



This is the **accepted version** of the journal article:

Corral, Isaac; Cardellach López, Esteve; Corbella i Cordomí, Mercè; [et al.]. «Origin and evolution of mineralizing fluids and exploration of the Cerro Quema Au-Cu deposit (Azuero Peninsula, Panama) from a fluid inclusion and stable isotope perspective». Ore Geology Reviews, Vol. 80 (January 2017), p. 947-960. DOI 10.1016/j.oregeorev.2016.09.008

This version is available at https://ddd.uab.cat/record/289476

under the terms of the **COBY-NC-ND** license

1	Origin and evolution of mineralizing fluids and exploration of the Cerro Quema
2	Au-Cu deposit (Azuero Peninsula, Panama) from a fluid inclusion and stable
3	isotope perspective
4	
5	Authors: Isaac Corral ^{1,2,*} , Esteve Cardellach ¹ , Mercè Corbella ¹ , Àngels Canals ³ , Albert
6	Griera ¹ , David Gómez-Gras ¹ , Craig A. Johnson ⁴ .
7	
8	Affiliation:
9	1) Departament de Geologia, Universitat Autònoma de Barcelona, 08193 Barcelona,
10	Spain.
11	2) EGRU (Economic Geology Research Centre), College of Science, Technology and
12	Engineering, James Cook University, Townsville, QLD 4811, Australia.
13	3) Facultat de Geologia, Universitat de Barcelona. 08028 Barcelona, Spain.
14	4) U.S. Geological Survey, Denver, CO 80225. USA.
15	
16	*Corresponding author: Isaac Corral
17	Postal address: EGRU (Economic Geology Research Centre), College of Science,
18	Technology and Engineering, James Cook University, Townsville, QLD 4811, Australia.
19	e-mail: I.Corral.Geo@gmail.com
20	Phone: +61 7 47815681
21	Cell: +61 473935727
22	
23	
24	Keywords: Cerro Quema, fluid inclusions, stable isotopes, alunite-pyrite, exploration.
25	
26	
27	
28	

29 Highlights

30	•	Fluid inclusion data reflect the hydrothermal fluid pathway through the Cerro
31		Quema Au-Cu deposit.

32 δ^{34} S, δ^{18} O and δ D isotope geochemistry, δ^{34} S alunite-pyrite geothermometry 33 and composition of mineralizing fluids (δ^{34} S_{Σ S}, X_{SO²⁻} and X_{H₂S}).

- Integration of fluid inclusions and stable isotopes as tool-generating exploration
 vectors in porphyry systems.
- 36

37 Abstract

38 Cerro Quema is a high sulfidation epithermal Au-Cu deposit with a measured, indicated and inferred resource of 35.98 Mt @ 0.77 g/t Au containing 893,600 oz Au (including 39 40 183,930 oz Au equiv of Cu ore). It is characterized by a large hydrothermal alteration 41 zone which is interpreted to represent the lithocap of a porphyry system. The innermost 42 zone of the lithocap is constituted by vuggy quartz with advanced argillic alteration 43 locally developed on its margin, enclosed by a well-developed zone of argillic 44 alteration, grading to an external halo of propylitic alteration. The mineralization occurs 45 in the form of disseminations and microveinlets of pyrite, chalcopyrite, enargite, 46 tennantite, and trace sphalerite, crosscut by quartz, barite, pyrite, chalcopyrite, 47 sphalerite and galena veins. 48 Microthermometric analyses of two phase (L+V) secondary fluid inclusions in igneous 49 quartz phenocrysts in vuggy quartz and advanced argillically altered samples indicate 50 low temperature (140-216 °C) and low salinity (0.5-4.8 wt % NaCl eq.) fluids, with 51 hotter and more saline fluids identified in the east half of the deposit (Cerro Quema 52 area). 53 Stable isotope analyses (S, O, H) were performed on mineralization and alteration minerals, including pyrite, chalcopyrite, enargite, alunite, barite, kaolinite, dickite and 54 vuggy quartz. The range of δ^{34} S of sulfides is from -4.8 to -12.7‰, whereas δ^{34} S of 55 sulfates range from 14.1 to 17.4‰. The estimated $\delta^{34}S_{\Sigma S}$ of the hydrothermal fluid is -56

0.5‰. Within the advanced argillic altered zone the δ^{34} S values of sulfides and sulfates 57 58 are interpreted to reflect isotopic equilibrium at temperatures of ~240 °C. The $\delta^{18}O$ values of vuggy guartz range from 9.0 to 17.5‰, and the δ^{18} O values estimated for the 59 vuggy guartz-forming fluid range from -2.3 to 3.0%, indicating that it precipitated from 60 mixing of magmatic fluids with surficial fluids. The δ^{18} O of kaolinite ranges from 12.7 to 61 62 18.1‰ and δD from -103.3 to -35.2‰, whereas the $\delta^{18}O$ of dickite varies between 12.7 63 and 16.3‰ and δD from -44 to -30. Based on $\delta^{18}O$ and δD , two types of 64 kaolinite/dickite can be distinguished, a supergene type and a hypogene type. 65 Combined, the analytical data indicate that the Cerro Quema deposit formed from 66 magmatic-hydrothermal fluids derived from a porphyry copper-like intrusion located at 67 depth likely towards the east of the deposit. The combination of stable isotope 68 geochemistry and fluid inclusion analysis may provide useful exploration vectors for 69 porphyry copper targets in the high sulfidation/lithocap environment.

70

71

72 **1. Introduction**

73 The Azuero Peninsula in southwestern Panama shows evidence of widespread 74 hydrothermal activity as indicated by the existence of several Au-Cu mineral 75 occurrences including the Cerro Quema high sulfidation epithermal Au-Cu deposit (Fig. 76 1). The Cerro Quema deposit was discovered in 1988 by the Compañía de Exploración 77 Minera S. A. (CEMSA) during geological and metallogenetic studies by the United 78 Nations Development Program (UNDP; 1965), and today is one of the most 79 prospective exploration projects in Panama. The Cerro Quema deposit consists of 80 several mineralized zones: from east to west, Cerro Quema, Cerro Quemita, and La 81 Pava (Fig. 2). The measured, indicated and inferred resources as of 2014 are 35.98 Mt 82 @ 0.77 g/t Au containing a total of 893,600 troy ounces Au of which 183,930 troy 83 ounces are gold equivalent of the contained copper (Sutcliffe et al., 2014). Additional 84 mineralized bodies are found to the east —the Cerro Idaida, East Quema Jungle, and

85 Cerro Pelona prospects— but the Au and Cu grade and content of these prospects
86 have not yet been determined.

87	The first geological studies of the Azuero Peninsula (e.g., Del Giudice and Recchi,
88	1969; Ferencic, 1970; Kesler et al., 1977) noted the potential for Au and Cu deposits in
89	the region. Later works specifically on the Cerro Quema deposit focused on geology
90	(Horlacher and Lehmann, 1993; Nelson, 1995; Corral et al., 2011), and metallogeny
91	(Leach, 1992; Corral, 2013; Corral et al., 2016). This report presents the first
92	documentation of the genesis of the deposit as indicated by the study of the origin and
93	evolution of the ore fluids.
94	Here we present the results of fluid inclusion and stable isotope studies undertaken to
95	determine the origin of the fluids responsible for ore deposition. The results are
96	integrated into a genetic model of high sulfidation Au-Cu deposits (c.f., Arribas, 1995).
97	The results also provide exploration vectors for underlying porphyry copper targets,
98	vectors that may be useful elsewhere in terranes with similar geology.
99	

100 ------ Insert Figure 1 near here-----

101

102 2. Regional Geology

Panama is underlain by the youngest segment of the land bridge that joins the North
and South American plates. The geology consists of a volcanic arc that developed in
Late Cretaceous to Paleocene time during subduction of the ancient Farallon plate
beneath the Caribbean plate. The resulting tectonic block lies at the junction of the
Caribbean, South American, Cocos, and Nazca plates (e.g., Duque-Caro, 1990;
Kellogg *et al.*, 1995; Harmon, 2005; Fig. 1). The Azuero Peninsula in southwestern
Panama represents the Panamanian portion of the arc. Major tectonic structures

110 include the northwest trending Soná-Azuero Fault zone and the east trending Ocú-111 Parita, Agua Clara, and Río Joaquín Fault Zones (Kolarsky et al., 1995; Buchs, 2008; 112 Corral et al., 2011, 2013; Fig. 1).

113 In the Azuero Peninsula the different geodynamic stages are reflected in seven main 114 lithostratigraphic units: 1) the Azuero Igneous Basement, a series of pillow and flow 115 basalts of Coniacian – Early Santonian age (~89-85 Ma; Kolarsky et al., 1995; 116 Lissinna, 2005; Buchs et al., 2009, 2010); 2) the Río Quema Formation, a volcano-117 sedimentary sequence of Campanian to Maastrichtian age (~ 71 to 66 Ma; Wegner et 118 al., 2011; Corral et al., 2013, 2016) that represents the initial stage of the arc; 3) the 119 Cretaceous El Montuoso I-type batholith (~ 69 to 65 Ma; Del Giudice and Recchi, 1969; 120 Kesler et al., 1977; Montes et al., 2012; Corral et al., 2016), which corresponds to arc-121 related plutonism; 4) the Lower Eocene volcanic arc, which developed on top of the 122 Cretaceous arc and corresponds to the Valle Rico I-type batholith (~55 to 49 Ma; Del 123 Giudice and Recchi, 1969; Lissinna, 2005; Montes et al., 2012; Corral et al., 2016); 5) 124 the Parita batholith, Middle Eocene volcanic arc located north of the Cretaceous-Lower 125 Eocene volcanic arc (~45 Ma; Lissinna, 2005; Montes et al., 2012; Corral et al., 2016) 126 (Fig. 1); 6) the Paleocene to Middle Eocene seamounts, oceanic plateaus and a 127 mélange accreted along the ancient subduction trench, which represents the 128 accretionary complex (Hoernle et al., 2002; Lissinna, 2005; Hoernle and Hauff, 2007; 129 Buchs et al., 2011); and 7) the Tonosí Formation, the youngest overlapping 130 sedimentary sequence of Middle Eocene - Early Miocene age (Recchi and Miranda, 131 1977; Kolarsky et al., 1995; Krawinkel et al., 1999). 132

- 133 ------ Insert Figure 2 near here------
- 134

135 3. Geology of the deposit 136 *3.1. Geology*

137 The Cerro Quema high sulfidation epithermal Au-Cu deposit is located in the center of the Azuero Peninsula, and covers an area of ~20 km² (Fig. 1, 3). The mineralization is 138 139 hosted in the dacite dome complex of the Río Quema Formation, a late Campanian to 140 Maastrichtian volcanic sequence that outcrops in the central and southeastern parts of 141 the Peninsula. The Rio Quema Formation is bounded to the north by a series of diorite 142 and quartz diorite batholiths (e.g., the El Montuoso and Valle Rico batholiths) and to 143 the south by the Azuero Igneous Basement. The Cerro Quema deposit is estimated to 144 be early Eocene in age and may be related to subvolcanic porphyry intrusions 145 contemporaneous with the Valle Rico batholith (~55-49 Ma; Corral et al., 2016). 146 In the study area, the Río Quema Formation overlies the Azuero Igneous Basement 147 and is discordantly overlapped by the Tonosí Formation. The unit is affected by an 148 extensive network of faults, predominantly normal faults with northwest- and northeast-149 trend, subvertical dips, and occasional strike slip components. In addition, mesoscale 150 southwest-plunging open folds with moderately dipping limbs are observed in the area 151 (Fig. 3). Deformation indicators (e.g., tension gashes, cataclasites, etc.) are observed 152 mostly in the northern area, whereas a kilometer-scale east-trending syncline characterizes the southern area (Fig. 3). Overall, the structures suggest dextral 153 154 transpression with dominant reverse dip-slip motion during late Campanian to middle 155 Eocene time (Corral et al., 2013).

156

157 ------ Insert Figure 3 near here------

158

159 3.2. Hydrothermal alteration

160 The hydrothermal alteration at Cerro Quema follows an eastward trend that is parallel 161 to secondary faults of the Río Joaquín Fault Zone. It defines concentric alteration halos 162 which, in cross-section, are mushroom-shaped (e.g., La Pava; Fig. 4). Water vapor 163 permeability and porosity of different volcanic rocks of the Río Quema Formation have been measured, obtaining values for dacite rocks (permeability of 2.0.10⁻¹¹ kg/s*m·Pa 164 and 10.8% of porosity), andesite rocks (permeability of 5.1.10⁻¹² kg/s*m.Pa and 7.4% of 165 166 porosity) and basalt rocks (permeability of 1.7.10⁻¹¹ kg/s*m.Pa and 6.4% of porosity). 167 Wall-rock alteration at Cerro Quema is mainly restricted to the dacite domes of the Río 168 Quema Formation, which have higher porosity and permeability than other rock types 169 of the volcano-sedimentary sequence.

170 Four distinct alteration zones have been identified at Cerro Quema on the basis of field 171 mapping, logging, and on analysis of surface and drill core samples by optical 172 microscopy, X-ray diffraction analysis, and scanning electron microscopy-energy 173 dispersive spectroscopy (Fig. 4). Several vuggy quartz centers (up to ~600 m in length) 174 and local advanced argillic alteration zone (up to ~250 m in length), are enclosed by an 175 argillic alteration zone (up to ~1,900 m in length). Propylitic alteration has been 176 observed only in a few drill-core samples, and forms a halo surrounding argillic 177 alteration zone.

178 Vuggy guartz consists of a groundmass of microcrystalline anhedral grains with 179 disseminated pyrite, chalcopyrite, enargite, tennantite, barite, minor rutile, and trace 180 sphalerite. The advanced argillic alteration zone is characterized by quartz, alunite, 181 natroalunite, aluminium-phosphate-sulfate minerals (APS), dickite, pyrophyllite, barite, 182 illite, and minor diaspore and rutile. Argillic alteration produced quartz, kaolinite, illite, 183 illite-smectite, and minor chlorite with local disseminated pyrite. The propylitic alteration 184 zone is characterized by chlorite, epidote, carbonate, rutile, pyrite, chalcopyrite, and 185 minor hematite and magnetite.

186	Intense weathering, typical of tropical latitudes, has affected fresh and hydrothermally
187	altered rocks in the Cerro Quema area to depths up to 150 m. Weathering of the
188	deposit produced a thick quartz- and iron oxide-rich zone (well-developed on the vuggy
189	quartz alteration) that overprints the primary sulfide-bearing zone.
190	
191	Insert Figure 4 near here
192	
193	3.3. Mineralization
194	The gold and copper mineralization is mainly associated with sulfides occurring in the
195	vuggy quartz and advanced argillic alteration zones. Sulfides (±Au-Cu) formed during
196	hydrothermal alteration, as indicated by: 1) the occurrence of disseminated sulfides in
197	the groundmass of hydrothermally altered rocks, 2) the coprecipitation of sulfides and
198	alteration minerals (e.g., alunite, dickite) in the cement of hydraulic breccias, and 3) the
199	occurrence of sulfides filling fissures and in stockworks and voids of hydrothermally
200	altered rocks (e.g., within the vuggy quartz).
201	Gold occurs as disseminated submicroscopic grains and as invisible gold within pyrite.
202	Copper is found in chalcopyrite, enargite, bornite, tennantite, covellite and chalcocite,
203	the latter two interpreted to be of supergene origin.
204	
205	Insert Figure 5 near here
206	
207	Mineralization can be divided into six stages (Corral et al., 2016; Fig. 5), where stages
208	3 and 4 are the main ore-forming stages. Stage 1 consists of disseminated pyrite along
209	with rutile and barite in voids and groundmass with minor chalcopyrite, enargite,

210 tennantite, and trace sphalerite (Fig. 6A). Stage 2 is defined by a disseminated pyrite 211 and chalcopyrite in the cement of hydraulic breccia associated with alunite-natroalunite 212 and dickite (Fig. 6B). Stage 3 consists of pyrite, chalcopyrite, enargite and tennantite 213 veinlets that crosscut Stages 1 and 2 (Fig. 6C). Replacement of pyrite by enargite, 214 enargite by tennantite, and tennantite by chalcopyrite can be observed in the veinlets. 215 Stage 4 is found in breccia bands (~5 cm thick) composed of pyrite and minor 216 chalcopyrite and enargite. These bands crosscut all previous stages (Fig. 6D). Stage 5 217 reflects intermediate sulfidation mineralization as compared to high sulfidation in the 218 previous stages. They are characterized by up to 10 cm wide veins containing pyrite. 219 guartz and barite with minor chalcopyrite, sphalerite and galena. Stage 6 is 220 represented by supergene mineralization which developed an oxide zone and an 221 enrichment zone. The oxide zone is characterized by hematite and goethite, replacing 222 the cement of hydrothermal breccias and filling voids in the vuggy quartz zone. 223 Residual hypogene pyrite, barite and rutile are trace minerals in the oxide zone. At the 224 base of the oxide zone secondary Cu-bearing minerals such as chalcocite and minor 225 covellite constitute a zone of supergene enrichment. Secondary Cu-sulfides replace 226 chalcopyrite, tennantite and enargite and also fill submillimeter-scale fractures. 227 228 229 230 231 ------ Insert Figure 6 near here-----232 233 4. Fluid inclusion study

234 4.1. Sampling and analytical methods

235 Fluid inclusion studies were carried out on six samples of the vuggy quartz and 236 advanced argillic alteration zones of the Cerro Quema deposit, from surface and drill 237 core samples. All microthermometric analyses were performed on fluid inclusions 238 trapped in igneous quartz phenocrysts from the dacites of the Río Quema Formation 239 (Fig. 7A). Due to small sizing, no fluid inclusions were found in the microcrystalline 240 guartz formed during hydrothermal alteration (e.g., vuggy guartz alteration). 241 Microthermometry was performed at the Universitat Autonoma de Barcelona fluid 242 inclusion laboratory. Measurements were made on doubly polished thin sections (about 243 100 µm thick) using a Linkam THMSG-600 heating-freezing stage. The equipment was 244 previously calibrated from the melting of synthetic H_2O-CO_2 fluid inclusions (-56.6°C), 245 melting of ice and critical homogenization at 220 bar (374.1°C) of synthetic H₂O 246 inclusions in quartz. Reproducibility was ± 0.2 °C below 0°C and ± 2°C at the fluid 247 inclusion homogenization temperatures (e.g., Navarro-Ciurana et al., 2016). A total of 248 224 inclusions were measured (Th + Tmi) by cycling down to $-180 \,^{\circ}$ C and heating to 249 the temperature of total homogenization to ensure stability of the inclusions and 250 representativeness of the determinations. The cycling was generally repeated several 251 times in order to avoid nucleation problems during freezing runs. 252 Homogenization of inclusions (Th) was by bubble disappearance (V \rightarrow L). Salinities are

252 Homogenization of inclusions (1n) was by bubble disappearance ($\sqrt{-2L}$). Satinities are 253 expressed as wt. % NaCl equivalent and were estimated from the ice melting 254 temperatures (*Tmi*) for two-phase fluid inclusions (e.g., Bodnar, 1993, 1994). Due to 255 the small inclusion sizes, no eutectic temperatures (*Te*) were measured. Salinity was 256 estimated using the program *Aqso5e*, included in the package FLUIDS (Bakker, 2003) 257 following the equation of Potter *et al.* (1978), for NaCl bearing aqueous solutions of low 258 temperature and low salinity.

- 259
- 260
- 261
- 262 ------ Insert Figure 7 near here-----
- 263

264 *4.2. Fluid inclusion types and occurrence*

265 The fluid inclusion study was carried out on samples affected by different hydrothermal 266 alteration types that are present throughout the deposit (Fig. 4): vuggy quartz from La 267 Pava, Cerro Quemita, Chontal Edge and Cerro Idaida, and advanced argillic from 268 Chontal Edge. Only one type of fluid inclusions was identified on the basis of the 269 number of phases and liquid-to-vapor ratio at room temperature. These inclusions are 270 two-phase (L+V) and are characterized by a dark vapor bubble generally less than 271 50% of the inclusion volume. They correspond to the two-phase liquid-rich (L>V) fluid 272 inclusions described by Shepherd et al. (1985).

The inclusions show a variety of shapes: rounded, elongate or irregular, a few with
negative crystal shapes (Fig. 7B and 7C). They are typically between 5 and 15 µm in
diameter, which hampers observation of phase changes during heating-freezing runs.
The studied fluid inclusions occur isolated, in clusters and following trails, all of them
trapped in igneous quartz phenocrysts from the dacites of the Río Quema Formation
(Fig. 7B). As fluid inclusions were trapped after the volcanic rock crystallization, during
the later hydrothermal alteration, they are interpreted to be secondary in origin.

280

281 4.3. Microthermometrical data

The microthermometry results are summarized in Table 1 and Figure 8. In general, heterogeneous trapping and postentrapment necking down are common features in inclusions from epithermal systems (Bodnar *et al.*, 1985). We studied fluid inclusions in trails, groups and clusters where all inclusions showed similar liquid-vapor phase ratios as well as similar shape.

287 During the microthermometric measurements, phase transitions characteristic of the 288 presence of volatiles (melting of CO₂ around -56.6 °C or presence of CO₂ hydrates) 289 were not observed, implying the absence of significant amounts of volatiles. However, 290 some ice melting temperatures above 0°C were observed (11 of 91 inclusions gave 291 between 0.1 and 1.1 °C), which could indicate small quantities of CO₂ in the 292 hydrothermal fluid. Calculated salinities should therefore be considered as maximum 293 values, as CO_2 depresses the ice melting temperature (Hedenguist and Henley, 1985; 294 Fall et al., 2011).

295

296 4.3.1. Vuggy quartz

- 297 Fluid inclusions in La Pava vuggy quartz sampled from outcrops (520 m above sea
- 298 level; Fig. 8) have an average Th of 140 °C (σ =10 °C) and an average Tmi of -0.9 °C
- 299 (σ = 0.6 °C). The calculated salinity is 1.6 wt% NaCl eq. Inclusions in La Pava vuggy
- 300 quartz sampled from drill core (359 m above sea level; Fig. 8) have an average Th of
- 301 178 °C (σ =11 °C) and an average Tmi of -2.2 °C (σ = 0.5 °C), corresponding to a salinity
- 302 of 3.7 wt% NaCl eq.
- 303 Vuggy quartz collected from Chontal Edge outcrops (334 m above sea level: Fig. 8)
- 304 contains fluid inclusions with an average Th of 196 °C (σ = 19 °C). No Tmi
- 305 measurements are available for this sample due to small inclusion size.
- 306 Further east, fluid inclusions of the vuggy quartz alteration zone from Cerro Quemita
- 307 drill core (791 m above sea level; Fig. 8) are characterized by an average Th of 210 $^{\circ}$ C
- 308 (σ =17 °C). The average Tmi is -0.3 °C (σ = 0.4 °C), equivalent to a salinity of 0.5 wt%
- 309 NaCl eq.
- 310 In the easternmost part of the mineralized area, drill core samples of vuggy quartz from 311 Cerrro Idaida (666 m above sea level; Fig. 8) are characterized by fluid inclusions with 312 an average Th of 216 °C (σ =46 °C). The average Tmi is -2.9 °C (σ = 2.3°C), equivalent 313 to a salinity of 4.8 wt% NaCl eq.
- Whether fluid temperature and salinity varied spatially during vuggy quartz alteration is difficult to determine unambiguously due to sparse data and high standard deviations for some Th and Tmi values. However, temperature increases are suggested from surface to depth in the La Pava deposit and from west to east from La Pava to Cerro Idaida.
- 319
- 320 ------ Insert Figure 8 near here-----
- 321
- 322 4.3.2. Advanced argillic

The scarcity of secondary fluid inclusions in igneous quartz phenocrysts precluded a better systematic study for the advanced argillic alteration-forming fluids. Only one drill core sample from the Chontal Edge could be studied (341 m above sea level, Fig. 8). Fluid inclusions from this sample have an average Th of 205 °C (σ =11 °C), and an average Tmi of -1.4 °C (σ =0.8 °C), which corresponds to a salinity of 2.4 wt% NaCl eq. 328

329 5. Stable Isotopes

330 5.1. Sampling and analytical methods

331 Stable isotope analyses (O, H and S) were carried out on 65 surface and drill core 332 samples from the Cerro Quema deposit at the laboratories of the USGS Crustal 333 Geophysics and Geochemistry Science Center in Denver (USA). Pyrite, chalcopyrite, 334 enargite, barite, vuggy quartz and igneous quartz phenocrysts were handpicked from 335 crushed and sieved samples. Kaolinite, dickite and alunite were separated by 336 decantation methods to obtain the clay fraction and then by centrifugation to separate 337 the different minerals. Mineral separates were checked by X-ray diffraction to assure 338 purity. Identification of the different clay minerals was performed by oriented X-ray 339 diffraction.

340 Sulfates and sulfides were combined with V_2O_5 and combusted in an elemental 341 analyzer, the resulting SO₂ flowed directly into a Thermo Delta mass spectrometer for 342 sulfur isotope measurement (δ^{34} S) according to the method of Giesemann *et al.* (1994), 343 with a precision of $\pm 0.5\%$ (2 σ). Oxygen isotope analyses of sulfates were performed 344 by online high-temperature carbon reduction, with a precision of $\pm 0.6\%$ (2 σ). Silicates 345 were reacted with BrF_5 (Clayton and Mayeda, 1963) and the resulting CO_2 gas was 346 analyzed using a FinniganMAT252 mass spectrometer, with a precision of $\pm 0.3\%$ (2 σ) 347 for δ^{18} O. H isotope ratios of kaolinite-dickite were determined with a FinniganMAT252 348 mass spectrometer, using the method of Vennemann and O'Neil (1993), with a 349 precision of $\pm 2\%$ (2 σ) for δD . Analytical precision was based on replicate analyses. The δ^{18} O and δ D analyses are reported in per mil values relative to V-SMOW, and the 350 351 δ³⁴S values are reported relative to the Vienna Cañon Diablo Troilite standard.

352

353 5.2 Sulfur isotopes

Sulfur isotope data were obtained on pyrite (n=22), enargite (n=8), chalcopyrite (n=1), barite (n=5) and alunite (n=6), (Table 2 and Fig. 9). The range of δ^{34} S of sulfides is from -4.8 to -12.7‰, whereas δ^{34} S of sulfates range from 14.1 to 17.4‰. The δ^{34} S values of sulfides and sulfates are similar to those reported in other high sulfidation epithermal deposits such as Summitville, USA (Rye *et al.*, 1990, 1992; Bethke *et al.*, 2005), Lepanto Phillipines (Hedenquist and Garcia, 1990) and Pueblo Viejo, Dominican Republic (Kesler *et al.*, 1981; Vennemann *et al.*, 1993). Comparing the sulfur

- 361 composition of sulfides and sulfates in the different orebodies of the Cerro Quema
- 362 deposit reveals no significant variability, suggesting a homogeneous sulfur source at
- deposit scale.
- 364
- 365 ------ Insert Figure 9 near here-----
- 366

367 5.3 Oxygen and hydrogen isotopes

- 368 Oxygen isotope analyses were performed on vuggy quartz (n=24), kaolinite (n=19)
- 369 dickite (n=4), alunite (n=6), barite (n=5) and igneous quartz phenocrysts (n=3).
- 370 Hydrogen isotope analysis were performed on kaolinite (n=19) and dickite (n=4).
- 371 Results are shown in Table 2.
- 372 The δ^{18} O values of vuggy quartz (9.0 to 17.5‰) are higher than those of quartz 373 phenocrysts of the Cerro Quema host dacite (8.6 to 8.8‰). In general, igneous quartz 374 phenocrysts in altered rocks retain their primary isotopic compositions (Taylor, 1968; 375 Fifarek and Rye, 2005). This is shown by igneous quartz phenocryst composition of 376 samples QA32 and LP218 (vuggy quartz altered; Table 2), and igneous quartz phenocryst composition of sample LP204 (fresh dacite; Table 2). Therefore, acid 377 leaching and vuggy guartz alteration resulted in a ~ 0.5 to 8.9‰ δ^{18} O enrichment of the 378 379 microcrystalline quartz with respect to the igneous quartz phenocrysts (e.g., sample 380 LP218: 17.5‰ and 8.8‰, respectively; Table 2). The δ^{18} O values of vuggy quartz show 381 a wide variability throughout the Cerro Quema deposit. They become higher from east 382 to west in surface samples (11.8% at Cerro Quema to 17.5% at La Pava) and lower 383 from surface to depth (17.5% at La Pava surface to 12.5% at La Pava at depth).
- 384 The δ^{18} O values of kaolinite range from 12.7 to 18.1% and δ D from -103 to -35%,
- 385 whereas δ^{18} O of dickite varies between 12.7 and 16.3‰ and δ D from -44 to -30‰.
- 386 Although δD values of kaolinite and dickite are highly variable in the Cerro Quema
- 387 deposit, δ^{18} O values, show a trend from higher values in the east to lower values in the
- 388 west (from 18‰ at Cerro Quema to 16‰ at La Pava), in agreement with δ^{18} O values of
- 389 vuggy quartz. Alunite and barite show a wide δ^{18} O range, with alunite varying from -1.6 390 to 9.8‰, and barite ranging from 2.7 to 11.6‰.
- 391

392 6. Discussion

393 6.1 Characteristics of the hydrothermal fluid

- In studying any hydrothermal system it is essential to determine the thermal history, as
 this relates to fluid flow characteristics and geochemical structure (Hedenquist *et al.*,
 1992). Two of the principal physical processes occurring in the epithermal environment
 are boiling and mixing (Giggenbach and Stewart, 1982). Fluid inclusion data (*Th* and
- 398 *Tml*) provide information on these processes at Cerro Quema.
- 399 Results of microthermometric measurements on secondary fluid inclusions from the
- 400 Cerro Quema deposit are shown on a *Tmi* vs. *Th* plot in Figure 10. The relationship
- 401 between homogenization and melting ice temperatures is complicated, probably
- 402 reflecting a complex sequence of fluid events. Only the study of individual samples or a
- 403 group of samples with the same characteristics helps to understand the processes
- 404 recorded by the fluid inclusions.
- 405 When fluid inclusions of the advanced argillic alteration zone (Chontal Edge) and of the 406 vuggy quartz (La Pava surface and depth) are plotted together (Fig. 10A), a trend 407 evolving from high Tmi and low Th towards low Tmi and high Th can be recognized. 408 The most likely interpretation is that they represent a heterogeneous fluid resulting from 409 isothermal mixing of fluids with different origins (e.g., Wilkinson, 2001; Morales-Ruano 410 et al., 2002). However, they can also be interpreted as indicative of a boiling process 411 with slight cooling (Sheppard et al., 1985; Hedenguist and Henley, 1985), although no 412 evidence of boiling were observed in the petrography study.
- 413 On the other hand, in Figure 10B, where fluid inclusions of the vuggy quartz (Cerro 414 Idaida and Cerro Quemita at depth) are plotted, two different trends can be 415 distinguished. A first trend is characterized by fluid inclusions with uniform Tmi and 416 variable Th (a). The second trend is characterized by fluid inclusions with high Tmi and 417 high Th evolving towards low Tmi and low Th (b). According to Sheppard et al. (1985) 418 and Hedenguist and Henley (1985), these trends would be indicative of simple cooling 419 and mixing or dilution of the fluid with cooler and a less saline fluid, respectively. Such 420 processes could explain the high standard deviation (σ) observed in some of the Cerro 421 Quema fluid inclusion data (Th and Tmi).
- 422
- 423 ------ Insert Figure 10 near here-----

425	On the basis of fluid inclusion data, the hydrothermal fluids at Cerro Quema varied from
426	140 to 212 °C, and had low salinity (from 0.5 to 4.3 wt% NaCl). <i>Th</i> vs. <i>Tmi</i> plots (Fig.
427	10) clearly suggest processes such as mixing of magmatic and meteoric fluids. These
428	processes have been documented in several other high sulfidation deposits (e.g.,
429	Arribas, 1995).
430	
431	
432	Insert Figure 11 near here
433	
434	
435	6.2 Sulfur source and geothermometry
436	The isotopic composition and evolution of total sulfur $(\delta^{34}S_{\Sigma S})$ in a hydrothermal fluid
437	may provide insights as to the provenance of sulfur and the conditions of mineral
438	formation. Coexisting sulfides and sulfates may, in turn, be useful for sulfur isotope
439	thermometry. Alunite is an important mineral component of the advanced argillic
440	alteration assemblage and is abundant in drill core samples from the Chontal Edge.
441	Textural and mineralogical relationships indicate that alunite was contemporaneous
442	with associated pyrite.
443	According to Field and Gustafson (1976), Kusakabe et al. (1984) and Field et al. (1983;
444	2005), the use of δ^{34} S of sulfide and sulfate vs. $\Delta^{34}S_{sulfate-sulfide}$ plot is a powerful tool to
445	estimate the $\delta^{34}S_{\Sigma S}$, $X_{S0_4^{-}}$ and X_{H_2S} of the hydrothermal fluid (assuming isotopic
446	equilibrium between sulfate and sulfide). The $\delta^{34}S$ of coexisting pyrite and alunite from
447	Chontal Edge (Table 2) are shown in a δ^{34} S- Δ^{34} S _{sulfate-sulfide} plot in Figure 11A. Linear
448	regression of alunite-pyrite pairs indicates two converging trends; the point of
449	convergence of these two lines on the y axis (δ^{34} S) defines the value for δ^{34} S _{2S} ,
450	whereas the slopes of the upper and lower regression lines approximate the Xso_4^2 and
451	XH ₂ s of the system, respectively.
452	Based on Figure 11A, the $\delta^{34}S_{\Sigma S}$ value of the hydrothermal fluid responsible for the
453	Cerro Quema deposit is -0.5‰, and calculated Xso_4^{2-} and X_{H_2S} are 0.3 and 0.7
454	respectively, with a H_2S/SO_4^{2-} (R) ratio of 2.2. These results suggest that: 1) sulfur in

455 the deposit is of magmatic origin, with a $\delta^{34}S_{\Sigma S}$ value similar to that reported in other

high sulfidation epithermal deposits and porphyry copper deposits associated with Itype granites (Ohmoto and Goldhaber, 1997; Hedenquist and Lowenstern, 1994;
Arribas, 1995); 2) fluids responsible for the pyrite and alunite alteration assemblages
have R value that is in the range of hydrothermal ore-forming fluids (4 ±2; Rye *et al.*,
1992; Hedenquist *et al.*, 1994; Arribas *et al.*, 1995; Arribas, 1995); 3) the magmatichydrothermal system at Cerro Quema was produced by a sulfide dominant
hydrothermal fluid.

463

464 ------ Insert Figure 12 near here-----

465

466 Comparison of the sulfur isotope composition of sulfides and sulfates in the different 467 ore bodies of Cerro Quema reveals no significant variability, suggesting that the sulfur source was homogeneous at the deposit scale. δ^{34} S values of sulfides coexisting with 468 sulfates reflect isotopic equilibrium between H_2S and SO_4^{2-} in the hydrothermal fluid 469 470 (Fig. 9). Equilibrium among these species is typical of magmatic-hydrothermal deposits 471 (Rye, 2005), and has been observed in world class high sulfidation epithermal deposits 472 (e.g., Field and Fifarek, 1985; Arribas, 1995). Using the equation of Rye et al., (1982), 473 calculated equilibrium temperatures for alunite-pyrite pairs range between 221 and 271 474 °C (n=6; Fig. 11B). The average equilibrium temperature is 239 °C (σ = 20), which is 475 slightly higher than the Th values measured in the advanced argillic alteration fluid 476 inclusions (205 ±11 °C).

This difference may be due to the pressure effect on the trapping temperatures of the fluids during the advanced argillic alteration event. Sulfur isotope temperatures are consistent with the disproportionation temperature of magmatic SO₂ to $H_2S + SO_4^{2-}$ in the hydrothermal solution, which is known to be below 400 °C (Sakai and Matsubaya, 1977; Bethke, 1984; Stoffregen, 1987; Rye *et al.*, 1992).

482

483 6.3 δ^{34} S / δ^{18} O of alunite and barite

484 δ^{34} S and δ^{18} Oso₄⁻⁻ values of alunite and barite from the advanced argillic alteration zone 485 at Cerro Quema are shown in Figure 11C. Both minerals present a narrow range of 486 δ^{34} S values (14.1 to 17.4‰), but variable δ^{18} O values, ranging from -1.6 to 11.6‰. 487 According to these isotopic characteristics, the Cerro Quema sulfates fall within the

488 magmatic-hydrothermal field defined by Rye et al. (1992) and Rye (2005). The 489 relatively high δ^{34} S values are consistent with sulfate derivation from sulfuric acid 490 produced by disproportionation of magmatic SO₂, which reacted with the wall rocks to 491 produce the acid-sulfate alteration (Holland, 1965; Stoffregen, 1987). However, the 492 variable δ^{18} O values, which define the vertical trend of the data, suggest that 493 disproportionation may have occurred in mixtures of magmatic and meteoric waters, 494 typically as the result of condensation of magmatic steam in meteoric groundwaters. The effect of mixing between magmatic and meteoric waters is to decrease the δ^{18} O by 495 an amount dependent on the δ^{18} O and the degree of involvement of meteoric waters 496 497 (Rye et al., 1992). This interpretation is in agreement with the microthermometry data 498 obtained from fluid inclusions in the vuggy quartz and the advanced argillic alteration 499 zone, which suggest mixing with cooler and less saline fluids (e.g., meteoric waters).

500

501

502 6.4 H and O isotope composition of hydrothermal fluids

Studies of fluid-mineral isotopic equilibrium in geothermal systems have shown that quartz is very resistant to isotopic exchange, preserving the original isotopic signature (Clayton *et al.*, 1968; Taylor, 1968; Blattner, 1975; Clayton and Steiner, 1975; Fifarek and Rye, 2005). If the hydrothermal fluids had uniform oxygen isotopic composition, the increase in δ^{18} O would reflect decreasing temperatures of mineral precipitation, which is in agreement with the decrease in the homogenization temperatures obtained from fluid inclusions.

Also, δ^{18} O values of vuggy quartz decrease with depth, suggesting an increase in temperature towards the deeper parts of the system, a characteristic of other high sulfidation epithermal deposits such as Summitville, USA (Larson and Taylor, 1987; Bethke *et al.*, 2005). Thus, the δ^{18} O values of vuggy quartz may be useful as an exploration vector in high sulfidation deposits and lithocaps indicating the direction of hotter conditions closer to the fluid source, presumably indicating the direction of a causative porphyry intrusion that may represent a target for Cu mineralization.

517

518 ------ Insert Figure 13 near here-----

520 As shown in Figure 12, the δ^{18} O values of vuggy quartz at Cerro Quema, overlap those 521 from other high sulfidation epithermal deposits such as Summitville (USA; Larson and 522 Taylor, 1987; Bethke *et al.*, 2005), Pueblo Viejo (Dominican Republic; Vennemann *et 523 al.*, 1993), Pierina (Peru; Fifarek and Rye, 2005), Nansatsu (Japan; Hedenquist *et al.*, 524 1994) and Rodalquilar (Spain; Arribas *et al.*, 1995).

525 Using temperatures of quartz formation from the of fluid inclusion data; 160°C for La

- 526 Pava (surface-depth average), 205°C for Chontal Edge, 210°C for Cerro Quemita,
- 527 212°C for Cerro Quema, and 212°C for Cerro Idaida, the calculated δ^{18} O values of the
- 528 vuggy quartz-forming fluid ranges from -2.3 to 3.0% (quartz-water fractionation of
- 529 Matsuhisa *et al.*, 1979). These values plot between arc type magmatic water δ^{18} O
- 530 values (Taylor, 1986; Giggenbach, 1992) and present day meteoric water δ^{18} O values
- 531 (Caballero, 2010). Thus, Cerro Quema vuggy quartz probably precipitated from
- 532 magmatic fluids diluted by variable amounts of meteoric water.
- 533 Kaolinite and dickite are widespread within the argillic and advanced argillic alteration 534 zones at Cerro Quema. The isotopic compositions of these minerals (δ^{18} O and δ D) may 535 reflect the geological conditions during the mineral deposition (e.g., Savin and Lee, 536 1988). For instance, δ^{18} O values of kaolinite of sedimentary origin usually vary from 19 537 to 23‰, whereas those of kaolinite from residual deposits (primary) range from 15 to 538 19‰ (Murray and Janssen, 1984). δ^{18} O and δ D values of dickite and kaolinite from 539 Cerro Quema range from 12.7 to 18.1‰, and from -30 to -103‰, respectively, and are 540 shown in Table 2 and plotted in Figure 13.
- 541 On the basis of mineralogy and $\delta^{18}O/\delta D$ relations, two populations of kaolinite-group
- 542 minerals can be distinguished (Fig. 13). One group lies between the
- 543 supergene/hypogene line (S/H; Sheppard *et al.*, 1969), and the 25°C kaolinite line
- 544 (Savin and Epstein, 1970; Sheppard and Gilg, 1996), corresponding to kaolinites of
- supergene origin. A second group plots between the S/H line and the Arc Type
- 546 Magmatic Waters box and are considered to be of hydrothermal origin.
- 547 At Cerro Quema kaolinite occurred in argillically altered rocks with little presence of
- 548 primary sulfides and similar oxidation overprint. No distinction could be made with
- 549 petrographic observations and X-ray diffraction analysis. In this case, isotope $\delta^{18}O/\delta D$
- values composition seems to be the best way to distinguish between meteoric andhydrothermal kaolinites.
- 552 In order to estimate the isotopic composition of fluids responsible for the kaolinite and 553 dickite precipitation, a temperature range based on fluid inclusion data, mineral

554 paragenesis and isotope geothermometry was used. The assumed temperatures were:

555 200 to 260°C for dickites, 150 to 200°C for hypogene kaolinites, and 25 to 30°C for

556 kaolinites of supergene origin. Calculations have been performed using the

557 fractionation equations of Gilg and Sheppard (1996), and Sheppard and Gilg (1996) for

558 oxygen and hydrogen, respectively.

559

560 ------ Insert Figure 14 near here ------

561

562 The results (Fig. 13) show fluid δD and $\delta^{18}O$ values corresponding to dickite of -28 to -

563 13‰ and 7.1 to 13.3‰, respectively. Fluid δD and $\delta^{18}O$ values corresponding to

564 hypogene kaolinite range from -32 to -15‰ and 6.5 to 12.4‰, respectively. These

565 ranges are compatible with an origin related to the arc-type magmatic waters defined

- 566 by Taylor (1986) and Giggenbach (1992).
- 567 On the other hand, fluid δD and $\delta^{18}O$ values corresponding to supergene kaolinite 568 range from -72 to -20‰ and -10.0 to -5.2‰, respectively (Fig. 13). These results imply 569 formation of supergene kaolinite from water with an isotopic composition similar to 570 present-day meteoric water (PDMW; Caballero, 2010).

571

572 7. Conclusions

573 The findings of the previous discussion are summarized in this section and illustrated in 574 Figure 14, which illustrates our interpretation of the magmatic-hydrothermal system at 575 Cerro Quema. From the perspective of stable isotope geochemistry (O, H, and S) and 576 fluid inclusion microthermometry, the main conclusions are as follows:

577 1) Fluid inclusion data reveal a hydrothermal fluid of low temperature (140 - 216 °C)

578 and low salinity (0.5 - 4.8 wt % NaCl eq.). The fluid was cooler in the western La Pava

579 area and hotter in the eastern Cerro Quema and Cerro Idaida areas, suggesting that

580 the eastern part of the deposit is more proximal to the fluid and heat source.

581 2) The fluid in the magmatic-hydrothermal system was sulfide dominant (X_{H_2} s=0.69)

and the sulfur was of magmatic origin ($\delta^{34}S_{\Sigma S}$ =-0.5‰). However, variable $\delta^{18}O$ at

583 constant δ^{34} S of alunite and barite, and δ^{18} O values of fluids in equilibrium with vuggy

quartz (-2.3-3.1‰), suggest dilution of magmatic fluid by meteoric waters (e.g.,

585 groundwater) during alteration/mineralization.

586 3) Alunite-pyrite sulfur isotope geothermometry indicates average equilibration

temperatures of 239 °C (σ = 20) for the advanced argillic alteration, slightly higher than

588 fluid inclusion measured in the same hydrothermal alteration, 205 °C (σ = 11).

589 4) δ^{18} O values of vuggy quartz decrease from western La Pava area to the eastern

590 Cerro Quema area and from surface to depth. This reflects progressively lower

temperatures of deposition and suggests that the fluid and heat source was situated to

the east of the Cerro Quema deposit and at depth.

593 5) Cerro Quema is a high sulfidation epithermal Au-Cu deposit generated by the

interaction of surface waters with fluids derived from a porphyry copper-like intrusion atdepth to the east of the deposit.

596 6) The combination of fluid inclusion and stable isotope data at Cerro Quema provide

597 vectors for exploration of porphyry Cu-Au deposits which may be applicable elsewhere.

598

599 8. Acknowledgments

600 This study is part of the first author's Ph.D., performed within the framework of the 601 Ph.D. program in Geology of the Universitat Autonoma de Barcelona. The research 602 was supported through the research project CGL2007- 62690/BTE (Spanish Ministry of 603 Science and Education), and a by a predoctoral grant from the "Departament 604 d'Universitats, Recerca i Societat de la Informació (Generalitat de Catalunya)". The 605 corresponding author would like to express his gratitude to the SEG Foundation and 606 the SEG Canada Foundation for the 2009, 2010 and 2011 Hugh E. McKinstry student 607 research grants, which paid for part of the field and related laboratory research 608 expenses. We thank Bellhaven Copper and Gold Inc. for access to mine samples and 609 drill cores used in this study. We really appreciate help and support of Cayce A 610 Gulbransen and George Breit in the understanding of the stable isotope analytical 611 techniques and alunite separation/analytical techniques respectively. We also thank the 612 helpful comments by Ryan Taylor (USGS reviewer) on the final manuscript draft. The

613 manuscript significantly benefited from a thorough review and many constructive

614 comments by Dr. Antonio Arribas and anonymous reviewer. Any use of trade, product,

- or firm names is for descriptive purposes only and does not imply endorsement by the
- 616 U.S. Government.
- 617

618 9. References

- 619 Arribas, A., 1995. Characteristics of high-sulfidation epithermal deposits, and their
- relation to magmatic fluid. In: Thompson, J.F.H. (Ed.), Magmas, Fluids, and Ore
- 621 Deposits, Mineral. Assoc. Can. Short Course vol. 23, pp. 419-454.
- 622 Arribas, A. Cunningham, C.G., Rytuba, J.J., Rye, R.O., Kelly, W.C., Podwysocky, M.H.,
- McKee, E.H., Tosdal, R.M., 1995. Geology, geochronology, fluid inclusions, and
 isotope geochemistry of the Rodalquilar gold alunite deposit, Spain. Econ. Geol., 90,
 795-822.
- Bakker, R.J., 2003. Package FLUIDS 1. Computer programs for analysis of fluid
 inclusion data and for modelling bulk fluid properties. Chem. Geol., 194, 3-23.
- Bethke, P.M., 1984. Controls on base- and precious-metal mineralization in deeper
 epithermal environments. U.S. Geol. Surv. Open-File Rep. 84-890.
- 630 Bethke, P.M., Rye, R.O., Stoffregen, R.E., Vikre, P.G., 2005. Evolution of the
- 631 magmatic-hydrothermal acid-sulfate system at Summitville, Colorado; integration of
- 632 geological, stable isotope, and fluid inclusion evidence. Chem. Geol., 215, 281-315.
- Blattner, P., 1975, Oxygen Isotopic composition of fissure-grown quartz, adularia, and
- 634 calcite from Broadlands geothermal field, New Zealand: Am. J. Sci., 275, 785-800.
- Bodnar, R.J., 1993. Revised equation and table for determining the freezing point
- depression of H₂O-NaCl solutions. Geochim. Cosmochim. Ac., 57, 683-684.
- 637 Bodnar, R.J., Reynolds, T.J., Kuehn, C.A., 1985. Fluid inclusion systematics in
- 638 epithermal systems. In: Berger, B.R., Bethke, P.M. (Eds.), Geology and
- 639 Geochemistry of Epithermal Systems. Rev. Econ. Geol., 2, 73-97.
- 640 Bodnar, R.J., Vityk, M.O., 1994. Interpretation of microthermometric data for H2O-NaCl
- fluid inclusions. In: De Vivo, B., Frezzotti, M.L. (Eds.), Fluid inclusions in Minerals,
 Methods and Applications, 117-130.
- Buchs, D.M., 2008. Late Cretaceous to Eocene geology of the South Central American
- 644 forearc area (southern Costa Rica and western Panama): Initiation and evolution of

- an intra-oceanic convergent margin. PhD thesis, Lausanne, Switzerland, Universitéde Lausanne, 230 pp.
- 647 Buchs, D.M., Baumgartner, P.O., Baumgartner-Mora, C., Bandini, A.N., Jackett, S.J.,

Diswrens, M.O., Stucki, J., 2009. Late Cretaceous to Miocene seamount accretion

- and melange formation in the Osa and Burica Peninsulas (southern Costa Rica);
- episodic growth of a convergent margin. Geol. Soc. Spec. Publ., vol. 328, pp. 411-456.
- Buchs, D.M., Arculus, R.J., Baumgartner, P.O., Baumgartner-Mora, C. and Ulianov,
- A.N, 2010. Late Cretaceous arc development on the SW margin of the Caribbean
 Plate: Insights from the Golfito, Costa Rica, and Azuero, Panama, complexes.
 Geochem. Geophy. Geosy. 11, Q07S24.
- Buchs, D.M., Baumgartner, P.O., Baumgartner-Mora, C., Flores, K. and Bandini, A.N.,
- 657 2011. Upper Cretaceous to Miocene tectonostratigraphy of the Azuero area
- 658 (Panama) and the discontinuous accretion and subduction erosion along the Middle659 American margin. Tectonophysics 512, 31-46.
- 660 Caballero, A., 2010. Exploración de aguas subterráneas en el arco seco de Panamá
- 661 (Sector de Las Tablas) mediante métodos geofísicos. PhD Thesis, Barcelona,
- 662 Spain, Universitat de Barcelona, 271 pp.
- 663 Clayton, R.N. and Mayeda, T.K., 1963. The use of bromine pentafluoride in the
- 664 extraction of oxygen from oxides and silicates for isotopic analysis. Geochim.
- 665 Cosmochim. Ac. 27, 43-52.
- 666 Clayton, R.N. and Steiner, A., 1975. Oxygen isotope studies of the geothermal system
 667 at Wairakei, New Zealand. Geochim. Cosmochim. Ac. 39, 1179-1186.
- 668 Clayton, R.N., Muffler, L.J.P., White, D.E., 1968. Oxygen isotope study of calcite and
 669 silicates of the River Ranch NO. 1 well, Salton Sea geothermal field, California. Am.
 670 J. Sci. 266, 968-979.
- 671 Corral, I., 2013. Geology and Metallogeny of the Cerro Quema Au-Cu deposit (Azuero
- 672 Peninsula, Panama). PhD thesis, Barcelona, Spain, Universitat Autònoma de673 Barcelona, 211 pp.
- 674 Corral, I. Griera, A., Gómez-Gras, D., Corbella, M., Canals, À., Pineda-Falconett, M.,
- 675 Cardellach, E., 2011. Geology of the Cerro Quema Au-Cu deposit (Azuero
- 676 Peninsula, Panama), Geol. Acta. 9, 481-498.
- 677 Corral, I., Gómez-Gras, D., Griera, A., Corbella, M., Cardellach, E., 2013.
- 678 Sedimentation and volcanism in the Panamanian Cretaceous intra-oceanic arc and
- 679 fore-arc: New insights from the Azuero Peninsula, (SW Panama), B. Soc. Geol. Fr.
- 680 184, 35-45.

- 681 Corral, I., Cardellach, E., Corbella, M., Canals, À., Gómez-Gras, D., Griera, A., Cosca,
- 682 M.A., 2016. Cerro Quema (Azuero Peninsula, Panama): Geology, Alteration,
- 683 Mineralization and Geochronology of a Volcanic Dome-Hosted High Sulfidation Au-

684 Cu Deposit, Econ. Geol. 111, 287-310

685 Del Giudice, D. and Recchi, G., 1969. Geología del área del Proyecto Minero de

686 Azuero. Informe técnico preparado para el gobierno de la República de Panamá por

687 las Naciones Unidas. Panama City (Panama), Gobierno de la República de688 Panamá. 48 pp.

689 Dirección General de Rercursos Minerales (DGRM), 1976. Panama Geologic Map
690 (Mapa Geológico de la República de Panamá), scale 1:250,000. , Panama City
691 (Panama).

692 Duque-Caro, H., 1990. Neogene stratigraphy, paleoceanography and

paleobiogeography in Northwest South America and the evolution of the PanamaSeaway. Palaeogeogr. Palaeocl. 77, 203-234.

- Fall, A.S., Tattitch, B., Bodnar, R.J., 2011. Combined microthermometric and Raman
 spectroscopic technique to determine the salinity of H₂O-CO₂-NaCl fluid inclusions
 based on clathrate melting. Geochim. Cosmochim. Ac. 75, 951-964.
- Ferencic, A., 1970. Porphyry copper mineralization in Panama. Miner. Deposita 5, 383-389.
- Field, C.W. and Fifarek, R.H., 1985. Light stable-isotope systematics in the epithermal
 environment. In: Berger, B.R., Bethke, P.M. (Eds.), Geology and Geochemistry of

702 Epithermal Systems, Rev. Econ. Geol. vol. 2, pp. 99-128.

Field, C.W. and Gustafson, L.B., 1976. Sulfur isotopes in the porphyry copper deposit
at El Salvador, Chile. Econ. Geol. 71, 1533-1548.

Field, C.W., Rye, R.O., Dymond, J.R., Whelan, J.F., Senechal, R.G., 1983.

706 Metalliferous sediments of the East Pacific. In: Shanks III, W.C. (Ed.), Cameron

Volume on Unconventional Mineral Deposits. Soc. Mining Eng. Am. Inst. MiningMetall. Petroleum Eng., 133-156.

Field, C.W., Zhang, L., Dilles, J.H., Rye, R.O., Reed, M.H., 2005. Sulfur and oxygen

isotopic record in sulfate and sulfide minerals of early, deep, pre-main stage

711 porphyry Cu-Mo and late main stage base-metal mineral deposits, Butte district,

712 Montana. Chem. Geol. 215, 61-93.

713 Fifarek, R.H. and Rye, R. ., 2005. Stable isotope geochemistry of the Pierina high-

sulfidation Au-Ag deposit, Peru; influence of hydrodynamics on SO_4^{2-} - H₂S sulfur

isotopic exchange in magmatic-steam and steam-heated environments. Chem.

716 Geol. 215, 253-279.

Giesemann, A., Jager, H.J., Norman, A.L., Krouse, H.R., Brand, W.A., 1994. On-line
sulfur isotope determination using an elemental analyzer coupled to a mass
spectrometer. Anal. Chem. 66, 2816-2819.

Giggenbach, W.F., 1992. Magma degassing and mineral deposition in hydrothermal
systems along convergent plate boundaries. Econ. Geol. 87, 1927-1944.

Giggenbach, W.F. and Stewart, M.K., 1982. Processes controlling the isotopic

composition of steam and water discharges from steam vents and steam-heated

pools in geothermal areas. Geothermics 11, 71-80.

- Gilg, H.A. and Sheppard, S.M.F., 1996. Hydrogen isotope fractionation between
 kaolinite and water revisited. Geochim Cosmochim. Ac. 60, 529-533.
- Harmon, R.S., 2005. Geological development of Panama. in Russell S.Harmon (Ed).
- The Rio Chagres, Panama. A muntidisciplinary Profile of a Tropical Watershed, 52,pp. 45-62.

Hedenquist, J.W. and Garcia, J.S., 1990. Sulfur isotope systematics in the Lepanto
mining district, northern Luzon, Phillippines. Mining Geol. 40, 67.

- Hedenquist, J.W. and Henley, R.W., 1985. The importance of CO₂ on freezing point
 measurements of fluid inclusions; evidence from active geothermal systems and
 implications for epithermal ore deposition. Econ. Geol. 80, 1379-1406.
- Hedenquist, J.W. and Lowenstern, J.B., 1994. The role of magmas in the formation ofhydrothermal ore deposits. Nat. Geosci. 370, 519-527.

737 Hedenquist, J.W., Reyes, A.G., Simmons, S.F., Taguchi, S., 1992. The thermal and

- geochemical structure of geothermal and epithermal systems; a framework for
 interpreting fluid inclusion data. Eur. J. Mineral. 4, 989-1015.
- 740 Hedenquist, J.W., Matsuhisa, Y., Izawa, E., White, N.C., Giggenbach, W.F., Aoki, M.,
- 741 1994. Geology, geochemistry, and origin of high sulfidation Cu-Au mineralization in
 742 the Nansatsu District, Japan. Econ. Geol. 89, 1-30.
- Hoernle, K., and Hauff, F., 2007. Oceanic Igneous Complexes. In: Bundschuh, J., and
 Alvarado, G., (Eds.), Central America, geology, resources, hazards vol. 1, pp. 523548.

Hoernle, K., van den Bogaard, P., Werner, R., Lissinna, B., Hauff, F., Alvarado, G.,

- 747 Garbe-Schonberg, D., 2002. Missing history (16-71 Ma) of the Galapagos hotspot:
- 748 Implications for the tectonic and biological evolution of the Americas. Geology 30,749 795-798.
- Holland, H.D., 1965. Some applications of thermochemical data to problems of ore
- deposits; Part 2, Mineral assemblages and the composition of ore forming fluids.
- 752 Econ. Geol. 60, 1101-1166.

- Horlacher, C.F. and Lehmann, J.H., 1993. Regional Geology, Geochemistry and
 Exploration potential of the central Cerro Quema concession, Panama. Tech. Rep.,
 36 pp.
- Kellogg, J.N., Vega, V., Stallings, T.C., Aiken, C.L.V., 1995. Tectonic development of
 Panama, Costa Rica, and the Colombian Andes; constraints from Global Positioning
- 758 System geodetic studies and gravity. Geol. S. Am. S. vol. 295, pp. 75-90.
- 759 Kesler, S.E., Sutter, J.F., Issigonis, M.J., Jones, L.M., Walker, R.L., 1977. Evolution of
- porphyry copper mineralization in an oceanic island arc; Panama. Econ. Geol. 72,1142-1153.
- Kesler, S.E. Russell, N., Seaward, M., Rivera, J., McCurdy, K., Cumming, G.L., Sutter,
 J.F., 1981. Geology and geochemistry of sulfide mineralization underlying the
 Pueblo Viejo gold-silver oxide deposit, Dominican Republic. Econ. Geol. 76, 10961117.
- Kolarsky, R.A., Mann, P., Monechi, S., Meyerhoff-Hull, D., Pessagno, E.A., 1995.
- 767 Stratigraphic development of southwestern Panama as determined from integration
 768 of marine seismic data and onshore geology. Geol. S. Am. S. vol. 295, pp. 159-200.
 769 Krawinkel, H., Wozazek, S., Krawinkel, J., Hellmann, W., 1999. Heavy-mineral analysis
- and clinopyroxene geochemistry applied to provenance analysis of lithic sandstones
- from the Azuero-Sona Complex (NW Panama). Sediment. Geol. 124, 149-168.
- Kusakabe, M. Nakagawa, S., Hori, S., Matsuhisa, Y., Jeda, J.M., Serrano, L., 1984.
- 773 Oxygen and sulfur isotopic composition of quartz, anhydrite, and sulfide minerals
- from the El Teniente and Rio Blanco porphyry copper deposits, Chile. Bull. Geol.
 Surv. Jap. vol. 35, pp. 583-614.
- Larson, P.B. and Taylor, H.P., 1987. Solfataric alteration in the San Juan Mountains,
 Colorado; oxygen isotope variations in a boiling hydrothermal environment. Econ.
 Geol. 82, 1019-1036.
- Leach, T.M., 1992. Petrological Evaluation of the High Sulphidation Systems in the La
 Pava and Cerro Quema Prospect Areas, Panama. Tech. Rep., 55 pp.
- Lissinna, B., 2005. A profile through the Central American Landbridge in western
- 782 Panama: 115 Ma Interplay between the Galápagos Hotspot and the Central
- American Subduction Zone. PhD thesis, Kiel, Germany, Christian-AlbrechtsUniversität zu Kiel, 102 pp.
- Matsuhisa, Y., Goldsmith, J.R., Clayton, R.N., 1979, Oxygen isotopic fractionation in
 the system quartz-albite-anorthite-water. Geochim. Cosmochim. Ac. 43, 1131-1140.
- 787 Montes, C., Bayona, G.A., Cardona, A., Buchs, D.M., Silva, C.A., Morón, S.E., Hoyos,
- 788 N., Ramírez, D.A., Jaramillo, C.A., Valencia, V., 2012. Arc-continent collision and

- orocline formation: Closing of the Central American seaway. J. Geophys. Res. 117,B04105.
- Morales-Ruano, S., Both, R.A., and Golding, S.D., 2002, A fluid inclusion and stable
 isotope study of the Moonta copper-gold deposits, South Australia: Evidence for
- fluid immiscibility in a magmatic hydrothermal system: Chem. Geol. 192, 211-226.
- Murray, H. and Janssen, J., 1984. Oxygen isotopes indicators of kaolin genesis? P.

795 Int. Geol. Cong. 15, 287-303.

- 796 Navarro-Ciurana, D., Corbella, M., Cardellach, E., Vindel, E., Gómez-Gras, D., Griera,
- A., 2016. Petrography and geochemistry of fault-controlled hydrothermal dolomites
 in the Riópar area (Prebetic Zone, SE Spain). Mar. Petrol. Geol. 71, 310-328.
- Nelson, C.E., 1995. Porphyry copper deposits of southern Central America. Arizona
 Geol. Soc. Digest 20, 553-565.
- 801 Ohmoto, H. and Goldhaber, M.B., 1997. Sulfur and carbon isotopes. In: Barnes, H.L.

802 (Ed.), Geochemistry of hydrothermal ore deposits. 517-612.

- Potter, R.W., Clynne, M.A., Brown, D.L., 1978. Freezing point depression of aqueous
 sodium chloride solutions. Econ. Geol. 73, 284-285.
- Recchi, G., Miranda, R., 1977. Calizas de los Planes-Guaniquito (Tonosí). Panama
 City, Dirección General de Recursos Minerales (DGRM). Tech. Rep. 27 pp.
- Rye, R.O., 2005. A review of the stable-isotope geochemistry of sulfate minerals in
 selected igneous environments and related hydrothermal systems. Chem. Geol.
 215, 5-36.
- 810 Rye, R.O., Stoffregen, R.E., Bethke, P.M., 1990. Stable isotope systematics and
- 811 magmatic hydrothermal processes in the Summitville, CO, gold deposit. U.S. Geol.
- 812 Surv. Open-File Rep. 90-626.
- Rye, R.O., Bethke, P.M., Wasserman, M.D., 1992. The stable isotope geochemistry of
 acid sulfate alteration. Econ. Geol. 87, 225-262.
- Sakai, H. and Matsubaya, O., 1977. Stable isotopic studies of japanese geothermal
 systems. Geothermics 5, 97-124.
- Savin, S.M. and Epstein, S., 1970. The oxygen and hydrogen isotope geochemistry of
 clay minerals. Geochim. Cosmochim. Ac. 34, 25-42.
- 819 Savin, S.M. and Lee, M., 1988. Isotopic studies of phyllosilicates. Rev. Mineral.
- 820 Geochem. 19, 189-223.
- Shepherd, T.J., Rankin, A.H., Alderton, D.H.M., 1985. A practical guide to fluid
 inclusion studies. Glasgow, United Kingdom, 239 pp.
- 823 Sheppard, S.M.F., 1986. Characterization and isotopic variations in natural waters.
- 824 Rev. Mineral. 16, 165-183.

- 825 Sheppard, S.M.F., Nielsen, R.L., Taylor, H.P., 1969. Oxygen and hydrogen isotope
- ratios of clay minerals from porphyry copper deposits. Econ. Geol. 64, 755-777.
- 827 Sheppard, S.M.F. and Gilg, H.A., 1996. Stable isotope geochemistry of clay minerals;
- The story of sloppy, sticky, lumpy and tough, Cairns-Smith (1971). Clay Miner. 31,1-24.
- Stoffregen, R.E., 1987. Genesis of acid-sulfate alteration and Au-Cu-Ag mineralization
 at Summitville, Colorado. Econ. Geol. 82, 1575-1591.
- 832 Sutcliffe, R., Kuchling, K., Burga, D., Armstrong, T., Yassa, A., Brown, F., Puritch, E.,
- 833 Tortelli, G., Lightwood, G., Brown, D., Gorman, M., 2014. Cerro Quema Project –
- 834 Pre-feasibility study on the La Pava and Quemita oxide gold deposits. Prepared for
- Preshimco Resources Inc., by P&E Mining Consultants Inc., Golder Associates Inc.,
 and Kappers, Cassiday and Associates. Tech. Rep. 280 pp.
- Taylor, B.E., 1986. Magmatic volatiles; isotopic variation of C, H, and S. Rev. Mineral.
 Geochem. 16, 185-225.
- 839 Taylor, H.P., 1968. The oxygen isotope geochemistry of igneous rocks. Contrib.
- 840 Mineral. Petr. 19, 1-71.
- Taylor, H.P., 1974. The Application of Oxygen and Hydrogen Isotope Studies to
- Problems of Hydrothermal Alteration and Ore Deposition. Econ. Geol. 69, 843-883.
- 843 Vennemann, T.W. and O'Neil, J.R., 1993. A simple and inexpensive method of
- hydrogen isotope and water analyses of minerals and rocks based on zinc reagent.Chem. Geol. 103, 227-234.
- 846 Vennemann, T.W., Muntean, J.L., Kesler, S.E., O'Neil, J.R., Valley, J.W., Russell, N.,
- 1993. Stable isotope evidence for magmatic fluids in the Pueblo Viejo epithermal
- acid sulfate Au-Ag deposit, Dominican Republic. Econ. Geol. 88, 55-71.
- 849 Wegner, W., Worner, G., Harmon, R.S., Jicha, B.R., 2011. Magmatic history and
- evolution of the Central American Land Bridge in Panama since Cretaceous times.Geol. Soc. Am. Bull. 123, 703-724.
- Wilkinson, J.J., 2001, Fluid inclusions in hydrothermal ore deposits: Lithos, v. 55, p.229-272.
- 854
- 855
- 856

857 Figure Captions

- Figure 1: A) Plate tectonic setting of south Central America. B) Simplified geological
- 860 map of the Azuero Peninsula and mineral occurrences. AAG: Azuero Arc Group, ACF:
- 861 Agua Clara Fault, PMF: Punta Mala Fault, RJFZ: Río Joaquín Fault Zone (After

- Dirección General de Recursos Minerales (DGRM), 1976; Buchs *et al.*, 2010; Corral *et al.*, 2011, 2013). Mineral ocurrences: 1) Cerro Quema, 2) Pitaloza, 3) Juan Díaz, 4) Las
- Minas, 5) Quebrada Barro, 6) Quebrada Iguana, 7) Cerro Viejo.
- 865

Figure 2: Overview of Cerro Quema including La Pava, Cerro Quemaita, Cerro Quemaand Cerro Idaida ore zones.

868

Figure 3: Simplified geologic map map of Central Azuero Peninsula, and location of the Cerro Quema deposit (modified from Corral *et al.*, 2011, 2013)

871

Figure 4: Cerro Quema hydrothermal alteration maps and cross section; A) La Pava

873 orebody and Chontal Edge zone. B) Cerro Quemita and Cerro Quema orebodies. C)

874 Interpretation of hydrothermal alteration at depth based on drill holes; 1: PDH 92-10, 2:

875 PDH 91-07, 3: PDH 93-16, 4: PDH 93-11, 5: PDH 93-22, 6: PDH 90-03, 7: PDH 93-12,

876 8: PDH 08-08, 9: PDH 07-08, 10: PDH 06-08, 11: PDH 04-08, 12: PDH 03-08, 13: PDH

877 09-08, 14: PDH 10-08, 15: PDH 91-04, 16: PDH 93-15.

878

879 Figure 5: Paragenetic sequence of ore minerals at Cerro Quema.

880

Figure 6: Microscope images of alteration assemblages at Cerro Quema. A)

B82 Dissemination of pyrite and minor enargite in the vuggy quartz groundmass (stage 1).

B) Alunite + pyrite ± dickite cemented hydraulic breccia (stage 2). C) Pyrite,

chalcopyrite, enargite and tennantite veinlets (stage 3). Note the replacement texturesof pyrite by enargite, enargite by tennantite and enargite-tennantite by chalcopyrite. D)

886 Breccia band (stage 4) composed of pyrite, chalcopyrite, minor enargite and secondary

copper-sulfides (chalcocite). Note the vuggy quartz altered clasts incorporated within
the breccia band. All images are reflected polarized light, excepting B (transmitted
crossed polarized light). alu: alunite, cc: chalcocite, cp: chalcopyrite, di: dickite, enr:
enargite, py: pyrite, qz: quartz, tn: tennantite.

891

892

Figure 7: Fluid inclusion types and occurrences. A) Igneous quartz phenocryst of a
vuggy quartz altered dacite. B) Two-phase (L>V) fluid inclusion trails within a quartz
phenocryst. C) Two-phase (L>V) fluid inclusions randomly distributed in a quartz
phenocryst. All images are taken with transmitted polarized light.

- Figure 8: Fluid inclusion frequency histograms. Homogenization temperature (*Th*) and
 melting ice temperature (*Tmi*) of Cerro Quema fluid inclusions. Discontinue line
 represent fluid inclusions that melted above 0 °C.
- 901

902 Figure 9: δ^{34} S frequency histogram of sulfides and sulfates of Cerro Quema.

903

Figure 10: Homogenization temperature (*Th*) – Melting ice temperature (*Tmi*) plots. A) *Th-Tmi* plot showing fluid inclusion data of vuggy quartz altered dacites form La Pava
(surface and drill core), and of advanced argillic altered dacite at Chontal Edge. B) *Th- Tmi* plot showing fluid inclusion data of vuggy quartz altered dacites from Cerro
Quemita and Cerro Idaida.

909

910 Figure 11: Summary of δ^{34} S and δ^{18} O data on alunite, barite and pyrite. A) δ^{34} S plot of 911 sulfate (alunite) and sulfide (pyrite) pairs vs. delta (Δ) value. The convergence and 912 slope of the two regression lines is an approximation of the bulk sulfur isotopic 913 composition ($\delta^{34}S_{\Sigma S}$) and the proportion of oxidized to reduced sulfur species (Xso₄⁻ and 914 X_{H_2S}) in the hydrothermal fluid (see text for explanation). B) $\delta^{34}S$ values of sulfides and 915 sulfates from the advanced argillic alteration zone. The temperature determined from 916 sulfide-sulfate mineral pairs is also shown. C) δ^{34} S and δ^{18} O of barite and alunite showing a vertical trend, indicating a mixing between magmatic sulfate and meteoric 917 918 waters (see text for explanation).

919

Figure 12: Diagram showing the range of vuggy quartz δ¹⁸O values and δ¹⁸O values of
vuggy quartz-froming fluids. For comparison data from Rodalquilar (Arribas *et al.*,
1995), Pierina (Fifarek and Rye, 2005), Nansatsu (Hedenquist *et al.*, 1994), Pueblo
Viejo (Vennemann *et al.*, 1993), and Summitville (Larson and Taylor, 1987; Bethke *et al.*, 2005) has been represented. Local meteoric waters field from Caballero (2010); arc
type magmatic waters field from Taylor (1986) and Giggenbach (1992). Rdl:
Rodalquilar, Pier: Pierina, Nan: Nansatsu, PV: Pueblo Viejo, Sum: Summitville.

927

Figure 13: Plot of δ^{18} O and δ D values of kaolinite and dickite, and the values of water in equilibrium with them. PDMW: Present day meteoric water (Caballero, 2010), PMW: Primary magmatic waters (Taylor, 1974; Sheppard, 1986), ATMW: Arc-type magmatic waters (Taylor, 1986; Giggenbach, 1992), S/H line: Supergene/Hypogene line (Sheppard *et al.*, 1969), Kaolinite line (25°C; Savin and Epstein, 1970; Sheppard and Gilg, 1996).

- 935 Figure 14: Conceptual model of Cerro Quema hydrothermal system integrating the fluid
- 936 inclusion (temperatures) and the stable isotope (processes) interpretations.
- 937 Microthermometric and geochemical data presented in this work suggest the location of
- a causative porphyry copper-like intrusion towards the east of the deposit at certaindepth.
- 940
- 941 Table 1: Summary of microthermometric results for fluid inclusions of the Cerro Quema
- 942 deposit. All measurements were performed in quartz phenocrysts (see text for
- 943 explanation). AAA: advanced argillic alteration, Th: homogenization temperature, Tmi:
- 944 melting ice temperature, VQ: vuggy quartz alteration.
- 945
- Table 2: Summary of the stable isotope data (S, O, H) from the different minerals
- 947 analyzed in this study. al: alunite, ba: barite, cpy: chalcopyrite, en: enargite, kaol-dck:
- 948 kaolinite-dickite, py: pyrite.
- 949