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1 **Plant secondary compounds in soil and their role in belowground species interactions**

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21 belowground interactions

22

23 **Abstract**

24

25 Knowledge about the effect of plant secondary compounds (PSCs) on belowground interactions in
26 the more diffuse community of species living outside the rhizosphere is sparse compared to what
27 we know about how PSCs affect aboveground interactions. We illustrate here that PSCs from foliar
28 tissue, root exudates, and leaf litter effectively influence such belowground plant-plant, plant-
29 microorganism, and plant-soil invertebrate interactions. Climatic factors can induce PSC production
30 and select for different plant chemical types. Climate change can therefore alter both quantitative
31 and qualitative PSC production, and how these compounds move in the soil. This can change the

32 soil chemical environments, with cascading effects on both the ecology and evolution of
33 belowground species interactions, and ultimately soil functioning.

34

35

36

37 **Plant chemicals mediate species interactions**

38 Secondary compounds in plants are major contributors to the chemical diversity of nature. They are
39 formed from **primary metabolites** (see Glossary) in specific pathways and omnipresent in plants.
40 The distribution of **plant secondary compounds** (PSCs) is heterogeneous across the plant kingdom,
41 and these compounds exhibit extensive variation both among and within species [1]. Though many
42 PSCs also play pivotal roles in the plant's primary metabolic processes (such as growth or
43 development), the large majority are known to serve as infochemicals mediating interactions between
44 plants and their biotic and abiotic environments. PSCs can profoundly modify the abiotic environment
45 as they are released into the atmosphere, soil, and water of an ecosystem, which in turn determines
46 the community structure of the biotic constituents. For example, PSCs that accumulate in stems and
47 foliage of the vegetation can promote bushfires, that profoundly alter the physicochemical properties
48 of soil and groundwater [2]. Many PSCs are also involved in ecologically crucial interactions for
49 plants, as they can form chemical barriers against pathogens and herbivores, attract pollinators and
50 predators of herbivores, and help disperse fruit and seeds [3]. To fine-tune the release of PSCs, plants
51 “eavesdrop” on their neighbours and increase their own defences when they sense volatiles released
52 from neighbouring plants damaged by leaf herbivores [4].

53 PSC-mediated interactions between plants and their associated organisms are best-documented for
54 aboveground compounds associated with airborne transport and signalling [3, 5, 6]. PSCs, however,
55 are also omnipresent in the soil, where they mediate interactions with soil organisms and
56 neighbouring plants. Compared to aboveground interactions, these soil PSCs represent a “hidden”
57 link between organisms. Recent reviews of PSCs and belowground interactions [7-9] have focused
58 on PSCs in root exudates and on species interactions, mainly in the **rhizosphere**. However, less
59 attention has been given to the impact of PSCs on the broader and more diffuse community of
60 associated species further away from roots, which also includes the effects of PSC from foliar tissue
61 and litter decomposition. Climate change may alter both the quantitative and qualitative production
62 of PSCs, which could cascade to affect belowground PSC-mediated community interactions. Given
63 the vital role PSCs play aboveground, we expect PSCs to be equally important to belowground plant-
64 organism interactions.

65 Here, we focus on four major groups of PSCs: terpenes, flavonoids, glucosinolates, and alkaloids
66 (Table 1) that enter the soil matrix. We discuss their origin and fate in the soil and their ecological

67 and evolutionary impacts on belowground interactions. Release of PSCs by plants alters soil
68 chemistry, and neighbouring species can either tolerate, adapt to or escape the chemical environment
69 this imposes on them. However, the production, release, and transport of PSCs by plants are greatly
70 affected by local conditions, such as microclimate, microbes, herbivore and pathogen pressure.
71 Therefore, changes in climate will affect the quantitative and qualitative production and release of
72 PSCs both directly through changes in temperature and humidity but also indirectly through potential
73 concomitant changes in herbivore and pathogen pressure. As the effects of PSCs on neighbour species
74 are often both compound-specific and dose-specific, there is the potential for cascading effects of
75 PSC-driven alteration of soil chemistry on belowground interactions. So far, however, very little is
76 known regarding such cascading effects in plant-soil community interactions. To encourage more
77 studies on this topic, we outline knowledge gaps and describe best practices for sampling and
78 detecting PSCs in soil, to designate directions for future studies advancing this field of research.

79

80 *Origin of PSCs in soils*

81 How do PSCs enter the soil? Those formed in roots can be excreted directly into the soil from living
82 and deteriorating tissues [10]. PSCs formed in aboveground organs are deposited into the soil from
83 belowground organs following internal transport [11] or during wet and dry deposition from the
84 atmosphere after volatilization and further chemical reactions [12-14]. PSCs can also be washed from
85 vegetation surfaces with water intercepted during precipitation [15], or they may outgas and leach
86 from fresh and decomposing litter on the soil surface and in the upper horizons [16, 17]. Many plant
87 species accumulate large quantities of PSCs in secretory organs that are either external, such as
88 trichomes, or internal, such as oil blisters or resin ducts, and a strong discontinuous release is expected
89 during the decline and decomposition of these organs.

90 Concentrations of PSCs can be high and persistent in the soil. For instance, mono- and sesquiterpenes
91 released from pine needles, litter and roots are detectable in forest soil throughout the year [18] and
92 can attain high concentrations especially in topsoil but also lower mineral soil, being highest in late
93 summer, and in the winter during snow cover [19]. Glucosinolates, produced by all cruciferous plants,
94 are hydrolysed in the soil to highly bioactive compounds, notably isothiocyanate, nitriles, and
95 thiocyanate. The degradation and hydrolyzation of glucosinolates depend on the presence of the
96 extracellular enzyme myrosinase [20]. Glucosinolates are water-soluble and can more easily be
97 washed out from soil, but isothiocyanates are hydrophobic and adsorb strongly to organic matter [20,
98 21], and may thus persist in the soil for periods of time. The isothiocyanate sinigrin, for example, is

99 persistent in the soil of forests invaded by *Alliaria petiolata* (garlic mustard), where even low
100 concentrations suppress germination and growth of arbuscular mycorrhiza. This suppression disrupt
101 the mycorrhizal association with native plant, negatively affecting the native plant's resource uptake
102 [22]. The persistence and slow degradation of many PSCs in soil emphasize the ecological and
103 evolutionary potential pressure PSCs can impose on belowground species.

104

105 *PSC production under climate change*

106 The production of PSCs improves the ability of plants to cope with different abiotic [23] and biotic
107 stressors [5, 6]. Various abiotic factors, such as CO₂ and temperature, affect PSC release by up- or
108 downregulating their production, although the specific PSC compounds that are affected may differ
109 among factors. For example, in northern hemisphere trees, high CO₂ levels generally increase the
110 foliar concentration of phenolic compounds such as flavonoids and tannins, and decrease the
111 production of terpenoid compounds, whereas elevated temperatures lead to the opposite trend [24-
112 26]. Drought can both increase and decrease PSC production dependent on the severity of drought
113 and on the compound. Terpenoids and phenolics in temperate and Mediterranean trees and herbs
114 increased under moderate drought stress but decreased under severe drought [25, 27, 28]. Although
115 severe drought overall decreased total terpenoid emission, some specific mono- and sesquiterpenes
116 either increased or remained unaffected, suggesting an adaptive role of specific compounds in
117 mitigating abiotic stress [27, 28].

118

119 In addition to environmental effects on the production of specific compounds or compound classes,
120 genetically determined variation in the PSC chemical phenotype (hereafter referred to as chemotype)
121 is present in many species. For example, a strong genetic basis for variation in the identity of leaf
122 chemical compounds has been demonstrated for terpenes produced in oak and pine trees [29] [30,
123 31], aromatic Lamiaceae [32], Asteraceae [33], for glucosinolates produced in the Brassicaceae [34],
124 and for flavonoids in Berberidaceae [35], illustrating that genetic variation for these compounds are
125 present in a wide range of plant families. Thus the (a)biotic environment can shape which compounds
126 are released to the environments not only through differential induction of PSCs, but also by natural
127 selection, favouring genotypes producing specific chemotypes (Figure 1)[36]. Differences in climatic
128 conditions and herbivore pressure favour specific chemotypes both within and among species [29,
129 37, 38]. A recent example of evolutionary changes in PSC production in response to climate change

130 is that of wild thyme (*Thymus vulgaris*), a small shrub widespread in the Mediterranean Basin. The
131 dominant monoterpene in thyme is genetically determined to be a phenolic or a non-phenolic type.
132 These two chemotypes are ecotypes with adaptation to warm dry summers and early winter frost,
133 respectively [39]. The lack of very cold winters in the last 25 years is associated with a significant
134 increase in phenolic chemotypes in regions that were previously dominated by non-phenolic types
135 [38]. In addition to differences in their ability to tolerate frost and drought, the toxicities of phenolic
136 and non-phenolic chemotypes differ towards associated plants, microorganisms and herbivores,
137 suggesting that climate-driven changes in the genetic composition of this widespread species may
138 have cascading effects on associated species interactions.

139 A correlation between variation in PSC production and environmental variation holds for many plant
140 species, but we need more long-term experimental studies to understand whether current and future
141 changes in climate and herbivore pressure will result in major induced or genetic changes in PSC
142 production. If so, cascading effects of such changes are expected for belowground species interactions
143 and decomposition of organic matter and ultimately nutrient cycling in soil [25, 40]. Moreover,
144 climate change may also affect the concentration of PSCs in soil by altering soil-water content,
145 temperature, and pH. As a result, the levels and circulation of liquid and gas among soil pores (see
146 below) will change and hence the movement and diffusion of PSCs.

147

148 *Diffusion of PSC in soil*

149 In contrast to atmospheric PSCs, volatile PSCs in the soil can form stable concentration gradients due
150 to the lack of air turbulence. The diffusion of PSCs in soil depends on their physicochemical
151 properties (Table 1) and the texture of the soil matrix. Fine-textured and clayey soils facilitate the
152 flow of volatile PSCs that are diluted in smaller air volumes due to the smaller pores of these soil
153 types. This generates gradients that are steeper and propagate faster than those in soils with large air
154 spaces. In well-ventilated soils, though, PSCs will rapidly fill large air volumes, albeit at lower
155 concentrations.

156 PSC transfer in soil is also affected by adsorption to the mineral and organic surfaces of soil particles,
157 whose fractions depend on soil type and layer, with A horizons (topsoil) generally containing more
158 organic material than the underlying B horizons (subsoil) [41, 42]. PSCs will adhere to and dissolve
159 on lipo- and hydrophilic surfaces during diffusion, thus forming intermediate pools in equilibrium

160 with the gas-phase concentrations [43, 44]. Soil organisms, therefore, receive PSCs more slowly and
161 with an attenuated amplitude compared to the actual variations in the emission strength of the PSC
162 source.

163 Abiotic factors such as temperature, soil moisture, and pH interfere with the persistence and diffusion
164 of PSCs in soil. The adsorption of volatile PSCs by the mineral and organic fractions of soil depends
165 on pH, with higher adsorption in alkaline than acidic soils [45]. The acidity of soil-pore water can
166 affect the uptake of volatiles on aqueous surfaces, e.g. by protonating highly reactive sesquiterpenes
167 that are otherwise poorly soluble in water [46]. Periods of rain and drought change the movement
168 patterns of PSCs through soil, with opposite behaviours expected for volatile and non-volatile water-
169 soluble PSCs (Table 1). During rainy weather, volatile PSCs are first concentrated in the soil pores
170 during water penetration and then move with air bubbles to the surface [47], whereas non-volatile
171 water-soluble PSCs are diluted and transported away from their sources [48]. High levels of soil
172 moisture also affect the availability of oxygen for roots and soil microorganisms, eventually
173 generating anaerobic conditions, which influence both plant PSC-producing and microbial PSC-
174 decomposing metabolisms [45, 49]. In contrast, drought may decrease the concentration of volatile
175 PSCs in the soil due to the increase in the volume of air in the soil and better ventilation at the soil
176 surface. Sesquiterpenes, for example, emitted by maize diffuse faster and farther at low moisture
177 levels [50].

178 Climate change can deeply impact the movement and lifetime of PSCs in soils, either by affecting
179 abiotic drivers directly or by affecting the macro- and micro-biotic communities involved in the
180 liberation, processing and transport of PSCs [51] (see also section “Plant-soil microorganism
181 interactions” below). For example, increasing soil temperature will enhance PSC diffusion and
182 exchange rates and alter state and partitioning equilibria between surfaces. Extreme weather events
183 such as thunderstorms and drought events are expected to increase in both frequency and intensity,
184 which will amplify the asymmetric transport and concentration build-up of polar/non-polar PSCs in
185 soils. Determining the presence, and the fate of PSCs in soil is an under-researched area but is highly
186 relevant to evaluate their ecological and evolutionary impact on the belowground interactions
187 discussed below. Understanding the main environmental factors driving these processes is necessary
188 to assess how climate change will alter belowground interactions mediated by PSCs (Figure 1).
189 Different methods may be needed for different types of PSCs, depending on their solubility and
190 volatility (see Box 1, Table 1).

191 **Ecological and evolutionary impacts of PSCs on belowground species interactions**

192

193 Similar to the functions of PSCs aboveground, key organismal interactions in soil, i.e. plant-plant,
194 plant-soil microorganism, and plant-soil invertebrate interactions, are also mediated by PSCs. The
195 identity, concentration, and composition of PSCs belowground play pivotal roles in species
196 interactions and community structuring, and ultimately affect entire ecosystem processes.

197

198 Plant-plant interactions

199 PSCs mediate both competitive and facilitative interactions among plants. **Allelopathic** effects of
200 PSCs released by a focal plant to the local soil environment can inhibit germination and growth of
201 competitors giving the focal plant a competitive advantage for space and nutrients [52]. The “Novel
202 Weapons” hypothesis [53] is based on the idea that invasive plants release **allelochemicals** which the
203 native community has not previously encountered and that give the invader a competitive advantage.
204 Allelopathic effects of PSC have helped some invasive plant species to create virtual monocultures
205 in the habitats they invade. For example, production of **phytotoxic** phenolic acids, and di-, and
206 sesquiterpene lactones likely aided the invasive success of Canadian goldenrod *Solidago canadensis*
207 [54, 55] and yellow star thistle *Centaurea solstitialis* [56]. Allelopathic effects are often less dramatic
208 within the native ranges of plants because coexisting species have evolved a tolerance to local PSCs,
209 or the ability to detoxify them [57].

210 PSCs released to the soil by one plant species may alter competitive hierarchies among neighbouring
211 plant species both directly when PSC inhibit growth of one species more than another, and indirectly
212 by PSC modifying the soil nutrient availability that favour some species over others. For example,
213 pine soil altered plant competition between spotted knapweed (*Centaurea stoebe*) and different grass
214 species in favour of grasses. Soil containing pine litter had a higher content of phosphorus (P), and
215 the higher availability of P made the grasses more competitive against *Centaurea* [58]. Phenolic
216 compounds and terpenes influence soil nitrogen (N) cycling as these compounds inhibit nitrification
217 and the net N mineralization [59]. This could differentially favour plant species that differ in their
218 preference for nitrate vs ammonium [60].

219 PSC mediated plant-plant interactions also facilitate neighbour plants via so-called associational
220 resistance [61] where neighbour plants benefit from reduced attack by herbivores and pathogens due

221 to the deterrent effects of the PSCs released by a focal plant. However, the PSCs in soil may also
222 increase susceptibility to herbivores and pathogens if these plant enemies are attracted rather than
223 repelled by the PSC [52, 61]. How these selective forces are balanced in the multispecies networks
224 present in a natural field needs further study.

225 PSCs can vary among species, genotypes within species, and damaged and healthy tissue, so they
226 contain information about the identity and state of the emitting plant [62]. Plants may therefore use
227 PSCs as a cue to recognize their neighbours and prepare for the competitive encounters by adjusting
228 their competitive growth towards or away from the neighbouring plant [63]. Root exudates and the
229 chemicals they contain have been shown to mediate recognition of neighbour plants [64-66]. The
230 competitive responses to such PSC-mediated neighbour recognition may depend on the history of co-
231 existence between interacting plants [67], and for some plant species also on the genetic relatedness
232 among interacting conspecifics [64, 68]. However, we lack knowledge on the identity of which
233 molecules in root exudates that mediate plant recognition and how plants perceive them. This
234 knowledge is relevant in order to understand if environmental changes in the soil may interfere with
235 plants ability to detect their neighbour.

236 Spatial consistency of PSC production allows PSCs to act as selective agents on other species and
237 favour the genotypes that respond most beneficially to the effects of frequently encountered PSCs.
238 For example, the grass *Bromus erectus* is a frequent neighbour of *T. vulgaris*, which releases either
239 phenolic or non-phenolic monoterpenes to the local soil environment. *Bromus* plants originating from
240 sites where their thyme neighbours produce a non-phenolic monoterpene germinate and grow better
241 in soil rich in non-phenolic thyme monoterpenes compared to similar soil rich in phenolic thyme
242 monoterpenes [69]. Hence, plants can adapt to the chemical environment imposed by frequent
243 neighbours, and this adaptation is highly compound-specific. Changes in the consistency of
244 frequently encountered PSCs could disrupt such adaptive responses.

245 *From pairwise plant-plant interaction to communities*

246 The PSCs released to the soil from different plant genotypes and species create a mosaic of soil
247 chemical environments. Because different chemical environments favour different species due to the
248 pairwise competitive and facilitative interactions described above, variation in chemical types both
249 within and among populations can contribute to local and regional species richness. For example, the
250 chemical diversity of Scots Pine (*Pinus sylvestris*) needle terpenes was positively correlated with
251 plant species richness in the vegetation under the trees, and trees with different needle chemistry

252 associated with a different plant species composition [70, 71]. Chemical diversity of *T. vulgaris* also
253 correlated with plant species richness and composition of the vegetation around thyme plants. Plant
254 species richness was higher, and species composition more similar around thyme plants, than in
255 thyme-free area only few meters away. However, species richness and species composition differed
256 in area with the different thyme chemical types [72]. These examples suggest that the chemical types
257 of dominant plants have an “extended phenotype” [73, 74] where the chemicals released from a focal
258 plant can shape the surrounding vegetation. However, we need more experimental manipulation
259 studies investigating the consistency of the effects of variation in PSC released by foundation plants
260 on richness and composition of associated plant communities. These studies will allow us to better
261 understand cause and effect of variation in PSC and to predict the potential cascading effects of
262 qualitative and quantitative changes in PSCs.

263

264 Plant-soil microorganism interactions

265

266 A large proportion of the nutrients present in the soil are not immediately available for plants but
267 require enzymes to degrade compounds into usable parts, and soil microorganisms are major suppliers
268 of these enzymes [80]. In addition, soil symbionts such as mycorrhizal fungi and rhizobacteria are
269 important regulators of plant productivity especially in nutrient poor ecosystems, where they are
270 responsible for 5 to 75% of nitrogen and up to 80% of phosphorous acquired by plants annually [81]
271 The importance and complexity of the interactions between plants and soil microbes have been
272 acknowledged for a long time, but only more recently it has been appreciated that many of these
273 interactions are mediated by PSCs [82]. Given the importance of such interactions, it is likely that
274 plants have evolved strategies to recognize friends from foes and to harness useful partners across
275 distances. For example, flavonoids produced by legumes have a fundamental role in attracting
276 rhizobia and promoting the production of Nod-factor molecules, which are critical for nodulation
277 [83]. In this process, both plants and microbes produce chemical signals that allow the plant to
278 recognize the rhizobia as a friend and prevent the activation of immune responses upon initiation of
279 the symbiosis [84]. When rhizobia are not found in the proximity of the rhizosphere and distances
280 between the roots of individual plants and their symbionts are too large to be covered by root
281 exudates, legumes can resort to the help of an intermediary to carry the symbionts to them [83, 85].
282 The production of volatile PSCs by the legume *Medicago truncatula* attracts individuals of the free-
283 living nematode *Caenorhabditis elegans*, which harbours the rhizobium *Sinorhizobium meliloti* either

284 on its cuticle or in its intestine. Isoflavonoids produced by the plant prime the rhizobium to initiate
285 the symbiosis when the nematode is near the roots [75].

286 Not only can plants affect the composition of specific soil microbial communities via PSC, but they
287 can also alter it when exposed to stress such as herbivore attacks [82]. In many plant species, the
288 presence of pathogens or root feedings fungi corresponds with a shift in the soil microbial
289 composition close to the rhizosphere. This is a consequence of both the production of specific
290 antimicrobial compounds by the plant rhizosphere and of recruitment of favourable bacteria [86] The
291 blend of PSCs produced by the graminoid *Carex arenaria* shifts toward a higher concentration of
292 terpenes and terpenoids [59] to attract specific bacteria with antifungal properties when the roots are
293 infected with the pathogenic fungus *Fusarium culmorum* [87]. Bacteria successfully rallied by the
294 plant can gain from this interaction via the additional nutrients obtained by feeding on the plant
295 pathogens [87]. Recent studies compared wild types of *Arabidopsis thaliana* with mutants that are
296 unable to synthesize PSCs involved in induced systemic resistance against soil pathogens.
297 Metagenome analysis of root microbiome associated with wild types vs. mutants showed substantially
298 different community composition, indicating the important role of PSC in modulating the
299 composition of root microbiome [76].

300 PSC-induced shifts in the composition of soil microbiota can also be activated by attacks to plant
301 leaves. For example, infestations of white flies (*Bemisia tabaci*) in pepper plants lead to a change in
302 belowground microflora enhancing anti-pathogen responses [88].

303 The interactions between plants and soil microorganisms are not unidirectional, as soil microbes can
304 also emit chemical compounds that can be detected and used by plants. For example, secondary
305 compound emissions from soil microflora may help the woodland strawberry, *Fragaria vesca*, to
306 detect nutrient-rich patches prior to rooting. Stolons can preferentially develop toward substrates
307 emitting unbranched fatty acids and avoid substrates with higher concentrations of terpenes and
308 aldehydes [89].

309 Many PSCs are used by bacteria and fungi as a source of nutrients [90, 91], and soil microorganisms
310 can alter and repurpose the PSCs released by plants. In the legume *Lotus japonicus*, the inoculation
311 with its symbiont *Mesorhizobium loti* leads to a change in the profile of the plant phenolic compounds
312 [60]. The soil bacterial community associated with vetiver, *Chrysopogon zizanioides*, uses the
313 sesquiterpenes produced by the grass as a source of carbon, and after metabolizing them releases
314 several compounds from vetiver oil [63]. Colonization of roots by arbuscular mycorrhizal fungi can

315 increase the production of glucosinolates, phenolic acids, and flavonoids in several plant species [64,
316 65].

317 On a large scale, the ability of soil microbial communities to process and degrade plant secondary
318 metabolites can have important consequences on plant communities and ecosystem stability. For
319 example, soil microorganisms can either favour or hinder the proliferation of invasive plant species,
320 depending on their ability to degrade the allelopathic compounds produced by invasive plants [92].

321 A high species-specificity of plant-microbe interactions suggest that single plants can shape their soil
322 bacterial communities, and microorganisms can respond to subtle genetic variation in plants. In
323 cottonwood trees *Populus angustifolia*, 70% of the variation in soil microbial community
324 composition was explained by plant genotype identity [77].

325 Although there is still much that we need to understand about the complex interactions between PSC
326 and roots microbiome, progress is being made in elucidating the factors that drive the composition of
327 root inhabiting bacterial communities [78, 79]. Future steps will need to go beyond pairwise
328 interactions between plants and a specific guild of soil microorganism, targeting instead how whole
329 communities of soil microorganisms are affected by PSC and in turn affect plant communities.
330 Climate change may mediate shifts in plant-soil microorganism interactions via shifts in the chemical
331 profile of plants and soil microorganisms. Such studies are highly valuable as the positive effect of
332 certain plant-soil microorganism interactions can be harnessed to either enhance agricultural practices
333 or as biocontrol agents. Understanding how to engineer beneficial soil communities for plant
334 community conservation, sustainable agriculture and preservation of native communities require
335 extensive experimentation and in depth understanding of potential unintended consequences of such
336 manipulations.

337

338 Plant-soil invertebrate interactions

339 Many of the interactions between plants and soil invertebrates are mediated by PSCs, often with
340 significant effects on soil processes that determine soil function and plant productivity. Two major
341 ways can be distinguished through which PSCs mediate the interactions between plants and soil
342 invertebrates. First, soil organisms can be exposed to substantial input of PSCs from aboveground
343 plant tissues, through PSCs leaching from foliar tissue and PSC-rich litter deposition. These PSCs
344 are mostly known as feeding repellents, and many of them with toxic effects to aboveground insect

herbivores [80], and the available evidence indicates that they also have toxic effects on non-target organisms such as soil invertebrates [81, 82]. Toxic effects can occur if PSCs diffuse across the cuticle or if animals ingest PSC-rich litter, with the main route of exposure determined by cuticle morphology and diet of the soil invertebrate and by the chemical properties of the PSC (see Table 1 for mode of toxicity). Nonpolar (lipophilic) volatile PSCs likely enter animal bodies primarily by simple diffusion across the cuticle and easily partition into cellular membranes [81]. High concentrations of nonpolar PSCs can occur regularly in soil, especially in older litter and near roots, as is the case for terpenoids (1-5 mg g⁻¹ fresh weight) [2]. The likelihood of exposure is increased because terpenoids will remain in the litter layer for several months until fully decomposed [83]. Exposure to polar PSCs is mainly through ingestion of litter or roots, for example in case of isothiocyanates. Isothiocyanates have a high toxicity towards non-target soil invertebrates, including beneficial macro-detritivores that promote nutrient cycling and soil fertility [84]. Currently, the relative importance of the two exposure routes, either via contact with volatile or water dissolved PSCs in pore water and diffusion across the cuticula or via ingestion of PSC-rich litters to soil fauna and how this affects soil fauna functional community composition is poorly known.

An unexplored aspect of the lipophilic PSCs is that their spread can be facilitated by animal transport. Soil animals can accumulate high internal concentrations of these PSCs in the vicinity of roots and litter, and subsequently, deliver them to other soil microenvironments with lower chemical activity when they move away from the PSC source. Aquatic Protozoa have demonstrated such animal-facilitated transport [85], but more research is needed to explore this phenomenon mediated by soil animals. In order to explore the toxicity of PSCs to soil invertebrates under field conditions, more precise analytical determinations of concentrations of PSCs and their degradation products in the environment are needed. This necessitates new experimental systems of exposure, e.g. by using the framework of chemical activity for volatile compounds [81].

The second way in which PSCs can mediate plant invertebrate interactions in the soil is by targeted release of PSCs from the roots. These PSCs can attract or deter soil invertebrates and serve as chemical signals released by the plant to control trophic interactions with plant-parasitic nematodes, root herbivores, and their predators. The use of PSCs in tri-trophic interactions between plants and belowground organisms may be similar to the extensively studied chemically mediated interactions between plants, foliar herbivores, and their natural aboveground predators. Attack by root herbivores has been shown to induce a change in the quality and quantity of PSCs released by plant roots, which can then function as a signal to attract soil-dwelling predators and reduce the pest population [9, 86,

87]. Entomopathogenic nematodes often function as indirect defences for plants [88, 89]. For example, larvae of the longhorn beetle *Tetraopsis tetraophthalmus* feed on the roots of the common milkweed *Asclepias syriaca* which releases volatiles into the soil that attract entomopathogenic nematodes, preventing the loss of plant biomass to herbivory [90]. Maize roots increase sesquiterpene exudation when attacked by insects, thus attracting entomopathogenic nematodes that feed on root herbivores [91]. However, notwithstanding this potential protective role of PSCs via attraction of beneficial nematodes, these PSCs can also attract the root herbivores themselves. The net effect of the attraction of both advantageous as well as detrimental nematodes on plant fitness has to be evaluated. Most examples of tri-trophic interactions belowground involve entomopathogenic nematodes, but one of the few papers that reports on other soil fauna in tri-trophic interactions is a case in which tulip bulbs infested by the rust mite *Aceria tulipae*, emits volatile signals to the predatory mite *Neoseiulus cucumeris* [92]. Given the importance of chemical cues in belowground signalling, the effect of PSCs on tri-trophic interactions is expected to be omnipresent, but currently, we do not know enough about the spatial dimensions of volatile and water-dissolved PSCs effects on plant-root herbivore or parasite-herbivore interactions [93]. Abiotic gradients in soils are rather short and steep compared to aboveground systems, but currently, we are ignorant of the spatial dimensions of volatile and water-dissolved PSCs effects on plant-root herbivore or parasite-predator interactions.

394

395 *PSCs and belowground co-evolutionary interactions*

396 The examples above demonstrate the different vital roles that PSCs play in belowground plant-plant, 397 plant-soil microorganism, and plant-soil invertebrate interactions. Some of these interactions are 398 examples of the evolutionary adaptation of species to frequently encountered PSCs. Except for 399 interactions specific to the rhizosphere, no studies have yet clearly shown whether adaptations of 400 organisms to specific PSCs reciprocally alter PSC composition and production in plants, i.e. if PSC- 401 mediated belowground interactions result in pairwise or diffuse co-evolution. Understanding if 402 belowground interactions are a source of selection on the chemical variation of PSC-producing plants 403 is not straightforward because of the unspecific nature of the interactions. Belowground interactions 404 often consist of multiple interactions, therefore, estimating the consequences for fitness in PSC- 405 producing plants is difficult. Also, for PSCs entering the soil via leaf leachates, the main selective 406 force determining which PSCs are produced and in what concentration may largely depend on 407 aboveground biotic interactions and local climate. In addition, the multifunctionality of PSCs extends

beyond biotic interactions, which is a further complication in the evaluation of fitness consequences in PSC-producing plants. For instance, maize produces benzoxazinoid compounds not only as a defence against generalist herbivores but also as chelating agents to acquire iron as a micronutrient. The larvae of western corn rootworm *Diabrotica virgifera*, are not injured by benzoxazinoids, but this specialist herbivore instead takes advantage of the presence of complexes between iron and these PSCs not only to detect the plant but also forage on these complexes [94].

Climate-induced shifts in the release of PSCs into the soil (Figure 1) may alter the outcome of within- and between-species interactions in ways that we still cannot predict. Long-term studies and the development of ecometabolomics [95] for elucidating the responses of PSC changes in plant communities due to warming and altered precipitation from climate change are particularly warranted for a better understanding of, and ability to predict, these shifts in PSCs production, diffusion in soil, and effects on belowground interactions.

Concluding remarks and perspectives

Soil is a theatre of facilitation, symbiosis and warfare deployed by plants and the various organisms living in it, and PSCs plays a major role mediating many of these interactions. Plants and soil organisms have adapted to withstand, detoxify or use the cocktail of PSCs originally meant to harm some of them. Understanding PSC mediated relationships at community scale and identifying the compounds involved in these interactions is therefore important for better insight into the functioning of these systems and their evolution, especially in changing environments. High-throughput technologies are increasingly becoming available and may help to identify and determine the concentrations, transport and modes of action of PSCs in soils. We encourage the creation of open, community-wide, curated, labelled, broad-spectrum PSC data sets across plant species and soils, as this would greatly increase the transfer of knowledge between scientist studying plants, microbes, and invertebrates in this biological belowground theatre. This could be a platform similar to, e.g. the TRY plant trait data base (<http://www.try-db.org>) or PSC data sets could be added and incorporated to such established well-functioning platforms. The warranted study of the quantitative and qualitative effects of climate change on PSCs and their interactions with the environment and organisms would also benefit from these PSC data sets. The alterations of the production of PSCs due to climate change are highly variable, depending on dose, timing, PSC, and species. These changes in the production of PSCs can lead to unforeseen consequences for soil structure and function

439 and can disturb biological feedbacks on soil chemistry and biology, perhaps even on atmospheric
440 chemistry and climate [96], with a direction and intensity that warrants in-depth investigation.

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676 **BOX 1.** Sampling of PSCs for identification and quantification.

677 The heterogeneous chemical and physical nature of soil creates spatial differences that complicate
678 representative sampling. In addition, PSCs possess a high diversity of functional groups, volatilities,
679 and polarities. Their belowground presence can thus be in air, in pore water, or adsorbed to solid
680 matter. Representative sampling may therefore require a combination of sampling techniques.

681 Sampling techniques for PSCs in soil include extraction (solvent or solid phase), head-space
682 analysis, and purge and trap techniques [7].

683 Solvent extraction relies on the principle of “like dissolves like”. Polar compounds are thus
684 extracted in polar solvents (e.g. water, methanol) and nonpolar compounds in nonpolar solvents
685 (e.g. n-hexane, dichloromethane). Several hundreds of grams of soil, however, may be needed for
686 the representative sampling of PSCs with low concentrations in soil [109]. Dilution beyond
687 detection limits may occur, so subsequent sample concentration by solid-phase extraction (SPE) or
688 evaporation (with the risk of loss of volatiles) may be necessary. Emission of PSCs directly from roots
689 have been investigated using primarily solid-phase micro extraction (SPME) or adsorption tubes (with e.g.
690 Tenax) coupled with GC-MS analysis in a number of studies as reviewed in [7].

691 Solid-phase micro extraction (SPME), head-space analysis, and purge and trap methods avoid the
692 dilution effect of solvents but are more selective in the compounds sampled. SPME relies on the
693 equilibrium between a fibre coated with a sorbing material and most often the gas phase (potentially
694 also water or solid phase). This technique allows in-situ sampling for evaluating small-scale spatial
695 differences. The sorbing material determines the selectivity of the compounds sampled [110],
696 whereas soil type, water content, and pH influence the equilibrium. Stir-bar sorptive extraction
697 (SBSE), where a magnetic rod is coated with sorbing material, may provide up to 1000-fold higher
698 sensitivity but requires thermal desorption for subsequent instrument introduction. SBSE is useful
699 for extraction from water, as demonstrated for many organic pollutants [111]. Identifying the mode
700 of transport of PSCs (i.e. in gas or water phase) could be accomplished by sampling the soil gas
701 phase using SPME and subsequently washing the soil and extracting using SBSE.

702 Collecting only head spaces using an air-tight syringe is a potential way to sample only volatile
703 PSCs (i.e. VOCs). The concentration of VOCs from soil, however, may be $<10 \mu\text{g m}^{-3}$, which is
704 below the detection limit for head-space analysis. This limitation may be overcome with purge and
705 trap methods recommended for analysis when concentrations are $<200 \mu\text{g kg}^{-1}$ [112], which have
706 recently been applied for sampling isoprene, monoterpenes, and sesquiterpenes in forest soil [19].

707 **Glossary box:**

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709 **Allelochemical** is a compound produced by an organism that can have a detrimental physiological
710 effect on individuals of another species when released into the environment.

711 **Allelopathic effect** is the interference effect of chemical compounds released by a plant on the
712 growth and performance of other plants. An allelopathic effect usually implies a direct negative
713 (harmful) effect, but some definitions also include positive effects.

714 **Narcotic effect** refers to the unspecific effects of lipophilic organic chemicals on membrane
715 fluidity. These chemicals tend to partition into the lipid compartments of the organism, causing
716 (nonpolar) narcosis or baseline toxicity.

717 **Phytotoxic** refers to chemical compounds toxic to plants, including any adverse effect on plants
718 such as delayed seed germination, inhibition of growth, death, or destruction of plant tissue.

719 **Plant secondary compounds** (PSCs) are organic compounds typically formed from primary
720 metabolites in specific pathways. Some PSCs are involved in plant's primary metabolic processes,
721 but the large majority of these are known to serve as infochemicals mediating interactions and are
722 important for plants to survive in the environment, e.g. by their repellent, attractant, or toxic effects
723 on other organisms.

724 **Primary metabolites** are compounds from metabolic pathways present in all plants and are
725 essential for growth, development, and reproduction. These metabolites comprise amino acids,
726 nucleic acids, peptides, various carbohydrates, and lipids.

727 **Rhizosphere** is the microecological zone of soil in direct proximity to plant roots and includes the
728 organisms in that soil that are directly affected by root exudates.

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734 Legend to Figure 1. Direct and indirect effects of climatic conditions (red box) and aboveground
735 herbivore pressure (green box) on plant secondary compound production (yellow boxes) and input
736 into the soil, and its belowground interactions (brown box). In addition, climate affects how PSCs
737 are transported and degrade in soil. Dashed arrows and boxes indicate processes and effects that are
738 less well understood and are elucidated in the main text.

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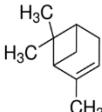
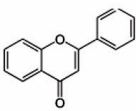
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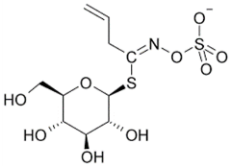
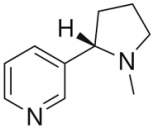
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758 Table 1. Overview of the important classes of PSCs involved in mediating belowground
759 interactions, and best practices for sampling and analysing them. The chemical properties of PSCs
760 affect their position and transport in the soil matrix, with polar PSCs dissolved in water, volatile
761 PSCs diffusing through air spaces in the soil, and less polar non-volatile PSCs adsorbed to the
762 organic fraction of soil.

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Compound class	Example of structure	Polarity	Water solubility	Volatility	Ecological interactions/ impacts and effects on associated soil organisms	The analytical instrument for detection in soil (see also Box 1)
<p>Terpenes</p> <p>Synthesized from isoprene units (C_5H_8). Includes monoterpenes ($C_{10}H_{16}$) and sesquiterpenes ($C_{15}H_{24}$).</p>	 <p>α-Pinene</p>	Low	Low	High	<p>Defensive and signalling compounds, many with a characteristic odour (e.g. α-pinene from pine, limonene from citrus). Antioxidant, antimicrobial, and phytotoxic properties [3]. Toxic to invertebrates due to baseline toxicity (“narcotic” effect) via interaction with cellular membranes [97]. Involved in formation of aerosol particles, thus affecting climate [98].</p>	<p>Monoterpenes and highly volatile sesquiterpenes: S with PDMS adsorbent, purge-and-trap with Tenax adsorbent and subsequent GC-MS [19], [99],</p> <p>On-line measurement with Proton-transfer-reaction MS [100].</p> <p>Nonpolar sesquiterpenes likely adsorbed to soil particles require extraction with organic solvents and evaporation followed by GC-MS [41].</p>
<p>Flavonoids</p> <p>General structural C-15 skeleton consisting of two phenyl rings</p>		Low to intermediate	Low to intermediate Presence of sugar moieties increases	Low	<p>Both defensive and signalling; promote symbiosis (used by legumes to attract rhizobia), various antimicrobial activities [101, 102]. Can interact with Ca^{2+}</p>	<p>Most likely present belowground in water and air. Extraction using polar solvents (water, methanol) and metals are formed [104].</p> <p>Extracts can be analysed with reversed-phase LC-MS.</p>

and a heterocyclic ring (C6-C3-C6).	Basic flavonoid structure		solubility in water		channels in cell membranes causing cytotoxic influx of Ca^{2+} [103].	Aglycones may be sampled using SPE [101] and analyzed by GC-MS after derivatization. Greater sensitivity may be obtained with SBSE and subsequent GC-MS.
Glucosinolates Derived from glucose and an amino acid. Most are found in the order Brassicales (30).	 Sinigrin	High	High	Low	Due to polarity and water solubility, are readily taken up by soil organisms, where they can be converted to highly reactive compounds such as isothiocyanate and thiocyanates. Toxicity pathways include inhibition of cytochrome P_{450} detoxification complex, induction of programmed cell death (apoptosis), and genotoxic effects [106].	Mostly present in the water phase due to high polarity. Extraction using polar solvents (water, methanol) followed by analysis with LC-MS [17] or GC-MS after derivatization for concentration in the sample.
Alkaloids Highly diverse, usually with a basic nitrogen atom either in	 Nicotine	Low to intermediate	Low to intermediate	Low	Mostly defensive compounds. Phytotoxic, neurotoxic (e.g. strychnine), and narcotic effects (e.g. morphine, codeine). Can interfere with gene	Low concentration in soil, so clean-up and pre-extraction may be necessary. Polarity can be adjusted with pH changes. Extraction with liquid-liquid extraction, SPE, or SPME [108] followed by analysis with both LC-MS and GC-MS.

cyclic rings or open chains.					transcription and protein synthesis [107].	
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Outstanding Questions

To what extent does the plant secondary compound (PSC) plume originally released into a soil compartment become chemically degraded during diffusion and mixed with secondarily formed and other ambient PSCs? Does this dilution and mixing with other PSCs impair the signal quality for the receiving soil organisms or have they adapted to and even benefited from signal noise, e.g. to improve their foraging efficiency?

How will climate change affect PSC production in both above- and belowground organs and how will these effects in turn affect the belowground interactions of associated communities of microbes, plants, and soil invertebrates?

What is the relative importance of specific PSCs for both plant-soil invertebrate and plant-plant interactions and can we understand their prevalence from knowledge on their mode of action? If the effects of PSC on associated species is very compound specific, quantitative and qualitative changes in the input of PSC to the soil can have cascading effects on richness and composition of belowground soil communities that may ultimately alter soil functioning.

What is the importance of a common co-evolutionary history between PSCs in soil and the organisms exposed to them for shaping the outcome of their interactions?

How important are belowground interactions in determining the composition and concentration of PSCs released by the producer? In particular, what is the impact of selection (if any) that belowground interactions can exert on PSC composition and production in leaves relative to aboveground interactions and climate?

HIGHLIGHTS:

- Plants produce a high diversity of secondary compounds that are released to their surroundings.
- The role of plant secondary compounds (PSCs) on aboveground interactions has been well studied, but less is known about the role of PSCs in mediating and determining the outcome of belowground interactions – especially interactions beyond the rhizosphere.
- PSCs enter the soil directly by excretion from belowground plant organs or indirectly by washing from the canopy and leaching from litter.
- Leaching of PSCs creates a heterogeneous soil chemical environment and mediate plant-plant interactions, plant-microbiome associations, and plant-soil invertebrate relationships. This affects and shape the diffuse community of species living in soil also further away from the rhizosphere.
- Climate change can substantially alter the quantitative and qualitative production of PSCs released to the soil, which may change belowground species interactions and affect species composition, richness, and ultimately soil functioning.

Figure

