



APPLICATION OF SULFATE REDUCING BACTERIA IN ACID MINE DRAINAGE ACTIVE REMEDIATION

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WHAT IS ACID MINE DRAINAGE?	PROBLEMS ASSOCIATED WITH AMD	AIMS	APPLICATION OF SULFATE REDUCING BACTERIA (SRB) FOR AMD REMEDIATION
<p>Acid mine drainage (AMD) is one of the main contaminating problems derived from the mining industry and metallurgy, as it is an acidic metal and sulfate-containing wastewater. It is mainly a consequence of the exposure of iron pyrite (FeS₂) among other sulfide minerals to both oxygen and water during mining and processing of metal ores and coal, which triggers the accelerated oxidation of these minerals.</p> $2 \text{FeS}_2 + 7 \text{O}_2 + 2 \text{H}_2\text{O} \rightarrow 2 \text{Fe}^{2+} + 4 \text{SO}_4^{2-} + 4 \text{H}^+$	<ul style="list-style-type: none"> » Contaminated drinking water. » Affected growth and reproduction of plants and animals. » Corroding effects of acid on infrastructures. 	<p>Among all the bioremediation strategies for AMD, this work focuses on active technologies. The two principal standards will be described and compared. Therefore, real examples will be analyzed.</p>	<p>Under anaerobic conditions, SRB catalyze the reduction of sulfate to sulfide, generating alkalinity by transforming a strong acid (sulfuric) into a relatively weak acid (hydrogen sulfide):</p> $\text{SO}_4^{2-} + 2 \text{CH}_2\text{O} + 2 \text{H}^+ \rightarrow \text{H}_2\text{S} + 2 \text{H}_2\text{CO}_3$ <p>Many metals react with the produced H₂S to form highly insoluble sulfides:</p> $\text{Fe}^{2+} + \text{H}_2\text{S} \rightarrow \text{FeS (s)} + 2 \text{H}^+$

AMD BIOREMEDIATION

- Decrease of sulfate concentration
- Removal and recovery of metals

SULFIDOGENIC BIOREACTORS

Electron donor for SRB	Reactor type
<p>SRB require the addition of an electron donor to achieve sulfate reduction. To this effort, they use several low molecular weight substrates, such as ethanol and hydrogen.</p> <div style="display: flex; justify-content: space-around;"> <div style="border: 1px dashed orange; padding: 5px;"> <p>Ethanol</p> <ul style="list-style-type: none"> » Small-scale applications. » High operational costs. $3 \text{SO}_4^{2-} + 2 \text{C}_2\text{H}_5\text{OH} \rightarrow 3 \text{HS}^- + 3 \text{H}_2\text{O} + 3 \text{HCO}_3^- + \text{CO}_2$ </div> <div style="border: 1px dashed orange; padding: 5px;"> <p>Hydrogen gas</p> <ul style="list-style-type: none"> » Large-scale applications. » Low production of biomass. » Produced on site by reforming natural gas. » High investment costs. $\text{SO}_4^{2-} + 4 \text{H}_2 + \text{H}^+ \rightarrow \text{HS}^- + 4 \text{H}_2\text{O}$ </div> </div>	<p>CIRCOX® gas-lift loop</p> <ul style="list-style-type: none"> » Optimum mixing and mass transfer of H₂. » 3-phase separator: safeguards biomass retention. Enhance gas/liquid separation. » Bacteria grow into aggregates. Excess must be removed.

THIOPAQ

- » Anaerobic sulfate reduction by SRB and metal precipitation in a single tank ①.
- » (Optional) Aerobic oxidation of any excess H₂S to elemental sulfur by sulfide-oxidizing bacteria ②.

- *Thiopaq* technology can treat effectively low metal concentration streams.
- However, the process has limitations, especially when it comes to the treatment of high metal concentration streams:
 - » Sensitivity of SRB to low pH and high metal concentration.
 - » Difficulty to maintain the optimum conditions due to seasonal fluctuations of the entering mine water.
 - » Plugging in the bioreactor, as a consequence of precipitation of sulfide sludge.
 - » Sludge contains a mixture of metals. } Greater volume of sludge increases disposal expense
 - » Sludge also contains biomass.

BIOSULFIDE

- » Chemical precipitation of metal sulfides ① separated from biological sulfate reduction ②.
- » In chemical tank, raw AMD contacts with H₂S generated in the biological tank.

- *Biosulfide* technology is useful with low and high metal concentration streams.
- It presents some advantageous aspects compared to *Thiopaq* technology:
 - » SRB receive very low concentration of metals.
 - » Optimum conditions easily maintained.
 - » Sulfide sludges isolated in chemical circuit.
 - » Selective separation and removal of metals by using several precipitation tanks.

Case of Study. Budel Zinc Refinery (Netherlands)

Treatment of AMD by neutralization with lime: production of gypsum

➔

Legislative restrictions for further production of residues

➔

Thiopaq technology: treatment of sulfate and recovery of zinc sulfide, which is recycled to the refinery

- » Industrial treatment plant (2000)
- » Influent = 40 m³/h
- » Electron donor: hydrogen

Component	Influent (mg/l)	Effluent (mg/l)
Sulfate	15,000	<250
Zinc	10,000	<0.3

Case of study. Kennecott Utah Copper (USA)

Biosulfide technology: Treatment of a contaminated groundwater stream with high levels of metals and sulfate.

- » Pilot treatment plant (1995)
- » Influent = 0.2 m³/h
- » Electron donor: hydrogen

Component	Influent (mg/l)	Effluent (mg/l)
Sulfate	30,000	<500
Copper	60	<0.1
Iron	675	<0.3
Zinc	65	<0.1
Manganese	350	0.3
Aluminium	2,200	<2

CONCLUSIONS

- » AMD remediation with sulfidogenic reactors can address the threat that mine wastes pose to the environment.
- » In addition to the environmental benefit, it can trigger economic advantages by recovering valuable metals.
- » *Biosulfide* technology has many upsides in comparison to *Thiopaq* technology. However, every situation requires an individualized design of the treatment process.
- » There is a lack of awareness in the mining industry and metallurgy in the aspect of implementing active and biological technologies.

REFERENCES

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