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1 **Chronic drought alters extractable concentrations of mineral elements in**
2 **Mediterranean forest soils.**

3

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23 **Abstract**

24 Soil mineral elements play a crucial role in ecosystem productivity and pollution
25 dynamics. Climate models project an increase in drought severity in the Mediterranean
26 Basin in the coming decades, which could lead to changes in the composition and
27 concentrations of mineral elements in soils. These changes can have significant impacts
28 on the fundamental processes of plant-soil cycles. While previous studies have
29 predominantly focused on carbon, nitrogen, and phosphorus, there is a notable lack of
30 research on the biogeochemical responses of other mineral elements to increasing
31 drought. In this study, we investigated the effects of chronic drought (15 years of
32 experimental rainfall exclusion) and seasonal drought (summer period) on the
33 extractable soil concentrations of 17 mineral elements in a Mediterranean holm oak
34 forest. We also explored the potential biotic and abiotic mechanisms underlying the
35 changes in extractable elemental concentrations under chronic drought conditions. Our
36 findings reveal that soil elemental concentrations varied significantly due to seasonal
37 changes and chronic drought, with soil microclimate, biological activity, and organic
38 matter being the main drivers of this variability. Levels of soil water content primarily
39 explained the observed variations in soil elemental concentrations. Most of the mineral
40 elements (13 out of 17) exhibited higher concentrations during winter-spring (wet
41 seasons) compared to summer-autumn (dry seasons). The chronic drought treatment
42 resulted in potassium limitation, increasing vegetation vulnerability to drought stress.
43 Conversely, the accumulation of sulfur in soils due to drought may intensify the risk of
44 sulfur losses from the plant-soil system. Under drought conditions, certain trace

45 elements (particularly manganese, vanadium, and cadmium) exhibited increased
46 extractability, posing potential risks to plant health and the exportation of these
47 elements into continental waters. Overall, our results suggest that alterations in mineral
48 element concentrations under future drier conditions could promote ecosystem
49 degradation and pollution dispersion in the Mediterranean Basin. Understanding and
50 predicting these changes are essential for effective ecosystem management and
51 mitigating the potential negative impacts on plant health and water quality.

52

53 **Keywords:** Chronic drought; Seasonality; Mediterranean holm oak forest; Soil mineral
54 elements; Extractable concentrations; Soil water content

55

56 **1. Introduction**

57 Spatial-temporal variability of mineral elements in soils is caused by a complex range
58 of factors including parent material, pedogenic processes, topography, climate and both
59 biotic and anthropogenic impacts (Alloway, 2013; Beygi and Jalali, 2019; Islam et al.,
60 2015; Salekin et al., 2021; Valladares et al., 2009). Soil mineral elements play critical
61 roles in ecosystem productivity and pollution as a result of biogeochemical processes
62 (Chorover et al., 2007; Huang et al., 2005; Sardans and Peñuelas, 2007a, b; Sardans et
63 al., 2008c, 2020; Zhou et al., 2021).

64 Drought events in the Mediterranean Basin are likely to be more frequent and
65 intense during the next decades (IPCC, 2014; Vicente-Serrano et al., 2014). Moreover,
66 the changing patterns of precipitation in the Mediterranean region are also expected to

67 intensify droughts in these ecosystems where water stress is already the principal
68 constraint (Cramer et al., 2018; Vicente-Serrano et al., 2014; Zhou et al., 2019). These
69 alterations are expected to affect the composition and concentrations of mineral
70 elements in Mediterranean soils, as shown by several studies of the biogeochemical
71 responses of soil carbon, nitrogen and phosphorus to greater drought over various time
72 scales (Asensio et al., 2021; Marañón-Jiménez et al., 2022; Sardans and Peñuelas, 2005;
73 Sardans et al., 2008a, 2020). However, much less attention has been paid to the effects
74 of drought on other mineral elements that also play a role in crucial biogeochemical
75 processes such as mineral nutrient cycling (Deng et al., 2021; Johnson and Turner, 2019;
76 Pastor and Bockheim, 1984) and the dispersion of trace element pollutants (Hooda,
77 2010; Purves, 2012; Szykowska et al., 2018).

78 Drought is likely to change the mobility and biological availability of mineral
79 elements along both direct and indirect pathways (Sardans and Peñuelas, 2007a, b;
80 Sardans et al., 2008c, d; Schlesinger et al., 2016), which can affect the cycling and loss
81 of elements and deeply influence ecosystem structure and function (Deng et al., 2021;
82 Sardans et al., 2008c). Water limitation and the consequent reductions in soil diffusion
83 capacity lead to greater accumulations of organic matter (OM), which, together with
84 less soil enzymatic activity and lower microbial mineralization rates (Manzoni et al,
85 2012; Peñuelas et al., 2018; Sardans and Peñuelas, 2005), slow down the release of
86 mineral nutrients and therefore decrease soil nutrient availability, the nutrient uptake
87 capacity of plants and microbes, and plant growth (Asensio et al., 2021; Marañón-
88 Jiménez et al., 2022; Sardans et al., 2008c, d). These alterations [can heighten the](#)

89 [likelihood of losing mineral elements through runoff](#) (Sardans et al., 2008c, d;
90 Schlesinger et al., 2016). On the other hand, less soil water also decreases leaching into
91 the soil solution and mineral weathering rates (Schlesinger et al., 2016; Slessarev et al.,
92 2016). Greater soil OM under chronic drought conditions conditions may increase
93 metal-OM complexes, which will improve soils' capacities to prevent the loss of
94 mineral elements from ecosystems (Sardans et al., 2008c; Solly et al., 2020). However,
95 on the long-term this effect can be offset by the decrease in total biomass and
96 consequently in soil organic matter. In addition, the drying-rewetting events induced by
97 droughts could also influence the status of mineral elements and, for example, a rise in
98 the intensity of the drying and rewetting cycles could enhance the transformation of
99 certain elements from available to unavailable forms (Ma and Uren, 1997; Sardans and
100 Peñuelas, 2007b).

101 Extractable concentrations (EC) are crucial variables for evaluating the mobility
102 and biological availability of mineral elements (Beygi and Jalali, 2019; Li et al., 2014;
103 Rees, et al., 2014; Valladares et al., 2009). Changes in EC caused by drought may
104 depend on the balance of input and output concentrations that represent the combined
105 effects of the biological and abiotic processes described above. The effects of drought
106 on the EC of individual elements differ in terms of the element and a range of factors
107 such as the type of ecosystem and the cause and degree of the drought (Deng et al.,
108 2021; Sardans et al., 2008c, d). Previous studies in the Mediterranean Basin reported
109 increases in the EC of iron, magnesium, sulphur, arsenic and titanium, but decreases in
110 the EC of potassium, nickel and lead in holm oak forest soils after a 6-year

111 experimentally induced continuous drought (Sardans and Peñuelas, 2007a, b; Sardans
112 et al., 2008c); conversely, there was no significant effect on the EC of mineral elements
113 in shrubland soils (Sardans et al., 2008b, d, e). However, our knowledge about how soil
114 mineral elements respond to long-term drought conditions is still highly incomplete. A
115 comprehensive analysis of the effects of drought on soil mineral elements can provide
116 insights into the complex mechanisms controlling the variability in their EC and how
117 drought reshapes soil element composition in the field. Any such analysis will thus
118 enhance our ability to accurately model and predict biogeochemical responses to future
119 climatic conditions. Therefore, we studied the effects of seasonal (summer drought) and
120 long-term chronic drought on the EC of mineral elements in soils in 2014, after 15 years
121 of continuous water exclusion treatment (started in 1999) and investigated the
122 relationship between EC and possible biotic (soil respiration, soil enzymatic activity,
123 microbial biomass) and abiotic environmental factors (soil moisture and temperature)
124 in a Mediterranean holm oak forest. We hypothesized that 1) both seasonal and chronic
125 drought affect the quantity and composition of soil mineral elements even though the
126 effects are more significant due to season than chronic drought and 2) the mechanisms
127 driving the responses of soil elements to changing environmental conditions differ from
128 one element to another because their concentrations involve different biological and
129 abiotic controls and have different physico-chemical traits such as distinct solubility,
130 responses to pH changes or chelating interactions with soil organic matter.

131

132 **2. Materials and methods**

133 *2.1. Study site and experimental design*

134 The study was conducted in a natural holm oak forest in the Prades Mountains in
135 Catalonia, NE Iberian Peninsula (41°21'N, 1°2'E). The study sites have been described
136 elsewhere in detail by Mu et al. (2018, 2019) and Ogaya et al. (2020). Briefly, the
137 climate is typically Mediterranean characterized by hot dry summers and moderately
138 wet and cold winters; springs and autumns are usually also wet. The average annual
139 temperature and rainfall are 12.2°C and 616.1 mm, respectively. The soil is a Dystric
140 Cambisol at a depth of 35–90 cm and the vegetation is dominated by *Quercus ilex* with
141 an abundance of shrubs such as *Phillyrea latifolia* and *Arbutus unedo*.

142 Eight plots (15×10 m) were established at the same altitude (950 m) along a slope
143 (25°), four of which were subject to the drought treatment and four left as control plots
144 under natural conditions. The drought treatment was carried out from 1999 to 2014 (15
145 years) and consisted of partial rain exclusion caused by the installation of PVC strips
146 above the ground that intercepted 30% of plot precipitation; moreover, water runoff was
147 intercepted by a 0.8-m deep ditch excavated along the entire top edge of the plot.
148 Further details of the plot location and drought treatment are given by Mu et al. (2018)
149 and Ogaya et al. (2020).

150 *2.2. Soil CO₂ efflux and temperature and core sampling*

151 Soil sampling in 2014 was performed mid-season, on 25–27 January (winter), 16–18
152 May (spring), 31 July–2 August (summer) and 24–26 October (autumn). Five soil cores
153 (5-cm diameter, 10-cm depth) were taken per plot per season (total n = 160).

154 The soil CO₂ efflux (SR) was measured between 09.00–15.00 and prior to soil core

155 sampling (see below) using five soil collars permanently inserted into the soil (2 cm)
156 of each plot in a closed system with a SRC-1 soil chamber connected to an EGM-4
157 portable system (PP-systems, Hitchin, UK), as described by Zuccarini et al. (2020) and
158 Mu et al. (2022). Simultaneously, soil temperature (ST) was measured at one point near
159 the collars at a depth of 10 cm using a TO 15 digital soil thermometer (Jules Richard
160 Instruments, Argenteuil, France). The soil cores were collected at around 15:00 and the
161 soil samples were transported to the laboratory, sieved to 2 mm and stored in
162 hermetically closed plastic bags.. The <2 mm fraction was selected and divided into
163 subsamples for the analysis of gravimetric soil water content, concentrations and
164 activities of soil extracellular enzymes, soil extractable element concentrations, and
165 microbial biomass element concentrations. The subsamples were stored in hermetic
166 plastic bags at 4°C prior to analysis. Gravimetric soil water content (SW) was
167 determined by measuring weight loss after drying at 105°C for 24 h.

168 *2.3. Extracellular enzyme activity*

169 Fresh soil subsamples (2 mm sieved) were used for colorimetric analyses of soil
170 extracellular enzyme activity, as described by Zuccarini et al. (2020) and Mu et al.
171 (2022). Briefly, acid and alkaline phosphomonoesterase (acp and akp), β -glucosidase
172 (bgl), protease (prot) and urease (ure) enzyme activity were determined and expressed
173 as the μ mol of substrate released per gram of dry soil and incubation time (h). These
174 values were used as enzymatic indicators of the biogeochemical cycling of C (bgl), N
175 (prot + ure) and P (acp + akp), which were labelled as enzC, enzN, and enzP respectively.

176 *2.4. Soil extractable elements*

177 Soil extractable organic carbon (EOC) and extractable total nitrogen (ETN) were
178 measured in 0.5 M K₂SO₄ extracts (4:1 v:w; fresh soil) using a total carbon/total
179 nitrogen (TC/TN) analyser (Multi N/C 3100; Analytik Jena AG, Jena, Germany).
180 Extractable total phosphorus (ETP) and extractable arsenic (As), calcium (Ca),
181 cadmium (Cd), chromium (Cr), copper (Cu), iron (Fe), mercury (Hg), potassium (K),
182 magnesium (Mg), manganese (Mn), molybdenum (Mo), nickel (Ni), lead (Pb), sulphur
183 (S), strontium (Sr), vanadium (V) and zinc (Zn) in 0.5 M NaHCO₃ extracts (10:1 v:w;
184 fresh soil) were measured using inductively coupled plasma/optical emission
185 spectrometry (ICP-OES Optima 4300DV, PerkinElmer, Wellesley, USA) after digestion
186 with 1% HNO₃, as described by Zuccarini et al. (2020). The elemental concentrations
187 in the extracts were expressed as µg of extractable element per unit of soil dry mass.
188 The mineral elements targeted in this study were As, Ca, Cd, Cr, Cu, Fe, Hg, K, Mg,
189 Mn, Mo, Ni, Pb, S, Sr, V and Zn, the EOC, ETN and ETP having been reported before
190 (Asensio et al., 2021) and were only used as explanatory variables for the
191 concentrations of the target elements.

192 The extraction with bicarbonate at pH 8.5 is considered the most suitable for the
193 analysis of inorganic P forms in calcareous to slightly acidic soils (Olsen et al., 1954;
194 FAO, 2021), although the low acidity of the sodium bicarbonate reduces the solubility
195 of other elements like Al, Co, Cu, Fe, Ni and Mn but increases that of Mo and S (Truog
196 1948). Other standard extraction methods such as Mehlich-3, the use of chelating agents
197 such as diethylenetriaminepentaacetic acid or monocalcium phosphate extractions can
198 be more suitable for specific elements (Antonangelo et al., 2022), but it would take

199 several extractions for the same sample and the use of hazardous chemicals as compared
200 to the simple bicarbonate extraction. Thus, since our aim was to perform a screening of
201 the elemental concentrations to determine biologically available elements in these soils
202 and their links to environmental parameters, we have selected the simplest method. It
203 should be noted that soil water content at sampling can affect the extractability of soil
204 elements, with lower concentrations measured under low soil water content (Ross et al.,
205 1989). Given that we did not adjust the soil water content of samples to equal values,
206 the elemental concentrations during the driest seasons could be underestimated.

207 *2.5. Microbial elemental concentrations*

208 Microbial biomass C, N and P were measured using chloroform fumigation-extraction
209 (Jenkinson and Powlson, 1976) as described by Marañón-Jiménez et al. (2022). Briefly,
210 each fresh soil sample was divided into two, with one subsample being fumigated for
211 24 h with ethanol-free chloroform and the other not fumigated. Extractable C, N and P
212 in both subsamples were measured as described above, and the microbial biomass C
213 (MicC), N (MicN) and P (MicP) concentrations were calculated as the difference
214 between the fumigated and unfumigated samples. All elemental concentrations in the
215 microbial biomasses were expressed as μg of element per unit of soil dry mass.

216 *2.6. Data analysis*

217 The effects of season and the chronic drought on SW, ST, SR and the target mineral
218 elements were analysed using two-way ANOVAs, with season and treatment as fixed
219 effects and plot as a random factor. For the target mineral elements, statistical
220 differences were further tested between treatments for each season with a Student's *t*-

221 test and within seasons using one-way ANOVAs. The correlations between target
222 mineral elements and SW for the year and each season were tested using Pearson's
223 correlations. The effects of season on the relationships between target mineral elements
224 and SW were tested using analyses of covariance (ANCOVAs), with season as a fixed
225 factor and SW as the covariate. The effects of season and treatment on the EC of the 17
226 target mineral elements considered as a whole were investigated in a PERMANOVA
227 using the *vegan* package in R (v. 4.1.1). Effects and differences were considered
228 significant at $P < 0.05$.

229 Pairwise Pearson correlations between the soil chemical-biological variables were
230 performed and represented in heatmaps using the *reshape2* package in R (v. 4.1.1) that
231 highlighted significant and strong correlations. A principal component analysis was
232 performed with the *FactoMineR* and *factoextra* packages to visualize the multivariate
233 correlations between the previously standardized soil chemical-biological variables.
234 Vector variables were marked as target mineral elements (red) or other soil variables
235 (blue), while cases were marked by treatment and season. Finally, a redundancy
236 analysis was performed to visualize the multivariate correlations of macroelements with
237 the explicable soil chemical-biological variables ($P < 0.1$) using the *vegan* and *ggrepel*
238 packages. The *ggplot2* package was used for visualization in R.

239

240 **3. Results**

241 *3.1. Effects of chronic drought and seasonal drought on soil water content, temperature,*
242 *and CO₂ efflux*

243 Soil water content was higher in winter coinciding with the lowest temperatures, while
244 the minimum soil water content was recorded in summer (Fig. 1). There were seasonal
245 variations in soil water content ($P < 0.001$), temperature ($P < 0.001$) and CO₂ efflux (P
246 < 0.001); variations in water content were also observed between the treatment plots (P
247 < 0.05 , Fig. 1A). The average soil water content over the sampling period was lower in
248 the chronic drought treatment plots ($12.31 \pm 1.87\%$) than in the control plots ($18.58 \pm$
249 3.24% , $P < 0.001$), with significant differences occurring in winter ($P < 0.01$), summer
250 ($P < 0.05$) and autumn ($P < 0.05$, Fig. 1A). The chronic drought treatment also provoked
251 a reduction in soil CO₂ efflux in all seasons ($P < 0.05$, Fig. 1), although no effects of
252 drought on soil temperature were observed (Fig. 1).

253 *3.2. Effects of chronic drought and seasonal drought on soil extractable element* 254 *concentrations*

255 Ca, K, Mg and S were the most abundant elements in water extracts during the sampling
256 periods, all with a mean EC above $100 \mu\text{g g}^{-1}$ (Table 1 and Fig. 2A), followed by Fe,
257 Sr, Mn and Cu, whose means were above $1 \mu\text{g g}^{-1}$ (Table 1 and Fig. 2B). The mean EC
258 of Zn, Mo, Ni, V, Pb, Cr, As, Cd and Hg were below $1 \mu\text{g g}^{-1}$ (Table 1). Season explained
259 46% ($P = 0.001$) of variation in EC of soil mineral elements when the 17 elements were
260 analysed as a whole in a PERMANOVA, while the effects of chronic drought treatment
261 and Season \times Treatment interaction were not significant (Table 2). The seasonal
262 changes observed on 13 of the studied elements were greater than the changes induced
263 by the chronic drought treatment (Fig. 2 and Table S1). The chronic drought treatment
264 significantly increased the EC of S, Mn, V and Cd but decreased those of K and Cu;

265 other than for Cu, these changes in elemental EC were consistent over the sampling
266 periods (Fig. 2 and Table S1).

267 The EC of K, Mg, Sr, and Cu were strongly positively correlated with soil water
268 content ($r > 0.60$ and $P < 0.001$), with correlations strongest – i.e. with steeper slopes
269 than in other seasons – in spring for K ($r = 0.93$ and $P < 0.001$) and Sr ($r = 0.83$ and P
270 < 0.05) (Fig. 3 and Table S2). The EC of S was negatively correlated with soil water
271 content ($r = 0.49$ and $P < 0.01$; Table S2). The responses of Zn and Mo to soil water
272 content varied seasonally (SW \times Season interaction: $P < 0.05$), with strong correlations
273 only found in one season (Table S2). The EC of Zn was positively correlated with water
274 content in spring ($r = 0.84$ and $P < 0.01$), while for Mo the correlation was negative in
275 autumn ($r = 0.78$ and $P < 0.05$).

276 The chronic drought treatments did not affect soil element EC significantly in any
277 of the seasons analysed separately ($P > 0.05$, Table S3), although the seasonality of Cu,
278 Zn and Hg may have been slightly affected by the treatment (Table S1, Season \times
279 Treatment interaction: $P < 0.05$). Thus, we used data combining control and drought
280 treatments to investigate the seasonal dynamics of soil element EC (Table 3), which
281 represent the average seasonal variation in response to current and future climate
282 conditions. There were similar seasonal patterns in soil element EC, the highest EC of
283 most elements being recorded in winter (Ca, K, Mg, Sr, Cu, Zn, Pb, Cr and Hg) or
284 spring (Fe, Ni, V, As), with minimums recorded in autumn (K, Fe, Sr, Mn, Cu, Pb, Ni,
285 V, As, Cd and Hg). The exception was S, whose EC was highest in summer and autumn
286 and lowest in spring (Table 3).

287 *3.3. Associations in the soil chemical-biological variables*

288 Pairwise correlation analysis identified 13 strong positive correlations in the soil
289 extractable elements ($r > 0.60$ and $P < 0.05$, Fig. 4 and S1). The EC of Ca, K, Mg and
290 Sr were correlated with each other, while Mn was correlated with Fe, V and Cd. Ni was
291 correlated with Fe and Hg, K with Cu, and Mg with Pb (Fig. 4).

292 The EC of some elements were also strongly correlated with other soil variables
293 ($r > 0.60$ and $P < 0.05$, Fig. 4 and S1). K, Mg, Sr and Cu were positively correlated
294 with soil water content. Mn was positively correlated with EOC and ETN, as was Cd.
295 There were also positive correlations between K and EnzP, MicN and MicP, and Cu and
296 MicP. The only negative correlation was between Cr and soil temperature (Fig. 4).

297 The first (PC1) and second (PC2) principal components of the PCA performed with
298 the soil chemical-biological variables explained 29.72 and 18.35% of the total variance,
299 respectively (Fig. 5). The elements K, Sr, Cu and Mg, and SW, ST and micN mainly
300 contributed to PC1, while the elements Mn, Cd, V and Fe, and EOC and ETN mainly
301 contributed to PC2 (Fig. 5C and Table S4A). The EC of Ca, K, Sr and Cu were
302 positively related to SW and microbial activities, while the EC of S showed the opposite
303 and was positively associated with ST (Fig. 5C). The EC of Mn and Cd were most
304 closely correlated with EOC and ETN. Samples differed between the chronic drought
305 and control treatments along PC1 ($P = 0.0358$) and PC2 ($P = 0.0163$). Samples also
306 differed seasonally along PC1 ($P < 0.0001$; Table S4B). The ellipses grouping the data
307 were not well separated for either the four seasons or between the chronic drought and
308 control treatments (Figs. 5A and B).

309 The redundancy analysis of eight macroelements showed that variation in mean
310 soil element EC was best explained by water content ($r^2 = 0.45$, $P < 0.01$; Table S5).
311 Selected soil chemical-biological variables explained 48.76 and 6.54% of variation in
312 macroelement EC along axes 1 and 2, respectively, above all for Ca, K, S and Fe (Fig.
313 6). The seasonal variation in soil element EC (winter-spring and summer-autumn) was
314 separated along the first axis by soil temperature, moisture and respiration: the EC in
315 winter-spring were positively associated with higher levels of soil respiration and
316 moisture, while those in summer-autumn were positively associated with higher soil
317 temperatures (Fig. 6). Overall, the EC of Ca was the dominant soil extractable element
318 in winter, whereas the EC of K and Fe were more abundant in winter and spring and
319 the EC of S was more abundant in summer and autumn (Fig. 6). The EC of Ca was
320 positively correlated with soil water content and microbial N and P concentrations,
321 while the EC of K was more associated with CO₂ efflux and the activity of N- and P-
322 degrading enzymes. The EC of S was positively correlated with soil temperature but
323 negatively correlated with soil water content and CO₂ efflux (Fig. 6).

324

325 **4. Discussion**

326 *4.1. Effects of chronic and seasonal drought on extractable element concentrations in* 327 *Mediterranean soils*

328 The response of individual mineral elements to water stress, as indicated by their
329 elemental concentrations (EC), varied depending on the element and season. While
330 chronic drought did not cause significant seasonal differences, it predominantly

331 affected element EC in spring. Notably, it resulted in decreased extractable Ca, K, Sr,
332 Zn, and Pb at lower levels of soil moisture, whereas extractable Ni and Mo tended to
333 increase during drier winter and autumn, respectively ($r > 0.70$ and $P < 0.05$, Table S2).
334 Additionally, chronic drought increased the EC of S, Mn, V, and Cd but decreased the
335 EC of K and Cu throughout the year ($P < 0.05$, Table S1), aligning with previous studies
336 on the effects of medium-term drought at the study site (Sardans and Peñuelas, 2007a;
337 Sardans et al., 2008c).

338 Our findings indicate that soil moisture strongly influenced the movement and
339 allocation of K in plant-soil compartments, which is closely related to biotic retention
340 processes (Deng et al., 2021; Sardans and Peñuelas, 2007a). Under drier conditions, K
341 leaching declines (Deng et al., 2021; Schlesinger et al., 2016; Slessarev et al., 2016)
342 partly due to increased binding by enzymes in dehydrated K forms (Deng et al., 2021;
343 Maathuis et al., 2009). We also observed lower microbial activity, indicated by reduced
344 soil CO₂ efflux ($P < 0.05$, Fig. 1C) and soil enzyme activity (Asensio et al., 2021;
345 Marañón-Jiménez et al., 2022), in drought plots, which may limit the release capacity
346 of soil soluble K by decreasing mineralization rates under low water availability
347 (Sardans and Peñuelas, 2007a). Consequently, the decrease in the EC of K could be
348 attributed to reduced leaching and microbial mineralization rates. However, other
349 factors, such as lower weathering rates under drought conditions, could also contribute
350 to the decline in soluble K concentrations at the study site (Sardans and Peñuelas,
351 2007a). Reduced soluble concentrations, combined with poorer diffusion capacity
352 under low soil moisture, further limit the availability of K and subsequently its uptake

353 by plants (Sardans et al., 2020). Considering that K plays a critical role in controlling
354 leaf water losses, K limitation can increase plant vulnerability to drought stress,
355 resulting in reduced tree growth and potential forest mortality (Gessler et al., 2017;
356 Sardans and Peñuelas, 2015).

357 Regarding extractable S, our results suggest that its increase is due to lower
358 microbial consumption activity. However, the decrease in plant uptake may be a more
359 significant factor, as previous studies have reported increased extractable S and reduced
360 S concentrations in aboveground biomass in response to drought at the study site
361 (Sardans and Peñuelas, 2007a). Leaf S concentrations have been found to be negatively
362 correlated with sclerophylly, which tends to increase in response to drier conditions in
363 Mediterranean ecosystems (Bacelar et al., 2004; Sardans et al., 2006, 2008c). This
364 could lead to a decline in plant S uptake and subsequent S accumulation in the soil.
365 Additionally, the upward transportation of salts from deep groundwater under drier
366 conditions may contribute to the increase in S (Luo et al., 2016) if there are greater net
367 mass gains of sulphate in drought plots than in control plots. The cation exchange
368 capacity of these soils is low (Table S6) due to the slightly acidic soil pH, which
369 decreases the number of negative charges on the colloids, and the high percentage of
370 the sand fraction, which has no electrical charge and therefore no capacity to exchange
371 cations. Thus, the extractable forms of S and other elements accumulated in the soil
372 during the driest periods are not ionically bonded to the soil surfaces or molecules in
373 the soil solution, being more prone to be lost by leaching after heavy rain events. This
374 makes the ecosystem susceptible to net S and other element losses under a scenario in

375 the Mediterranean Basin of ever more frequent drought crises and a rise in the amount
376 of torrential rainfall (IPCC, 2014; Vicente-Serrano et al., 2014).

377 The increases in EC of Mn, Cd, and V (Table S1) were primarily attributed to
378 higher levels of metal-organic matter (OM) complexes, as evidenced by their strong
379 correlations with extractable organic carbon (EOC) and extractable total nitrogen (ETN)
380 (Figs. 4 and 5C). Previous studies conducted at the study site have demonstrated
381 increased soil EOC and ETN under progressively drier conditions (Asensio et al., 2021;
382 Marañón-Jiménez et al., 2022). The formation and destruction of complexes, influenced
383 by drying-rewetting events, are likely to have a minor role in this process, as the
384 intensified drying and rewetting cycles induced by drought would have insignificant or
385 unclear effects on their availability (Sardans and Peñuelas, 2007b; Sardans et al.,
386 2008b).

387 In the case of Cu, its response to drought exhibited different patterns but is
388 expected to be similar to that of K, considering the close correlation between K and Cu
389 (Fig. S1). Consequently, lower leaching and microbial mineralization levels may have
390 contributed to the decrease in extractable Cu. Apart from Mn, V, and Cd, the EC of
391 other elements such as Zn, Cr, and Ni also showed tendencies of increase (Table 1),
392 indicating a generally higher accumulation of trace elements in soluble forms. This
393 could exacerbate the negative effects of drought on plant productivity and accelerate
394 the export of trace elements from soils to continental waters after torrential rainfall, as
395 observed in similar studies (Sardans and Peñuelas, 2007b; Sardans et al., 2008b).

396 *4.2. Biological and abiotic controls of extractable elemental concentrations in*

397 *Mediterranean soils*

398 Increases in soil temperature may enhance the solubility of all the elements in soil
399 solutions but may also decrease soil water content to a greater extent, as shown by the
400 strong negative relationship between temperature and moisture ($r = 0.82$ and $P < 0.05$,
401 Fig. S1). In addition, soil CO₂ efflux was positively correlated with soil water content
402 but not with soil temperature (Fig. 5C). As a result, the variations in EC were best
403 explained by levels of soil water content (Fig. 5C and Table S5) in this Mediterranean
404 forest.

405 The EC of Ca, K, Mg, Sr, Cu, Pb and Cr were positively correlated with soil water
406 content (Figs. 4 and 5C), which suggests that these elements are more prone to leaching
407 into the soil solution than others and so have a higher risk of being lost to the ecosystem.
408 By contrast, the EC of S was negatively correlated with moisture and CO₂ efflux (Figs.
409 4 and 5C), indicating that increases in microbial and/or root activities could promote S
410 uptake to a greater degree than physical passive leaching and/or weathering at high soil
411 moisture levels. It is likely that our results reflect the complex biophysical interactions
412 taking place in soils: soil enzyme activity releases soluble S from OM, which enhances
413 the sulphate reduction to sulphide via microbial respiration under higher levels of soil
414 water content (Hao et al., 1996; Muyzer and Stams, 2008), thereby indirectly leading
415 to decreases in the concentrations of soluble S in the soil solution. The EC of Mn, Cd,
416 V and Mo were positively correlated to levels of OM (EOC and ETN, Fig. 4 and 5C),
417 probably due to the formation of dissolved metal-OM complexes (Filipović et al., 2018;
418 Li et al., 2013). The strong soil retaining power of Mn and Cd (Figs. 4 and 5C) coincides

419 with previous findings suggesting that the biological availability of Mn (Li et al., 2021;
420 Tian et al., 2016; Wang et al., 2017) and Cd (Li et al., 2013; Wang and Zhou, 2017;
421 Yang et al., 2020) increases with rising concentrations of soluble C and N in the soil.
422 Their relationship with ETN may also indicate that soil acidification caused by
423 excessive soil N enhances the leaching of heavy metals (Atafar et al., 2010; Ata-ul-
424 karim et al., 2020; Mitchell et al., 2000; Raza et al., 2020), especially by increasing the
425 concentrations of soluble Mn and Cd.

426 Soil Cu and K were related with the storage of N and P in the microbial biomass
427 pool, as indicated by the positive relationship of their EC with microbial variables (Figs.
428 4, 5C and 6). Higher extractable K levels boost enzyme activities, thereby increasing
429 the potential mineralization of N and P and their acquisition by both microbes and plants
430 (Figs. 4 and 6), although increased microbial storage of N and P may reduce their
431 availability to plants under higher extractable Cu levels (Fig. 4).

432 The EC of Mn, Cd, Ni, V, Hg and Fe were all positively correlated among
433 themselves, which probably indicates the strong adsorption of these heavy metals to the
434 oxides of Mn and/or Fe-Mn in soils (Tack et al., 2006; Violante et al., 2010), as shown
435 by the close correlations between Mn and the other elements (Figs. 5C and S1). Ca, Mg,
436 Cu, Sr, Pb and K constitute another closely correlated group of elements (Figs. 5C and
437 S1), probably due to their high levels of cation-exchange reactions (Liu et al., 2019;
438 Martins et al., 2014; Miller et al., 1993; Ross et al., 2008; Sharma et al., 2015;
439 Sanderson et al., 2015). In addition, the strongly synergistic behaviour in EC between
440 Ca, Mg and Sr (Fig. S1) could be attributed to their geochemical similarity since they

441 are all naturally occurring alkaline earth metal elements (Bunde et al., 1997; Frei et al.,
442 2020; Scott et al., 2020).

443 *4.3. Seasonal variability in extractable elemental concentrations in Mediterranean* 444 *soils*

445 Although the seasonality of EC varied with the elements, the most common pattern was
446 characterized by higher EC in winter-spring (wet seasons) than in summer-autumn (dry
447 seasons) (Table 3), which indicates that temporal variations in EC were mainly driven
448 by soil water content.

449 Throughout the study period the EC generally tracked seasonal variations in soil
450 microclimate, biological activity and OM. Extractable Ca, Mg, Sr and Cr showed
451 similar seasonal patterns of change, whereby the EC were higher in winter than in other
452 seasons, probably due to the greater soil moisture in winter (Fig. 1A and Table 3). Like
453 these previous four elements, the EC of K, Cu and Pb were also strongly correlated with
454 soil water content (Fig. 4), although these latter elements had relatively higher EC in
455 spring (Table 3) given that high levels of microbial and/or root activities may increase
456 their availability during the growing season (spring) (Fig. 1C). By contrast, the EC of
457 S was higher in summer and autumn when biological activity is lower, suggesting that
458 less uptake by microbes and/or plants causes an accumulation of soluble S in the dry
459 seasons (Fig. 1C and Table 3). Fe, Mn and As had lower EC in autumn than in the other
460 seasons (Table 3), probably linked to the multiple effects of lower levels of water
461 content and biological activity in the dry seasons (Fig. 1), and lower levels of OM in
462 autumn than in summer (Asensio et al., 2021). The EC of V was seasonally irregular,

463 with maxima in spring and minima in autumn (Table 3). The close association between
464 V and Mn, however, may indicate that these two elements are involved in
465 biogeochemical processes in similar ways, although their lower magnitude may ensure
466 that the EC of V is highly variable. The EC of the other elements (Zn, Ni, Mo, Cd and
467 Hg) did not oscillate seasonally (Table 3), which suggests that they are less affected by
468 seasonality in climate and phenology and more dependent on parent material and soil
469 mineralogy.

470

471 **5. Conclusions**

472 The study revealed that variations in EC were primarily associated with soil water
473 content. The EC of elements exhibited seasonal responses linked to changes in soil
474 microclimate, biological activity, and organic matter levels. Generally, the EC of most
475 elements was higher during winter-spring (wet seasons) compared to summer-autumn
476 (dry seasons). Chronic drought had notable effects on soil element composition,
477 resulting in decreased concentrations of extractable potassium (K) but increased
478 concentrations of sulfur (S) and, to a lesser extent, other trace elements studied. These
479 alterations have implications for the cycling and loss of soil mineral elements,
480 potentially contributing to ecosystem degradation and pollution dispersion under future
481 drier conditions in the Mediterranean Basin. By assessing EC, the fundamental
482 mechanisms governing soil mineral element dynamics can be identified, enabling
483 projections of soil elemental composition, as well as the bioavailability of nutrients and
484 trace elements, under scenarios of global environmental change.

485

486 **Credit authorship contribution statement**

487 **Zhaobin Mu**: Data curation, Formal analysis, Software, Investigation, Validation,
488 Writing – original draft, Writing – review & editing. **Dolores Asensio**:
489 Conceptualization, Methodology, Data curation, Investigation, Validation, Writing –
490 original draft, Writing – review & editing. **Jordi Sardans**: Conceptualization,
491 Methodology, Validation, Writing – review & editing. **Romà Ogaya**: Methodology,
492 Investigation, Writing – review & editing. **Joan Llusà**: Methodology, Writing – review
493 & editing. **Iolanda Filella**: Writing – review & editing. **Lei Liu**: Writing – review &
494 editing. **Xinming Wang**: Writing – review & editing. **Josep Peñuelas**:
495 Conceptualization, Methodology, Validation, Project administration, Supervision,
496 Funding acquisition, Writing – original draft, Writing – review & editing.

497

498 **Data availability**

499 Data will be made available on request.

500

501 **Declaration of Competing Interest**

502 The authors declare that they have no known competing financial interests or
503 personal relationships that could have appeared to influence the work reported in this
504 paper.

505

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513

514 **Appendix A. Supplementary material**

515 Supplementary data to this article can be found online.

516

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767 **Table 1**

768 Mean (\pm SEM) extractable concentrations of soil mineral elements ($\mu\text{g g}^{-1}$ dry soil),
769 from control and chronic drought plots over the study period ($n = 4$ seasons \times 4 plots).

770 The elements were classified based on the concentrations of control plots in this study,
771 as ‘Macroelement’ ($\geq 1 \mu\text{g g}^{-1}$) or ‘Microelement’ ($< 1 \mu\text{g g}^{-1}$). The elements were
772 indicated as nutrients and/or pollutants and classified into different categories.

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Element	Treatment		Concentration class	Nutrient class	Pollutant class
	Control	Drought			
Ca	479.87 ± 58.24	473.31 ± 47.40	Macroelement	Macronutrient	(-)
K	349.66 ± 28.10	297.56 ± 12.90	Macroelement	Macronutrient	(-)
Mg	132.30 ± 9.13	136.73 ± 6.17	Macroelement	Macronutrient	(-)
S	107.51 ± 17.48	151.66 ± 17.98	Macroelement	Macronutrient	(-)
Fe	36.77 ± 3.07	41.24 ± 5.26	Macroelement	Micronutrient	Minor heavy metal pollutant
Sr	4.03 ± 0.46	3.60 ± 0.34	Macroelement	(-)	Radioactive metal pollutant
Mn	2.26 ± 0.34	3.53 ± 0.45	Macroelement	Micronutrient	Minor heavy metal pollutant
Cu	1.72 ± 0.51	0.72 ± 0.12	Macroelement	Micronutrient	Minor heavy metal pollutant
Zn	0.334 ± 0.069	0.468 ± 0.069	Microelement	Micronutrient	Minor heavy metal pollutant
Pb	0.314 ± 0.050	0.313 ± 0.040	Microelement	(-)	Major heavy metal pollutant
Cr	0.179 ± 0.025	0.203 ± 0.027	Microelement	(-)	Major heavy metal pollutant
Ni	0.157 ± 0.014	0.179 ± 0.021	Microelement	Micronutrient	Minor heavy metal pollutant
V	0.134 ± 0.031	0.180 ± 0.039	Microelement	Non-essential nutrient	Minor heavy metal pollutant
As	0.080 ± 0.010	0.068 ± 0.007	Microelement	(-)	Major heavy metal pollutant
Mo	0.019 ± 0.001	0.021 ± 0.001	Microelement	Micronutrient	Minor heavy metal pollutant
Cd	0.006 ± 0.001	0.008 ± 0.001	Microelement	(-)	Major heavy metal pollutant
Hg	0.006 ± 0.001	0.006 ± 0.001	Microelement	(-)	Major heavy metal pollutant

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779 **Table 2**

780 Results of a PERMANOVA (season × treatment) for the extractable concentrations of
781 the 17 soil mineral elements treated as covariables (permutations = 999; *** $P < 0.01$).

	<i>Df</i>	SumsOfSqs	MeanSqs	Model <i>F</i>	<i>R</i> ²	Pr(> <i>F</i>)	
Season	3	0.311	0.104	7.612	0.460	0.001	***
Treatment	1	0.021	0.021	1.533	0.031	0.213	
Season:Treatment	3	0.017	0.006	0.421	0.025	0.895	
Residuals	24	0.327	0.014		0.484		
Total	31	0.676			1.000		

782 **Table 3**

783 Mean (±SEM) seasonal concentrations of soil mineral elements ($\mu\text{g g}^{-1}$ dry soil) ($n = 2$
784 treatments × 4 plots). Different letters indicate significant differences among seasons

785 according to one-way ANOVAs.

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Element	Season			
	Winter	Spring	Summer	Autumn
Ca	709.29 ± 108.75 ^a	383.27 ± 36.75 ^b	418.50 ± 10.85 ^b	395.30 ± 21.69 ^b
K	399.13 ± 37.11 ^a	346.85 ± 34.24 ^{ab}	289.57 ± 9.58 ^b	258.90 ± 13.75 ^b
Mg	164.66 ± 11.73 ^a	122.16 ± 12.04 ^b	124.12 ± 5.63 ^b	127.14 ± 5.83 ^b
S	92.48 ± 11.57 ^a	58.81 ± 10.77 ^a	186.96 ± 24.63 ^b	180.09 ± 19.23 ^b
Fe	43.03 ± 6.74 ^{ab}	52.44 ± 6.76 ^a	35.41 ± 2.02 ^{ab}	25.14 ± 2.99 ^b
Sr	5.71 ± 0.75 ^a	3.17 ± 0.32 ^b	3.36 ± 0.15 ^b	3.02 ± 0.14 ^b
Mn	2.57 ± 0.54 ^{ab}	3.58 ± 0.75 ^a	3.97 ± 0.34 ^a	1.47 ± 0.30 ^b
Cu	2.37 ± 0.81 ^a	1.72 ± 0.49 ^{ab}	0.46 ± 0.03 ^b	0.34 ± 0.02 ^b
Zn	0.535 ± 0.082	0.495 ± 0.093	0.257 ± 0.094	0.317 ± 0.110
Pb	0.445 ± 0.073 ^a	0.383 ± 0.059 ^{ab}	0.239 ± 0.036 ^b	0.187 ± 0.028 ^b
Cr	0.303 ± 0.044 ^a	0.173 ± 0.015 ^b	0.138 ± 0.007 ^b	0.150 ± 0.034 ^b
Ni	0.167 ± 0.028	0.193 ± 0.033	0.166 ± 0.010	0.146 ± 0.024
V	0.079 ± 0.009 ^a	0.302 ± 0.020 ^b	0.199 ± 0.020 ^c	0.019 ± 0.008 ^d
As	0.107 ± 0.029 ^{ab}	0.094 ± 0.012 ^a	0.080 ± 0.013 ^{ab}	0.044 ± 0.007 ^b
Mo	0.017 ± 0.001	0.019 ± 0.001	0.022 ± 0.001	0.022 ± 0.002
Cd	0.007 ± 0.001	0.007 ± 0.001	0.009 ± 0.001	0.005 ± 0.001
Hg	0.007 ± 0.001	0.005 ± 0.001	0.006 ± 0.001	0.005 ± 0.001

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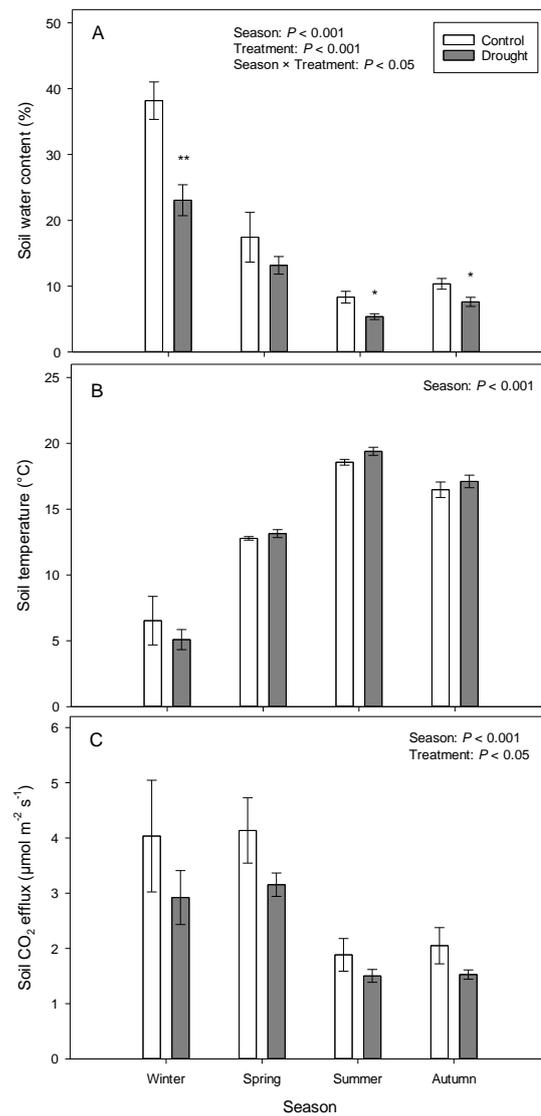
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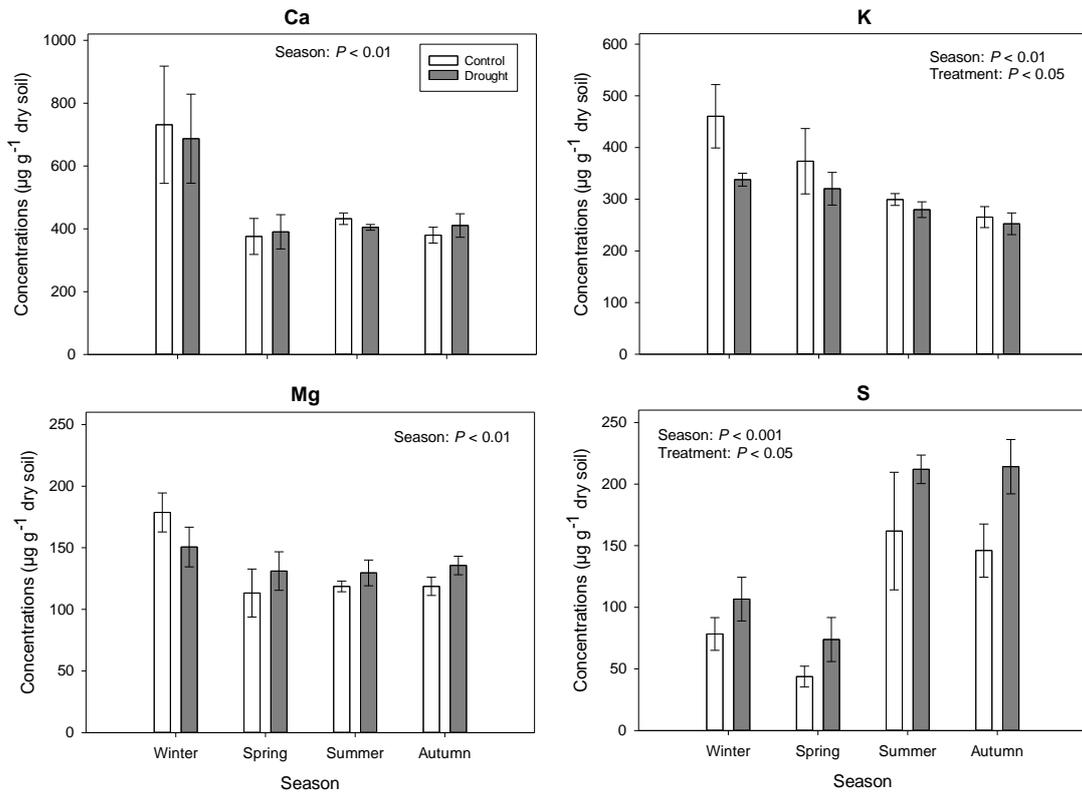
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825 **Fig. 1.** Seasonal soil (A) water content, (B) temperature, and (C) CO₂ efflux in control
 826 and drought plots. Data are means ± standard error (n = 4 plots). Treatment differences
 827 between seasons are indicated with asterisks (Student's *t*-tests, **P* < 0.05 and ***P* <
 828 0.01). The effects of season and treatment (two-way ANOVAs) are depicted in the
 829 panels when significant.

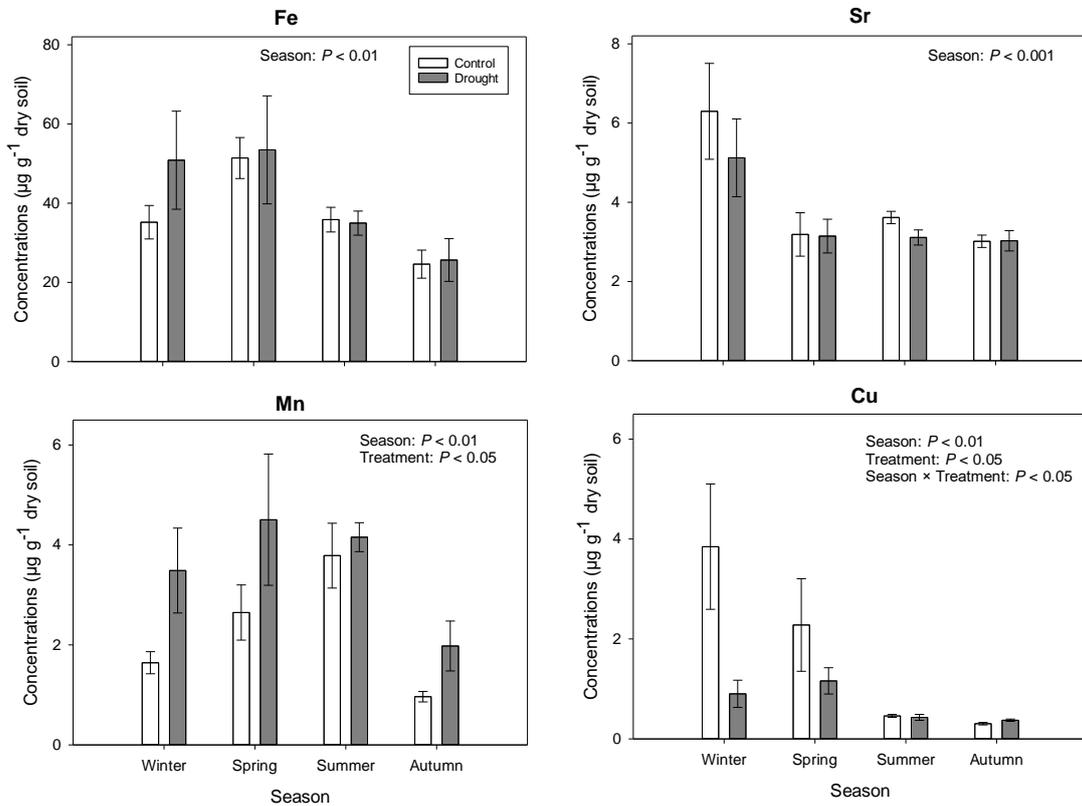
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838 **Fig. 2.** Seasonal extractable concentrations of soil (A) Ca, K, Mg, and S, and (B) Fe,
839 Sr, Mn, and Cu in control and drought plots. Data are means \pm standard error ($n = 4$
840 plots). The effects of season and treatment (two-way ANOVAs) are depicted in the
841 panels when significant. The P values (Student's t -tests) between treatments for each
842 season are shown in Table S3.

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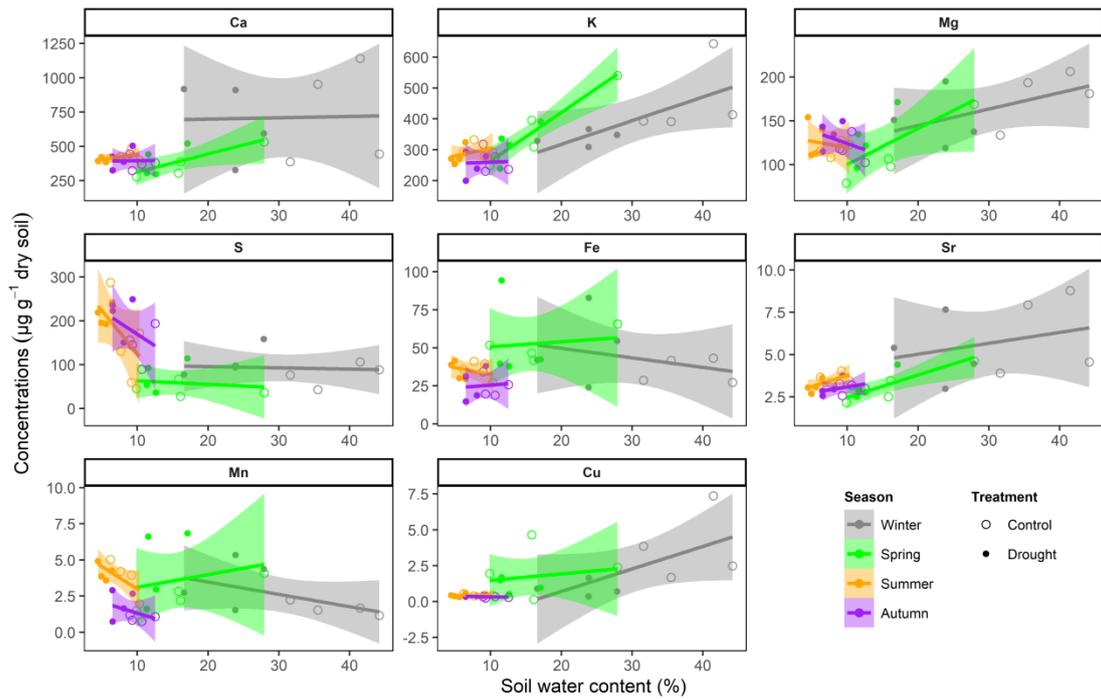
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862 **Fig. 3.** Correlations between extractable Ca, K, Mg, S, Fe, Sr, Mn and Cu

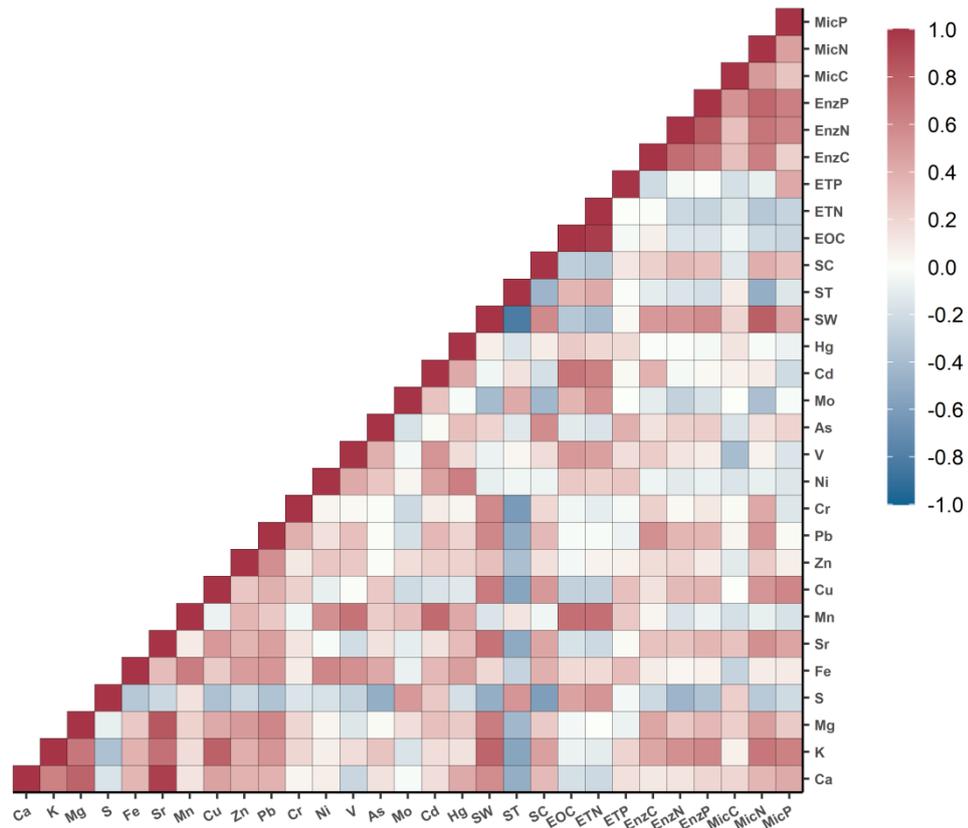
863 concentrations and soil water content. Shaded areas are the 95% confidence intervals

864 of the regression lines. Chronic drought samples are represented by filled dots, control

865 samples are represented by empty dots. Different seasons are represented by different

866 colors. *r* coefficients and *P* values for each season are shown in Table S2.

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869 **Fig. 4.** Heat map of pairwise correlation coefficient (r) between the soil chemical-
 870 biological variables. The colors represent the coefficient of correlation between each
 871 pair of variables, where red represents a positive correlation and blue represents a
 872 negative correlation. The r values are shown in Figure S1.

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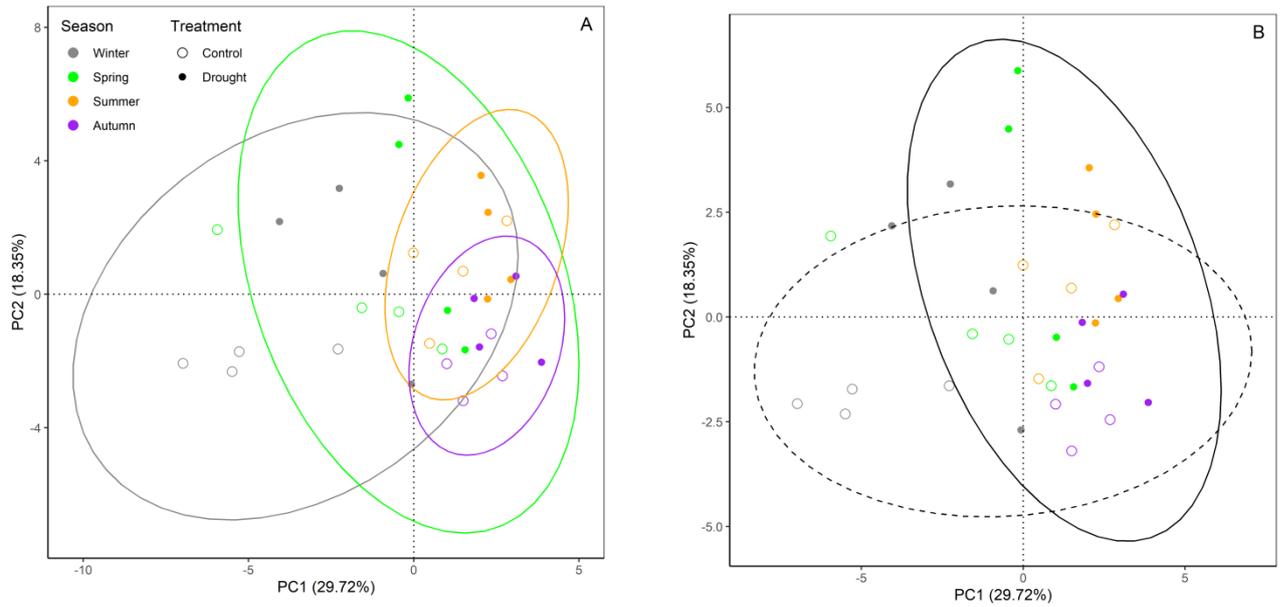
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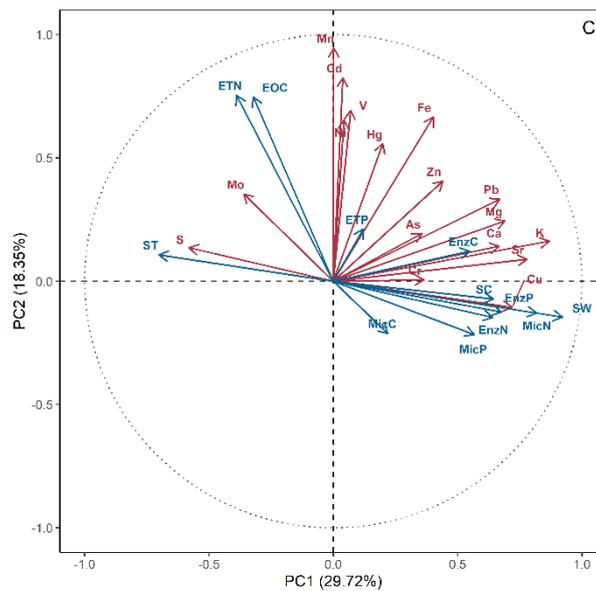
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898 **Fig. 5.** Two-dimensional representation of the areas defined by the first two principal

899 components (PC1 and PC2) of the principal component analysis of season, treatment,

900 and soil chemical-biological variables. A) Values assigned to each soil sample for PC1

901 and PC2 grouped by season. B) Values grouped by treatment. C) Relative contribution

902 of each variable to PC1 and PC2. Chronic drought samples are represented by filled

903 dots, control samples are represented by empty dots. Different seasons are represented
904 by different colors. Vectors are divided into target mineral elements (red) and other soil
905 variables (blue). The percentages contributions to the total variability are in parentheses.

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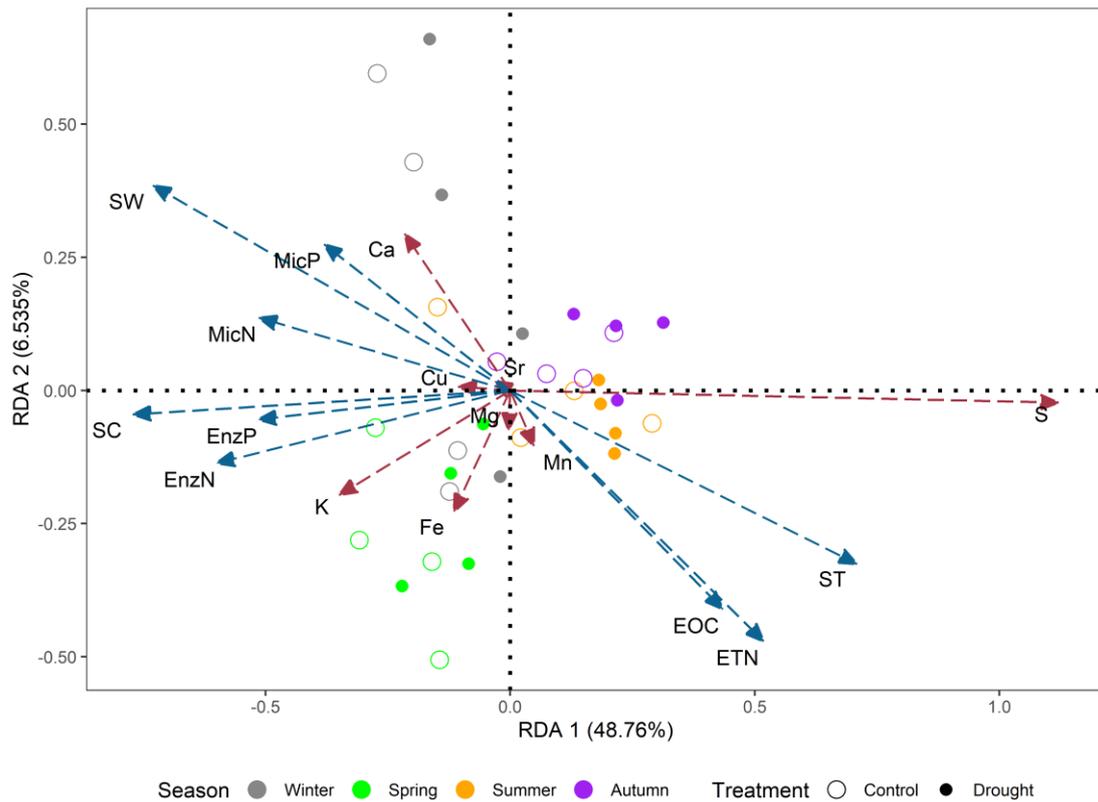
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928 **Fig. 6.** Redundancy analysis (RDA) based on correlations between macroelement
929 concentrations (red) and explicable soil chemical-biological variables ($P < 0.1$) (blue)
930 in this study. Chronic drought samples are represented by filled dots, control samples
931 are represented by empty dots. Different seasons are represented by different colors.
932 Correction of R squares and significance test for the constrained axes (permutations =
933 999) of soil variables are shown in Table S5.

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