Ma et al., 2012), as well as the  
88 selection of appropriate water management approaches (Ma et al., 2013), cultivation  
89 methods (Liu et al., 2013), and fertilisation schemes (Linguist et al., 2012; Liang et al.,  
90 2013), in an attempt to boost up rice yield and mitigate greenhouse gas emissions.  
91 Meanwhile, the application of exotic materials, for example, biochar (Zhang et al.,  
92 2010), steel slag (Wang et al., 2012), and straw rice (Müller-Stöver et al., 2012), is  
93 also a typical method of improving the soil quality and productivity of paddy fields.  
94 Steel slag, a residual product of the steel industry, contains high concentrations of  
95 electron acceptors such as active and free oxide forms of iron. While slag application

96 has been proven to be effective in reducing CH<sub>4</sub> emissions from the temperate paddy  
97 fields (Furukawa and Inubushi, 2002; Ali et al., 2008), its effectiveness in mitigating  
98 N<sub>2</sub>O emissions is not clear. Moreover, steel slag is thus far less commonly applied in  
99 the subtropical region compared to the temperate counterpart. With 90% of the paddy  
100 fields in China being located in the subtropics, such as in Fujian, Jiangxi, and Hunan  
101 provinces, there is a need to develop a better understanding of the effects of steel slag  
102 additions on the yield and greenhouse gas emissions from the subtropical Chinese  
103 paddy fields. Furthermore, it is important to assess the impacts of steel slag  
104 applications on the heavy metal contents in the rice grains and paddy soils, which are  
105 largely unknown at the moment but can have considerable health implications.

106 In a previous study, we found that steel slag was an effective amendment in  
107 reducing CH<sub>4</sub> flux and increasing rice yields of a subtropical paddy field in Fujian  
108 Province of China over a short growing season (Wang et al., 2014). However, it is not  
109 known whether slag application would affect N<sub>2</sub>O emissions and whether the  
110 beneficial effects arising from such addition would persist for more than one growing  
111 season and would not negative short-term impacts on the concentrations of heavy  
112 metals in the soil or the rice grains. This study aims to fill this knowledge gap by: (1)  
113 determining the response of CH<sub>4</sub> and N<sub>2</sub>O emissions to steel slag application over  
114 multiple growing seasons; (2) assessing the impacts of slag addition on crop  
115 productivity; and (3) determining the heavy metal concentrations in paddy soils and  
116 rice grains following slag application. The steel slag used in this study is derived from  
117 the steel industry, and contains high levels of iron that can serve as an alternative  
118 electron acceptor and potentially reduce CH<sub>4</sub> and N<sub>2</sub>O production (Huang et al., 2009).  
119 It is also rich in silicon, calcium, and potassium, which are essential nutrients for rice  
120 growth (Luo et al., 2002).

## 121 **2. Materials and methods**

### 122 *2.1 Experimental site*

123 All field experiments were carried out in the Wufeng Agronomy Field of the Fujian  
124 Academy of Agricultural Sciences (26.1°N, 119.3°E; Fig. 1) in the subtropical region  
125 of southeastern China. This field was managed following the common practice of  
126 growing one crop in each of the three growing seasons over a year, including two  
127 successive rice crops (early rice and late rice) followed by a vegetable (lettuce) crop,  
128 with intervening periods of drainage. The first early rice crop, the late rice crop, the  
129 vegetable crop, and the second early rice crop were grown during the period of 16  
130 April-17 July 2011, 1 August-5 November, 2011, 17 December 2011-8 March 2012,  
131 and 11 April-13 July 2012, respectively. The whole study period lasted for 448 days.  
132 The site was flooded and drained during the growth of rice and vegetable,  
133 respectively.

134 The soil of the paddy field was moist, poorly drained, and had a ratio of sand :  
135 silt : clay content of 28:60:12 (Wang et al., 2013). The bulk density of the soil before  
136 the start of this study was  $1.1 \text{ g cm}^{-3}$ . Moreover, the soil had a pH value (1:5 with  
137  $\text{H}_2\text{O}$ ) of 6.5, and concentrations of organic carbon, total nitrogen, and total  
138 phosphorus of  $18.1 \text{ g kg}^{-1}$ ,  $1.2 \text{ g kg}^{-1}$ , and  $1.1 \text{ g kg}^{-1}$ , respectively (Wang et al., 2012).  
139 The water level was maintained at 5-7 cm above the soil surface during the rice  
140 growth periods by means of an automatic water-level controller, and the paddy field  
141 was drained two weeks before harvesting.

142

143 *2.2 Experimental design and treatment application*

144 The experimental field had three independent replicate blocks, with each of them  
145 containing four treatment plots (50 m<sup>2</sup> each) being arranged in a randomised block  
146 design. We thus had three replicates for each treatment. The steel slag used in this  
147 study was granular, smaller than 2 mm in diameter, had a pH of 8.5, and was  
148 composed mainly of CaO (34.9%), SiO<sub>2</sub> (40.7%), and Fe<sub>2</sub>O<sub>3</sub> (4.8%) (Wang et al.,  
149 2012), which was similar to those used in previous studies (Ali et al., 2008). The  
150 slag was applied to the paddy field at 0 (control), 2, 4, and 8 Mg ha<sup>-1</sup>, which was  
151 equivalent to the addition of 0, 67.2, 134, and 269 kg Fe ha<sup>-1</sup>, respectively, two days  
152 before rice transplantation for the first early rice crop. All control and treatment plots  
153 followed the same scheme of crop management, including conventional fertilisation.  
154 “For fertilization, the common practice among farmers in Fujian, China, was  
155 followed. Applied chemical fertilizers consisted of using a mix of complete fertilizer  
156 (N : P<sub>2</sub>O<sub>5</sub> : K<sub>2</sub>O=16% : 16% : 16%, Keda Fertilizer Co., Ltd., Shandong, China) and  
157 urea (46% N). Fertilizers were applied at a rate of 95, 70, and 70 kg ha<sup>-1</sup> (N, P<sub>2</sub>O<sub>5</sub>,  
158 and K<sub>2</sub>O respectively) to the rice crops in each one of the three studied phases:  
159 before transplantation, at the tillering stage, and at the panicle-formation stage. In  
160 each phase the fertilizer was applied split in three times. Fertilizer was applied in 13  
161 April, 2011, 27 April, 2011, 15 June, 2011 for the first early rice crop, in 29 July,  
162 2011, 10 August, 2011, 9 October, 2011 for the late rice crop, and in 8 April, 2012,  
163 20 April, 2012, 12 June, 2012 for the second early rice crop. Chemical fertiliser (N,  
164 P<sub>2</sub>O<sub>5</sub>, and K<sub>2</sub>O at a rate of 200, 158, and 141 kg ha<sup>-1</sup>, respectively) was applied once

165 to the vegetable (lettuce) crop on 17 December, 2011. Fertilizer was applied on dry  
166 soil and incorporated and puddled, and the other two additional fertilizer broadcasts  
167 were applied on flooded water. For the vegetable the fertilizer was applied on dry  
168 soil and incorporated and puddled only one time. The field was ploughed to a depth  
169 of 15 cm with a mound board plough. The plough dates for the first early rice crop,  
170 the late rice crop, the vegetable crop, and the second early rice crop were 12 April,  
171 2011, 28 July, 2011, 10 December 2011, 8 April, 2012, and the puddling dates for  
172 the first early rice crop, the late rice crop, and the second early rice crop were 14  
173 April, 2011, 30 July, 2011, 9 April, 2012. The rice and lettuce varieties were  
174 Hesheng 10 and Kexing 5, respectively, and the spacing of the individual rice and  
175 lettuce plants was 14 x 28 cm and 40 x 60 cm, respectively. The yields of rice and  
176 lettuce were recorded after harvesting.

### 177 178 *2.3 Measurements of CH<sub>4</sub> and N<sub>2</sub>O emissions*

179 Static closed chambers were used to measure the emissions of CH<sub>4</sub> and N<sub>2</sub>O (Datta  
180 et al., 2013). The chambers were made of PVC and were constructed in two sections:  
181 a removable upper transparent compartment (100 cm in height, 30 cm in width, and  
182 30 cm in length) placed on a permanently installed bottom collar. The bottom collars  
183 were 10 cm in height, 30 cm in width, and 30 cm in length, and inserted into the soil  
184 leaving a 2-cm collar protruding above the soil surface. Each chamber was equipped  
185 with a circulating fan for mixing gases, a thermometer to monitor temperature  
186 changes during the period of gas sampling, and a gas sampling port with a septum.



187 Three chambers were deployed in each of the plots for all four crops, thus a total of  
188 twelve chambers were deployed for the four treatments. The chambers were installed  
189 at the same places every time that GHG sampling was conducted. Each chamber  
190 covered three rice plants and one lettuce plant for the rice and vegetable crops,  
191 respectively. Wooden bridges were constructed in the study area to minimise soil  
192 disturbance during flux measurement.

193 Gas samples from the chambers were collected once a week during the first early  
194 paddy and late rice crops, and every two weeks during other periods over the study  
195 period. A 100-ml plastic syringe equipped with a three-way stopcock was used to  
196 collect gas samples through the septum of the sampling port at 0, 15, and 30 minutes  
197 after chamber deployment. Three replicate samples were collected at each time from  
198 each chamber. To represent the average daily flux of CH<sub>4</sub> and N<sub>2</sub>O, samples were  
199 collected twice a day, at 08:00h and 12:00h (Wang and Shangguan, 1995). The  
200 collected gas samples were immediately transferred to 100-ml air-evacuated  
201 aluminium foil bags (Delin gas packaging Co., Ltd., China), sealed with a butyl  
202 rubber septum, and transported to the laboratory for analysis.

203

#### 204 *2.4 Determination of CH<sub>4</sub> and N<sub>2</sub>O concentrations*

205 The concentrations of CH<sub>4</sub> and N<sub>2</sub>O in the collected air samples were measured by  
206 gas chromatography (Shimadzu GC-2014, Japan) packed with a flame ionisation  
207 detector (FID) and an electron capture detector (ECD). The protocol is described in  
208 detail by Tong et al. (2013).

209

## 210 *2.5 Calculation of CH<sub>4</sub> and N<sub>2</sub>O emissions*

211 CH<sub>4</sub> and N<sub>2</sub>O emission rates in the paddy fields were calculated from the increase in  
212 CH<sub>4</sub> and N<sub>2</sub>O concentrations per unit surface area of the chamber for a specific time  
213 interval. A closed-chamber equation (Ali et al., 2008) was used to estimate CH<sub>4</sub> and  
214 N<sub>2</sub>O fluxes from the three replicates of each treatment:

$$215 \quad F = \frac{M}{V} \cdot \frac{dc}{dt} \cdot H \cdot \left( \frac{273}{273 + T} \right)$$

216 where  $F$  is the CH<sub>4</sub> or N<sub>2</sub>O flux (mg/μg CH<sub>4</sub>/N<sub>2</sub>O m<sup>-2</sup> h<sup>-1</sup>),  $M$  is the molecular weight  
217 of the gas (16 and 44 g mol<sup>-1</sup> for CH<sub>4</sub> and N<sub>2</sub>O, respectively),  $V$  is the molar volume  
218 of gas in a standard state (22.4 l mol<sup>-1</sup>),  $dc/dt$  is the variation ratio of CH<sub>4</sub> and N<sub>2</sub>O  
219 concentrations (μmol mol h<sup>-1</sup>),  $H$  is the height of the chamber above the water surface  
220 (m), and  $T$  is the air temperature inside the chamber (°C).

221

## 222 *2.6 Global warming potential (GWP)*

223 To estimate GWP, CO<sub>2</sub> is typically taken as the reference gas, and a change in the  
224 emission of CH<sub>4</sub> or N<sub>2</sub>O is converted into “CO<sub>2</sub>-equivalents” (Hou et al., 2012). The  
225 GWP for CH<sub>4</sub> is 34 (based on a 100-year time horizon and a GWP for CO<sub>2</sub> of 1), and  
226 the GWP for N<sub>2</sub>O is 298 (Myhre et al., 2013). The GWP of the combined emission of  
227 CH<sub>4</sub> and N<sub>2</sub>O was calculated using the equation (Ahmad et al., 2009):

$$228 \quad \text{GWP} = \text{cumulative CH}_4 \text{ emission} \times 34 + \text{cumulative N}_2\text{O emission} \times 298$$

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## 230 *2.7 Measurement of soil properties*

231 On each sampling date, *in situ* measurements of soil salinity (mS cm<sup>-1</sup>), soil redox  
232 potential (Eh, in mV), soil pH, and soil temperature at a depth of 20 cm were also

233 made. Soil Eh, pH, and temperature were measured with a Eh/pH/Temperature meter  
234 (IQ Scientific Instruments, USA), and soil salinity was measured using a 2265FS EC  
235 Meter (Spectrum Technologies Inc., USA). Soil chemical properties were measured  
236 following Lu et al. (1999). The iron content was measured every two weeks. Fresh  
237 soil was digested with 1M HCl, and the Fe<sup>2+</sup> and total Fe contents were determined  
238 using the 1,10-phenanthroline and spectrometric method. The Fe<sup>3+</sup> content was  
239 calculated as the difference between the Fe<sup>2+</sup> and total Fe content. The  
240 concentrations of Mn, Ni, Zn, Cr, Pb, Cu, and Co in the grains and soils of the first  
241 early rice crop after harvesting were determined by atomic absorption  
242 spectrophotometry (Soylak and Aydin, 2011). All soil properties were measured by  
243 three repeats for each treatment.

#### 245 2.8 Statistical analysis

246 All statistical analyses were performed using PASW Statistics 18.0 software (IBM  
247 SPSS Inc., Chicago, USA). The data were checked for assumptions of normality and  
248 homogeneity of variance, and if necessary, were log-transformed before further  
249 analyses. Pearson correlation analysis was used to determine the relationships of  
250 CH<sub>4</sub> and N<sub>2</sub>O emissions with soil properties. Repeated-measures analysis of variance  
251 (RM-ANOVAs) was conducted to determine the effect of steel slag amendment and  
252 measurement dates on CH<sub>4</sub> and N<sub>2</sub>O emissions and other environmental parameters.  
253 We used one-way ANOVA with Tukey's post-hoc test to assess the treatment effects  
254 on the concentrations of heavy metals in the rice grains and the soil.

## 255 **3. Results**

256

### 257 *3.1 CH<sub>4</sub> emissions*

258 The emission of CH<sub>4</sub> varied significantly among measurement dates as well as among  
259 the amount of steel slag added ( $P < 0.001$ , Table 1). The rate of CH<sub>4</sub> emission in the  
260 initial period (1-22 days after rice transplanting) was consistently low for either the  
261 first early paddy, late paddy, or the second early paddy (0.04-0.55 mg m<sup>-2</sup> h<sup>-1</sup>), but  
262 then increased quickly (1.03-16.90 mg m<sup>-2</sup> h<sup>-1</sup>) following the development of  
263 anaerobic soil conditions and the growth of rice (Fig. 2). Regardless of the amount of  
264 steel slag applied, average CH<sub>4</sub> emission was considerably higher during the periods  
265 of rice crop growth (0.90-3.66 mg m<sup>-2</sup> h<sup>-1</sup>) compared to that during the periods of  
266 fallowing and vegetable crop growth ( $< 0.12$  mg m<sup>-2</sup> h<sup>-1</sup>) (Table 2).

267 The average CH<sub>4</sub> emission rate for each of the growing seasons was significantly  
268 higher in the controls compared to the plots with steel slag application ( $P < 0.05$ ),  
269 except during the growth of vegetable crop (Table 2). Moreover, the mean CH<sub>4</sub>  
270 emission over the whole study period was 36.3, 52.1, and 56.0% lower in the plots  
271 amended with 2, 4, and 8 Mg ha<sup>-1</sup> of slag, respectively, when compared to that in the  
272 controls (Table 2). The cumulative CH<sub>4</sub> emission rate over the study period was  
273 significantly lower in the plots receiving 4 and 8 Mg ha<sup>-1</sup> of slag compared to the  
274 controls and plots receiving only 2 Mg ha<sup>-1</sup> of slag ( $P < 0.01$ , Table 2).

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### 276 *3.2 N<sub>2</sub>O emissions*

277 The emission rates of N<sub>2</sub>O varied significantly among measurement dates and  
278 quantity of slag applied ( $P < 0.001$ , Table 1), with no interactions between the two  
279 factors. For both early and late rice crops, the rates of N<sub>2</sub>O emission in the treatments

280 plots reached a maximum ( $147\text{-}359 \mu\text{g m}^{-2} \text{h}^{-1}$ ) during the initial period after rice  
281 transplanting, and thereafter decreased quickly to low values ( $-89$  to  $-75 \mu\text{g m}^{-2} \text{h}^{-1}$ )  
282 until the rice was harvested (Figure 3). For the controls, mean  $\text{N}_2\text{O}$  fluxes were much  
283 higher during the fallow periods ( $119 \mu\text{g m}^{-2} \text{h}^{-1}$ ) than in other periods when rice and  
284 vegetable crops were grown ( $5\text{-}36 \mu\text{g m}^{-2} \text{h}^{-1}$ ) (Table 2).

285 The average rates of  $\text{N}_2\text{O}$  emission for each of the growing seasons were not all  
286 following the same patterns between the controls and treatment plots (Table 2).  
287 Meanwhile, when pooling all the data over the study period together, the cumulative  
288  $\text{N}_2\text{O}$  emission was significantly lower ( $0.21 \text{g m}^{-2}$ ) in the plots amended with  $8 \text{Mg ha}^{-1}$   
289 of slag than in those amended with 2 and  $4 \text{Mg ha}^{-1}$  of slag ( $9.82$  and  $10.02 \text{g m}^{-2}$ ,  
290  $P < 0.05$ ). The controls had a significantly higher cumulative  $\text{N}_2\text{O}$  flux of  $17.06 \text{g m}^{-2}$   
291 than all other treatment plots. The mean rates of  $\text{N}_2\text{O}$  emission over the study period  
292 were thus 42.4, 41.2, and 98.8% lower in the plots amended with 2, 4, and  $8 \text{Mg ha}^{-1}$   
293 of slag, respectively than in the control plots ( $P < 0.05$ , Table 2).

### 295 3.3 GWPs of $\text{CH}_4$ and $\text{N}_2\text{O}$ emissions

296 The GWPs for  $\text{CH}_4$  and  $\text{N}_2\text{O}$  emissions varied considerably with the growth of  
297 different crops as well as the level of slag application. The GWPs caused by  $\text{CH}_4$  was  
298 higher than that of  $\text{N}_2\text{O}$  emissions. When both  $\text{CH}_4$  and  $\text{N}_2\text{O}$  emissions were  
299 combined, the overall GWPs showed a decreasing trend with increasing levels of slag  
300 application, with a drop of 42.8, 52.5, and 63.4% in plots amended with 2, 4, and 8  
301  $\text{Mg ha}^{-1}$  of slag respectively, when compared to the controls (Table 3).

### 303 3.4 Soil parameters and their relationships with $\text{CH}_4$ and $\text{N}_2\text{O}$ emissions

304 Soil ferric concentration, temperature, salinity, Eh, and pH varied significantly  
305 throughout the year ( $P<0.001$ ; Fig. 4, Table 4). Soil  $\text{Fe}^{3+}$  concentrations and salinity  
306 increased significantly in the plots with slag amendment compared to those without  
307 ( $P<0.001$ ). Only soil temperature was significantly and positively correlated with  $\text{CH}_4$   
308 emissions in all the plots ( $r = 0.41-0.43$ ,  $P<0.05$ , Table 5). Soil salinity was  
309 significantly correlated with  $\text{CH}_4$  emission only in the control plots ( $r = 0.49$ ,  $P<0.01$ ,  
310 Table 5). No significant correlations were observed between  $\text{N}_2\text{O}$  emissions and any  
311 environmental variables, including soil ferric concentration, temperature, salinity, pH  
312 and Eh ( $P>0.05$ ).

313

### 314 *3.5 Response of crop production to slag application*

315 The yield of the first early rice crop was significantly higher ( $8.43\pm 0.09 \text{ Mg ha}^{-1}$ )  
316 when slag was applied at a rate of  $8 \text{ Mg ha}^{-1}$  than when lower rates were applied  
317 (yield 4.2% higher than the controls;  $P<0.05$ , Table 6). The yields of the late rice crop  
318 with slag added at a rate of 4 and  $8 \text{ Mg ha}^{-1}$  were  $8.08\pm 1.08$  and  $8.14\pm 0.48 \text{ Mg ha}^{-1}$ ,  
319 respectively, which were significantly higher than that in the controls by 8.3 and 9.1%,  
320 respectively ( $P<0.05$ ). Meanwhile, no significant changes in rice yields were observed  
321 when steel slag was applied at a rate of  $2 \text{ Mg ha}^{-1}$  only ( $P>0.05$ ). Slag application had  
322 no significant effects on the yields of both the second early paddy and the vegetable  
323 crops.

324

### 325 *3.6 Effects to slag application on heavy metal concentrations in rice grains and soils*

326 The mean concentration of Mn in the rice grains was 26.2% lower in the plots treated  
327 with  $8 \text{ Mg ha}^{-1}$  of slag than in the control plots ( $P<0.05$ , Table 7). The mean Zn  
328 concentrations in the grains were 41.4 and 39.9% lower in the plots treated with 4 and

329 8 Mg ha<sup>-1</sup> of slag, respectively, than in the control plots ( $P<0.05$ ). No significant  
330 differences in the concentrations of other heavy metals were observed between the  
331 treatment and control plots ( $P>0.05$ ).

332 In the soils, the mean concentrations of Mn were significantly higher in the plots  
333 treated with 4 and 8 Mg ha<sup>-1</sup> of slag by 35.5 and 45.1%, respectively, than in the  
334 controls ( $P<0.05$ ). The application of steel slag also increased the mean soil Co  
335 concentrations by 32.3, 35.9, and 30.5% when added at a rate of 2, 4, and 8 Mg ha<sup>-1</sup>,  
336 respectively ( $P<0.05$ ) (Table 7). We found no significant impacts of steel slag  
337 amendments on the concentrations of other heavy metals in the soils ( $P>0.05$ ) (Table  
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354 **4. Discussion**

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356 *4.1 The effect of steel slag on CH<sub>4</sub> emission*

357 In the control plots, the mean CH<sub>4</sub> emission was higher during the period of paddy  
358 growth than during the period of fallowing and vegetable crop growth, which was  
359 likely a result of water management. During the rice cultivation period, the paddy  
360 field was submerged in water, which facilitated the development of an anaerobic  
361 environment in soil that subsequently enhanced the production and emissions of CH<sub>4</sub>.  
362 In contrast, as the plots were drained during the fallow and vegetable crop growth  
363 periods, CH<sub>4</sub> emission was substantially reduced. This was in agreement with the  
364 results reported in previous studies (Minamikawa et al., 2014), in which a lowering of  
365 water table would lead to a decrease in the abundance of methanogenic archaeal  
366 population and hence CH<sub>4</sub> production, as well as an increase in the total abundance of  
367 methanotrophs and thus CH<sub>4</sub> oxidation (Ma and Lu, 2011). Furthermore, the higher  
368 mean CH<sub>4</sub> emission observed during the growth of late rice crop compared to the  
369 early rice crop could be attributed to the effects of temperature. In our study, the mean  
370 soil temperature during the late paddy period was higher than that of the early paddy  
371 (26.0 vs. 24.2 °C). We found a significant and positive correlation between soil  
372 temperature and CH<sub>4</sub> emission in the control plots (Table 5). When the temperature is  
373 low, CH<sub>4</sub> flux from the paddy field is constrained by the reduction of both microbial-  
374 mediated CH<sub>4</sub> production as well as CH<sub>4</sub> transport by physical and biological  
375 processes (Gaihre et al., 2013).

376 Our results strongly support the use of steel slag as an amendment to reduce CH<sub>4</sub>  
377 emissions from paddy fields, in accordance with the findings of similar studies  
378 conducted in other rice-producing countries (Furukawa and Inubushi, 2002; Ali et al.,



379 2008; Ali et al., 2014). The application of steel slag reduces CH<sub>4</sub> emissions mainly  
380 through increasing the Fe<sup>3+</sup> concentrations in soils during multiple growing seasons  
381 (Fig. 6, Table 3). When we applied 2, 4, and 8 Mg ha<sup>-1</sup> of steel slag in our plots, there  
382 was a corresponding supply of additional Fe at a rate of 67.2, 134, and 269 kg ha<sup>-1</sup>,  
383 respectively, mainly in the oxidised form such as Fe<sub>2</sub>O<sub>3</sub>. Given that Fe<sup>3+</sup> is  
384 thermodynamically more favourable alternative electron acceptor than CO<sub>2</sub> or acetate  
385 (Chidthaisong and Conrad, 2000), the increased availability of Fe<sup>3+</sup> helps to suppress  
386 the activity of methanogens, the addition of Fe<sup>3+</sup> increase the soil reduction capacity  
387 and becomes the main electron acceptor (van Bodegom and Stams, 1999), and iron  
388 reducing bacteria thus tend to outcompete methanogens (Andrews et al., 2013).  
389 Moreover, in addition to reducing methane synthesis, Fe<sup>3+</sup> can increase existing  
390 methane oxidation and the slag porous structure increases soil microbial activity of  
391 methanotrophs. Previous studies have also observed that the slag addition reduces the  
392 methane production and increases the methane oxidation thus reducing methane  
393 emission (Ali et al., 2008; Wang et al., 2014).

394 We also found that the addition of increasing amount of steel slag to our plots  
395 had a greater effect on reducing CH<sub>4</sub> flux during the rice crop period (first and second  
396 early rice crop and late rice crop) than in the vegetable crop period (Table 2). This  
397 could be attributed to the greater initial CH<sub>4</sub> emission from the flooded sites during  
398 rice cultivation than from the drained sites during the growth of vegetable crop (Wang  
399 et al., 2012; Ali et al., 2014). The additional supply of Fe<sup>3+</sup> was able to suppress  
400 methanogenesis to a much greater extent in the anaerobic soils that were predominant  
401 during paddy growth. Moreover, we found that steel slag was more effective in  
402 reducing CH<sub>4</sub> emission from our subtropical paddy fields (a reduction of 52.1% with  
403 4 Mg hm<sup>-2</sup> steel slag amendment) compared to those located in the temperate regions

404 of Japan (a reduction of less than 35% with over 10 Mg hm<sup>-2</sup> steel slag amendment,  
405 Furukawa and Inubushi, 2002) and South Korea (a reduction of 16-20% with 4 Mg  
406 hm<sup>-2</sup> steel slag amendment, Ali et al., 2008), which might be related to the difference  
407 in mean temperature between these regions. Lovley (1991) found via experiments  
408 with soil cultures that the optimum temperature for the reduction of Fe<sup>3+</sup> was between  
409 32 and 41°C. While the mean temperature of our study site in the subtropical Fujian  
410 province in China was within the optimum temperature range for iron reduction, the  
411 average temperature during the growing season was much lower in South Korea and  
412 Japan, leading to lower rates of iron reduction and hence less efficient suppression of  
413 CH<sub>4</sub> emissions from these temperate sites following slag amendments (Furukawa and  
414 Inubushi, 2002; Ali et al., 2008).

415 Furthermore, in our study, the average CH<sub>4</sub> emissions from rice growth period in  
416 control plots (1.62-3.66 mg m<sup>-2</sup> h<sup>-1</sup>) were lower than in a of Japanese paddy field (24.8  
417 mg m<sup>-2</sup> h<sup>-1</sup>, Lou et al., 2008), and similar than in an India paddy field(1.9-5.7 mg m<sup>-2</sup>  
418 h<sup>-1</sup>, Bhattacharyya et al., 2012).

#### 419 420 *4.2 The effect of steel slag on N<sub>2</sub>O emission*

421 N<sub>2</sub>O emissions from paddy fields varied considerably among the measurement dates  
422 and crop periods. In the control plot, the highest N<sub>2</sub>O emissions were observed in the  
423 fallow period, which might be related to the increased oxidation of NH<sub>4</sub><sup>+</sup> in response  
424 to the drainage of paddy fields. As the nitrogenous fertilizer that we applied to our  
425 plots contained mostly NH<sub>4</sub><sup>+</sup> rather than NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup> was the major form of nitrogen  
426 found in our paddy soils. Also, the prolonged submergence of soil during crop  
427 cultivation would have reduced most of the soil NO<sub>3</sub><sup>-</sup> present, leading to a further  
428 increase in NH<sub>4</sub><sup>+</sup> concentration. Hence, when the field was drained for harvesting,

429 there existed an abundant supply of  $\text{NH}_4^+$  in the soils that served as substrates for the  
430 production of  $\text{N}_2\text{O}$  (Rochette et al., 2010). Drainage during the fallow period would  
431 facilitate the diffusion of oxygen from the atmosphere into the soil, which promoted  
432 the oxidation of  $\text{NH}_4^+$  and hence production of  $\text{N}_2\text{O}$  as reported in previous studies  
433 (Yu et al., 2006; Rosamond et al., 2011).

434 We found significant differences in mean and cumulative  $\text{N}_2\text{O}$  emissions over  
435 the study period among the plots amended with different amounts of steel slag, with  
436 the highest and lowest emission rates being associated with a slag application rate of 0  
437 and  $8 \text{ Mg ha}^{-1}$ , respectively (Fig. 5, Table 2). The effect on reductions was so great  
438 that at a rate of  $8 \text{ Mg ha}^{-1}$  we observed negative values of  $\text{N}_2\text{O}$  emissions, showing  
439 that soil absorbed  $\text{N}_2\text{O}$ , which is consistent with the limiting role of N observed in  
440 these soils (Wang et al., 2014). Alternative periods of soil  $\text{N}_2\text{O}$  emission and  
441 absorption in wetlands have been observed in other studies, being related to  
442 fluctuations in environmental traits such as water content and temperature (Liu et al.,  
443 2003; Hao et al., 2006). Yet, in a single growing season, the variation in  $\text{N}_2\text{O}$  flux  
444 within a treatment group was quite large, and it was difficult to identify a discernible  
445 effect of slag application on mean  $\text{N}_2\text{O}$  fluxes. While Zhu et al. (2013) suggested soil  
446  $\text{Fe}^{3+}$  concentration being one of the most sensitive factors in regulating  $\text{N}_2\text{O}$  emissions  
447 from paddy fields, we failed to observe any significant correlations between  $\text{Fe}^{3+}$   
448 concentration and  $\text{N}_2\text{O}$  emissions at all levels of slag application. The absence of a  
449 consistent effect of steel slag addition on  $\text{N}_2\text{O}$  flux from the paddy field could  
450 possibly be attributed to the following reasons. Firstly, higher  $\text{Fe}^{3+}$  concentrations  
451 could enhance  $\text{N}_2\text{O}$  release to the atmosphere by inhibiting the enzymatic reduction of  
452  $\text{N}_2\text{O}$  in soils (Huang et al., 2009). Secondly, higher  $\text{Fe}^{3+}$  concentration is known to  
453 increase the production of hydroxylamine through the biological oxidation of

454 ammonia, which then further reacts chemically with  $\text{Fe}^{3+}$  to generate  $\text{N}_2\text{O}$  (Bengtsson  
455 et al., 2002). Thirdly, an increase in  $\text{Fe}^{3+}$  concentrations can in turn lead to an increase  
456 in  $\text{Fe}^{2+}$  concentrations through microbial reduction (Ali et al., 2008), which then  
457 further promote the reduction of nitrites to  $\text{N}_2\text{O}$  (Hansen et al., 1994). On the other  
458 hand, Noubactep (2011) found that an increase in  $\text{Fe}^{3+}$  concentration could lead to the  
459 suppression of microbe activities, including  $\text{N}_2\text{O}$  production. A previous study has  
460 reported both positive and negative correlations between  $\text{Fe}^{3+}$  concentrations and  $\text{N}_2\text{O}$   
461 production, which was a function of different soil conditions and hence the presence  
462 of various forms of  $\text{Fe}^{3+}$  (active,  $\text{Fe}^{3+}$ , and complex ferric oxide,  $\text{Fe}_2\text{O}_3$ ) (Zhu et al.,  
463 2013).

464 Furthermore, in our study, the average  $\text{N}_2\text{O}$  emissions from rice growth period in  
465 control plots, in the first early rice crop period ( $36.09 \mu\text{g m}^{-2} \text{h}^{-1}$ ) were higher than in  
466 the India paddy field, and in the late rice crop period ( $7.71 \mu\text{g m}^{-2} \text{h}^{-1}$ ) and second  
467 early rice crop period ( $5.04 \mu\text{g m}^{-2} \text{h}^{-1}$ ) the emissions were similar to the  $\text{N}_2\text{O}$   
468 emissions from the India paddy field ( $5\text{-}10 \mu\text{g m}^{-2} \text{h}^{-1}$ , Bhattacharyya et al., 2012).  
469 Despite the rice crop management in other countries such as Indonesia and Japan is  
470 different and not directly comparable, the  $\text{CH}_4$  and  $\text{N}_2\text{O}$  emissions during rice crop  
471 season are similar than those observed in our study (Hadi et al., 2010).

472 Overall, we found that  $\text{CH}_4$  emissions had a greater influence than  $\text{N}_2\text{O}$   
473 emissions on the total GWP of crop cultivation in our paddy fields. When both  $\text{CH}_4$   
474 and  $\text{N}_2\text{O}$  emissions were combined, the total GWPs showed a decreasing trend with  
475 increasing levels of slag application, which suggested that steel slag application at the  
476 level of  $8 \text{ Mg hm}^{-2}$  was best for reducing paddy field influence on climate warming.

477

478 *4.3 The effect of steel slag on rice and vegetable productivity and concentrations of*  
479 *heavy metals in rice grains and the soil*

480 An application of 8 Mg ha<sup>-1</sup> of slag significantly increased grain yield from 8.09 ±  
481 0.15 to 8.43 ± 0.09 Mg ha<sup>-1</sup> for the first early rice crop and from 7.46 ± 0.11 to 8.14 ±  
482 0.48 Mg ha<sup>-1</sup> for the late rice crop (Table 6), which could be explained by the  
483 following reasons. Firstly, the steel slag used in our study composed mainly of CaO  
484 (34.9%), SiO<sub>2</sub> (40.7%), and Fe<sub>2</sub>O<sub>3</sub> (4.8%), which are the sources of many essential  
485 nutrients for crop growth. In our previous work, we found an increase in the  
486 concentrations of soil nutrients, including available P<sub>2</sub>O<sub>5</sub> from 53 to 96 mg kg<sup>-1</sup>, and  
487 SiO<sub>2</sub> from 254 to 1232 mg kg<sup>-1</sup>, at the end of the first rice crop growth period in plots  
488 receiving 8 Mg ha<sup>-1</sup> of steel slag when compared to the controls (Wang et al., 2014).  
489 The increase in the availability of SiO<sub>2</sub> could increase crop yield by promoting  
490 photosynthesis, enhancing the resistance of crops to attacks by fungi and insects,  
491 increasing the tolerance of crops to drought and frost, and decreasing mineral toxicity  
492 (Yoshida, 1981; Deren et al., 1994). Secondly, since the steel slag used in our study  
493 had a pH of 8.5, the soil pH increased from 6.48 in the controls to 7.16 in the plots  
494 with steel slag amendments at a rate of 8 Mg ha<sup>-1</sup> (Wang et al., 2014). The close to  
495 neutral soil pH greatly increased the availability of phosphates for uptake by crop  
496 plants (Wang et al., 2014). Thirdly, the presence of silicate ions plays a role in  
497 displacing or desorbing phosphate from soils, which again increases the available  
498 phosphate concentrations (Shariatmadari et al., 1999; Lee et al., 2004). Overall, an  
499 increase in soil nutrient content was the main cause of improved crop yield following  
500 the application of steel slag in paddy fields (Ali et al., 2008). Moreover, the slag also  
501 can absorb and retain nutrients, slowing the nutrient release, preventing the leaching  
502 and consequent water eutrophication and improving the soil capacity to provide

503 nutrients to plant uptake (Kostura et al., 2005; Zhao, 2012). However, the yields of  
504 lettuce and the second early rice crop were not significantly enhanced by the  
505 application of slag at any levels (Table 5), which implied that steel slag might only  
506 provide a short-term improvement in soil nutrient availability and crop yield during  
507 the initial period after application, and that more frequent application of slag (at least  
508 twice a year) would be needed to improve the yield in multiple growing seasons.  
509 Moreover, this fertilization method could be applied at large scale at low cost because  
510  $7.82 \times 10^8$  t of steel were produced in China in 2013, and the amount of generated  
511 steel slag was 0.46 t per each ton steel produced (Xie and Xie, 2003). Thus, the total  
512 steel slag production amount was  $3.60 \times 10^8$  t in 2013, making very cheap its  
513 application in the paddy field.

514 Furthermore, we have determined the heavy metal contents in grain and soil  
515 subsequent to slag application. While the concentrations of Mn and Zn in rice grains  
516 were significantly lower in plots treated with  $8 \text{ Mg ha}^{-1}$  of slag, the difference in  
517 concentrations when compared to the controls was actually small and could be a result  
518 of dilution effect arising from an increase in biomass production. The observed  
519 concentrations of Zn in rice grains were lower than typical concentrations reported in  
520 the literature (Cakmak et al., 2004; Wissuwa et al., 2008). The observed  
521 concentrations of Mn in rice grains were also within the range reported in a previous  
522 study (Wang et al., 2009) and very much lower than the threshold values to be  
523 considered toxic to humans (Dube et al., 2002). The considerable loads of steel slag  
524 did not affect the status of heavy metals in rice grains in a way that could pose a risk  
525 of toxicity. The daily intake of rice necessary to reach the threshold of toxicity for Co,  
526 Ni, Cr, Cu, and Zn would be at least 5-10 kg (at an absorption efficiency of 100%),  
527 based on the data reported by international agencies (EPA, 1992; EFSA, 2006).

528 Similarly, the changes in total concentrations of heavy metals in the soil in  
529 response to slag amendment were small. Only the total Mn and Co concentrations in  
530 soils increased, especially in the plots treated with 4 and 8 Mg ha<sup>-1</sup> of slag, whereas no  
531 significant changes were observed in total soil Cr, Cu, Ni, Pb, and Zn concentrations.  
532 The increase in Mn and Co concentrations might have been due to the sorption of Mn  
533 and Co by the slag, which had a porous structure (He et al., 2013). Furthermore, the  
534 slag itself contained heavy metals, especially Mn, and hence served to increase the  
535 supply of these metals to the soil (Wang et al., 2012). The observed maximum values  
536 of total soil Mn and Co concentrations (314 and 12 mg kg<sup>-1</sup>, respectively), though, are  
537 well below the thresholds deleterious for growth in plants (Jugsujinda and Patrick,  
538 1993; Kapustka et al., 2006; Shanahan et al., 2007; Mico et al., 2008; Binner and  
539 Schenk, 2013).

540

## 541 **5. Conclusions**

542 This study has demonstrated the effectiveness of steel slag as an amendment to  
543 mitigate CH<sub>4</sub> and N<sub>2</sub>O emissions and increase grain yields over multiple growing  
544 seasons in the subtropical paddy fields in China, without causing adverse short-term  
545 impacts on the concentrations of heavy metals in the soil and grains. The application  
546 of steel slag had a significant and positive effect on crop yield especially in the initial  
547 period of addition, probably as a result of the increased availability of inorganic  
548 nutrients such as silicates and calcium. Also, the Fe concentrations in paddy soils  
549 significantly increased with the rate of steel slag application. The cumulative CH<sub>4</sub>  
550 emission rate over the study period was 52.1% and 56.0% lower in the plots receiving  
551 4 and 8 Mg ha<sup>-1</sup> of slag compared to the controls, while the cumulative N<sub>2</sub>O emission  
552 was lower in the plots amended with 8 Mg ha<sup>-1</sup> of slag than in any other plots. Based

553 on our findings, the addition of steel slag at a rate of 8 Mg ha<sup>-1</sup> would be able to  
554 provide the maximum environmental and economic benefits. Moreover, it might be  
555 better to apply steel slag again after two growing seasons in order to have a  
556 sustainable enhancement of crop yield because while in the first early rice crop the  
557 yield increase in response to the application of 8 Mg ha<sup>-1</sup> of steel slag was 4.2%, in the  
558 late rice crop yield increased 8.3% and 9.1% in response to the application of 4 and 8  
559 Mg ha<sup>-1</sup> of steel slag, respectively. Further studies should be conducted to determine if  
560 even higher dose of steel slag than that used in our study could lead to further  
561 reduction of greenhouse gas fluxes and increase in yield. Moreover, despite in this  
562 study at short-medium term no negative effects on heavy elements have been  
563 observed, the suitability of the application of this practice at large scale requires  
564 further studies to assess the long-term effects and also the economical cost/benefit  
565 dimension of the method.

566

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576

### 577 **Conflicts of Interest**

578 The authors declare no conflicts of interest.

579



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785 **Tables**

786 **Table 1**

787 Summary of the RM-ANOVAs for CH<sub>4</sub> and N<sub>2</sub>O flux in the different studied periods

788 and for the different steel slag amendments.

789

	<i>df</i>	<i>F</i>	<i>P</i>
CH <sub>4</sub> flux			
Steel-slag quantity	3, 8	19.22	<0.001
Days after amendment	48, 384	7.70	<0.001
Steel-slag quantity × Days after amendment	144, 384	1.28	0.212
N <sub>2</sub> O flux			
Steel-slag quantity	3, 8	2.89	<0.05
Days after amendment	48, 384	3.61	<0.001
Steel-slag quantity × Days after amendment	144, 384	0.94	0.532

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792 **Table 2**793 Mean of CH<sub>4</sub> and N<sub>2</sub>O flux in the different stages of the experimental period at the

794 various levels of steel slag application. Negative values indicate absorption by the soil.

795 The data on CH<sub>4</sub> flux in the first early rice crop are from Wang et al. (2014).

	Application level (Mg ha <sup>-1</sup> )			
	0	2	4	8
CH <sub>4</sub> flux (mg m <sup>-2</sup> h <sup>-1</sup> )				
First early rice crop	3.11±0.90a	2.29±0.64b	1.76±0.53b	1.59±0.42c
Late rice crop	3.66±1.38a	2.04±0.75b	1.32±0.37c	1.24±0.33c
Second early rice crop	1.62±0.60a	0.90±0.39b	1.07±0.48b	1.09±0.51b
Fallow periods	0.12±0.05a	0.09±0.04b	0.07±0.03b	0.07±0.05b
Vegetable crop	0.00±0.01a	0.00±0.01a	0.01±0.01a	0.01±0.01a
Total average	2.34±0.53a	1.49±0.32b	1.12±0.22c	1.03±0.19c
N <sub>2</sub> O flux (μg m <sup>-2</sup> h <sup>-1</sup> )				
First early rice crop	36.09±30.58a	28.54±18.30b	26.97±16.95b	12.14±19.08c
Late rice crop	7.71±8.16a	31.72±6.90b	4.35±5.27c	6.48±3.63a
Second early rice crop	5.04±30.62a	-15.43±15.41b	18.71±12.13c	-12.33±6.38b
Fallow periods	118.96±98.23a	9.94±15.72b	12.21±21.87b	-1.41±18.46c
Vegetable crop	27.41±24.25a	8.83±14.11b	38.01±31.15c	-20.73±29.03d
Total average values	32.43±15.57a	18.62±6.66b	19.04±7.62b	0.41±7.42c

796

797 Different letters within a row indicate statistical differences ( $P < 0.05$ )

798

799 **Table 3.**

800 Global warming potentials (GWPs, mean  $\pm$  SD) of CH<sub>4</sub> and N<sub>2</sub>O in the different  
801 stages of the experimental period at the various levels of steel slag application.  
802 Negative values indicate N<sub>2</sub>O absorption by the soil.

	Application level (Mg ha <sup>-1</sup> )			
	0	2	4	8
<b>CH<sub>4</sub> GWPs (CO<sub>2</sub>-eq)</b>				
First early rice crop	2648 $\pm$ 219a	1946 $\pm$ 155b	1497 $\pm$ 129b	1342 $\pm$ 101c
Late rice crop	3253 $\pm$ 350a	1817 $\pm$ 191b	1177 $\pm$ 94c	1103 $\pm$ 84c
Second early rice crop	1030 $\pm$ 109a	574 $\pm$ 71b	684 $\pm$ 88b	694 $\pm$ 93b
Fallow periods	119 $\pm$ 14a	92.6 $\pm$ 11.8b	71.8 $\pm$ 8.8b	76.8 $\pm$ 15.7b
Vegetable crop	-2.42 $\pm$ 0.09a	-0.44 $\pm$ 0.03a	3.20 $\pm$ 0.91a	4.52 $\pm$ 1.29a
Total	7045 $\pm$ 456a	4428 $\pm$ 272b	3434 $\pm$ 193c	3220 $\pm$ 170c
<b>N<sub>2</sub>O GWPs (CO<sub>2</sub>-eq)</b>				
First early rice crop	237 $\pm$ 57a	188 $\pm$ 34b	177 $\pm$ 32b	79.9 $\pm$ 15.8c
Late rice crop	60.1 $\pm$ 18.2a	247 $\pm$ 15b	33.9 $\pm$ 11.7c	50.5 $\pm$ 8.1a
Second early rice crop	28.1 $\pm$ 28.8a	-85.7 $\pm$ 24.4b	104 $\pm$ 19c	-68.4 $\pm$ 10.1b
Fallow periods	1075 $\pm$ 254a	89.6 $\pm$ 40.5b	110 $\pm$ 26.3b	-12.7 $\pm$ 3.4c
Vegetable crop	163 $\pm$ 41a	52.4 $\pm$ 23.9b	225 $\pm$ 15c	-123 $\pm$ 19d
Total	1563 $\pm$ 214a	491 $\pm$ 50b	651 $\pm$ 74b	-73.3 $\pm$ 13.9c
<b>CH<sub>4</sub> and N<sub>2</sub>O GWPs (CO<sub>2</sub>-eq)</b>				
First early rice crop	2885 $\pm$ 206a	2134 $\pm$ 145b	1674 $\pm$ 119b	1422 $\pm$ 97c
Late rice crop	3313 $\pm$ 344a	2064 $\pm$ 169b	1212 $\pm$ 92c	1153 $\pm$ 81c
Second early rice crop	1058 $\pm$ 107a	488 $\pm$ 79b	788 $\pm$ 67c	625 $\pm$ 102bc
Fallow periods	1194 $\pm$ 229a	182 $\pm$ 26b	182 $\pm$ 38b	64.1 $\pm$ 9.4c
Vegetable crop	161 $\pm$ 23a	52.0 $\pm$ 7.8b	228 $\pm$ 52c	-118 $\pm$ 21d
Total	8608 $\pm$ 412a	4919 $\pm$ 248b	4085 $\pm$ 174b	3147 $\pm$ 182c

803

804 Different letters within a row indicate statistical differences ( $P < 0.05$ )

805

806



807 **Table 4**  
 808 Summary of the RM-ANOVAs for soil properties in the different studied periods and  
 809 for the different steel-slag amendments.

	<i>df</i>	<i>F</i>	<i>P</i>
Ferric concentration (Fe <sup>3+</sup> )			
Steel-slag quantity	3, 8	8.26	<0.001
Days after amendment	48, 384	36.8	<0.001
Steel-slag quantity × Days after amendment	144, 384	1.11	0.369
Temperature			
Steel-slag quantity	3, 8	0.06	0.981
Days after amendment	48, 384	83.2	<0.001
Steel-slag quantity × Days after amendment	144, 384	0.02	1.002
pH			
Steel-slag quantity	3, 8	1.79	0.182
Days after amendment	48, 384	18.1	<0.001
Steel-slag quantity × Days after amendment	144, 384	0.51	0.953
Redox potential			
Steel-slag quantity	3, 8	1.34	0.282
Days after amendment	48, 384	20.0	<0.001
Steel-slag quantity × Days after amendment	144, 384	0.57	0.914
Salinity			
Steel-slag quantity	3, 8	15.1	<0.001
Days after amendment	48, 384	22.6	<0.001
Steel-slag quantity × Days after amendment	144, 384	0.76	0.741

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812 **Table 5**813 Pearson correlations of CH<sub>4</sub> and N<sub>2</sub>O fluxes with soil parameters during all periods.

Application level (Mg ha <sup>-1</sup> )	Index	Ferric concentration	Temperature	pH	Redox potential	Salinity
0	CH <sub>4</sub> flux	0.156	0.433*	0.021	0.024	0.485**
	N <sub>2</sub> O flux	0.153	0.133	-0.110	-0.009	0.104
2	CH <sub>4</sub> flux	0.007	0.421*	0.112	0.026	0.300
	N <sub>2</sub> O flux	-0.160	-0.125	-0.239	-0.196	0.119
4	CH <sub>4</sub> flux	0.007	0.410*	-0.035	-0.081	0.323
	N <sub>2</sub> O flux	0.054	-0.051	-0.089	-0.100	0.094
8	CH <sub>4</sub> flux	0.020	0.427*	-0.046	-0.001	0.226
	N <sub>2</sub> O flux	-0.313	0.001	0.022	-0.025	-0.095

814 \**P* < 0.05 \*\**P* < 0.01

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816 **Table 6**  
 817 Yields (mean  $\pm$  SD) of crops amended with various rates of steel slag. The data on  
 818 yield in the first early rice crop are from Wang et al. (2014).  
 819

Crop yield (Mg ha <sup>-1</sup> )	Application level (Mg ha <sup>-1</sup> )			
	0	2	4	8
First early rice crop				
fieldfields	8.09 $\pm$ 0.15a	8.22 $\pm$ 0.13ab	8.33 $\pm$ 0.12ab	8.43 $\pm$ 0.09b
Late rice crop	7.46 $\pm$ 0.16a	7.49 $\pm$ 0.12a	8.08 $\pm$ 0.34b	8.14 $\pm$ 0.28b
Vegetable crop	26.9 $\pm$ 2.8a	26.9 $\pm$ 1.9a	27.0 $\pm$ 3.2a	27.0 $\pm$ 4.2a
Second early rice crop	7.87 $\pm$ 0.09a	7.78 $\pm$ 0.11a	7.93 $\pm$ 0.12a	7.84 $\pm$ 0.09a

820 Different letters within a row indicate statistical differences ( $P < 0.05$ ).

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**Table 7**

Heavy-metal concentrations in grain and soil (mean  $\pm$  SD) of plots amended with various rates of steel slag.

Grain	Application level (Mg ha <sup>-1</sup> )			
	0	2	4	8
<b>Total Mn (mg kg<sup>-1</sup>)</b>	<b>56.6 <math>\pm</math> 2.3b</b>	<b>50.2 <math>\pm</math> 1.6b</b>	<b>50.4 <math>\pm</math> 2.5b</b>	<b>41.8 <math>\pm</math> 1.1a</b>
Total Ni (mg kg <sup>-1</sup> )	0.48 $\pm$ 0.09	0.49 $\pm$ 0.15	0.51 $\pm$ 0.14	0.47 $\pm$ 0.06
<b>Total Zn (mg kg<sup>-1</sup>)</b>	<b>17.7 <math>\pm</math> 2.0b</b>	<b>13.7 <math>\pm</math> 1.4 b</b>	<b>10.3 <math>\pm</math> 1.8a</b>	<b>10.6 <math>\pm</math> 2.1a</b>
Total Cr (mg kg <sup>-1</sup> )	0.66 $\pm$ 0.19	0.69 $\pm$ 0.25	0.66 $\pm$ 0.13	0.57 $\pm$ 0.12
Total Pb (mg kg <sup>-1</sup> )	0.13 $\pm$ 0.02	0.11 $\pm$ 0.03	0.10 $\pm$ 0.03	0.12 $\pm$ 0.02
Total Cu (mg kg <sup>-1</sup> )	4.70 $\pm$ 2.12	4.56 $\pm$ 1.98	4.48 $\pm$ 1.23	4.33 $\pm$ 2.51
Total Co (mg kg <sup>-1</sup> )	0.36 $\pm$ 0.08	0.37 $\pm$ 0.06	0.41 $\pm$ 0.09	0.38 $\pm$ 0.03
<b>Soil</b>				
<b>Total Mn (mg kg<sup>-1</sup>)</b>	<b>217 <math>\pm</math> 2b</b>	<b>245 <math>\pm</math> 7b</b>	<b>293 <math>\pm</math> 18a</b>	<b>314 <math>\pm</math> 25a</b>
Total Ni (mg kg <sup>-1</sup> )	25.5 $\pm$ 3.1	26.1 $\pm$ 2.2	26.4 $\pm$ 3.6	25.9 $\pm$ 3.5
Total Zn (mg kg <sup>-1</sup> )	100 $\pm$ 9	107 $\pm$ 9	126 $\pm$ 13	121 $\pm$ 14
Total Cr (mg kg <sup>-1</sup> )	45.5 $\pm$ 6.3	53.9 $\pm$ 7.1	54.4 $\pm$ 5.2	53.7 $\pm$ 8.8
Total Pb (mg kg <sup>-1</sup> )	8.98 $\pm$ 2.23	9.26 $\pm$ 1.11	9.96 $\pm$ 2.21	10.5 $\pm$ 2.53
Total Cu (mg kg <sup>-1</sup> )	29.9 $\pm$ 5.7	32.5 $\pm$ 4.6	31.9 $\pm$ 6.9	33.1 $\pm$ 6.4
<b>Total Co (mg kg<sup>-1</sup>)</b>	<b>8.81 <math>\pm</math> 1.89b</b>	<b>11.7 <math>\pm</math> 2.2a</b>	<b>12.0 <math>\pm</math> 3.7a</b>	<b>11.5 <math>\pm</math> 2.5a</b>

Different letters within a row indicate statistical differences ( $P < 0.05$ ).

## Figure legends

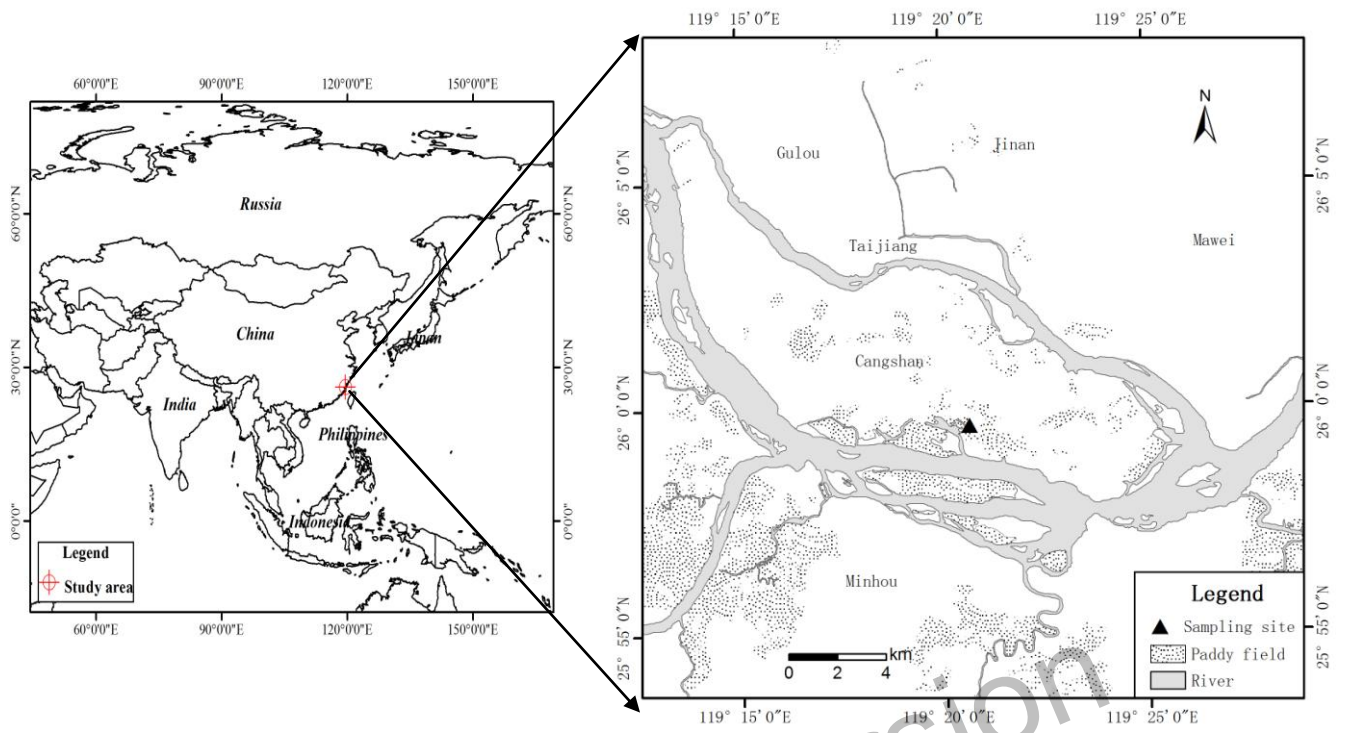
**Figure 1** The location of the study area and sampling site (▲) in southeastern China.

**Figure 2** Seasonal dynamics of CH<sub>4</sub> emissions from plots amended with different amounts of steel slag. ER, LR, F, and V represent early rice crop, late rice crop, fallow period, and vegetable crop, respectively.

**Figure 3** Seasonal dynamics of N<sub>2</sub>O emissions from plots amended with different amounts of steel slag. ER, LR, F, and V represent early rice crop, late rice crop, fallow period, and vegetable crop, respectively.

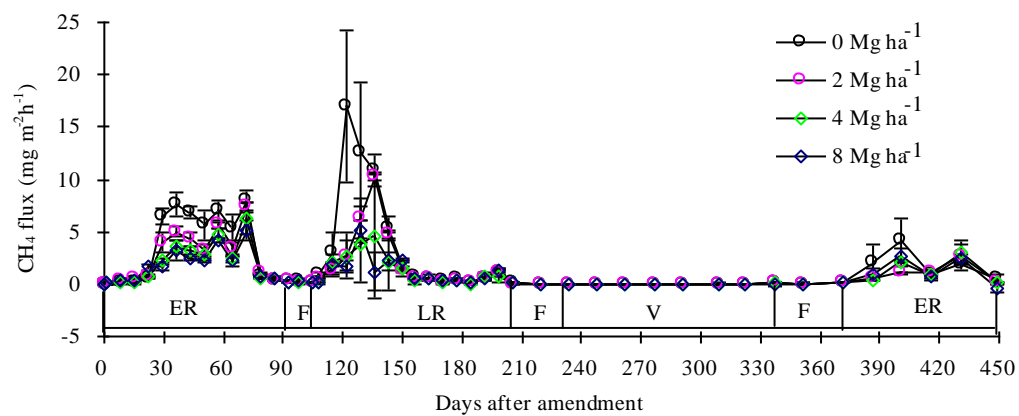
**Figure 4** Seasonal dynamics of soil ferric concentration, temperature, salinity, Eh, and pH in plots amended with different amounts of steel slag. ER, LR, F, and V represent early rice crop, late rice crop, fallow period, and vegetable crop, respectively.

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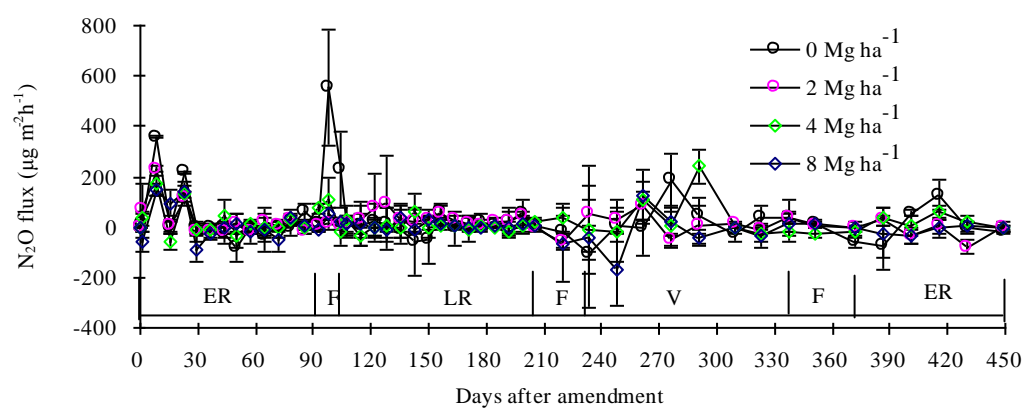
**Fig. 1**

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**Fig. 2**

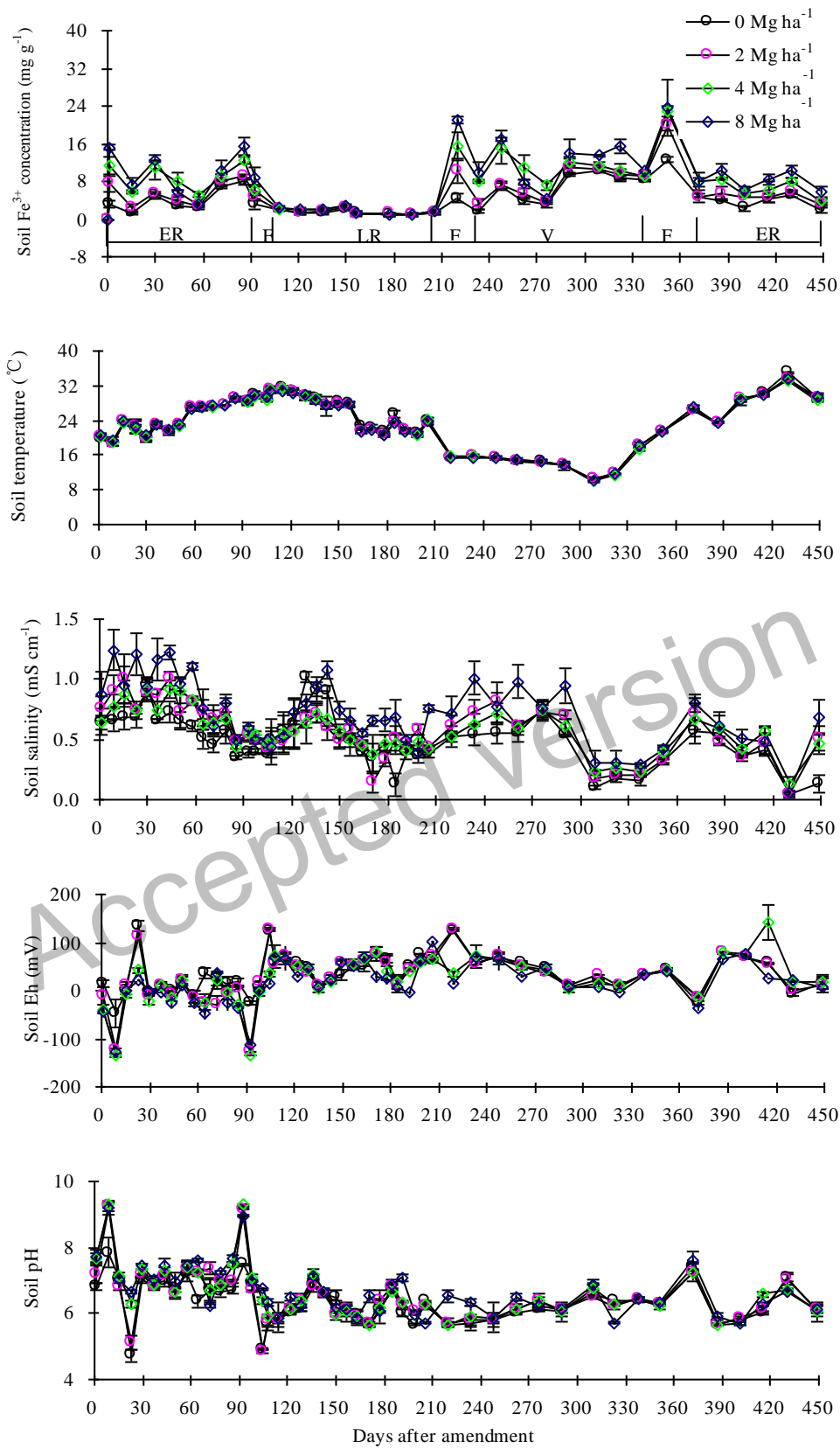
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**Fig. 3**

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**Fig. 4**