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1 *Short title: Using wastes to reduce jasmine plantation greenhouse-gas emissions*

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3 **EFFECTS OF STRAW COMBINED WITH INDUSTRIAL AND**  
4 **AGRICULTURAL WASTES ON GREENHOUSE-GAS EMISSIONS IN A**  
5 **SUBTROPICAL JASMINE PLANTATION**

6  
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18  
19 **SUMMARY**

20  
21 The effects of straw alone or combined with industrial and agricultural wastes as  
22 fertilizers on GHG emissions in cropland areas are still poorly known. Here, we studied  
23 the effects of 3.5 Mg ha<sup>-1</sup> straw and 3.5 Mg ha<sup>-1</sup> straw combined with 8 Mg ha<sup>-1</sup> of  
24 diverse wastes on GHG emission in a subtropical *J. sambac* plantation in southeastern  
25 China. There were five treatments in a completely randomized block design: control;  
26 straw only; straw + biochar; straw + steel slag; and straw + gypsum slag. Emissions of  
27 CO<sub>2</sub> were generally higher in the treatments with waste than in the control or straw-  
28 only treatments, whereas the contrary pattern was observed in CH<sub>4</sub> and N<sub>2</sub>O emission  
29 rates. Moreover, the total global-warming potentials (GWPs) were no significantly  
30 higher in most of the amended treatments as compared to the control and straw-only  
31 treatments. In relation to the treatment with only straw, GWPs was 9.4% lower when  
32 steel slag was used. This finding could be a consequence of Fe amount added by steel  
33 slag, which would limit and inhibit the emissions of GHGs and their transport from soil  
34 to atmosphere. Our results showed that the application of slags did not increase the  
35 emission of GHGs and the combination of straw with steel slag or biochar could be  
36 more effective than straw alone for controlling GHGs emission and improve soil C and  
37 nutrient provision.

38  
39 **Keywords:** CH<sub>4</sub> flux; CO<sub>2</sub> flux; N<sub>2</sub>O flux; climate change; straw; biochar; steel slag;  
40 gypsum slag; jasmine plantation

## INTRODUCTION

Global warming is caused by emission of large amounts of greenhouse gases (GHGs) to the atmosphere from the combustion of fossil fuels, land-use changes and other human activities (IPCC, 2013). Global GHG emissions from agricultural activities are about 5.1-6.1 Pg CO<sub>2</sub>-eq y<sup>-1</sup> (Smith *et al.*, 2007), which can contribute to approximately 20% of the current emissions (Hütsch, 2001). The emissions of carbon dioxide (CO<sub>2</sub>), methane (CH<sub>4</sub>) and nitrous oxide (N<sub>2</sub>O) from agricultural lands are especially relevant (Myhre *et al.*, 2013). Minimizing the emission of GHGs is an important task for mitigating their adverse impacts on climate change. Many methods have been developed to reduce and/or control the emission of GHGs, through water, fertilizers and tillage management (Peyron *et al.*, 2016; Gupta *et al.*, 2016; Malhi *et al.*, 2016).

Recycling industrial and agricultural waste is becoming an effective way to solve environmental and resource availability issues, with the application of industrial and agricultural waste on agricultural lands gradually increasing (Wang *et al.*, 2015a,b). The effects of waste application on soil nutrients and properties (Prendergast Miller *et al.*, 2014), plant growth (Wang *et al.*, 2015a, b) and GHG emission (Zhang *et al.*, 2012) have been evaluated. Waste materials such as biochar (Zhang *et al.*, 2010) and steel slag (Wang *et al.*, 2012a) have been widely studied for increasing crop yields and mitigating GHG emissions. However, the effect of mixed waste application on the emission of GHGs is more complex and has been rarely and inconclusively studied. Industrial and agricultural wastes contain high concentrations of electron acceptors such as the active and free-oxide forms of iron (Fe), sulfur, nitrogen (N) and phosphorus (P).

Recently, biochar application has been found to significantly increase soil CO<sub>2</sub> fluxes by 22%, while decreasing N<sub>2</sub>O fluxes by 31% without any effect in CH<sub>4</sub> fluxes (He *et al.*, 2017). Such effects were dependent on latitude, soil type and soil use (Fan *et al.*, 2017; He *et al.*, 2017; Wang *et al.*, 2017a). The use of steel and gypsum slag application to reduce GHGs emissions has been less studied, despite some positive effects (reduction in GHGs emissions) in rice fields (Susilawati *et al.*, 2015; Wang *et al.*, 2017b). Few studies have provided an overall evaluation of the total global-warming potential (GWP) of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O, the three main GHGs (Wang *et al.*, 2015a,b). The combined application of biochar and steel slag has been associated with decreases in GHG emissions by rice crops (Wang *et al.*, 2016; 2017).

At the best of our knowledge no information is available on the effects of industrial and agricultural wastes on GHG emissions from subtropical *Jasmine sambac* plantations. *J. sambac* tea is unique and more than half of the *J. sambac* tea in China is produced in Fuzhou Province (Yang *et al.*, 2008). Developing effective strategies to enhance the yield of *J. sambac* flowers without increasing GHG emissions is considered as an important policy for minimizing future adverse conditions induced by climate change. On the other hand, use of waste materials for the management of *J. sambac* plantations and GHG emission would be an alternative way to recycle waste without environmental risks.

Here, we evaluated the effects of various waste materials (straw, and straw

85 combined with steel slag, biochar or gypsum slag) as fertilizers on the GHG emissions  
86 under field conditions, taking into account soil traits and the relationships between CO<sub>2</sub>,  
87 CH<sub>4</sub> and N<sub>2</sub>O emissions and soil traits.

## 89 MATERIALS AND METHODS

### 91 *Study site and experimental design*

92 A field experiment was conducted in a *Jasminum sambac* (L.) field in Fujian Province,  
93 China (Suppl. material Figure S1, 25°59'10"N, 119°20'7"E, ~~XXX-220~~ m a.s.l), during *J. sambac* growing season from April to October 2015. This  
94 region has a subtropical monsoonal climate, with a mean air temperature of about 25  
95 °C during the study period and a mean annual precipitation of approximately 1400 mm.  
96 About 80% of the total rainfall was concentrated in the rainy season between May and  
97 October (Figure S1). The soil contained 25%, 59% and 16% sand, silt and clay,  
98 respectively. Bulk density was 1.2 g cm<sup>-3</sup>, pH 4.4, salinity of 0.15 mS cm<sup>-1</sup> and  
99 concentrations of total carbon, total N, total P and total potassium were 11.7, 1.1, 0.5  
100 and 13.3 g kg<sup>-1</sup>, respectively. The study site has been cultivated with *J. sambac* since  
101 2008.

102  
103 *J. sambac* was cultivated using a ridge and ditch system, with a ridge height of 20  
104 cm, ridge width of 100 cm and ditch width of 30 cm. Double-valve *J. sambac* branches  
105 (10 cm long) were transplanted by hand at a distance among them of 3 cm, along the  
106 ridges and 20 cm from the nearest plant in the side ridges in April 2008. ~~Double-valve *J.*~~  
107 ~~*sambac* branches (10 cm long) were transplanted by hand in the ridges at a spacing of~~  
108 ~~3×20 cm in April 2008.~~ The *J. sambac* was cut at the end of March or early April each  
109 year when the air temperature was about 20 °C. About 3.5 Mg ha<sup>-1</sup> of branches and  
110 leaves were returned to the field, which was ridged but not plowed each year. *J. sambac*  
111 branches and leaves began to grow from early April to early May. ~~Growth period last~~  
112 ~~from early May to the end of May.~~ Flowering occurred from early June to late September, when the final growth  
113 period began. A complete fertilizer (N:P<sub>2</sub>O<sub>5</sub>:K<sub>2</sub>O, 16:16:16) was applied in two  
114 applications: the first application was 130 kg ha<sup>-1</sup> one day after *J. sambac* was  
115 pruned ~~was applied in two splits: the first application was 130 kg ha<sup>-1</sup> one day after *J.*~~  
116 ~~*sambac* was pruned; and the second one was 100 kg ha<sup>-1</sup> one day after the first *J. sambac*~~  
117 ~~flowers were collected and the second one was 100 kg ha<sup>-1</sup> one day after the first *J.*~~  
118 ~~*sambac* flowers were collected.~~

119  
120 Triplicate plots (20 m<sup>2</sup> each) were established for five treatments in a completely  
121 randomized block design: 1) no straw or waste (control); 2) straw only (straw); 3) straw  
122 combined with steel slag (straw+steel slag); 4) straw combined with biochar  
123 (straw+biochar); and 5) straw combined with gypsum slag (straw+gypsum slag). Straw  
124 was added at 3.5 Mg ha<sup>-1</sup>, and the biochar and steel and gypsum slags were added at 8  
125 Mg ha<sup>-1</sup> as granules (2 mm in diameter). The steel slag was collected from the Jinxing  
126 Iron & Steel Co., Ltd in Fujian (China). The biochar was collected from the Qinfeng  
127 Straw Technology Co., Ltd in Jiangsu Province. The gypsum slag was collected from  
128 building waste (from the indoor decoration of buildings). The industrial and agricultural  
129 wastes used in this study were rich in Si, Ca and K, which are essential nutrients for  
130 plant growth (Prendergast - Miller et al., 2014) (Table 1). All treatments received the

131 ~~same water management~~The industrial and agricultural wastes used in this study were  
132 rich in Si, Ca and K, which are essential nutrients for plant growth (Prendergast – Miller  
133 et al., 2014) (Table 1). All treatments received the same water management.

#### 134 135 *Measurement of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O emissions*

136 The experimental period was from April 2015 to March 2016. Static closed chambers  
137 were used to measure CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O emissions, as described by Wang et al.  
138 (2015a). The chambers were made of rigid PVC and consisted of two parts, an upper  
139 transparent compartment (100 cm height, 30 cm width, 30 cm length) placed on a  
140 permanently installed bottom collar (10 cm height, 30 cm width, 30 cm length). Each  
141 chamber had two battery-operated fans to mix the air inside the chamber headspace, an  
142 internal thermometer to monitor temperature changes during gas sampling and a gas-  
143 sampling port with a neoprene rubber septum at the top of the chamber for collecting  
144 gas samples. Three replicate chambers in each treatment were used. A wooden  
145 boardwalk was built for accessing the plots to minimize disturbance of the soil during  
146 gas sampling. The chambers had a vent to avoid pressure buildup.

147 Gas flux was measured for all chambers twice weekly during the growing season  
148 and four times a week during the other seasons. The temperature in chamber was not  
149 significantly changed during sampling process (30 min). Gas samples were collected  
150 from the chamber headspace using a 100-mL plastic syringe with a three-way stopcock  
151 0, 15 and 30 min after chamber deployment. The samples were immediately transferred  
152 to 100-mL air-evacuated aluminum-foil bags (Delin Gas Packaging Co., Ltd., Dalian,  
153 China) sealed with butyl rubber septa and transported immediately to the laboratory for  
154 evaluation of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O concentrations.

155 CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O concentrations were determined by gas chromatography  
156 (Shimadzu GC-2010 and Shimadzu GC-2014, Kyoto, Japan) using a stainless steel  
157 Porapak Q column (2 m length, 4 mm OD, 80/100 mesh). A methane-conversion  
158 furnace, flame ionization detector (FID) and electron-capture detector (ECD) were used  
159 for determining CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O concentrations, respectively. The operating  
160 temperatures of the column, injector and detector for quantifying CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O  
161 were adjusted to 45, 100 and 280°C; to 70, 200 and 200°C and to 70, 200, and 320°C,  
162 respectively. Helium (99.999% purity) was used as a carrier gas (30 mL min<sup>-1</sup>), and a  
163 make-up gas (95% argon and 5% CH<sub>4</sub>) was used for the ECD. The gas chromatograph  
164 was calibrated before and after each set of measurements using 503, 1030 and 2980 μL  
165 CO<sub>2</sub> L<sup>-1</sup> in He; 1.01, 7.99 and 50.5 μL CH<sub>4</sub> L<sup>-1</sup> in He and 0.2, 0.6 and 1.0 μL N<sub>2</sub>O L<sup>-1</sup>  
166 in He (CRM/RM Information Center of China) as standards. CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O fluxes  
167 were then calculated as the rate of change in the mass of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O per unit of  
168 surface area and per unit of time. Three injections were used for each analysis. One  
169 sample was injected to the GC for each analysis. The detection limits of the instrument  
170 for CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O were 1, 0.1 and 0.05 ppm, respectively.

#### 171 172 *Global warming potential (GWP)*

173 GWP is typically estimated using CO<sub>2</sub> as the reference gas, and changes in emission of  
174 CH<sub>4</sub> or N<sub>2</sub>O are converted to “CO<sub>2</sub>-equivalents” (Hou *et al.*, 2012). The GWP of the

175 combined emission of CH<sub>4</sub> and N<sub>2</sub>O was calculated as (Ahmad *et al.*, 2009):  $GWP =$   
176  $(\text{cumulative CO}_2 \text{ emission} \times 1) + (\text{cumulative CH}_4 \text{ emission} \times 34) + (\text{cumulative N}_2\text{O}$   
177  $\text{emission} \times 298)$

178

#### 179 *Measurement of soil properties*

180 Three replicates of soil samples were collected from each treatment. The samples were  
181 transported to the laboratory and stored at 4°C until analysis. The temperature, pH,  
182 salinity and water content of the top 15 cm of soil were measured *in situ* at each plot on  
183 each sampling day. Soil pH was measured with a pH/temperature meter (IQ Scientific  
184 Instruments, Carlsbad, USA), while salinity was measured using a 2265FS EC meter  
185 (Spectrum Technologies Inc., Paxinos, USA) and water content measured using a TDR  
186 300 meter (Spectrum Field Scout Inc., Aurora, USA). Soil samples were collected from  
187 the 0-15 cm layer from each plot for quantifying ferric, ferrous and total Fe  
188 concentrations. Total Fe concentration was determined by digesting fresh soil samples  
189 with 1 M HCl. Ferrous ions were extracted using 1,10-phenanthroline and measured  
190 spectrometrically (Lu, 1999). Ferric concentration was calculated by subtracting the  
191 ferrous concentration from the total Fe concentration. After 60 DAT total soil C, N and  
192 P were analyzed. Total soil C and N were determined by Elementar Vario MAX CN  
193 Analyzer (Elementar Scientific Instruments, Hanau, Germany). Total soil P  
194 concentration was quantified by perchloric-acid digestion followed by ammonium-  
195 molybdate colorimetry, using a UV-2450 spectrophotometer (Shimadzu Scientific  
196 Instruments, Kyoto, Japan).

197

#### 198 *Statistical analysis*

199 Differences in soil properties and CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O emissions among the treatments  
200 were tested for statistical significance by repeated-measures analyses of variance. The  
201 relationships between mean GHG emissions and soil properties were evaluated by  
202 Pearson correlation analysis. These statistical analyses were performed using SPSS  
203 Statistics 18.0 (SPSS Inc., Chicago, USA). The effects of treatments on total soil C, N  
204 and P concentrations were analyzed by one-way ANOVA, with Tukey post hoc test  
205 using Statistica 8.0 (StatSoft, Inc. Tule, Oklahoma, USA).

206

207

## 208 RESULTS

209

#### 209 *CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O emissions and plant growth*

210 CO<sub>2</sub> emission varied significantly among sampling dates but not among treatments (  
211 Tables S1, S2). In general, CO<sub>2</sub> flux remained low (<1040 mg m<sup>-2</sup> h<sup>-1</sup>) during the first  
212 16 days after treatment (DAT) and then increased to a seasonal peak (>1610 mg m<sup>-2</sup> h<sup>-1</sup>)  
213 between 16 and 91 DAT (Figure 2A1A). *J. sambac* was nearly mature by 106 DAT,  
214 with a corresponding decrease in CO<sub>2</sub> emission.

215 ~~CH<sub>4</sub> emission did not vary among sampling dates and treatments~~ N<sub>2</sub>O emission  
216 varied significantly among sampling dates but not among treatments (Tables S1, S2).  
217 Maximum fluxes were earliest in the straw+gypsum treatment at 16 DAT (2.11 mg m<sup>-2</sup>  
218 h<sup>-1</sup>). The N<sub>2</sub>O flux peaked at 46 DAT in the control (2.56 mg m<sup>-2</sup> h<sup>-1</sup>), straw (2.01 mg

219 m<sup>-2</sup> h<sup>-1</sup>), straw+biochar (1.78 mg m<sup>-2</sup> h<sup>-1</sup>) and straw+steel slag (1.57 mg m<sup>-2</sup> h<sup>-1</sup>)  
220 treatments (Figure 2E1C). N<sub>2</sub>O emissions from the *J. sambac* plantation presented a  
221 decreasing trend until 316 DAT.

222

223 The flowering period lasts from 91 DAT to the sampling date. During this period  
224 plants had reached their maximum height and did not grow more (Figure 2D1D). The  
225 little changes in plant height from 91 DAT to the harvest (286 DAT) observed in Figure  
226 2D-21D were not significant.

227

#### 228 *Cumulative emissions*

229

230 The cumulative CO<sub>2</sub> emissions during the study period were higher for straw + gypsum  
231 slag than the control. Cumulative CO<sub>2</sub> emission were not statistically different in steel  
232 slag, the straw+biochar and straw+gypsum slag treatments than in the straw treatment  
233 (Figure 3A2A). The cumulative CH<sub>4</sub> emissions were not statistically different in straw,  
234 steel slag, the straw+biochar and straw+gypsum slag treatments than in the control  
235 (Figure 3B2B). Cumulative N<sub>2</sub>O emissions were not statistically different in straw, steel  
236 slag, the straw+biochar and straw+gypsum slag treatments than in the control (Figure  
237 2C) (Tables S1 and S2).

238

#### 239 *Soil properties*

240 Soil pH, temperature, water content and ferrous, ferric and total Fe concentrations  
241 varied among the treatments (Figure 43, Tables S3, S4). Soil pH was higher in the  
242 straw+biochar, straw+steel slag and straw+gypsum slag treatments than in the control  
243 and straw treatments ( $P<0.05$ ). Soil salinity did not differ significantly among the  
244 different experimental treatments but all the treatments were significantly higher than  
245 in the control ( $P<0.05$ ), and straw+biochar and straw gypsum slag treatments had more  
246 effect on the soil salinity than that of straw treatment ( $P<0.05$ ). Soil-water content was  
247 also higher in the straw+biochar and straw+gypsum slag treatments than the in control  
248 and was higher in the straw+biochar treatment than in the straw treatment ( $P<0.05$ ).  
249 Soil Fe<sup>2+</sup> concentration was significantly higher in all amended treatments than the  
250 control ( $P<0.05$ ), while Fe<sup>3+</sup> concentration was significantly lower in the straw  
251 treatment than in the control, straw+biochar, straw+steel slag and straw+gypsum slag  
252 treatments ( $P<0.05$ ). Total Fe concentration was significantly higher in the straw+steel  
253 slag treatment than in the control ( $P<0.05$ ).

254

#### 255 *Relationships between CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O emissions and soil properties*

256 Seasonal CO<sub>2</sub> and N<sub>2</sub>O emissions were correlated positively ( $P<0.01$ ) with soil  
257 temperature (Table S5) and negatively with soil Fe<sup>3+</sup> and total Fe concentrations  
258 ( $P<0.01$ ) in all treatments. Soil salinity and water content were also positively  
259 correlated with CO<sub>2</sub> emission in all treatments ( $P<0.01$ ). Seasonal CH<sub>4</sub> emission was  
260 negatively correlated with soil Fe<sup>2+</sup> concentration ( $P<0.05$ ) in the straw and straw+steel  
261 slag treatments. Salinity and water content were also positively correlated and soil pH  
262 was negatively correlated with N<sub>2</sub>O emission ( $P<0.05$ ) in all treatments.

263

264 *Global warming potential (GWP)*  
265 GWP was significantly higher for CO<sub>2</sub> than for CH<sub>4</sub> and N<sub>2</sub>O emissions. However, total  
266 GWP was similar in among treatments (Table 2).

267  
268 *Total soil C, N and P concentrations*  
269 Soil total C, N and P concentrations after 60 days (growing period) of treatments  
270 application were higher in treatment plots than in control plots (Figure 54). Total soil  
271 C and N concentrations followed the order of straw+biochar > straw+steel slag =  
272 straw+gypsum slag > straw > control (Figure 4A34A,B). Total soil P concentration  
273 values followed the order straw = straw+steel slag > straw+biochar = straw+gypsum  
274 slag > control (Figure 54C).

## 275 DISCUSSION

276 *Effects of the treatments on soil properties*

277 Soil pH was higher in the straw+biochar, straw+steel slag and straw+gypsum slag  
278 treatments than the control and straw treatments, likely due to the high alkaline contents  
279 of the amendment material. Many cations are released due to the increase of pH (Wang  
280 et al., 2015a). Soil salinity was significantly higher in the amended treatments than the  
281 control (P<0.05). The straw, biochar, steel and gypsum amendment materials used in  
282 this study contained some elements such as K and Ca that can easily been released into  
283 the soil solution increasing the salinity (Wang et al., 2015a). Soil-water content was  
284 also higher in the straw+biochar treatment than in the control and straw treatments.  
285 Biochar can absorb water, because it has high total porosity, and can thus increase soil-  
286 water content (Asai et al., 2009).

287 Soil Fe<sup>2+</sup> concentration was significantly higher in all amended treatments than the  
288 control. The wastes are rich in minerals containing Fe<sup>3+</sup> and straw decomposition would  
289 add more carbon substrates, such as short-chain fatty acids that can act as electron  
290 donors, favoring altogether the conditions for Fe<sup>3+</sup> reduction to Fe<sup>2+</sup> (Li et al., 2011).  
291 At this regard it is important to highlight the important inputs of iron that the treatments  
292 suppose (207, 16 and 17 kg ha<sup>-1</sup>, for steel slag, biochar and gypsum slag, respectively).  
293 Higher soil concentrations of organic matter and CO<sub>2</sub> production and Fe<sup>2+</sup> under  
294 treatments should also favor methane production (da Silva et al., 2010), mostly in the  
295 soil water contents between 40 and 60% observed in the studied period. At these levels  
296 of soil moisture, several soil meso- and micro-pores should be flooded favoring  
297 anaerobic conditions and methane production (Wagner, 2017).In our study, the Fe<sup>3+</sup>  
298 concentration was lower in the straw treatment than the control, supporting the premise  
299 that straw or straw combined with waste could increase the reduction of Fe<sup>3+</sup> to Fe<sup>2+</sup>  
300 and increase the Fe<sup>2+</sup> concentration.

301 All treatments increased total soil C, N and P concentrations respect to control  
302 after a year of application, which suppose a general increase of soil nutritional capacity,  
303 especially in the soils receiving the application of straw + biochar that presented the  
304 highest total soil C and N concentrations. The increases of soil N concentrations would



305 be especially important considering that those wetlands have been proved to be N-  
306 limited (Wang *et al.*, 2015a).

307  
308 *Effects of the treatments on CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O emissions*

309 CO<sub>2</sub> emission changed significantly during the *J. sambac* growing season, and seasonal  
310 CO<sub>2</sub> emission was positively correlated with soil temperature in all treatments. CO<sub>2</sub>  
311 emission was negatively correlated with soil Fe<sup>3+</sup> and total Fe concentrations. Iron  
312 plaques form around *J. sambac* roots at high Fe concentrations, which would limit the  
313 transport of CO<sub>2</sub> from the soil to the atmosphere by the internal system of  
314 interconnected gas lacunae of the plants. Soil-water content was also positively  
315 correlated with CO<sub>2</sub> emission. When soil is not water saturated, soil respiration can  
316 increase during wetter periods, so soil CO<sub>2</sub> emissions are then positively related to soil  
317 water content (McElligott *et al.*, 2017).

318 Straw application can increase CO<sub>2</sub> emissions by several mechanisms. First, the  
319 decomposition of straw directly releases CO<sub>2</sub> (Curtin *et al.*, 1998). Second, the  
320 decomposition of straw adds carbon substrates to the soil, which will increase the  
321 amount and activity of soil microbes and then increasing emission of CO<sub>2</sub> (Curtin *et al.*,  
322 1998). Despite not statistically significantly, mean CH<sub>4</sub> emissions were generally higher  
323 in the straw, straw+biochar and straw+gypsum slag treatments than the control, most  
324 likely due to the input of carbon substrates from the straw. The decomposition of straw  
325 would increase the carbon concentration of the soil and promote the emission of CH<sub>4</sub>  
326 (Wang *et al.*, 2015b). Contrarily mean CH<sub>4</sub> emissions tend to be lower in the straw+steel  
327 slag treatment than the control. The steel slag is rich in Fe<sup>3+</sup> and would increase the  
328 concentration of soil Fe<sup>3+</sup>, which is an alternative electron acceptor that can use C  
329 substrates before methanogens (Jiang *et al.*, 2013) and thereby decrease the amount of  
330 CH<sub>4</sub> production and emission (Gauci *et al.*, 2008) by competing with methanogens for  
331 C substrates (Jiang *et al.*, 2013).

332 Mean N<sub>2</sub>O emissions tended to be lower in the straw+biochar and straw+steel slag  
333 treatments than the straw treatment. Biochar can decrease N<sub>2</sub>O emission (Wang *et al.*,  
334 2011) and is rich in alkaline material, so it can increase soil pH, stimulate N<sub>2</sub>O reductase  
335 activity and thereby induce the reduction of N<sub>2</sub>O to N<sub>2</sub> (Yanai *et al.*, 2007). Biochar  
336 may also improve soil aeration and impede the function and diversity of denitrifying  
337 bacteria, thereby decreasing N<sub>2</sub>O emission (Cavigelli and Robertson, 2001). Steel slag  
338 is also rich in Fe<sup>3+</sup>, and an increase in soil Fe<sup>3+</sup> concentration could suppress microbial  
339 activity, including N<sub>2</sub>O production (Noubactep, 2011).

340  
341 **CONCLUSIONS**

342 **The total GWPs for all emissions were not different in the amended treatments than the**  
343 **control.**

345 All treatments increased total soil C, N and P concentrations after one-year of their  
346 application. The high increase of total soil N concentration under straw + biochar was  
347 especially important due to the limiting role of N in these wetlands. Moreover, biochar  
348 and steel slag are rich in nutrients, such as Si, Ca and magnesium, which can all improve  
349 soil fertility

350

351 The alkalinity of the biochar and the steel and gypsum slags also improved the soil  
352 quality in this *J. sambac* producing area with high levels of acid deposition. The soil-  
353 water content also increased. Our results also suggested that the application of straw  
354 combined with biochar, steel slag or gypsum slag effectively reduced the adverse  
355 impacts of *J. sambac* agriculture on plant growth and soil properties, even increasing  
356 the soil N and P stocks without increasing significantly the GWPs.

357

358 Another positive aspect is that the wastes used as fertilizer materials in this study are in  
359 abundant supply for application to agriculture. They also have a low cost and contribute  
360 to the recycling of wastes.

361

362 Summarizing, our results thus provide strong evidence for several benefits from the  
363 application of these fertilizers: recycling of several waste materials and improved *J.*  
364 *sambac* production without affecting GHG emissions. Thus, the application of straw  
365 combined with steel slag, and especially with biochar is a potentially positive and  
366 sustainable agricultural practice that moreover is largely available and have a low  
367 economical cost.”

368

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519

## Tables

Table 1. Chemical properties of different waste amendments in this study.

Treatments	Chemical properties										
	Fe <sub>2</sub> O <sub>3</sub> (%)	Fe (%)	SO <sub>3</sub> (%)	S (%)	SiO <sub>2</sub> (%)	C (%)	N (%)	P (%)	K (%)	Mg (%)	Ca (%)
Steel slag	4.8	-	-	-	40.7	0.79	0.01	0.01	0.5	0.36	24.9
Biochar	-	0.2	-	0.6	-	56.6	1.4	1.0	1.8	1.0	0.5
Gypsum slag	0.4	-	54.4	-	0.7	0.7	0.01	0.01	0.1	0.3	30.6

Table 2. Effect of the amended treatments on the global-warming potential.

Treatment	Global warming potential (kg CO <sub>2</sub> -eq ha <sup>-1</sup> )			Global-warming potential (kg CO <sub>2</sub> -eq ha <sup>-1</sup> )
	CO <sub>2</sub>	CH <sub>4</sub>	N <sub>2</sub> O	
Control	52572.44±1226.13a	2.65±2.89a	41.34±13.66a	52616.43±1321.25a
Straw	66162.44±11791.43a	4.11±4.40a	31.56±4.90a	66198.12±11790.94a
Straw+biochar	66714.37±8762.01a	25.14±22.09a	28.99±2.45a	66768.50±8753.62a
Straw+steel slag	59953.90±7961.67a	1.71±2.66a	20.41±7.89a	59976.01±7971.45a
Straw+gypsum slag	71004.78±5828.47a	15.20±10.37a	32.21±7.10a	71052.19±5822.38a

Different letters within a column indicate significant differences between the amended and control treatments ( $P < 0.05$ ) obtained by Bonferroni's post hoc test.

1 **Figure legends**

2 Figure 1. CO<sub>2</sub> (A), CH<sub>4</sub> (B), N<sub>2</sub>O (C) emissions and Plant height (D) in the control and  
3 treatment. Error bars indicate one standard error of the mean of triplicate measurements.  
4 Different letters indicate significant differences ( $P<0.05$ ) between treatments.

5 Figure 2. Cumulative emissions of CO<sub>2</sub> (A), CH<sub>4</sub> (B) and N<sub>2</sub>O (C) in the amended and  
6 control treatments during the study period. Error bars indicate one standard error of the  
7 mean of triplicate measurements. Different letters indicate significant differences  
8 ( $P<0.05$ ) between treatments.

9 Figure 3. Soil pH (A), temperature (B), salinity (C), water content (D), Fe<sup>2+</sup>  
10 concentration (E), Fe<sup>3+</sup> concentration (F) and total Fe concentration (G) in the control  
11 and treatment. Error bars indicate one standard error of the mean of triplicate  
12 measurements. Different letters indicate significant differences ( $P<0.05$ ) between  
13 treatments.

14 Figure 4. Total soil C (A), N (B) and P (C) concentrations observed in each treatment  
15 plots after one-year of treatments application. Error bars indicate one standard error of  
16 the mean of triplicate measurements. Different letters indicate significant differences  
17 ( $P<0.05$ ).

18