

## Bactericidal efficacy of UV activated TiO<sub>2</sub> nanoparticles against Gram-positive and Gram-negative bacteria on suspension

Carolina Ripolles-Avila , Maria Martinez-Garcia, Anne-Sophie Hascoët and José Juan Rodríguez-Jerez 

Hygiene and Food Inspection Unit, Department of Food and Animal Sciences, Facultat de Veterinària, Universitat Autònoma de Barcelona, Barcelona, Spain

### ABSTRACT

Two different TiO<sub>2</sub> nanoparticles, NM101 and NM105, were evaluated against a range of Gram-positive (*Staphylococcus aureus*, *Bacillus cereus*, *Lactobacillus casei*, *Lactobacillus bulgaricus*, *Lactobacillus acidophilus* and *Lactobacillus lactis*) and Gram-negative (*Salmonella enterica* var. Enteridis and *Escherichia coli*) bacteria. Both NM101 and NM105 TiO<sub>2</sub> nanoparticles (UV-exposed or none) had a significant antibacterial activity when the concentration of TiO<sub>2</sub> suspension was 100 µg mL<sup>-1</sup>. The activation of the TiO<sub>2</sub> NPs led, in all cases, to a shift in the growth curve, revealing lower counts as the concentration increased. *E. coli* was the most significantly affected pathogen by both TiO<sub>2</sub> nanoparticles reaching among 2–3 log CFU.mL<sup>-1</sup> reduction. In addition, in the case of the probiotic bacteria, NM105 TiO<sub>2</sub> nanoparticles had similar effects as the bacterial density was reduced by 2–3 log CFU.mL<sup>-1</sup>. These results may be applied as a potent technology to be included in the formulation of new disinfectants.

### ARTICLE HISTORY

Received 30 October 2018  
Accepted 27 February 2019

### KEYWORDS

TiO<sub>2</sub>; nanoparticles;  
photocatalytic disinfection;  
UV; food hygiene

### PALABRAS CLAVE

TiO<sub>2</sub>; nanopartículas;  
desinfección fotocatalítica;  
UV; higiene alimentaria

### Eficacia bactericida de nanopartículas de TiO<sub>2</sub> activadas por luz UV contra bacterias Gram-positivas y Gram-negativas en suspensión

### RESUMEN

Se evaluaron dos nanopartículas diferentes de TiO<sub>2</sub>, denominadas NM101 y NM105, contra una batería de bacterias Gram-positivas (*Staphylococcus aureus*, *Bacillus cereus*, *Lactobacillus casei*, *Lactobacillus bulgaricus*, *Lactobacillus acidophilus* and *Lactobacillus lactis*) y Gram-negativas (*Salmonella enterica* var. Enteridis and *Escherichia coli*). Ambas nanopartículas de TiO<sub>2</sub>, NM101 y NM105 (expuestas o no a luz UV) tuvieron una actividad antibacteriana significativa cuando la concentración de suspensión de TiO<sub>2</sub> alcanzó los 100 µg mL<sup>-1</sup>. La activación de las nanopartículas de TiO<sub>2</sub> condujo, en todos los casos, a un desplazamiento en la curva de crecimiento microbiano, revelando recuentos más bajos a medida que concentración de TiO<sub>2</sub> aumentaba. *E. coli* fue el patógeno más afectado, alcanzando entre 2-3 log UFC.mL<sup>-1</sup> de reducción. Además, en el caso de las bacterias probióticas, las nanopartículas NM105 TiO<sub>2</sub> tuvieron efectos similares, ya que la densidad bacteriana se redujo en 2-3 log UFC.mL<sup>-1</sup>. Estos resultados podrían aplicarse como una potente tecnología a incluir en la formulación de nuevos desinfectantes

## 1. Introduction

Food safety is an important topic of food industrial development (Wang et al., 2016). This is mainly related to food-borne diseases which have been a widespread health problem in the contemporary world and an important cause of reduced economic productivity (WHO, 2015).

Even though disinfection methods commonly employed in the food industry can control microbial pathogens, researchers have shown that there is an increase of bacteria resistant to conventional treatments (González-Rivas, Ripolles-Avila, Fontecha-Umaña, Ríos-Castillo, & Rodríguez-Jerez, 2018; Zhao, Zhao, Wang, & Zhong, 2017). Consequently, there is a necessity to reevaluate traditional methods of disinfection, as well as, to consider new approaches oriented to improve the reliability and robustness of disinfection whilst eluding the resistance of bacteria (Li et al., 2008; Özkaleli & Erdem, 2017; Ripolles-Avila, Ríos-Castillo, & Rodríguez-Jerez, 2018). In this sense, a technology that has stimulated a significant interest in finding new possibilities in

this area is nanotechnology. In fact, it has been shown that many nanoparticles (NPs) such as copper oxide, silver and zinc oxide possess antimicrobial properties and could be a possible promising solution (Bondarenko et al., 2013; Duffy, Osmond-McLeod, Judy, & King, 2018; Sirelkhatim et al., 2015). However, there are some concerns about the residues liberation of these nanomaterials to the environment, and how the different organisms present in the environment may be affected (Bondarenko et al., 2013). With this respect, Kahru and Dubourguier (2010) demonstrated that over seven nanoparticles analyzed (ZnO, Ag, CuO, TiO<sub>2</sub>, C60-fullerenes, SWCNTs and MWCNTs), TiO<sub>2</sub> NPs were the least environmentally harmful. Furthermore, International Agency for Research on Cancer (IARC) classified TiO<sub>2</sub> as a human carcinogen group 2B although this category is used for chemicals for which less than sufficient evidence of carcinogenicity in experimental animals and also there is limited evidence of carcinogenicity in humans (Jovanović, 2015a). Furthermore, something to have into consideration is the lack

of consistent toxicity data of TiO<sub>2</sub> NPs due to the diversity of materials and test conditions followed by the different researchers, which makes a real challenge the fact of conducting risk assessment (Jovanović, 2015b).

TiO<sub>2</sub> is a semiconductor which upon UV light excitation, it can react with H<sub>2</sub>O or hydroxide ions adsorbed on the surface to produce hydroxyl radicals ( $\cdot\text{OH}$ ), or reduce O<sub>2</sub> to produce superoxide ions (Banerjee, Muraleedharan, Tyagi, & Raj, 2006). In addition, it has been shown that there is a production of reactive oxygen species (ROS), such as hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) and singlet oxygen, which are closely related to cell damage, when TiO<sub>2</sub> is activated by UV light (Carp, Huisman, & Reller, 2004; Foster, Ditta, Varghese, & Steele, 2011). All of those ROS can cause damage to living cells. Three principal polymorphs of TiO<sub>2</sub> exist: anatase, rutile and brookite (Landmann, Rauls, & Schmidt, 2012). In fact, some studies have revealed that anatase form is the most active photocatalyst and, rutile normally less effective (Foster et al., 2011). Nevertheless, Miyagi, Kamei, Mitsunashi, Ishigaki, and Yamazaki (2004) have shown that rutile and anatase mixtures have a higher photocatalytic effect than only anatase and also are more efficient to generate ROS.

Matsunaga (1985) reported the effect of photocatalytic inactivation of several microorganisms such as *E. coli* and *L. acidophilus* using nano-TiO<sub>2</sub>. Since then, research on nano-TiO<sub>2</sub> photocatalytic effect has been intensively conducted on a wide range of microorganisms to evaluate photocatalytic disinfection activities. Furthermore, it would be also worth to evaluate the effects of TiO<sub>2</sub> embedded in surfaces. As it has been proposed by other authors, surfaces made up of light-guiding materials, coated with specific semiconductors (i.e. TiO<sub>2</sub> NPs) and stimulated by UV light could provide a new way of disinfection (Barthomeuf et al., 2017; Jalvo, Faraldos, Bahamonde, & Rosal, 2017; Yemmireddy & Hung, 2017). In fact, the ability of TiO<sub>2</sub> for microbial elimination on photocatalytic self-cleaning/self-disinfecting surfaces may contribute an additional mechanism to the control of disease transmission. Thus, photooxidation mediated by TiO<sub>2</sub> appears to be promising for microbial elimination in areas where the employment of biocides or chemical cleaning agents is not effective or is restricted by food industry regulations (Yemmireddy & Hung, 2015). However, the bactericidal effect of TiO<sub>2</sub> NPs seems to be higher when applied in suspension rather than embedded in surfaces (Kim, Kim, Cho, & Cho, 2003), although comparative effects need to be investigated. Foster et al. (2011) have shown that a closer contact within the TiO<sub>2</sub> and the organism enhances the amount of oxidative damage. On this sense, it has been demonstrated that on water suspended TiO<sub>2</sub> has a greater effect than TiO<sub>2</sub> immobilized on surfaces like thin films (Alhaji et al., 2017; Sun, Tay, & Tan, 2003). This is presumably because of extended contact between the microbial cells in suspension and the TiO<sub>2</sub> NPs, augmenting the surface area for ROS production. A number of studies have demonstrated the significance of such contact (Caballero, Whitehead, Allen, & Verran, 2009; Cheng et al., 2009; Horie, David, Taya, & Tone, 1996).

This study aims to consider the potential of TiO<sub>2</sub> for the development of new surface disinfection alternatives to be employed in the food industry. The main purpose was to compare the bactericidal effect of photoactivated and non-photoactivated TiO<sub>2</sub> NPs on a range of bacterial indicators of

contamination in food, both pathogenic and spoilage microorganisms, tested on suspension, as a way to reduce the use of other disinfectants and contemplate it as an alternative. Two different forms of TiO<sub>2</sub> NPs were used in this study, one containing purified anatase form and the other containing a mixture of anatase/rutile phase. The ability of all the microorganisms to survive photo-inactivation with the UV light was assessed by varying the periodicity of irradiation. Furthermore, the turbidity of the two NPs in the different media used in the experiment and the photoactivation of the TiO<sub>2</sub> NPs was also evaluated.

## 2. Materials and methods

### 2.1. Nanoparticles

Two different types of TiO<sub>2</sub> nanoparticles named NM101 and NM105 were employed in this study. The NM101 NPs were obtained from the nanomaterials repository at the Institute for Health and Consumer Protection at the European Commission Joint Research (Ispra, Italy) and NM105 NPs were purchased from Degussa (Frankfurt, Germany). Sample NM101 consisted of anatase crystal phase with 7 nm size. Besides, sample NM105 was a combination of anatase-rutile phase (80:20 wt/wt) with 21 nm size. The main characteristics of the nanoparticles provided by the individual supplier are listed in Table S1.

Suspensions of TiO<sub>2</sub> nanoparticles at concentration of 1 mg mL<sup>-1</sup> using Nutrient Broth (NB; Oxoid, Hampshire, United Kingdom) and MRS (Oxoid, Hampshire, United Kingdom) were prepared and then vortexed (SLS, Southampton, United Kingdom) to mix the preparations in order to have a homogeneous mixture. Fresh suspensions were prepared prior to each experiment.

### 2.2. Bacterial cultures

*Salmonella enterica* var. Enteridis (ATCC 13076), *Escherichia coli* (ATCC 25922), *Staphylococcus aureus* (ATCC 29247) and *Bacillus cereus* (ATCC 29247) were purchased as a Culti-Loops (Remel Europe, Dartford, United Kingdom). The Culti-Loops were then streaked onto sterile plate count agar (PCA; Oxoid, Hampshire, United Kingdom) to obtain pure cultures on plates. After streaking, the tips of the Culti-Loops which contained the microorganisms were separated and deposited in sterile nutrient broth. Both the inoculated plates and broths were incubated at 37 ± 2 °C during 24 h. After that, the inoculated plates were sealed with Parafilm and kept as stock cultures at 4 ± 2 °C for a maximum period of 1 month. The broth cultures were subcultured into new sterile nutrient broth every day by transferring 200 µL from the initial broth culture and incubated at 37 ± 2 °C for 24 h.

Probiotic strains *Lactobacillus casei* (ATCC 393), *Lactobacillus delbrueckii subsp. bulgaricus* (ATCC 11842), *Lactobacillus lactis subsp. lactis* (ATCC 19435) and *Lactobacillus acidophilus* (ATCC 4356) were provided by Nutraceutical, Montreal, Canada. Individual strains were cultivated in MRS liquid medium (Oxoid, Hampshire, United Kingdom) and incubated at 37 ± 2 °C as static cultures for 48 h. Then, broths were subcultured into 20 mL of new sterile MRS liquid medium every 48 h by transferring 200 µL from the initial broth culture and then were kept at 37 ± 2 °C for 48 h more.

### 2.3. Inoculum preparation

Seed cultures of the four pathogens were inoculated into 100 mL nutrient broth and the four probiotic seed cultures were inoculated into 20 mL MRS. Cell concentration was adjusted at approximately  $10^7$  colony forming units (CFU).  $\text{mL}^{-1}$  by measuring the absorbance of bacterial suspension at 650 nm using a UV/Vis spectrophotometer (Helios Epsilon, Thermo Scientific, Loughborough, United Kingdom). This cell adjustment was done using the calibration curves obtained by each microorganism.

### 2.4. Impact of the NPs on the turbidity of the media

A turbidity testing of NM101 and NM105  $\text{TiO}_2$  was done to investigate whether nano- $\text{TiO}_2$  would change the turbidity of the media through time. A range of 8 different concentrations from 0.78 to  $100 \mu\text{g mL}^{-1}$  of both NPs were prepared and measured in nutrient broth and MRS. Figure 1 shows representative results for NM101  $\text{TiO}_2$  in NB; which are consistent with the results observed for the NM105  $\text{TiO}_2$  nanoparticle in NB media (supplementary materials).  $\text{TiO}_2$  showed a different turbidity between concentrations but it did not change the turbidity of the media over the time. In concordance, non-significant differences ( $P > 0.05$ ) were found between the turbidity achieved by the diverse concentrations of the nanoparticles tested.

### 2.5. Photocatalytic experiments

The photocatalytic bactericidal tests were performed in sterile microtitre plates. Twenty microliters of the aqueous suspensions of  $\text{TiO}_2$  nanoparticles (both NM101 and NM105) were added to each well in triplicate along with 180  $\mu\text{L}$  of bacterial cultures to give final nanoparticle concentration of  $0.78\text{--}100 \mu\text{g mL}^{-1}$ . In all the tests, four different conditions were examined: (i) presence of  $\text{TiO}_2$  and UV light, (ii) no  $\text{TiO}_2$  and presence of UV light, (iii) presence of  $\text{TiO}_2$  under dark conditions, (iv) no  $\text{TiO}_2$  under dark conditions.

The microtitre plates with bacteria-nanoparticles suspension were located in a noncommercial chamber which was fitted with a UVSPOT 1000 UV lamp (UV Technologies,

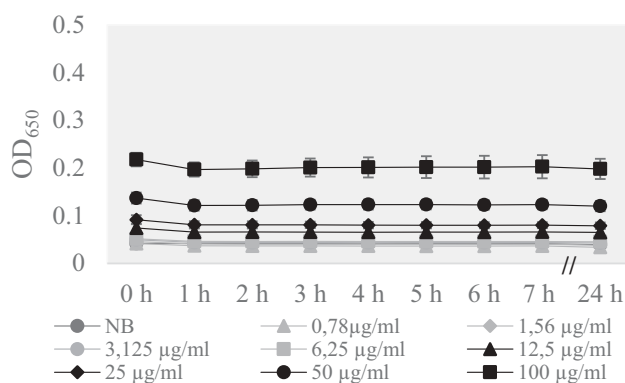
Germany), emitting radiation over 315–400 nm. The distance within the samples and the lamp was set up to acquire  $6.9 \text{ mW cm}^{-2}$  of UV light incident on the samples. As *S. aureus* was more sensitive to UV light than the rest of microorganisms (see section 3.2), the time of UV light exposure was set up at 6 min, which was the time period that did not compromise the growth of the microorganism. Furthermore, UV intensities higher than  $1 \text{ mW cm}^{-2}$  were always applied, which has been shown to be sufficient to begin the photocatalytic reaction at the established exposure time (Oka et al., 2008). After the UV exposure, samples were shaken to resuspend the cultures and the optical density was measured at 650 nm wavelength on a microplate reader (Spectra MAX M5, Molecular Devices, California, United States). The microtitre plates were incubated at  $37 \pm 2^\circ\text{C}$  during 24 h. After the incubation, samples were shaken again and the optical density was measured at 650 nm wavelength on the same microplate reader.

### 2.6. Correlation between optical density (OD) values and $\text{CFU mL}^{-1}$

Serial dilutions of all microorganisms (ranging from  $10^{-1}$  until  $10^{-9}$ ) were prepared in 200  $\mu\text{L}$  of nutrient broth and MRS broth using sterile microtitre plates and the optical density was measured at 650 nm wavelength on a microplate reader. From the dilution  $10^{-3}\text{--}10^{-9}$ , 100  $\mu\text{L}$  of aliquots were withdrawn and plated in sterile petri dishes with plate count agar (PCA and MRS, respectively, for pathogens and probiotic bacteria). The plates were incubated at  $37 \pm 2^\circ\text{C}$  for 24 h (*S. enterica* var. Enteridis, *E. coli*, *S. aureus* and *B. cereus*) and 48 h (*L. casei*, *L. bulgaricus*, *L. acidophilus* and *L. lactis*). All microbial counts were reported as  $\text{CFU mL}^{-1}$  values.

### 2.7. Dichloro-dihydro-fluorescein diacetate (DCFH-DA) assay

The level of ROS generated by  $\text{TiO}_2$  NPs was detected by carboxy-2',7'-dichloro-dihydro-fluorescein diacetate (DCFH-DA) staining. DCFH-DA is a fluorescent probe that can be deacetylated to form DCFH. DCFH is non-fluorescent unless oxidized by ROS to transform DCF. The fluorescence intensity of DCF can indicate the quantity of ROS produced (Sun et al., 2011). Ten microliters of DCFH-DA probe (Sigma-Aldrich, Poole, UK) and 40  $\mu\text{L}$  of sodium NaOH (Sigma-Aldrich, Poole, UK) solution were added to the wells of a microtitre plate in triplicate and incubated for 30 min at room temperature. After this period, 68  $\mu\text{L}$  of sterile phosphate buffered saline (PBS, Sigma-Aldrich, Poole, UK) solution was added to the mixture. In parallel, suspensions of  $\text{TiO}_2$  NPs (both NM101 and NM105) in nutrient broth and MRS broth were exposed to UV light during 6, 12 and 24 min. After the UV exposure, 118  $\mu\text{L}$  of the  $\text{TiO}_2$  nanoparticle suspension was added to the DCFH-DA+NaOH+PBS mixture and incubated during 10 min at room temperature. The investigated range of nanoparticle concentrations was between 0.78 and  $100 \mu\text{g mL}^{-1}$ . After the incubation, fluorescence was measured every 1 min until 15 min and then at 30, 45 and 60 min at 485 nm (excitation) and 535 nm (emission) wavelengths on a microplate reader.



**Figure 1.** Optical density measurements of different concentrations of NM101  $\text{TiO}_2$  in nutrient broth over time. Each value corresponds to the mean of three replicates performed on three separated days ( $n = 9$ ). The error bars represent standard deviation.

**Figura 1.** Densidad óptica obtenida a partir de diferentes concentraciones de NM101  $\text{TiO}_2$  en caldo nutritivo a lo largo del tiempo. Cada valor corresponde a la media de tres repeticiones realizadas en tres días separados ( $n = 9$ ). Las barras de error representan la desviación estándar.

## 2.8. Statistical analysis

All experiments were performed three times on three different days ( $n = 9$ ). Data from all experiments were analyzed using the Minitab statistical package using ANOVA general linear model with subsequent analysis of variance and Tukey's test. A  $P$  value of  $<0.05$  was considered to be significant. The global differences were examined followed by an analysis of the individual data sets from the two  $\text{TiO}_2$  nanoparticles.

## 3. Results

### 3.1. Effect of UV light over microorganisms

A preliminary study was conducted to select the time of exposure of the samples (bacteria + NPs) and to confirm that the time of UV exposure would not have impact on the growth of the bacteria. Suspensions of *S. enterica* var. Enteridis, *E. coli*, *B. cereus* and *S. aureus* in nutrient broth and suspensions of *L. casei*, *L. bulgaricus*, *L. acidophilus* and *L. lactis* in MRS broth were exposed to 3, 6, 12, 24, 36 and 48 min of UV light with an intensity of  $6.9 \text{ mW cm}^{-2}$  incident on the samples. The microorganism that showed to be more sensitive to UV light was *S. aureus*, increasing its sensitivity when the time of exposure was raised (Figure 2). Only an exposure period of 3 and 6 min did not have an impact on the growth of *S. aureus*. Because of that, 6 min of UV exposure was selected in order to maximize the potential for photoactivation of nano- $\text{TiO}_2$ . The microorganism less affected on its growth curve by UV light was *S. enterica* var. Enteridis (supplementary materials). In addition, it is also important to highlight that UV light did not have any impact on the turbidity of the culture broths employed in the study (i.e. NB and MRS). As can be observed in Figure 2, the increase of the exposure time to UV led to no change in the obtained  $\text{OD}_{650}$  values on NB and MRS broths.

### 3.2. Photoactivation test with a fluorescent probe

Results showed that DCF fluorescence intensity slightly increased through the time, increasing in parallel the ROS level with the increase of the concentration of  $\text{TiO}_2$  NPs from  $0.78$  to  $100 \mu\text{g mL}^{-1}$  although non-significantly ( $P > 0.05$ ) (Figure 3). Moreover, no significant differences ( $P > 0.05$ ) were found between the ROS level produced after exposure of the UV samples with respect to those subjected to dark conditions, suggesting that there is no need to expose the samples with UV so that the  $\text{TiO}_2$  NPs could play an antimicrobial role.

### 3.3. Effect of photocatalytic $\text{TiO}_2$ NPs on different microorganisms

The inhibitory effects of both  $\text{TiO}_2$  NPs, NM101 and NM105, on pathogenic bacteria is shown in Figure 4. All the pathogens in nutrient broth decreased their growth in a dose-dependent manner following exposure to NM101 and NM105  $\text{TiO}_2$ . The bacterial culture of *S. enterica* var. Enteridis showed a significant decrease ( $P < 0.01$ ) in the  $\text{OD}_{650}$  value at concentrations of 50 and  $100 \mu\text{g mL}^{-1}$  of NM101 and NM105  $\text{TiO}_2$ , compared to the control suspension. The pathogen *E. coli* showed a significant growth

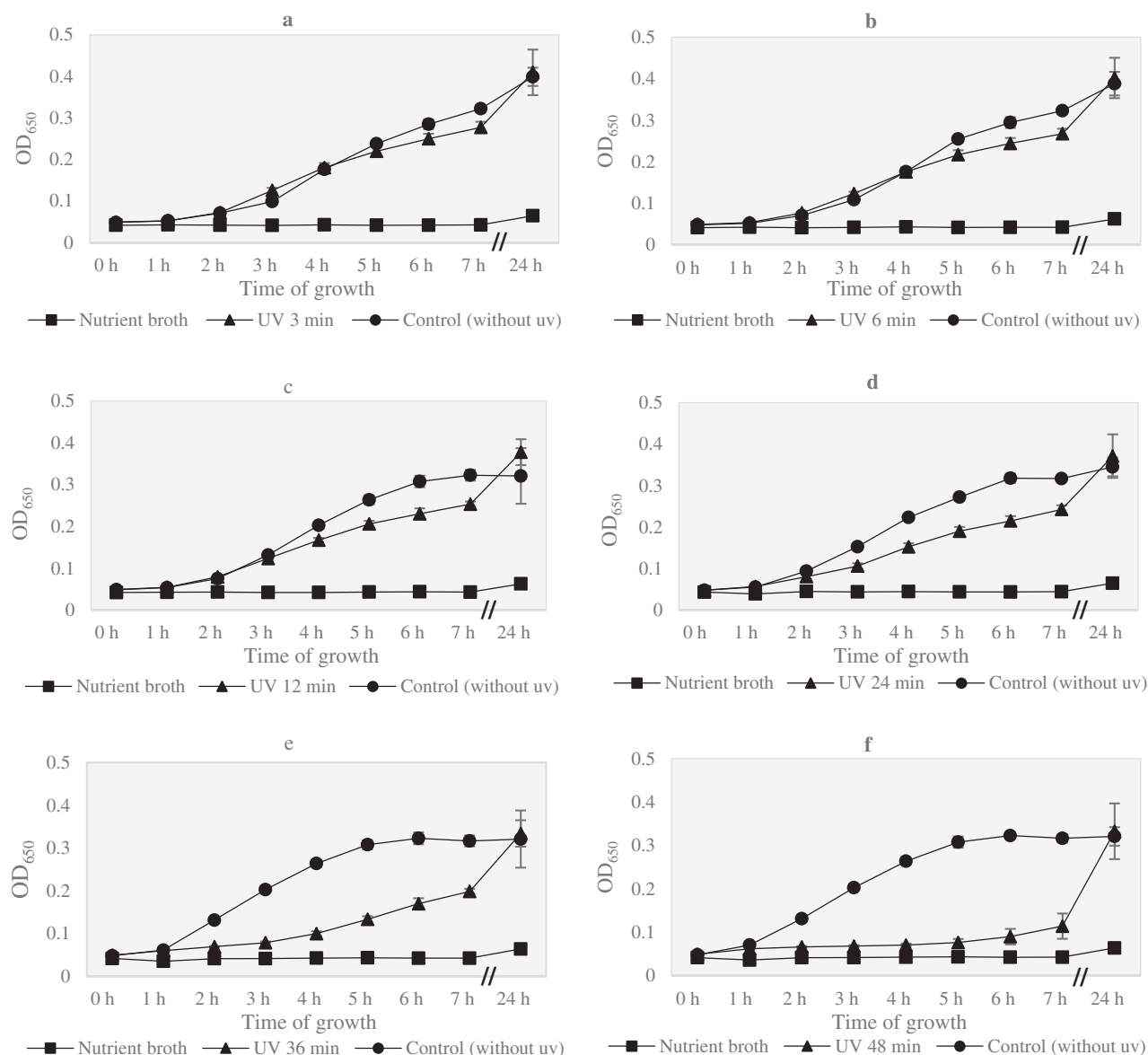
decrease ( $P < 0.01$  and  $P < 0.001$ ) in the  $\text{OD}_{650}$  at concentrations of  $12.5 \mu\text{g mL}^{-1}$  when compared to the control. Using the standard curves obtained from the relationship between absorbance and  $\text{CFU.mL}^{-1}$ , an estimated 2 log  $\text{CFU.mL}^{-1}$  reduction of the initial concentration of *E. coli* was obtained when NM101 and NM105  $\text{TiO}_2$  was activated by 6 min UV light exposure. In contrast, when both  $\text{TiO}_2$  NPs were not UV activated ( $\text{TiO}_2 + \text{UV}^-$ ), *E. coli* was 2.3–3 log  $\text{CFU.mL}^{-1}$  reduced, respectively, for NM101 and NM105. In the case of *S. aureus* and *B. cereus*, a significant decrease in the  $\text{OD}_{650}$  value at concentrations of 50 and  $100 \mu\text{g mL}^{-1}$  of non-photoactivated ( $P < 0.001$ ) and photoactivated ( $P < 0.05$ ) NM101  $\text{TiO}_2$  was observed when compared with the control. However, when *S. aureus* and *B. cereus* were exposed to NM105  $\text{TiO}_2$  showed a significant ( $P < 0.01$  for *S. aureus* and  $P < 0.05$  for *B. cereus*) decrease in the  $\text{OD}_{650}$  value just at the highest concentration (i.e.  $100 \mu\text{g mL}^{-1}$ ). Again, estimating the reduction in terms of  $\text{CFU.mL}^{-1}$  using the standard curves for each microorganism, in the presence of  $\text{TiO}_2$  and dark conditions, NM101 at  $100 \mu\text{g mL}^{-1}$  reduced 2 log  $\text{CFU.mL}^{-1}$  the initial microbial load of *S. aureus* and *B. cereus*, and NM105 reduced the load of them by 2.69 log  $\text{CFU.mL}^{-1}$ . With all these results, growth curves of the different pathogens (including Gram-positive and Gram-negative) have been observed to be significantly affected in a directly proportional way as applied  $\text{TiO}_2$  concentration increased.

From the data of the impact of photoactivated and non-photoactivated  $\text{TiO}_2$  NPs on probiotic bacteria, it was evident that there was a dose-dependent decrease in the  $\text{OD}_{650}$  values after 48 h of incubation (Figure 5). There was no significant difference between photoactivated and non-photoactivated  $\text{TiO}_2$ , showing in all cases, the same effects on probiotic bacteria ( $P > 0.05$ ). Furthermore, the  $\text{OD}_{650}$  values of the probiotic cultures showed no significant changes during the tests under  $0.79$  to  $25 \mu\text{g mL}^{-1}$  NM101 and NM105  $\text{TiO}_2$  ( $P > 0.05$ ). However, using the standard curves to obtain an estimated result of  $\text{CFU.mL}^{-1}$ , it was apparent that NM105  $\text{TiO}_2$  UV activated and non-activated (i.e. dark conditions) induced a very significant 3 log  $\text{CFU.mL}^{-1}$  reduction in the cases of *L. casei*, *L. bulgaricus* and *L. lactis*, and 2 log reduction in the case of *L. acidophilus*. Significant, but less pronounced, decreases on the  $\text{OD}_{650}$  values were also detected in the presence of photoactivated and non-photoactivated NM101  $\text{TiO}_2$  (around 1 log  $\text{CFU.mL}^{-1}$  reduction in all the cases at the highest concentration). NM101 and NM105  $\text{TiO}_2$  had a significant difference on decreasing the number of probiotic cultures between them, being NM105  $\text{TiO}_2$  the one that had the strongest effect ( $P < 0.01$ ).

## 4. Discussion

This study was aimed to evaluate the antibacterial effect of photoactivated and non-photoactivated  $\text{TiO}_2$  nanoparticles on a range of Gram-negative (*S. enterica* var. Enteridis and *E. coli*) and Gram-positive (*S. aureus*, *B. cereus*, *L. casei*, *L. bulgaricus*, *L. acidophilus* and *L. lactis*) bacteria.

Preliminary tests for the correlation between  $\text{OD}_{650}$  values and  $\text{CFU.mL}^{-1}$  (data not shown) were performed in nutrient broth and MRS, the growth media used for the cultivation of the pathogens and probiotic bacteria, respectively. This was performed to set up a correlation curve as a basis to



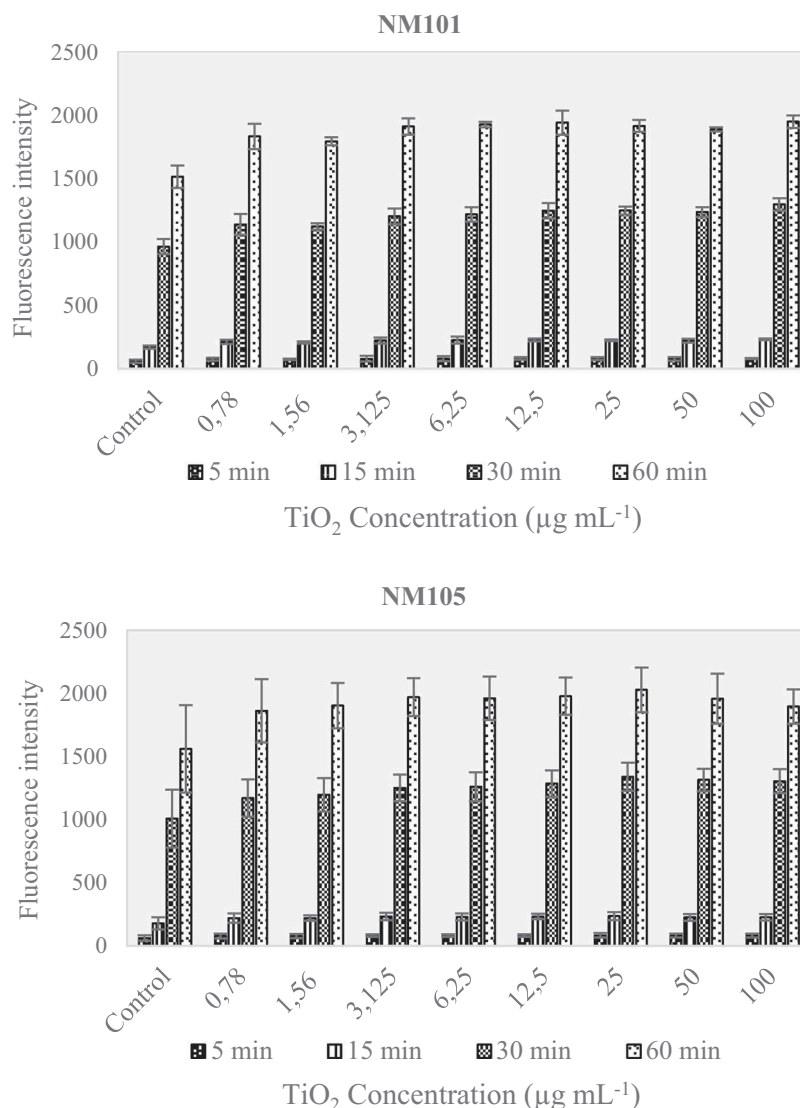
**Figure 2.** Effect of UV light at different exposure times on the growth curve of *S. aureus*. The graphs represent: (a) 3 min; (b) 6 min; (c) 12 min; (d) 24 min; (e) 36 min; and (f) 48 min of UV exposure. An every hour measurement was taken until 7 h of incubation and after that, another reading at 24 h of incubation was measured. Each value corresponds to the mean of 3 replicates performed on three separated days ( $n = 9$ ). The error bars represent standard deviation.

**Figura 2.** Efecto de la luz UV, a distintos tiempos de exposición, sobre la curva de crecimiento de *S. aureus*. Los gráficos representan: (A) 3 minutos; (B) 6 minutos; (C) 12 minutos; (D) 24 minutos; (E) 36 minutos; y (F) 48 minutos de exposición a luz UV. Se tomó una medición cada hora hasta las 7 horas de incubación y, posteriormente, se realizó otra lectura a las 24 horas de incubación. Cada valor corresponde a la media de 3 repeticiones realizadas en tres días separados ( $n = 9$ ). Las barras de error representan la desviación estándar.

compare the OD values from the experiments with estimates of the numbers of bacteria that were represented by these OD values, as other studies did (Kim, 2012; Ripolles-Avila, Ríos-Castillo, Guerrero-Navarro, & Rodríguez-Jerez, 2018; Yemmireddy & Hung, 2015). Similarly, we described a high correlation was observed between OD<sub>650</sub> values and CFU. mL<sup>-1</sup> for the different bacteria employed for the study.

Investigation of the effect of UV light on the different microorganisms used in this study was performed in the second instance to identify whether the exposure time of UV light on the bacteria had an impact on their growth. By establishing a UV light exposure time for the activation of the NPs that did not affect the growth of none of the bacteria, the bactericidal effects were able to be attributed to the effects of TiO<sub>2</sub> photocatalytic activation. Under the tested conditions, *S. aureus* was shown to be the most sensitive bacteria. When the time of UV light exposure

increased, the growth curve of *S. aureus* was affected. UV light can cause DNA damage (Hijnen, Beerendonk, & Medema, 2006). So the use of UV irradiation alone can have an impact on the growth of the bacteria but is incapable of bacterial disinfection; only strong UV irradiation (UV-B and UV-C) can have a disinfecting effect on bacteria (Pham & Lee, 2015; Yu, Lee, Lin, & Huang, 2008). The UV irradiation employed in the study was between the range 315–400 nm, which has also been used by other authors (Adhikari, Thind, Chen, Schraft, & Chen, 2018). The microorganism that we observed that had greater resistance to UV light was *S. enterica* var. Enteridis, which can be due to the range of UV light used was considered to be UV-A, and as it is not strong it has not had a bactericidal effect against this pathogen. Nevertheless, the TiO<sub>2</sub> NPs are photoactive and produce reactive oxygen species under natural sunlight (UV-A and UV-B) (Jovanović, 2015b).



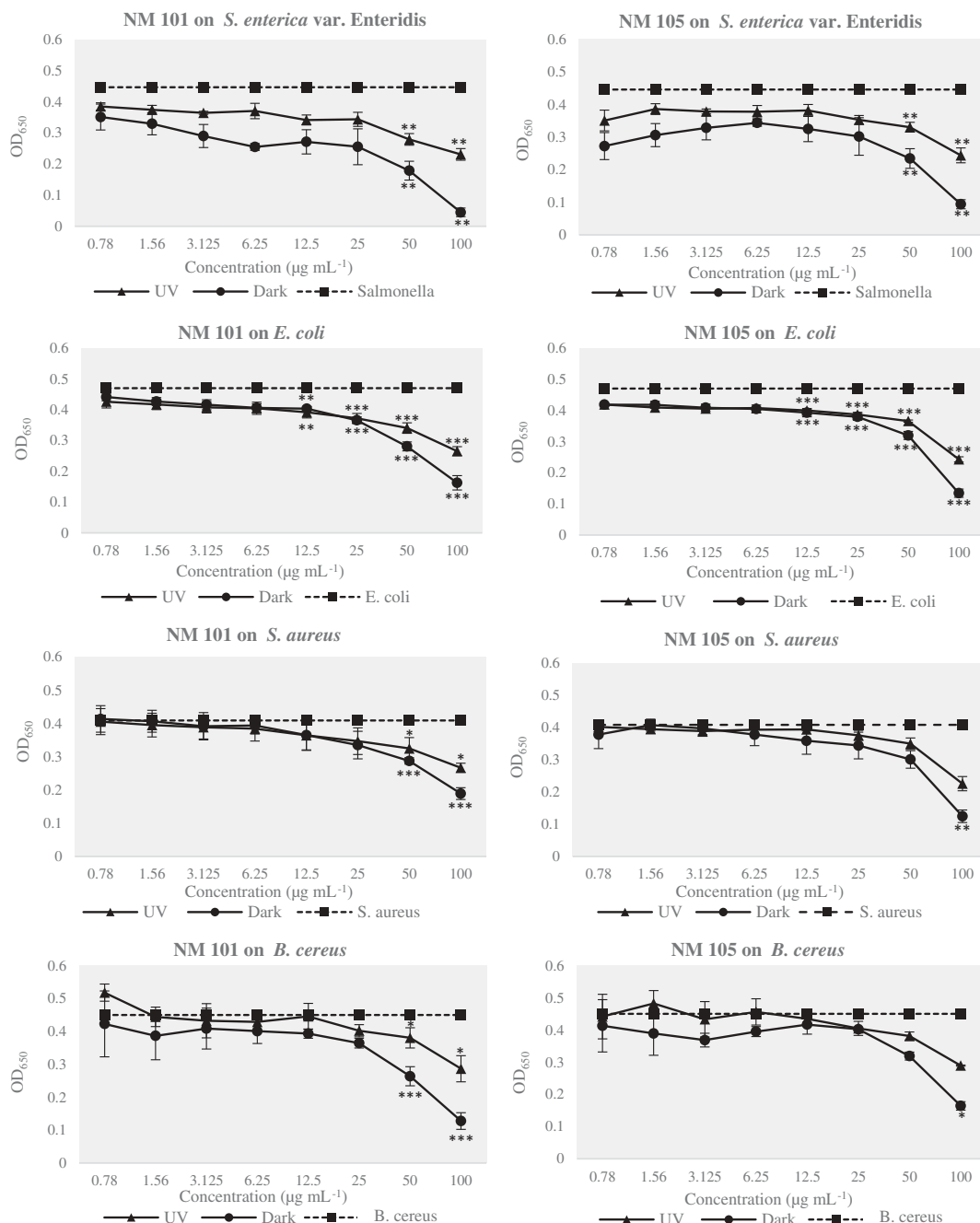
**Figure 3.** DCF fluorescence intensity produced by  $\text{TiO}_2$  NPs at different concentrations ranging from 0.78 to 100  $\mu\text{g mL}^{-1}$ . The UV exposure time was 6 min, but the results are representative for 12 and 24 min of exposure. The control sample did not contain  $\text{TiO}_2$ . Each value corresponds to the mean of 3 replicates performed on three separated days ( $n = 9$ ). The error bars represent standard deviation.

**Figura 3.** Intensidad de fluorescencia de DCF producida por nanopartículas de  $\text{TiO}_2$  a diferentes concentraciones que varían de 0.78 a 100  $\mu\text{g mL}^{-1}$ . El tiempo de exposición a luz UV fue de 6 minutos, pero los resultados son representativos de 12 y 24 minutos de exposición. La muestra de control no contenía  $\text{TiO}_2$ . Cada valor corresponde a la media de 3 repeticiones realizadas en tres días separados ( $n = 9$ ). Las barras de error representan la desviación estándar.

The ability of nano- $\text{TiO}_2$  to be photoactivated in the bacterial cultures was investigated using a DCFH-DA fluorescent emission assay. This fluorometric probe is primarily utilized for oxidative stress measurements. Our assay was based on the ability of the photoactivated  $\text{TiO}_2$  NPs to create ROS on the surface of the NPs. Aranda et al. (2013) and Griffiths et al. (2011) have already demonstrated the interference of different types of NPs with DCFH-DA fluorescence emission. A modification of the DCFH-DA test from the original protocol was done in order to adapt the assay to the experimental conditions, for example when the NPs were exposed to UV light the media used in the assay was culture medium rather than water or PBS. Although increasing the period of UV light exposure increased the fluorescence emission observed in the assay, non-significant enhancement with increasing concentrations of either  $\text{TiO}_2$  NPs was observed. Therefore, as the fluorescence of the control was not different to the fluorescence produced by the different concentration of NPs, the increased fluorescence observed

in the assay might be associated to NPs-culture media interactions rather than ROS production. For example, the enhancement of fluorescence emission could be caused due to agglomerate formation of the particles in the cell culture medium. Some studies have had similar problems (Aranda et al., 2013; Qi et al., 2012; Sabatini, Pereira, & Gehlen, 2007). Consequently, this photoactivation test using DCFH-DA as a fluorometric probe did not provide any evidence of  $\text{TiO}_2$  being photoactivated. Although some studies have not used any test to prove photoactivation of  $\text{TiO}_2$  (e.g. Long et al., 2014; Yemmireddy & Hung, 2015), some others have shown that a degradation test on an organic dye, rhodamine B, was very efficient to show if the NP was being photoactivated by UV light (e.g. Koizumi & Taya, 2002; Li et al., 2008; Polo et al., 2011).

$\text{TiO}_2$  NPs are a highly effective disinfectant once they are activated by UV light (Kermanizadeh et al., 2013; Kim et al., 2003). In water and air systems,  $\text{TiO}_2$  absorbs photons, resulting in the formation of active hydroxyl ( $\cdot\text{OH}$ ) and other ROS ( $\text{O}_2^-$ ,  $\text{H}_2\text{O}_2$ , among others) in the presence of  $\text{O}_2$  and  $\text{H}_2$

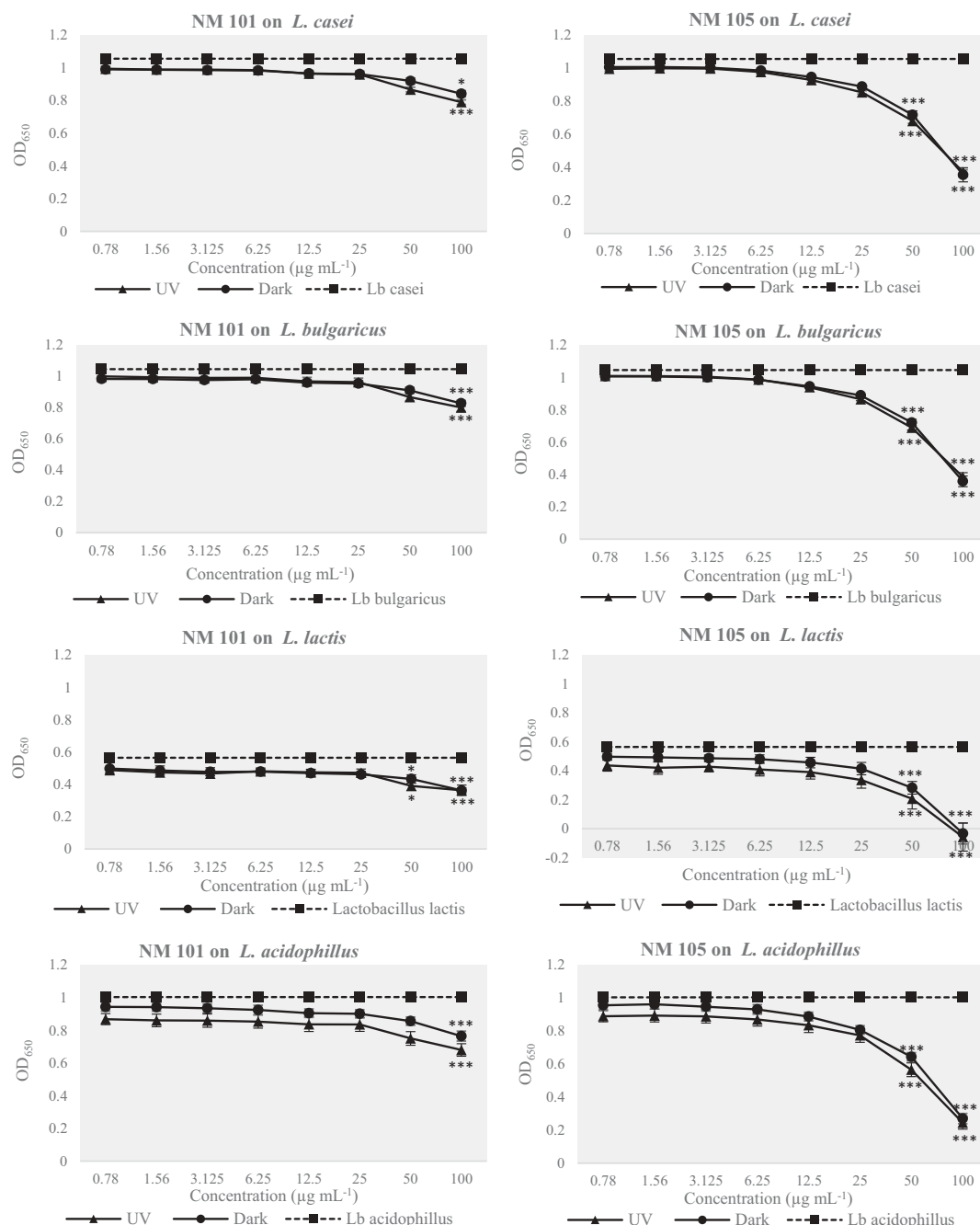


**Figure 4.** OD<sub>650</sub> values obtained in the tests of *S. enterica* var. Enteridis, *E. coli*, *S. aureus* and *B. cereus* with photocatalytic TiO<sub>2</sub> NPs in nutrient broth. In the graphs, y-axis shows the OD<sub>650</sub> values for each test condition and x-axis shows the different concentrations ranging from 0.78 to 100 µg mL<sup>-1</sup> of TiO<sub>2</sub> tested (NM101 and NM105). Samples were subjected to UV activation (UV) or non-UV activation (Dark). Each value corresponds to the mean of three replicates (n = 9). The error bars represent standard error of the mean. The graphs provide the P-values obtained by ANOVA analysis. The statistical analysis was performed separately for each experimental condition. Significance indicated by: \* = P < 0.05, \*\* = P < 0.01, and \*\*\* = P < 0.001.

**Figura 4.** Valores OD<sub>650</sub> obtenidos para *S. enterica* var. Enteridis, *E. coli*, *S. aureus* y *B. cereus* después de aplicar un tratamiento con nanopartículas de TiO<sub>2</sub> en caldo nutritivo. En los gráficos, el eje y muestra los valores de OD<sub>650</sub> para cada condición y el eje x muestra las diferentes concentraciones que van desde 0.78 a 100 µg mL<sup>-1</sup> de TiO<sub>2</sub> (NM101 y NM105). Las muestras se sometieron a activación por UV (UV) o no se activaron (Oscuridad). Cada valor corresponde a la media de tres repeticiones realizadas en tres días separados (n = 9). Las barras de error representan el error estándar de la media. Los gráficos proporcionan los valores de P obtenidos por análisis ANOVA. El análisis estadístico se realizó por separado para cada condición experimental. Significancia indicada por: \* = P < 0.05, \*\* = P < 0.01, y \*\*\* = P < 0.001.

O. Those ROS produced, particularly H<sub>2</sub>O<sub>2</sub>, cannot only transverse the cell membranes, but also produce oxidative active hydroxyl through a reaction inside bacterial cells that finally leads to the disinfection (Fujishima, Rao, & Tryk, 2000; Kikuchi, Sunada, Iyoda, Hashimoto, & Fujishima, 1997). There is overwhelming evidence that the mechanism of killing bacteria is due to membrane and cell wall damage (Foster et al., 2011). Some studies have demonstrated that once TiO<sub>2</sub> NPs are mixed with the bacterial cultures, TiO<sub>2</sub> is attached to the bacterial cells and damages the cell wall

under UV light (Dalrymple, Stefanakos, Trotz, & Goswami, 2010; Demidova & Hamblin, 2005; Kubacka, Ferrer, & Fernández-García, 2012). The damaged cell wall allows TiO<sub>2</sub> NPs to enter into the cells and releases the cell contents from the cells, resulting in cell death (Long et al., 2014). In this sense, we described TiO<sub>2</sub> NPs significantly reduced the populations of all tested Gram-positive and Gram-negative bacteria in the suspension solutions, being more effective NM105 than NM101. Although TiO<sub>2</sub> demonstrated antibacterial activity for each of the evaluated bacterial strains in



**Figure 5.** OD values obtained in the tests of *L. casei*, *L. bulgaricus*, *L. lactis* and *L. acidophilus* with photocatalytic TiO<sub>2</sub> NPs in MRS. In the graphs, y-axis shows the OD<sub>650</sub> values for each test condition and x-axis shows the different concentrations ranging from 0.78 to 100 µg mL<sup>-1</sup> of TiO<sub>2</sub> tested (NM101 and NM105). Samples were subjected to UV activation (UV) or non-UV activation (Dark). Each value corresponds to the mean of three replicates (n = 9). The error bars represent standard error of the mean. The graphs provide the P-values obtained by ANOVA analysis. The statistical analysis was performed separately for each experimental condition. Significance indicated by: \* = P < 0.05, \*\* = P < 0.01, and \*\*\* = P < 0.001.

**Figura 5.** Valores OD<sub>650</sub> obtenidos para *L. casei*, *L. bulgaricus*, *L. lactis* and *L. acidophilus* después de aplicar un tratamiento con nanopartículas de TiO<sub>2</sub> en caldo MRS. En los gráficos, el eje y muestra los valores de OD<sub>650</sub> para cada condición y el eje x muestra las diferentes concentraciones que van desde 0.78 a 100 µg mL<sup>-1</sup> de TiO<sub>2</sub> (NM101 y NM105). Las muestras se sometieron a activación por UV (UV) o no se activaron (Oscuridad). Cada valor corresponde a la media de tres repeticiones realizadas en tres días separados (n = 9). Las barras de error representan el error estándar de la media. Los gráficos proporcionan los valores de P obtenidos por análisis ANOVA. El análisis estadístico se realizó por separado para cada condición experimental. Significancia indicada por: \* = P < 0.05, \*\* = P < 0.01, y \*\*\* = P < 0.001.

a dose-dependent way, there were no significant differences ( $P > 0.05$ ) between the activity of the UV-exposed and non-UV-exposed TiO<sub>2</sub>. After 24 h of incubation, the biocidal effect of TiO<sub>2</sub> NPs was evident in the microorganisms tested. These results are in agreement with studies reported by Kim et al. (2003), Rincón and Pulgarin (2003), Rizzo (2009) and Swetha, Santhosh, and Geetha Balakrishna (2010), where, in similar conditions, *S. aureus*, *E. coli*, *Salmonella* spp., *Clostridium perfringens* spores, *Pseudomonas stutzeri*, *Pseudomonas aeruginosa*, *Listeria monocytogenes*, *Serratia marcescens*, *Vibrio*

*parahaemolyticus*, *Actinomyces viscosus*, *L. acidophilus*, *Streptococcus* spp., and coliform bacteria were killed by photoactivated TiO<sub>2</sub> NPs.

However, according to our results, UV-exposed and non-UV-exposed TiO<sub>2</sub> did not show significant differences in their bactericidal activity. As an inhibition of the bacteria was produced with the presence of TiO<sub>2</sub> NPs in dark conditions, this suggests that the NPs were effective and showing a dose-dependent response even when TiO<sub>2</sub> was not exposed to UV-A light. The decrease in CFU.mL<sup>-1</sup> associated

with the non-photoactivated TiO<sub>2</sub> has been shown in other studies (Lalucat, Bennasar, Bosch, García-Valdés, & Palleroni, 2006; Polo et al., 2011; Rincón & Pulgarin, 2004) where they suggested that the reduction of the bacterial cells can be put down to a fall in pH as a result of the addition of the photocatalyst. This fall in pH would be associated to both conditions (UV activated TiO<sub>2</sub> and non-UV activated TiO<sub>2</sub>) but with the difference that UV-exposed TiO<sub>2</sub> would have both inhibitory effects, the reduction of the pH and the photocatalytic activity. Nevertheless, this finding cannot be justified by the drop in the pH because of UV-exposed and non-UV-exposed TiO<sub>2</sub> did not show significant differences in their bactericidal activity. A possible explanation might be that TiO<sub>2</sub> nanoparticles are sensitive to photo-oxidation and when oxidized they lose their photocatalytic activity (Hu, Song, Jiang, & Wei, 2015; Lin et al., 2014). The fact that the TiO<sub>2</sub> NPs had been in use during some time suggests that light could have damaged the NPs. However, the results obtained demonstrated that although NPs were slightly activated by UV light, they presented an important role as antimicrobial for both Gram-positive and Gram-negative bacteria used in the study, as growth curves were compromised revealing a bacteriostatic effect, which is something interesting to continue investigating. In posterior studies, it would be interesting to include microbial viability tests to observe the real impact of TiO<sub>2</sub> NPs as potential disinfectants. On this sense, the applications of international standards such as UNE-EN 1040 or UNE-EN 1276 could be relevant. Furthermore, it is also interesting to evaluate the effectivity of these nanoparticles once they are embedded on surfaces in order to know the impact on biofilms, employing *in vitro* models to form mature structures and have results similar to real conditions (Ripolles-Avila, Hascoët, Guerrero-Navarro, & Rodríguez-Jerez, 2018).

Additionally, NM105 TiO<sub>2</sub> was more effective than NM101TiO<sub>2</sub>. Similar findings have been reported by Miyagi et al. (2004) who stated that anatase and rutile mixtures have a greater photocatalytic effect than 100% anatase. The increased activity of NM105 TiO<sub>2</sub> is commonly associated to interactions between the two forms, reducing the recombination between electrons and holes (Foster et al., 2011). In this regard, Yemmireddy and Hung (2015) reported that among the three TiO<sub>2</sub> tested (NM105 and two purified anatase crystal phase with different size measurements), NM105 was found to be the most efficient photocatalyst.

Furthermore, no significant differences were observed between Gram-negative (*Salmonella* spp. and *E. coli*) and Gram-positive (*S. aureus*, *B. cereus* and the probiotic bacteria) in their resistance to nano-TiO<sub>2</sub> treatment. Some studies have suggested that Gram-negative bacteria are more sensitive to photocatalytic disinfection than Gram-positive bacteria (Backhaus, Marugán, Van Grieken, & Sordo, 2010; B. Kim et al., 2003; Pal, Pehkonen, Yu, & Ray, 2007). This difference on sensitivity is normally associated to the differences in the cell surface structure such as cell wall and cell membrane, between Gram-negative and Gram-positive bacteria. However, several researchers have also stated that they found no difference within the bactericidal effect of photoactivated TiO<sub>2</sub> on Gram-negative and Gram-positive bacteria. For example, Wong et al. (2006) showed that the sensitivity of some human pathogens (*S. aureus*, *V. parahaemolyticus* and *L. monocytogenes*) to TiO<sub>2</sub> was not affected whether the target was Gram-negative or Gram-positive bacteria. Another investigation reported

analogous survival ratios when testing different bacteria (*E. coli*, *S. aureus* and *P. aeruginosa*) with a TiO<sub>2</sub> coated implant (Tsuang et al., 2008).

Regarding the impact of these NPs to human health, it is considered that toxicity and exposure data on TiO<sub>2</sub> NPs is not well defined until now in emerging nanomaterials, which makes the risk assessment more challenging (Tsang et al., 2017). However, TiO<sub>2</sub> is a food additive approved by different agencies such as the FDA and can be used in foods up to a concentration of 1% (Food and Drug Agency [FDA], 2015). In this case, the development of integrative technologies to respond to the food industry needs for the cleaning and disinfection protocols would include concentrations of these NPs lower than this established percentage for food matrices.

Considering the results obtained, the potential field of application of TiO<sub>2</sub> NPs can be related to the elimination of pathogenic bacteria that cause food poisoning or bacteria that cause food spoilage such as *Lactobacilli* spp. in meat and brewing industries. For example, Matsunaga (1985) demonstrate that *Lactobacillus* spp. was very sensitive to a Pt-doped TiO<sub>2</sub> catalyst and this could be exploited as an alternative to eliminate these microorganisms. In this time, Joost et al., (2015) suggested that nanosized TiO<sub>2</sub> based novel surface coating materials might be very effective for the elimination of pathogenic bacteria such as *E. coli*. In addition, (Long et al., 2014) suggested that TiO<sub>2</sub> NPs may be used as an effective antimicrobial treatment to improve food safety of meat products. Recently, UV-TiO<sub>2</sub> photocatalysis has been studied as an alternative for disinfection of fruits with a delicate texture, demonstrating highly effectivity against *E. coli* K12 on blueberry fruit surface (Lee et al., 2018).

## 5. Conclusions

All together, our results obtained in this study indicate that both UV-exposed and non-UV-exposed TiO<sub>2</sub> NPs (both NM101 and NM105) show, in a dose-dependent manner, an antibacterial activity against a range of suspended Gram-negative (*S. enterica* var. Enteridis, *E. coli*) and Gram-positive bacteria (*S. aureus*, *B. cereus*, *L. casei*, *L. bulgaricus*, *L. acidophilus* and *L. lactis*) as growth curves are compromised. The data demonstrate that the optimal content of TiO<sub>2</sub> NPs in a suspension for the disinfection of the bacterial cultures was 100 µg mL<sup>-1</sup> which was able to reduce an estimated amount of 2–3 log the bacterial population after a 24 h incubation. We suggest that further research into the comparative effects between TiO<sub>2</sub> embedded in surfaces and suspended TiO<sub>2</sub> need investigating. Additionally, the toxicological influence of TiO<sub>2</sub> NPs should be evaluated to determine the consequences of using these NPs in food safety. On the foundation of the results acquired and the literature survey, it can be concluded that the technology has the potential to provide, particularly in view of the development of visible light-activated catalysts, a forceful weapon to combat both, the transmission of food-borne diseases and the spoilage of food originated in, for example, meat and brewing industries by *Lactobacillus* spp.

## Disclosure statement

No potential conflict of interest was reported by the authors.

## ORCID

Carolina Ripolles-Avila  <http://orcid.org/0000-0003-0032-8893>

José Juan Rodríguez-Jerez  <http://orcid.org/0000-0001-7798-8529>

## References

- Adhikari, B. R., Thind, S. S., Chen, S., Schraft, H., & Chen, A. (2018). Efficient bacterial disinfection based on an integrated nanoporous titanium dioxide and ruthenium oxide bifunctional approach. *Journal of Hazardous Materials*, 356(May), 73–81.
- Alhaji, M. H., Sanaullah, K., Khan, A., Hamza, A., Muhammad, A., Ishola, M. S., ... Bhawani, S. A. (2017). Recent developments in immobilizing titanium dioxide on supports for degradation of organic pollutants in wastewater – A review. *International Journal of Environmental Science and Technology*, 14(9), 2039–2052.
- Aranda, A., Sequedo, L., Tolosa, L., Quintas, G., Burello, E., Castell, J. V., & Gombau, L. (2013). Dichloro-dihydro-fluorescein diacetate (DCFH-DA) assay: A quantitative method for oxidative stress assessment of nanoparticle-treated cells. *Toxicology in Vitro*, 27(2), 954–963.
- Backhaus, K., Marugán, J., Van Grieken, R., & Sordo, C. (2010). Photocatalytic inactivation of *E. faecalis* in secondary wastewater plant effluents. *Water Science and Technology*, 61(9), 2355–2361.
- Banerjee, J., Muraleedharan, P., Tyagi, A. K., & Raj, B. (2006). Physics and chemistry of photocatalytic titanium dioxide: Visualization of bactericidal activity using atomic force microscopy. *Current Science*, 90(10), 1378–1383.
- Barthomeuf, M., Raymond, P., Policarpo, N., Castel, X., Le Gendre, L., Denis, M., & Pissavin, C. (2017). Bactericidal efficiency of UVA-active titanium dioxide thin layers on bacteria from food industry environments. *Materials Technology*, 32(13), 782–791.
- Bondarenko, O., Juganson, K., Ivask, A., Kasemets, K., Mortimer, M., & Kahru, A. (2013). Toxicity of Ag, CuO and ZnO nanoparticles to selected environmentally relevant test organisms and mammalian cells *in vitro*: A critical review. *Archives of Toxicology*, 87(7), 1181–1200.
- Caballero, L., Whitehead, K. A., Allen, N. S., & Verran, J. (2009). Inactivation of *Escherichia coli* on immobilized TiO<sub>2</sub> using fluorescent light. *Journal of Photochemistry and Photobiology A: Chemistry*, 202 (2–3), 92–98.
- Carp, O., Huisman, C. L., & Reller, A. (2004). Photoinduced reactivity of titanium dioxide. *Progress in Solid State Chemistry*, 32(1–2), 33–177.
- Cheng, C.-L., Sun, D.-S., Chu, W.-C., Tseng, Y.-H., Ho, H.-C., Wang, J.-B., ... Chang, H.-H. (2009). The effects of the bacterial interaction with visible-light responsive titania photocatalyst on the bactericidal performance. *Journal of Biomedical Science*, 16, 7.
- Dalrymple, O. K., Stefanakos, E., Trotz, M. A., & Goswami, D. Y. (2010). A review of the mechanisms and modeling of photocatalytic disinfection. *Applied Catalysis B: Environmental*, 98(1–2), 27–38.
- Demidova, T. N., & Hamblin, M. R. (2005). Effect of cell-photosensitizer binding and cell density on microbial photoinactivation effect of cell-photosensitizer binding and cell density on microbial photoinactivation. *Antimicrobial Agents and Chemotherapy*, 49(6), 2329–2335.
- Duffy, L. L., Osmond-McLeod, M. J., Judy, J., & King, T. (2018). Investigation into the antibacterial activity of silver, zinc oxide and copper oxide nanoparticles against poultry-relevant isolates of *Salmonella* and *Campylobacter*. *Food Control*, 92, 293–300.
- FDA (Food and Drug Agency). (2015). *Summary of color additives for use in the United States in foods, drugs, cosmetics, and medical devices* [Online]. <https://www.fda.gov/forindustry/coloradditives/coloradditiveinventories/ucm115641.htm>
- Foster, H. A., Ditta, I. B., Varghese, S., & Steele, A. (2011). Photocatalytic disinfection using titanium dioxide: Spectrum and mechanism of antimicrobial activity. *Applied Microbiology and Biotechnology*, 90(6), 1847–1868.
- Fujishima, A., Rao, T. N., & Tryk, D. A. (2000). TiO<sub>2</sub> photocatalysts and diamond electrodes. *Electrochimica Acta*, 45(28), 4683–4690.
- González-Rivas, F., Ripolles-Avila, C., Fontecha-Umaña, F., Ríos-Castillo, A. G., & Rodríguez-Jerez, J. J. (2018). Biofilms in the spotlight: Detection, quantification, and removal methods. *Comprehensive Reviews in Food Science and Food Safety*, 17(5), 1261–1276.
- Griffiths, S. M., Singh, N., Jenkins, G. J. S., Williams, P. M., Orbaek, A. W., Barron, A. R., ... Doak, S. H. (2011). Dextran coated ultrafine super-paramagnetic iron oxide nanoparticles: Compatibility with common fluorometric and colorimetric dyes. *Analytical Chemistry*, 83(10), 3778–3785.
- Hijnen, W. A. M., Beerendonk, E. F., & Medema, G. J. (2006). Inactivation credit of UV radiation for viruses, bacteria and protozoan (oo)cysts in water: A review. *Water Research*, 40(1), 3–22.
- Horie, Y., David, D. A., Taya, M., & Tone, S. (1996). Effects of light intensity and titanium dioxide concentration on photocatalytic sterilization rates of microbial cells. *Industrial and Engineering Chemistry Research*, 35(11), 3920–3926.
- Hu, Y., Song, X., Jiang, S., & Wei, C. (2015). Enhanced photocatalytic activity of Pt-doped TiO<sub>2</sub> for NO<sub>x</sub> oxidation both under UV and visible light irradiation: A synergistic effect of lattice Pt<sup>4+</sup> and surface PtO. *Chemical Engineering Journal*, 274(x), 102–112.
- Jalvo, B., Faraldos, M., Bahamonde, A., & Rosal, R. (2017). Antimicrobial and antibiofilm efficacy of self-cleaning surfaces functionalized by TiO<sub>2</sub> photocatalytic nanoparticles against *Staphylococcus aureus* and *Pseudomonas putida*. *Journal of Hazardous Materials*, 340, 160–170.
- Joost, U., Juganson, K., Visnapuu, M., Mortimer, M., Kahru, A., Nömmiste, E., Joost, U., Kisand, V., & Ivask, A. (2015). Photocatalytic antibacterial activity of nano-tio2 (anatase)-based thin films: effects on escherichia coli cells and fatty acids. *Journal Of Photochemistry and Photobiology B*, 142, 178–185.
- Jovanović, B. (2015a). Critical review of public health regulations of titanium dioxide, a human food additive. *Integrated Environmental Assessment and Management*, 11(1), 10–20.
- Jovanović, B. (2015b). Review of titanium dioxide nanoparticle phototoxicity: Developing a phototoxicity ratio to correct the endpoint values of toxicity tests. *Environmental Toxicology and Chemistry*, 34 (5), 1070–1077.
- Kahru, A., & Dubourguier, H. C. (2010). From ecotoxicology to nanoeotoxicology. *Toxicology*, 269(2–3), 105–119.
- Kermanizadeh, A., Pojana, G., Gaiser, B. K., Birkedal, R., Bilančová, D., Wallin, H., ... Stone, V. (2013). *In vitro* assessment of engineered nanomaterials using a hepatocyte cell line: Cytotoxicity, pro-inflammatory cytokines and functional markers. *Nanotoxicology*, 7(3), 301–313.
- Kikuchi, Y., Sunada, K., Iyoda, T., Hashimoto, K., & Fujishima, A. (1997). Photocatalytic bactericidal effect of TiO<sub>2</sub> thin films: Dynamic view of the active oxygen species responsible for the effect. *Journal of Photochemistry and Photobiology A: Chemistry*, 106(1–3), 51–56.
- Kim, B., Kim, D., Cho, D., & Cho, S. (2003). Bactericidal effect of TiO<sub>2</sub> photocatalyst on selected food-borne pathogenic bacteria. *Chemosphere*, 52(1), 277–281.
- Kim, D.-J. (2012). Relation of microbial biomass to counting units for *Pseudomonas aeruginosa*. *African Journal of Microbiology Research*, 6 (21), 4620–4622.
- Koizumi, Y., & Taya, M. (2002). Kinetic evaluation of biocidal activity of titanium dioxide against phage MS2 considering interaction between the phage and photocatalyst particles. *Biochemical Engineering Journal*, 12(2), 107–116.
- Kubacka, A., Ferrer, M., & Fernández-García, M. (2012). Kinetics of photocatalytic disinfection in TiO<sub>2</sub>-containing polymer thin films: UV and visible light performances. *Applied Catalysis B: Environmental*, 121–122, 230–238.
- Lalucat, J., Bennasar, A., Bosch, R., García-Valdés, E., & Palleroni, N. J. (2006). Biology of *Pseudomonas stutzeri*. *Microbiology and Molecular Biology Reviews : MMBR*, 70(2), 510–547.
- Landmann, M., Rauls, E., & Schmidt, W. G. (2012). The electronic structure and optical response of rutile, anatase and brookite TiO<sub>2</sub>. *Journal of Physics Condensed Matter*, 24(19), 1–6.
- Lee, M., Shahbaz, H. M., Kim, J. U., Lee, H., Lee, D. U., & Park, J. (2018). Efficacy of UV-TiO<sub>2</sub> photocatalysis technology for inactivation of *Escherichia coli* K12 on the surface of blueberries and a model agar matrix and the influence of surface characteristics. *Food Microbiology*, 76(July), 526–532.
- Li, Q., Mahendra, S., Lyon, D. Y., Brunet, L., Liga, M. V., Li, D., & Alvarez, P. J. J. (2008). Antimicrobial nanomaterials for water disinfection and microbial control: Potential applications and implications. *Water Research*, 42(18), 4591–4602.
- Lin, Y.-T., Weng, C.-H., Hsu, H.-J., Huang, J.-W., Srivastav, A. L., & Shiesh, C.-C. (2014). Effect of oxygen, moisture, and temperature on the photo oxidation of ethylene on N-doped TiO<sub>2</sub> catalyst. *Separation and Purification Technology*, 134, 117–125.
- Long, M., Wang, J., Zhuang, H., Zhang, Y., Wu, H., & Zhang, J. (2014). Performance and mechanism of standard nano-TiO<sub>2</sub> (P-25) in

- photocatalytic disinfection of foodborne microorganisms – *Salmonella* Typhimurium and *Listeria monocytogenes*. *Food Control*, 39(1), 68–74.
- Matsunaga, T. (1985). Sterilization with particulate photosemiconductor. *Journal of Antibacterial Antifungal Agents*, 13, 211–220.
- Miyagi, T., Kamei, M., Mitsuhashi, T., Ishigaki, T., & Yamazaki, A. (2004). Charge separation at the rutile/anatase interface: A dominant factor of photocatalytic activity. *Chemical Physics Letters*, 390(4–6), 399–402.
- Oka, Y., Kim, W. C., Yoshida, T., Hirashima, T., Mouri, H., Urade, H., ... Kubo, T. (2008). Efficacy of titanium dioxide photocatalyst for inhibition of bacterial colonization on percutaneous implants. *Journal of Biomedical Materials Research. Part B, Applied Biomaterials*, 86, 530–540.
- Özkaleli, M., & Erdem, A. (2017). Factors promoting *Staphylococcus aureus* disinfection by TiO<sub>2</sub>, SiO<sub>2</sub> and AG nanoparticles. *The Online Journal of Science and Technology*, 7(2), 51–55.
- Pal, A., Pehkonen, S. O., Yu, L. E., & Ray, M. B. (2007). Photocatalytic inactivation of Gram-positive and Gram-negative bacteria using fluorescent light. *Journal of Photochemistry and Photobiology A: Chemistry*, 186(2–3), 335–341.
- Pham, T.-D., & Lee, B.-K. (2015). Disinfection of *Staphylococcus aureus* in indoor aerosols using Cu–TiO<sub>2</sub> deposited on glass fiber under visible light irradiation. *Journal of Photochemistry and Photobiology A: Chemistry*, 307–308, 16–22.
- Polo, A., Diamanti, M. V., Bjarnsholt, T., Høiby, N., Villa, F., Pedferri, M. P., & Cappitelli, F. (2011). Effects of photoactivated titanium dioxide nanopowders and coating on planktonic and biofilm growth of *Pseudomonas aeruginosa*. *Photochemistry and Photobiology*, 87(6), 1387–1394.
- Qi, X., Li, H., Lam, J. W. Y., Yuan, X., Wei, J., Tang, B. Z., & Zhang, H. (2012). Graphene oxide as a novel nanopatform for enhancement of aggregation-induced emission of silole fluorophores. *Advanced Materials*, 24(30), 4191–4195.
- Rincón, A. G., & Pulgarin, C. (2003). Photocatalytic inactivation of *E. coli*: Effect of (continuous-intermittent) light intensity and of (suspended-fixed) TiO<sub>2</sub> concentration. *Applied Catalysis B: Environmental*, 44(3), 263–284.
- Rincón, A. G., & Pulgarin, C. (2004). Effect of pH, inorganic ions, organic matter and H<sub>2</sub>O<sub>2</sub> on *E. coli* K12 photocatalytic inactivation by TiO<sub>2</sub>: Implications in solar water disinfection. *Applied Catalysis B: Environmental*, 51(4), 283–302.
- Ripolles-Avila, C., Hascoët, A. S., Guerrero-Navarro, A. E., & Rodríguez-Jerez, J. J. (2018). Establishment of incubation conditions to optimize the *in vitro* formation of mature *Listeria monocytogenes* biofilms on food-control surfaces. *Food Control*. doi:10.1016/j.jclepro.2018.01.232
- Ripolles-Avila, C., Ríos-Castillo, A. G., Guerrero-Navarro, A. E., & Rodríguez-Jerez, J. J. (2018). Reinterpretation of a classic method for the quantification of cell density within biofilms of *Listeria monocytogenes*. *Journal of Microbiology & Experimentation*, 6(2), 70–75.
- Ripolles-Avila, C., Ríos-Castillo, A. G., & Rodríguez-Jerez, J. J. (2018). Development of a peroxide biodeceptor for a direct detection of biofilms produced by catalase-positive bacteria on food-contact surfaces. *CyTA - Journal of Food*, 16(1), 506–515.
- Rizzo, L. (2009). Inactivation and injury of total coliform bacteria after primary disinfection of drinking water by TiO<sub>2</sub> photocatalysis. *Journal of Hazardous Materials*, 165(1–3), 48–51.
- Sabatini, C. A., Pereira, R. V., & Gehlen, M. H. (2007). Fluorescence modulation of acridine and coumarin dyes by silver nanoparticles. *Journal of Fluorescence*, 17(4), 377–382.
- Sirelkhatim, A., Mahmud, S., Seeni, A., Kaus, N. H. M., Ann, L. C., Bakhori, S. K. M., ... Mohamad, D. (2015). Review on Zinc oxide nanoparticles: Antibacterial activity and toxicity mechanism. *Nano-Micro Letters*, 7, 219–242.
- Sun, D. D., Tay, J. H., & Tan, K. M. (2003). Photocatalytic degradation of *E. coli* in water. *Water Research*, 37(14), 3452–3462.
- Sun, L., Li, Y., Liu, X., Jin, M., Zhang, L., Du, Z., Guo, C., Huang, P., & Sun, Z. (2011). Cytotoxicity and mitochondrial damage caused by silica nanoparticles. *Toxicology in Vitro*, 25, 1619–1629.
- Swetha, S., Santhosh, S. M., & Geetha Balakrishna, R. (2010). Synthesis and comparative study of nano-TiO<sub>2</sub> over degussa P-25 in disinfection of water. *Photochemistry and Photobiology*, 86(3), 628–632.
- Tsang, M. P., Hristozov, D., Zabeo, A., Koivisto, A. J., Østerskov Jensen, A. C., Alstrup Jensen, A., ... Sonnemann, G. (2017). Probabilistic risk assessment of emerging materials: Case study of titanium dioxide nanoparticles. *Nanotoxicology*, 11(4), 558–568.
- Tsuang, Y.-H., Sun, J.-S., Huang, Y.-C., Lu, C.-H., Chang, W. H.-S., & Wang, C.-C. (2008). Studies of photokilling of bacteria using titanium dioxide nanoparticles. *Artificial Organs*, 32(2), 167–174.
- Wang, S., Weller, D., Falardeau, J., Strawn, L. K., Mardones, F. O., Adell, A. D., & Moreno Switt, A. I. (2016). Food safety trends: From globalization of whole genome sequencing to application of new tools to prevent foodborne diseases. *Trends in Food Science and Technology*, 57, 188–198.
- WHO. (2015). *Who estimates of the global burden of foodborne diseases*. WHO Press.
- Wong, M. S., Chu, W. C., Sun, D. S., Huang, H. S., Chen, J. H., Tsai, P. J., ... Chang, H. H. (2006). Visible-light-induced bactericidal activity of a nitrogen-doped titanium photocatalyst against human pathogens. *Applied and Environmental Microbiology*, 72(9), 6111–6116.
- Yemmireddy, V. K., & Hung, Y. C. (2017). Using photocatalyst metal oxides as antimicrobial surface coatings to ensure food safety — Opportunities and challenges. *Comprehensive Reviews in Food Science and Food Safety*, 16(4), 617–631.
- Yemmireddy, V. K., & Hung, Y.-C. (2015). Selection of photocatalytic bactericidal titanium dioxide (TiO<sub>2</sub>) nanoparticles for food safety applications. *LWT - Food Science and Technology*, 61(1), 1–6.
- Yu, K. P., Lee, G. W. M., Lin, S. Y., & Huang, C. P. (2008). Removal of bioaerosols by the combination of a photocatalytic filter and negative air ions. *Journal of Aerosol Science*, 39(5), 377–392.
- Zhao, X., Zhao, F., Wang, J., & Zhong, N. (2017). Biofilm formation and control strategies of foodborne pathogens: Food safety perspectives. *RSC Advances*, 7(58), 36670–36683.