Ultraviolet pulsed laser crystallization of Ba_{0.8}Sr_{0.2}TiO₃

films on LaNiO₃-coated silicon substrates

Albert Queraltó $*^{a,b}$, Angel Pérez del Pino a , María de la Mata a,c , Mar Tristany a , Xavier

Obradors^a, Teresa Puig^a, and Susan Trolier-McKinstry^b

^aInstitut de Ciència de Materials de Barcelona, Consejo Superior de Investigaciones Científicas

(ICMAB-CSIC), Campus UAB, 08193 Bellaterra, Catalonia, Spain.

^bMaterials Research Institute and Materials Science and Engineering Department, The

Pennsylvania State University, University Park, Pennsylvania 16802, USA

^cInstitut Català de Nanociència i Nanotecnologia (ICN2), Campus UAB, 08193 Bellaterra,

Catalonia, Spain

CORRESPONDING AUTHOR INFORMATION

* Dr. Albert Queraltó

ICMAB - CSIC, Campus UAB, 08193 Bellaterra, Catalonia, Spain

Tel. +34 935 801 853

Fax. +34 935 805 729

E-mail: albert.queralto.lopez@gmail.com

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ABSTRACT: In this work, Ba_{0.8}Sr_{0.2}TiO₃ (BST) films on LaNiO₃-buffered SiO₂/Si (LNO/SiO₂/Si) substrates were crystallized by pulsed laser irradiation. Solution-derived amorphous barium-strontium-titanate precursor layers were crystallized with a KrF excimer laser in oxygen ambient at fluences ranging from 50 to 75 mJ cm⁻². With the substrate temperature set to 500 °C, the number of pulses and film thickness were varied until high-quality crystallinity could be achieved. It was found that films with a thickness of 40 nm are fully crystallized with a uniaxial {00l} orientation which is predetermined by the LaNiO₃ orientation. On the other hand, for 160 nanometer thick films, crystallization was observed after 12000 pulses in the 70 nm close to the surface, while the rest of the film remained amorphous. The large temperature difference between the film surface and interface due to the low thermal conductivity of the amorphous BST is suggested as the origin of this behavior. Films thicker than 80 nm cracked on crystallization due to the stress caused by the different thermal expansion coefficients of film and substrate, as well as the large temperature variations within the BST film.

KEYWORDS

A. Sol-gel processes; Laser crystallization; D. BaTiO3 and titanates; A. Films

INTRODUCTION

Perovskite oxide films are being explored for numerous functional electronic devices [1-4]. As one example, there is on-going interest in barium strontium titanate (Ba_{1-x}Sr_xTiO₃ or BST) for tunable radio-frequency (RF) and microwave components, flexoelectrics, pyroelectrics, and capacitors [4-9]. The dielectric and ferroelectric properties of BST can be tuned through variations in composition, stress, temperature or applied dc electric field [10-13]. Furthermore,

the high dielectric permittivity of BST is suitable for capacitors where high storage densities and low leakage currents are needed [9, 14, 15]. Finally, the high pyroelectric coefficients of BST have also attracted much attention for room temperature infrared detectors and thermal imaging devices [8, 16].

Among the techniques that have been employed to produce BST films [17-19], chemical solution deposition (CSD) is a simple, versatile and low-cost methodology for the development and largescale implementation of BST films, providing good control over film homogeneity and stoichiometry [10, 20]. Typically, high temperature thermal treatments in conventional furnaces or rapid thermal annealers are used to induce film crystallization. Nevertheless, this methodology cannot be used in applications as wearable electronics which often requires the growth of oxides on temperature sensitive substrates such as polymers. In this sense, photo-irradiation techniques appear as a novel strategy for low-temperature processing due to the photo-induced heating being highly confined in depth [21]. For instance, ultraviolet (UV) lamps have been employed directly or assisting rapid thermal annealing for the crystallization of oxides such as SiO₂, TiO₂ or PbTiO₃-based compounds at low temperatures [22-24]. Although UV lamps ensure the growth at temperatures lower than only-thermal treatments, the process is still time consuming requiring tens of minutes to hours of irradiation. In pulsed laser annealing (PLA) of CSD deposited layers, the use of a pulsed source delivers a large amount of energy in a very short time (i.e. nanoseconds) allowing a significantly fast crystallization as recently demonstrated numerous oxide materials such as epitaxial VO₂, Ce_{0.9}Zr_{0.1}O_{2-y}, Ba_{0.8}Sr_{0.2}TiO₃ and perovskite manganites (LMO, LSMO) [25-30], or polycrystalline TiO2, In2O3, Ba0.7Sr0.3TiO3 and (Pb,La)(Zr,Ti)O₃ [31-34]. Moreover, the spatially-localized heating effect of lasers enables high temperature crystallization of materials such as BaTiO3, silicon or MoS2 with a minimal

properties and show great potential for the development of flexible electronic devices at large scales [35-37]. PLA can also be readily incorporated into industrial fabrication processes confering this technique a high interest for emerging technologies as macroelectronics. Despite these promising benefits, the use of highly energetic laser sources requires a careful control over different processing parameters to prevent damaging (e.g. amorphization, ablation, etc) the constitutive materials [38]. The precursor film thickness also plays a crucial role in PLA since photo-induced heating is generally highly confined in depth. Thus, the non-homogeneous temperature distribution inside thick film material can provoke its partial crystallization and the development of high thermal stress leading to cracking [33]. In contrast, too thin precursor films or coatings that are transparent to the laser radiation do not develop the needed temperature for initiating crystallization. Therefore, specific photothermal considerations should be taken into account in the design of the treatments.

Laser crystallization of BST deposited by CSD has been reported previously on Pt/Ti/SiO₂/Si substrates [33, 39]. Platinum electrodes have a good electrical conductivity, excellent oxidation resistance, and allow the integration of different oxides with silicon [18, 40]. However, use of LaNiO₃ (LNO) enables a cost-effective route for texture control on wide variety of substrates [41]. In this article, an in-depth analysis of the crystallization process of Ba_{0.8}Sr_{0.2}TiO₃ films by PLA on LaNiO₃ bottom electrodes was performed. The influence of different experimental parameters on crystallization, such as the fluence, number of pulses and film thickness was investigated to develop optimized conditions. Solid-state crystallization was prioritized, in order to avoid melting and resolidification phenomena by using lower fluences. Also, a substrate temperature above room temperature and larger number of pulses were used to decrease

temperature variations between film surface and interface, and to improve film crystallization. Systematic investigations including two-dimensional X-ray diffraction (2D-XRD) were used to further analyze the crystallization process. Numerical simulations based on finite element methods were employed to support the experimental results, providing insight on the thermal evolution of the system and a better appreciation of the photoinduced effects involved during laser crystallization. Finally, efforts were made to evaluate film cracking at high film thicknesses and expand the current understanding about their origin.

MATERIALS AND METHODS

The thin films in this work were prepared using chemical solution deposition (CSD) on oxidized Si wafers. The thickness of the SiO_2 layer was $\sim 1~\mu m$. Firstly, 0.2~M LaNiO₃ (LNO) precursor solutions, prepared as described in ref [27], were deposited by spin-coating at 6000 rpm for 2 min on thoroughly cleaned 4-inch SiO_2/Si wafers (NOVA Electronic Materials, LLC). Each layer was pyrolyzed at 350 °C for 10 min on a hot-plate and crystallized at 700 °C for 1 min, with a heating ramp of 20 °C s⁻¹ in oxygen ambient using a rapid thermal annealer (RTP-600S, Modular Process Technology Corp.). Four repetitions were needed to reach a thickness of $\sim 150~nm$. The final LNO films had a root mean square (RMS) roughness of $\sim 0.8~nm$, and some uniaxial texture with the (001) orientation as shown in the Supporting Information (Figures S1 and S2). In particular, LNO buffer layers have a uniaxial fraction of $61\pm 8~\%$. The LNO-coated wafers were cut into pieces of approximately $5\times 5~mm^2$.

Stoichiometric Ba_{0.8}Sr_{0.2}TiO₃ (BST) precursor solutions with a 0.3 M concentration were prepared as described elsewhere [27]. These solutions were spun onto the LaNiO₃-coated

substrate using the same conditions as for the LaNiO₃, and were fully decomposed at 450 °C for 10 min on a hot-plate after each coating. The thickness of the amorphous pyrolyzed film was varied between 40 and 160 nm by repeating the deposition and pyrolysis steps.

Pulsed laser annealing of the BST films was carried out using a KrF excimer laser (COMPex Pro 102, Coherent Inc.) working at 248 nm with pulses of 20 ns in duration and a repetition rate of 20 Hz. The energy density at the film surface was controlled by adjusting the incoming laser energy, and varying the spot size with a plano-convex lens. Films were irradiated inside a vacuum chamber brought down to a residual pressure of 10^{-7} Torr, which was then filled with pure oxygen at a pressure of 0.1 Torr. Samples were mounted on a substrate heater using silver paint. The substrate temperature was controlled within ± 3 °C by employing a Eurotherm Digital Temperature Controller 847.

The morphological characterization of film surfaces was conducted using an Agilent 5100 atomic force microscope (AFM) and a FEI NanoSEM 630 scanning electron microscope (SEM). The micrographs were analyzed with the MountainsMap 7.0 software (Digital Surf). The crystallographic structure of the films was measured by X-ray diffraction (XRD) using a Cu-Kα radiation source with a PANalytical X'Pert Pro MPD diffractometer and a Bruker GADDS system equipped with a 2-dimensional detector. More detailed analysis of the film crystallization was attained by means of high resolution transmission electron microscopy (HRTEM) of cross-sectional specimens produced by mechanical polishing and ion milling. The HRTEM images were acquired with FEI Tecnai F20 and JEOL J2010F microscopes both equipped with field-emission guns working at 200 kV and achieving lateral resolutions of 0.14 nm. Electron energy loss spectroscopy (EELS) analyses were conducted in a FEI Tecnai F20 and FEI Titan S/TEM both equipped with post-column Gatan Image Filter energy spectrometers.

Finally, the laser-induced thermal behaviour in the samples were simulated by solving the transient heat equation with the finite element method by employing COMSOL 4.4 Multiphysics software [26]. Additional information on simulations can be found in the supporting information. The optical and thermophysical properties of films and substrate used in the calculations were reported in refs [21, 27, 42-45] and are summarized in the Supporting Information (Table S1).

RESULTS AND DISCUSSION

1. Study at different fluences and number of pulses

Numerical simulations were used to estimate appropriate irradiation conditions. The amount of radiation absorbed is characterized by the optical absorption coefficient, α , as described by the Beer-Lambert law. In particular, amorphous BST has an optical absorption coefficient α_{BST} at λ =248 nm of 28.1×10⁶ m⁻¹ [27], which corresponds to an optical penetration length of $l_{\alpha}^{BST}\sim36$ nm. Thus, a film with a thickness of 40 nm will absorb about a 68% of the incoming radiation, and the remaining intensity will be absorbed through the 150 nm thick LNO buffer layer $(\alpha_{LNO}=23.6\times10^6 \,\mathrm{m}^{-1})$ and $l_{\alpha}^{LNO}\sim42$ nm at λ =248 nm) [27].

It can be assumed that the amorphous BST film is mostly composed of Ba-O, Sr-O and Ti-O bonds with dissociation energies between 454 and 662 kJ mol⁻¹ (4.7 and 6.9 eV/bond) [46], whereas the laser photon energy is ~5 eV at 248 nm. Two major phenomena can be induced upon absorption of the laser radiation, i.e. photochemical and photothermal interactions. Photochemical mechanisms associated with the direct dissociation of chemical bonds may occur given that the laser photon energy is slightly higher than the lower limit for bond dissociation

energies. However, photothermal effects have been reported to be predominant for nanosecond-pulsed lasers [21]. Thus, for the sake of simplicity, it has been assumed that the interaction between laser radiation and films is essentially photothermal. The film pyrolysis procedure followed in the experimental section ensured the elimination of organic residues as reported for other propionate derived films within a detection limit of approximately 0.8 wt% [47, 48]. If any C-C, C-O and C-H bonds remained, they would be photochemically decomposed during initial stages of irradiation since their energies are significantly lower than the laser photon energy (3.6-4.3 eV/bond) [48].

Figure 1a shows the temperature profiles obtained from photothermal simulations of 40 nanometer thick BST films on LNO/SiO₂/Si substrates for fluences of 50, 65 and 75 mJ cm⁻², a substrate temperature of 500 °C and a single laser pulse. The chosen conditions allow similar maximum temperatures to the ones reported in ref [27], i.e. between ~1100-1450 °C. According to simulations, laser pulses provoke rapid thermal cycles without exceeding the melting point of the system constituents - (Supporting Information, Table S1). The high temperatures induced lead to heating/cooling rates up to 10^9 °C s⁻¹. The temperature profiles developed in the BST film last from hundreds to thousands of nanoseconds. As a result, large temperature differences are induced between the film surface and the BST/LNO interface. These variations lead to maximum temperature gradients inside the film of 10^8 - 10^9 °C m⁻¹ during tens of nanoseconds. Additionally, the effective heating time for a single pulse t_{eff} (here defined as the time the film is above 600 °C, which is used as the minimum temperature needed for BST crystallization [49]) can be extracted. In particular, it can be seen that t_{eff} increases with the fluence from approximately 0.8 (50 mJ cm⁻²) to 1.1 μ s (75 mJ cm⁻²). It is also important to highlight that these values of effective heating time are one to two orders of magnitude longer than those reported elsewhere [26, 27, 39, 50].

These differences mainly arise due to the different laser systems employed, experimental conditions (fluence and substrate temperature) and thermophysical parameters of the constituent materials.

Laser irradiation of amorphous BST (after the pyrolysis of the precursors) on LNO/SiO₂/Si was performed at fluences of 50, 65 and 75 mJ cm⁻²; samples were irradiated with 12000 pulses while held at a substrate temperature of 500 °C. XRD diffractograms in Figure 1b show the peaks corresponding to the silicon substrate and LNO buffer layer. Also, the BST (001) and (002) peaks are identified at ~22.2° and ~45.4°, respectively, and the polycrystalline BST (011) reflection is found at ~31.7°. No crystallization of BST is detected at 50 mJ cm⁻², whereas BST peaks are observed at 65 and 75 mJ cm⁻². These results indicate that the temperature reached at 50 mJ cm⁻² (Figure 1a) may be insufficient to promote crystallization for the very short laser annealing times employed. Thus, either higher fluences or larger number of pulses should be used.

Figure 2 shows the percentage of {100} texture fraction for BST films with thicknesses of 40 and 160 nm, fluences of 65 and 75 mJ cm⁻², and number of pulses between 1200 and 72000, as indicated. The methodology used (Supporting Information) is an improvement from semi-quantitative methods like the Lotgering factor [51]. The values for 40 nm-thick BST samples treated at 65 and 75 mJ cm⁻² after 12000 pulses (substrate at 500 °C) obtained from the Lotgering factor method would have been 0.63±0.03 and 0.77±0.04, respectively. In contrast, the proposed quantification methodology shows uniaxial fractions of 21±7 % and 26±5 %, respectively. Therefore, these films essentially have a random orientation although some degree of texture is detected along the {001} orientation, which seems to increase slightly with the laser fluence. The same method has been previously employed to evaluate the epitaxial content of ceria thin films

produced by laser annealing, giving epitaxial fraction values close to 100 % [26]. The surface morphology of films irradiated at 50 and 65 mJ cm⁻² as shown in Figures 1c and d, reveals granular and porous surfaces. Furthermore, grains with tens of nanometers in size are also seen, which led to RMS roughnesses of 3.0 and 5.2 nm, respectively. The sample irradiated at 75 mJ cm⁻² (Figure 1e) has a RMS roughness of 8.3 nm and displays larger groups of grains with sizes of hundreds of nanometers. Moreover, the grains are separated by boundaries that could be associated with grain boundary grooving, similar to those reported in ref [26].

A fluence of 65 mJ cm⁻² was selected for a substrate temperature of 500 °C since those conditions allow laser crystallization of BST with a rather homogeneous surface with low roughness. To investigate how the film surface morphology and crystallization is affected, the number of laser pulses was varied between 1200 and 72000 pulses. After accumulating 1200 pulses (Figure 3a), the surface shows two sets of features, one with compact grains smaller than 100 nm and the other larger than ~300 nm. The RMS roughness of the film is ~3.5 nm. The larger features could be associated with residual amorphous material, as they seem to disappear as the number of pulses increases to 12000 and 72000 (Figures 1d and 3b). The RMS roughness increased to 5.2 and 6.9 nm, respectively, because the smaller grains enlarge and the surface morphology becomes porous as mentioned earlier. XRD analyses of films irradiated for 1200, 12000 and 72000 pulses (Figure 3c) illustrate the presence of (001) and (011) orientations of BST, as reported previously. Based on the increase in the BST (002) peak area, the degree of crystallization increases with the number of pulses. Presumably, this is caused by the longer cumulative heating time achieved at higher number of pulses as it has been reported for other oxide films on different substrate architectures [26, 27, 33, 50]. Accordingly, longer heating times would allow the film constituents to rearrange into the final phase and also form larger crystallites through grain coarsening, which should lead to sharper and more intense XRD peaks. The 2D-XRD pattern of a film irradiated at 65 mJ cm⁻², a substrate temperature of 500 °C and 72000 pulses (Figure 3d) reveals that the BST film has some $\{001\}$ oriented material, as evidenced by the higher peak intensity for values close to χ =0° (depicted with a dashed white line). The percentages of $\{100\}$ orientation for films irradiated with 1200, 12000 and 72000 pulses at 65 mJ cm⁻² are 13 ± 5 %, 21 ± 7 % and 35 ± 6 %, respectively, as shown in Figure 2. The degree of orientation increases with the number of pulses, suggesting that the BST is trying to mimic the (001) texture of the LNO buffer layer (uniaxial fraction of 61 ± 8 %) as more pulses are applied.

2. Effect of film thickness on crystallization

Usually, the strategy followed for the fabrication of BST devices from chemical solutions is based on the deposition of multiple coatings until the desired thickness is reached [4, 13, 16]. In this section, laser crystallization of films with thicknesses between 40 and 160 nm is described. The methodology followed to obtain thicker films is based on several deposition and pyrolysis steps, as described in the experimental section.

SEM analyses in Figures 4a and b illustrate the resulting surface morphologies for films (laser annealed at 65 mJ cm⁻² with a substrate temperature of 500°C for 72000 pulses). It was found that the sample with a thickness of 40 nm is homogeneous with grains/pores of 100 nm in size and a RMS roughness of 6.9 nm, as already described. On the other hand, the film with a thickness of 160 nm presents a heterogeneous surface with lateral grain sizes exceeding 1 μm, resulting in a RMS roughness of 18.9 nm. Cracks also appear. Figure 4c depicts the XRD

measurements of BST films with thicknesses of 40 and 160 nm. Particularly, the θ -2 θ scans reveal the increasing intensity of the BST reflections with the film thickness because of the larger sampled volume. The uniaxial fraction values for films with thicknesses of 40 and 160 nm and irradiated at 65 mJ cm⁻² and 72000 pulses are 35±6 % and 29±7 %, respectively, revealing the presence of a large amount of polycrystalline material in the films.

HRTEM investigations of BST films after laser treatment at 65 mJ cm⁻² for 12000 pulses at a substrate temperature of 500°C were conducted. Figures 5a and b show a STEM image (performed under Annular Dark Field, ADF, conditions) and the corresponding intensity profile of the sample. The 40 nanometer thick BST film can be identified along with the LNO multilayered film and the amorphous SiO_2 layer. Figure 5b depicts an HRTEM image and power spectrum (fast Fourier transform, FFT) of a zoomed area into the square in Figure 5a, showing two different BST grains with different in-plane orientations. In contrast, Figure 5c shows the TEM characterization of a ~160 nm thick BST film, depicting a partially crystallized layer. More precisely, Figures 5d and e show a ~70 nanometer thick region close to the surface comprised of polycrystalline (randomly oriented) material, whereas the remaining film closer to the interface with the LNO (Figure 5f) is amorphous. The percentage of uniaxial fraction in this sample is around 15-30 % (Figure 2).

The transfer of texture from the LNO layer seems improbable provided nucleation is induced from their surface. Despite that, the homogeneity of this crystalline/amorphous layer cannot be completely evaluated due to the local nature of TEM. Thus, thicker films could present areas with a tendency to crystallize down to the interface, while other regions may have amorphous regions like the one depicted in Figure 5f. In addition, *ab initio* calculations of the surface energies for BaTiO₃ indicate that a lower energy is required to form a {001} oriented surface [52,

53]. Therefore, it could also be more favorable thermodynamically to form grains with a {00l} orientation. The differences observed between the two samples can be explained in terms of the absorbed radiation intensity, which is higher at the surface and decays exponentially inside the film as described by the Beer-Lambert law. As mentioned before, the heating process depends to a great extent on the optical penetration depth l_{α} of each material. It should be remembered that 68% of the incoming radiation is absorbed through the 40 nanometer BST film $(l_a^{BST} \sim 36 \text{ nm at})$ λ ~248 nm). The rest of the radiation is transmitted to (and absorbed in) the LNO buffer layer. In contrast, the laser radiation is completely absorbed in the 160 nanometer thick BST layer. As a result, the temperature profiles extracted from simulations for 40 and 160 nanometer thick BST films on LNO/SiO₂/Si substrates (Figure 6a) reveal comparable temperature profiles at the film surface. However, there is a large difference in the temperature reached at the interface with LNO, i.e. the surface-interface temperature differences are around 50 and 300 °C for 40 and 160 nm thick films, respectively. These results are similar to those reported previously [39, 50]. It is notable that the interface temperature for the 160 nm film, around 1000 °C, may be insufficient to promote the laser crystallization for a limited number of short pulses (Figure 5c). These results are in agreement with previous observations for 40 nanometer films where it was shown that crystallization is not detected for laser fluences as low as 50 mJ cm⁻² (equivalent to temperatures of ~ 1100 °C).

Cracks were observed in 160 nm thick films after being irradiated at 65 mJ cm⁻² for 72000 pulses at a substrate temperature of 500 °C. Comparable reports of the presence of cracks for laser-crystallized BST films on Pt/SiO₂/Si substrates have been given elsewhere [33, 39]. Specifically, it was determined that cracks appeared for thicknesses above 130 nm, even for laser energy densities as low as 50 mJ cm⁻². In the same way, in this work, cracking was observed for films

with thicknesses of ~160 nm even after laser annealing at 50 mJ cm⁻², 72000 pulses and a substrate temperature of 500 °C (Figure S4). No cracks could be observed for a thickness of 80 nm. Film cracking is a form of stress relaxation that occurs when elastic energy accumulates in the film until it exceeds a critical value. There are several physical mechanisms that lead to stress accumulation and crack formation. First, we considered the thermal stress derived from different thermal expansion coefficients (TECs) of the materials forming the system. The respective TECs of BST and LNO are approximately $12.9 \times 10^{-6} \text{ K}^{-1}$ and $8.2 \times 10^{-6} \text{ K}^{-1}$ [54, 55], and even a larger difference is attained between these values and the thermal expansion coefficient of SiO₂ $(0.5 \times 10^{-6} \text{ K}^{-1} \text{ [56]})$, which will certainly lead to stress development in the system. Temperature simulations in Figure 6b indicate that the silicon $(1.4 \times 10^{-6} \text{ K}^{-1} \text{ [55]})$ is at the initial substrate temperature.

Because of its thickness, it is expected to impose stress on the whole BST/LNO/SiO₂/Si structure. Also, there is the large temperature gradients associated laser heating/cooling. Simulations in Figure 6 show a temperature variation of ~300°C between the surface and interface for 160 nanometer thick BST films giving rise to a temperature gradient of ~2×10° °C m⁻¹. This film cracks. However, the temperature gradient is reduced down to 1x10° °C m⁻¹ for the 40 nm films, which did not crack. Therefore, the film thickness seems to be a critical parameter controlling crack formation in the current system as stress builds up with the increase in thickness. Finally, it is important to note that different degrees of crystallization exist between the surface and interface in thick films, i.e. the film is polycrystalline at the surface and still amorphous at the interface (Figure 5c-f). Thus, stress may be generated during film growth related to the densification process occurring during the transformation from amorphous to crystalline material, as reported previously [33]. The stress caused by lattice mismatch is

irrelevant here since there is no epitaxial growth. The results provided here would suggest that either films with thicknesses below 160 nm should be grown or, instead, a strategy with consecutive pyrolysis and crystallization steps should be followed until the desired thickness is reached. This should prevent excessively large temperature differences and film densification, hence diminishing the development of cracks.

CONCLUSIONS

Solution-derived amorphous BST films were crystallized on LNO-coated silicon wafers at low substrate temperatures mainly due to photothermal interactions derived from the absorption of laser radiation. The experimental conditions lead to a solid state crystallization, i.e. the maximum simulated temperatures in the BST layer are below the melting point. The films have a certain degree of uniaxial {001} crystalline texture that could be attributed to the lower surface energy needed to form these grains, as well as the presence of some crystalline regions that reach the interface in thicker films. Moreover, the amount of crystalline material increased with the fluence and number of pulses since higher temperatures and longer cumulative annealing times are developed in the system. Temperature simulations point toward cumulative effective heating times in the range of milliseconds. Careful control of the thickness is mandatory since it has also been found to affect the crystallinity. In particular, layers with a thickness of 40 nm are fully crystallized, while only a ~70 nm depth near the surface is crystallized in 160 nanometer thick films. Temperature inhomogeneity through the thickness coupled with the very short effective heating times of the laser treatments are proposed as the main sources for partially crystallized layers. Additionally, these phenomena together with the different thermal expansion coefficients

of materials and film thickness can generate stress in the system that result in the formation of cracks. Consequently, careful control of the processing parameters must be done to prevent

noncrystallized areas and crack formation from hindering the functionality of the films.

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SUPPLEMENTARY MATERIAL

AFM and XRD characterization of LNO/SiO₂/Si substrates, description of the quantitative

method for the evaluation of film texture from 2D-XRD, description of the numerical model

used for photothermal simulations, thermophysical and optical parameters of BST, LNO, SiO₂

and Si, in SC and YSZ/SS, SEM characterization of BST films irradiated at 50 mJ cm⁻² and

500°C of substrate temperature and a thickness of 80 nm.

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FIGURE CAPTIONS

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Figure 1. Analysis of 40 nm thick BST films on LNO/SiO₂/Si substrates irradiated with 12000 pulses at different laser fluences with a substrate temperature of 500°C. (a) Numerical simulations illustrating the temporal evolution of temperature at the BST film after irradiation at 50, 65 and 75 mJ cm⁻². The laser pulse dependence with time is also given for comparison purposes. The black-dotted line corresponds to the minimum temperature used to define the effective heating time. (b) XRD measurements of BST films treated at 50, 65 and 75 mJ cm⁻². AFM images for samples irradiated at (c) 50, (d) 65, and (e) 75 mJ cm⁻².

Figure 2. Percentage of {100} texture present in BST films (thicknesses of 40 and 160 nm) laser crystallized on LNO/SiO₂/Si. Fluences of 65 and 75 mJ cm⁻² were used during irradiation while the substrate was held at a temperature of 500 °C, and the pulses applied were set to 1200, 12000 and 72000.

Figure 3. 40nm thick BST films on LNO/SiO₂/Si irradiated at 65 mJ cm⁻² at a substrate temperature of 500°C. AFM characterization after (a) 1200 and (b) 72000 pulses. (c) θ -2 θ scans of the BST films presented in (a) and (b). (d) 2D-XRD analysis of a BST film laser treated at 65 mJ cm⁻², a substrate temperature of 500°C and 72000 pulses.

Figure 4. BST films on LNO/SiO₂/Si with different thicknesses irradiated at 65 mJ cm⁻² for 72000 pulses at a substrate temperature of 500°C. SEM images of BST films with thicknesses of (a) 40 nm and (b) 160 nm. (c) XRD measurements of the BST films described in (a) and (b).

Figure 5. (S)TEM characterization of BST films with different thicknesses on LNO/SiO₂/Si substrates after laser irradiation at 65 mJ cm⁻², 12000 pulses and a substrate temperature of 500°C. BST film with a thickness of 40 nm: (a) low magnification cross-sectional ADF STEM image, and (b) HRTEM image of the orange framed region in (a) and its FFT. BST film with a thickness of 160 nm: (c) low magnification cross-sectional TEM image, (d) HRTEM detail corresponding to polycrystalline grains (blue indicated in (c)) and (e) the associated power spectrum, and (f) HRTEM image of an area close to the interface with LNO (indicated by the green rectangle in (c)) depicting amorphous BST.

Figure 6. Temperature simulations of amorphous BST films on LNO/SiO₂/Si substrates for a fluence of 65 mJ cm⁻², and a substrate temperature of 500°C. (a) Temperature evolution with time for BST films with thicknesses of 40 and 160 nm. The solid and dashed lines are representative of the surface and interface temperature profiles, respectively. (b) Temperature distribution with depth inside the heterostructure for BST films with thicknesses of 40 and 160 nm at 30 ns.