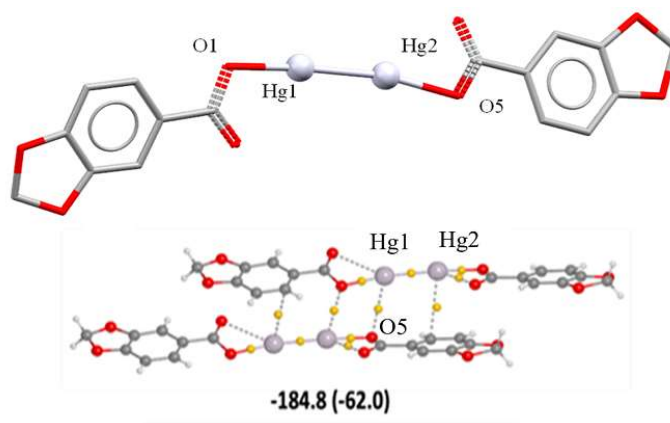


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The chemistry of Hg (I): a structural study



Mercury (Hg) is a metal capable of forming various polycations with various geometries. In this work of the Area of Inorganic Chemistry, the synthesis and characterization of $\text{Hg}_2(\text{Pip})_2$, a Hg (I) compound with aromatic ligands and carboxylate groups, is presented. Its characterization has been carried out by elemental analysis, spectroscopic techniques, single crystal X-ray diffraction and DFT calculations.

Crystal structure and interaction energies of the Hg(I) dimer - Josefina Pons

Hg is a metal that has the particularity of being capable of forming a variety of divalent, trivalent, or tetravalent polycations arranged either into linear, dimeric $[\text{Hg}_2]^{2+}$, trimeric $[\text{Hg}_3]^{2+}$, tetrameric $[\text{Hg}_4]^{2+}$, and chains $[\text{Hg}]_n$ or triangles $[\text{Hg}_3]^{4+}$. All of them present differences in the formation conditions, connectivity, geometry, and Hg-Hg bond lengths.

The formation of the $[\text{Hg}_2]^{2+}$ is stabilized by the lower solubility compared to those of their Hg(II) analogues. This preferred linear arrangement facilitates the formation of $\text{Hg}\cdots\pi$ interactions.

The coordination chemistry and structural arrangement of Hg(I) with aromatic carboxylates have not been extensively explored. In this paper, we provide a compound of Hg(I) with an aromatic carboxylate $[\text{Hg}_2(\text{Pip})_2]$ (Pip = piperonylic acid). The obtention of the compounds is due to decomposition of the $[\text{Hg}(\text{Pip})_2(4,4'\text{-bipy})]_n$ (4,4'-bipy = 4,4'-bipyridine) to give Hg(0) that have driven the formation of the $[\text{Hg}_2]^{2+}$ cation. This reaction depends on the solvent, temperature, and time (in this case, dimethylformamide, 105 °C and 1h).

The compound obtained ($[\text{Hg}_2(\text{Pip})_2]$), has been characterized by elemental analysis, spectroscopic techniques (IR, ^1H NMR), single crystal X-ray diffraction, and DFT calculations have been performed. The compound is dimeric and the $[\text{Hg}_2(\text{Pip})_2]$ units are joined together in tetrameric $[\text{Hg}_4(\text{Pip})_4]$ assemblies by Pip ligands. The structure is expanded into a 2D supramolecular structure by C-H \cdots O and Hg \cdots π interactions. DFT calculations have been performed to analyze the interactions between $[\text{Hg}_2(\text{Pip})_2]$ dimers.

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References

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